

Predicting the Critical Temperature of Superconductors

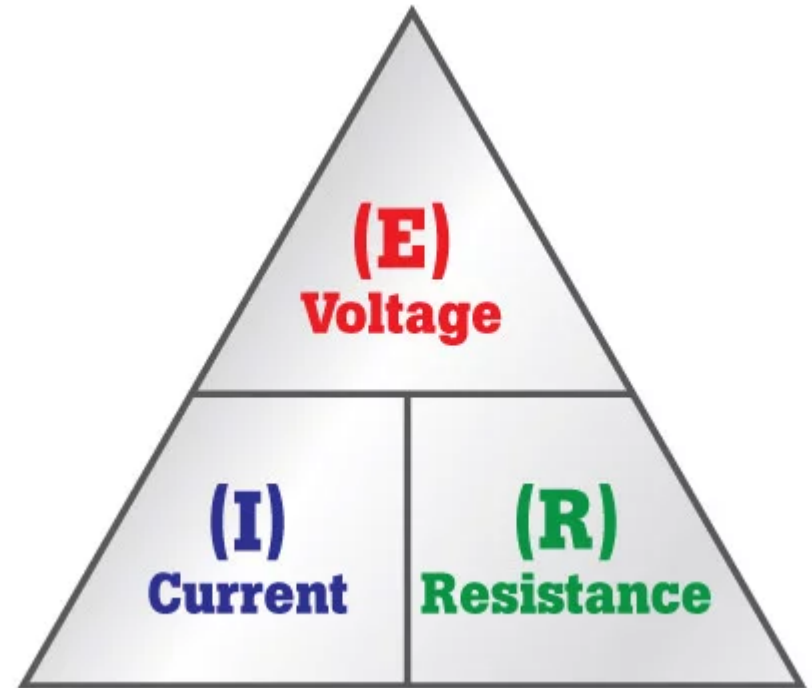
Yonaton Heit

Objective

- Predict the critical temperature for superconductors.
- Data Source:
 - UCI Machine Learning Repository database of superconductors and extracted properties
 - <http://archive.ics.uci.edu/ml/datasets/Superconductivity+Data>

What is a Superconductor

- Superconductor are materials with zero resistance.
- With zero resistance, electronic current can be maintained without external voltage
- With zero resistance, an electronic current can be maintained indefinitely.

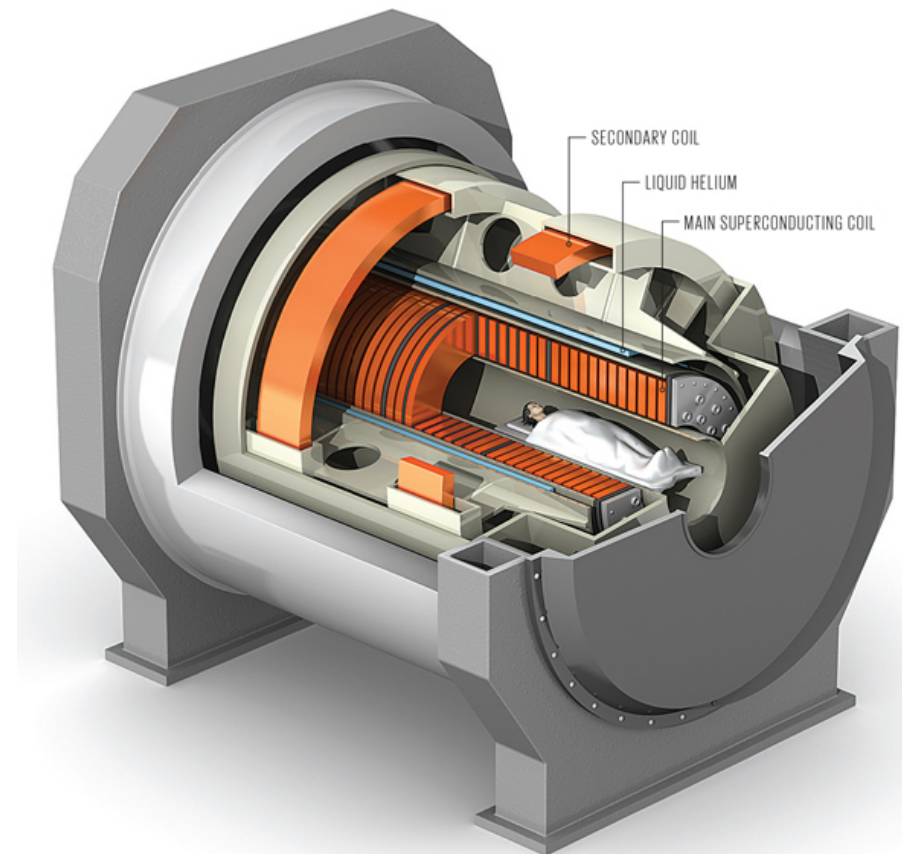


Source: <https://www.fluke.com/en-us/learn/best-practices/measurement-basics/electricity/what-is-ohms-law>

Applications for Superconductors

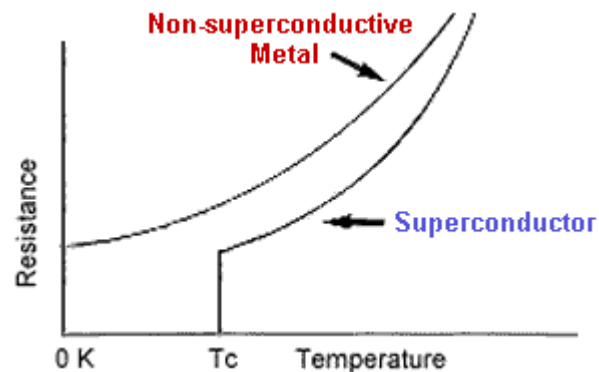
- Superconducting magnetic in Magnetic Resonance Imaging. (MRI)
- Superconducting coils in the Large Hadron Collider
- Superconductors could components in electronic powered systems [1]

MRI machine



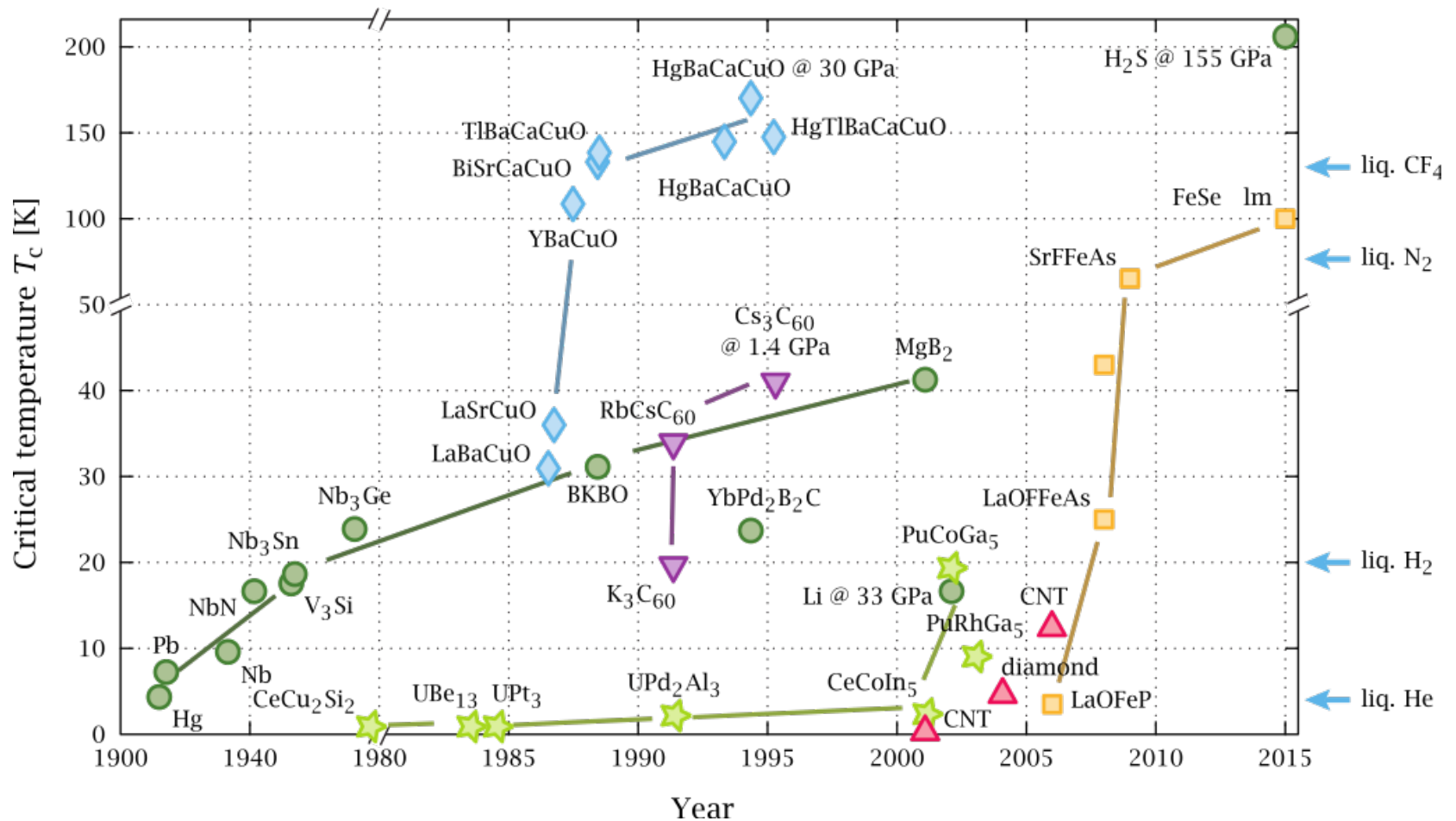
Critical Temperature

- Superconductivity can only be maintained below a certain temperature.
- This is called the critical temperature.
- Superconductors have refrigerated in order to maintain superconductivity in
 - Liquid Helium (4 K)
 - Liquid Nitrogen (77 K)



Source: http://www.superconductors.org/tc_graph.gif

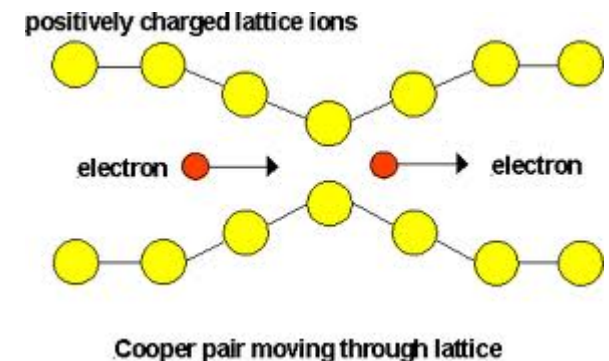
Critical Temperatures



Source: <https://en.wikipedia.org/wiki/Superconductivity>

Superconductor Theory

- There is not universal theory for superconductivity.[1]
 - In 1957, Bardeen, Cooper, Schrieffer (BCS) theory was proposed.[2]
 - Electrons are bound as Cooper Pairs.
 - Works well for low temperature superconductors (type I)
 - Not so well for higher temperatures superconductors which were later discovered (type II)
 - Other theories include
 - Resonating-valence-bond theory[3]
 - Spin fluctuation theory [4]



Source: <https://physics.stackexchange.com/questions/126742/do-all-the-electrons-form-cooper-pairs-at-absolute-zero>

[1] A. Mann, *Nature*, **475**, (21), 280-282, 2011

[2] J. Bardeen, et al. *Phys. Rev.* **106** (5), 162–164, 1957

[3] P.W. Anderson, *Science*, **235** (4793), 1196–1198, 1987

[4] P. Monthoux, et al. *Phys. Rev. Lett.*, **67** (24), 3448–3451, 1991

Data

- With no universal theory to predict critical temperature, we have to derive them from regressions.
- Data set contains:
 - 21,263 superconductors
 - 80 attributes
 - 8 properties derived from the elemental components
 - 10 statistical values determined from the 8 properties
 - $8 \times 10 = 80$
 - A complete list of the ratios elements for each superconductor (Not used by the original analysis by Hamidieh)

Attributes

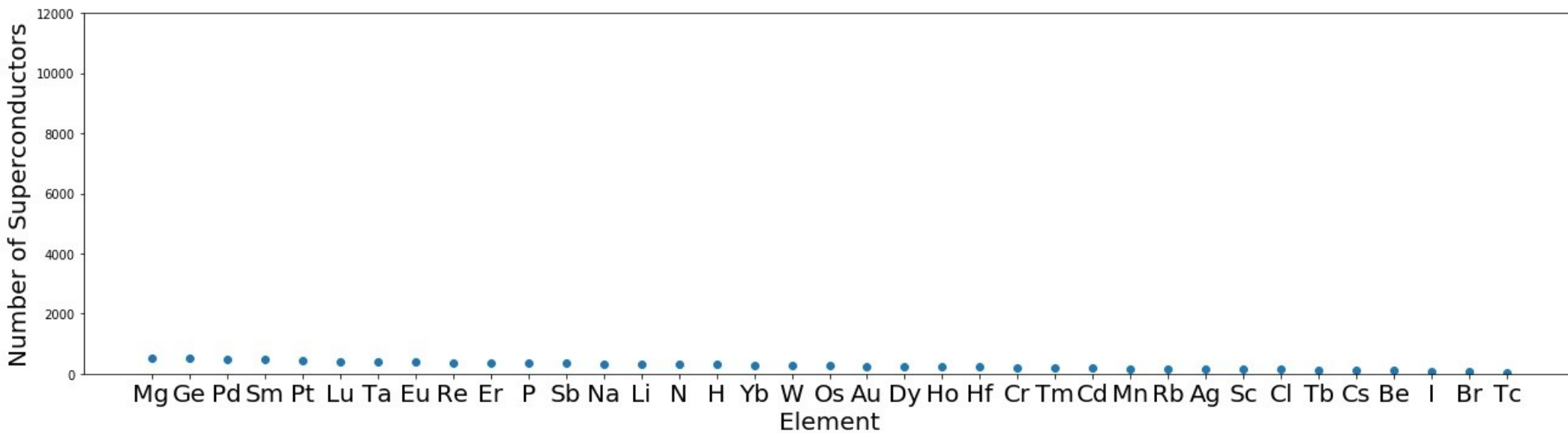
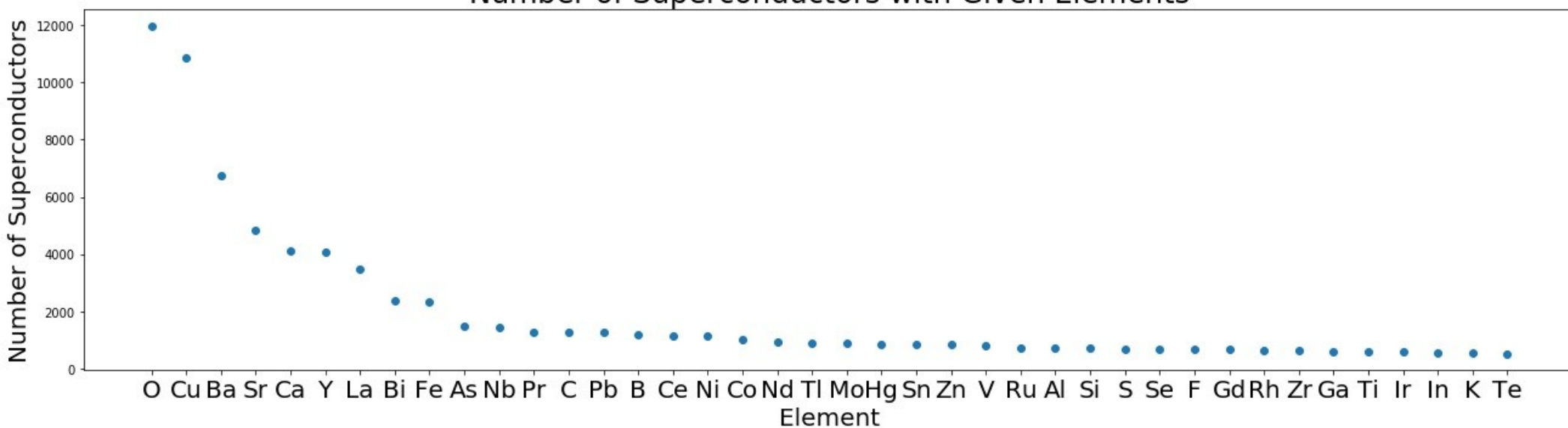
Variable	Units	Description
Atomic Mass	atomic mass units (AMU)	total proton and neutron rest masses
First Ionization Energy	kilo-Joules per mole (kJ/mol)	energy required to remove a valence electron
Atomic Radius	picometer (pm)	calculated atomic radius
Density	kilograms per meters cubed (kg/m ³)	density at standard temperature and pressure
Electron Affinity	kilo-Joules per mole (kJ/mol)	energy required to add an electron to a neutral atom
Fusion Heat	kilo-Joules per mole (kJ/mol)	energy to change from solid to liquid without temperature change
Thermal Conductivity	watts per meter-Kelvin (W/(m × K))	thermal conductivity coefficient κ
Valence	no units	typical number of chemical bonds formed by the element

Feature & Description	Formula
Mean	$= \mu = (t_1 + t_2)/2$
Weighted mean	$= \nu = (p_1 t_1) + (p_2 t_2)$
Geometric mean	$= (t_1 t_2)^{1/2}$
Weighted geometric mean	$= (t_1)^{p_1} (t_2)^{p_2}$
Entropy	$= -w_1 \ln(w_1) - w_2 \ln(w_2)$
Weighted entropy	$= -A \ln(A) - B \ln(B)$
Range	$= t_1 - t_2 \ (t_1 > t_2)$
Weighted range	$= p_1 t_1 - p_2 t_2$
Standard deviation	$= [(1/2)((t_1 - \mu)^2 + (t_2 - \mu)^2)]^{1/2}$
Weighted standard deviation	$= [p_1(t_1 - \nu)^2 + p_2(t_2 - \nu)^2]^{1/2}$

K. Hamidieh, *Computational Materials Science*, **154**, 346-354, 2018

Element analysis

Number of Superconductors with Given Elements



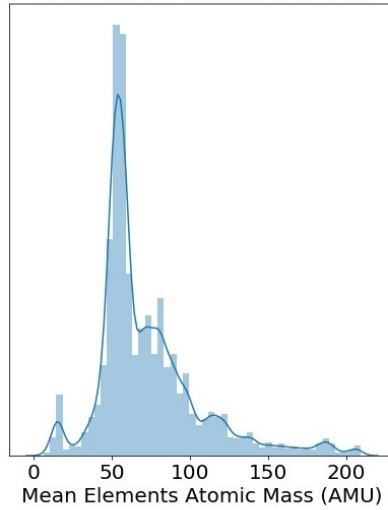
Element analysis

- There are 77 elements in the database.
- 60 elements appear in less than 5% of superconductors.
- Oxygen and Copper are the most common element
- 58.92% of superconductors have oxygen or copper.
- It would be interesting to see how well a model does using only elements.

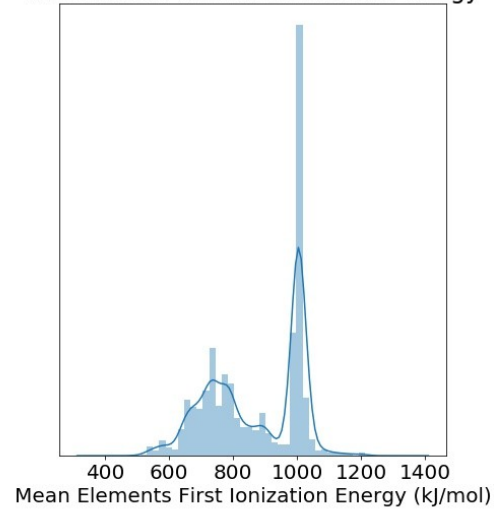
Superconductors with Element		
	Percent	Number
Oxygen	56.27%	11964
Copper	50.97%	10838
Barium	31.75%	6751
Strontium	22.82%	4852
Calcium	19.34%	4112
Yttrium	19.16%	4075
Lanthanum	16.29%	3463
Bismuth	11.24%	2389
Iron	11.00%	2339
Arsenic	7.06%	1502

Property Distribution of the Weighted Mean

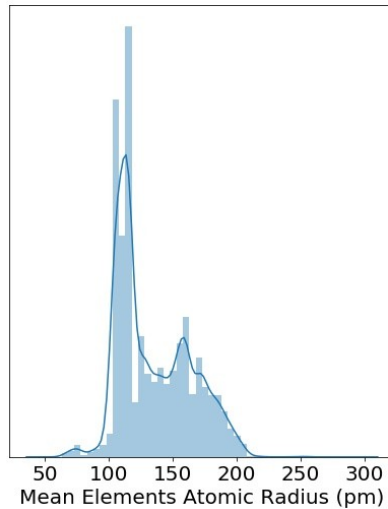
Distribution of Atomic Mass



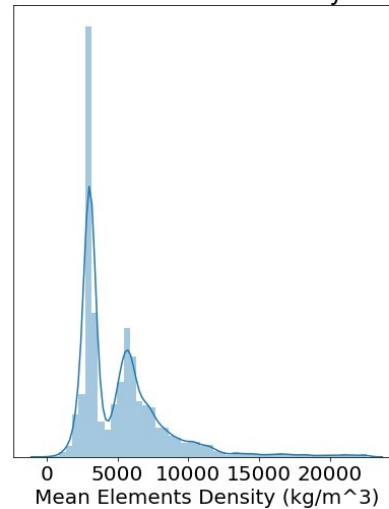
Distribution of First Ionization Energy



Distribution of Atomic Radius

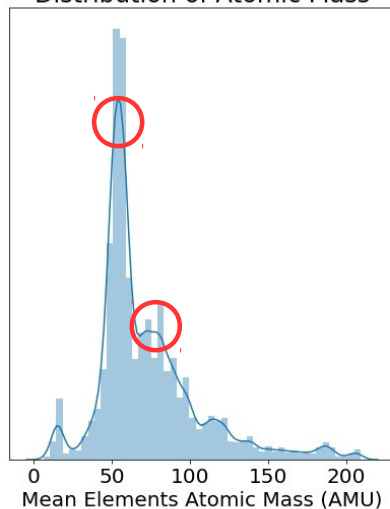


Distribution of Density

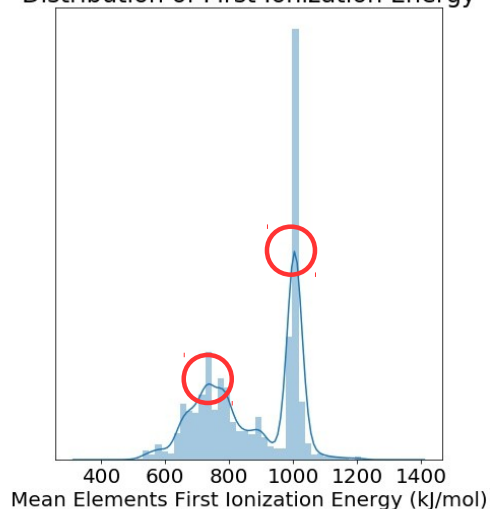


Property Distribution of the Weighted Mean

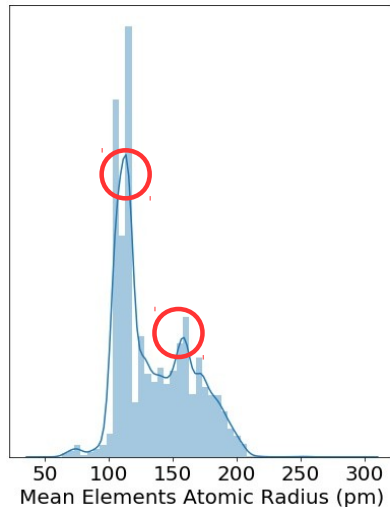
Distribution of Atomic Mass



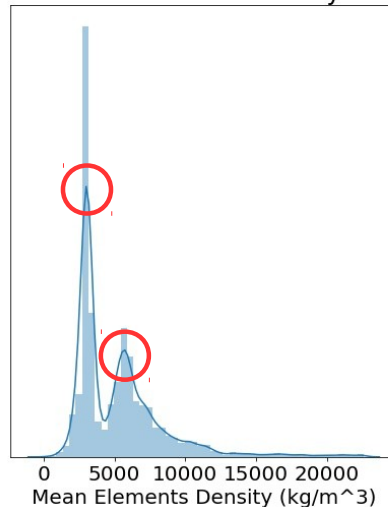
Distribution of First Ionization Energy



Distribution of Atomic Radius

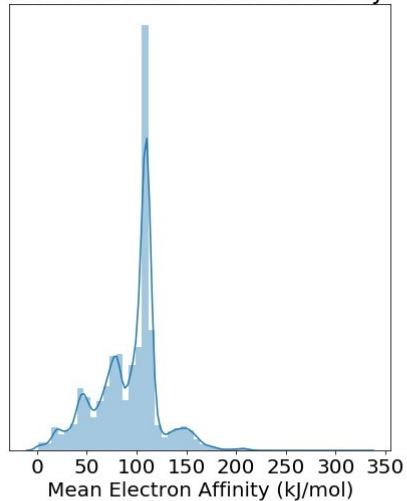


Distribution of Density

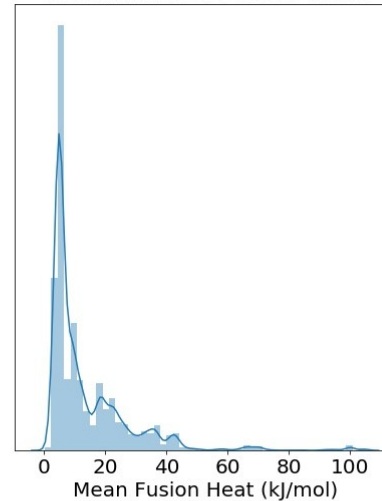


Property Distribution of the Weighted Mean

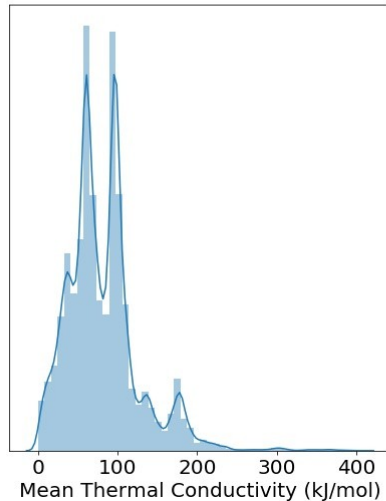
Distribution of Electron Affinity



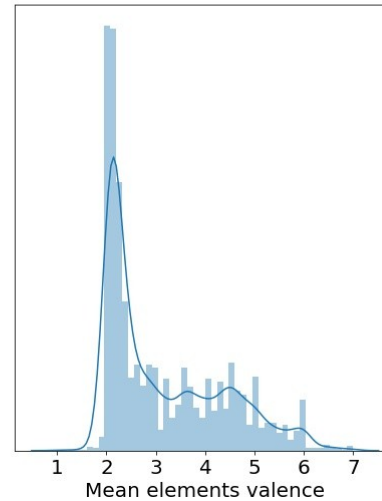
Distribution of Fusion Heat



Distribution of Thermal Conductivity

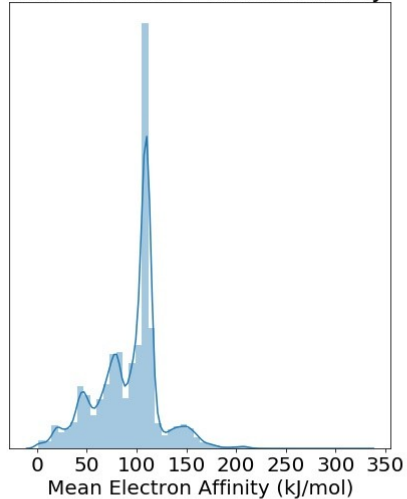


Distribution of Valence

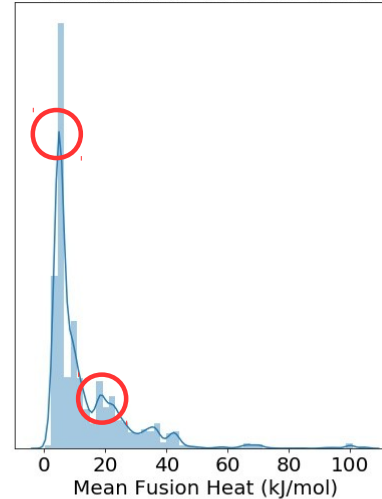


Property Distribution of the Weighted Mean

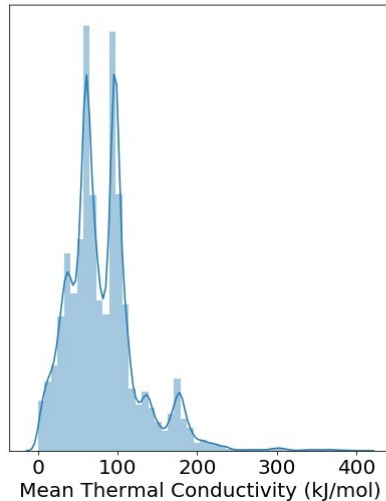
Distribution of Electron Affinity



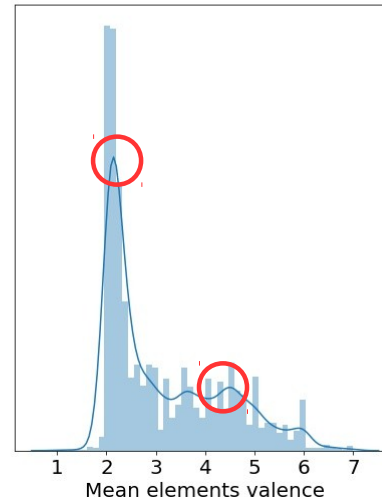
Distribution of Fusion Heat



Distribution of Thermal Conductivity



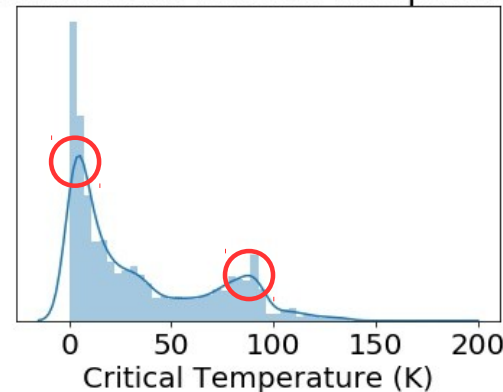
Distribution of Valence



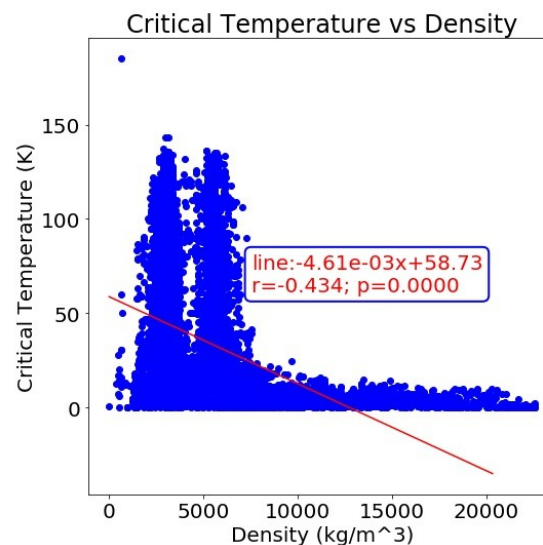
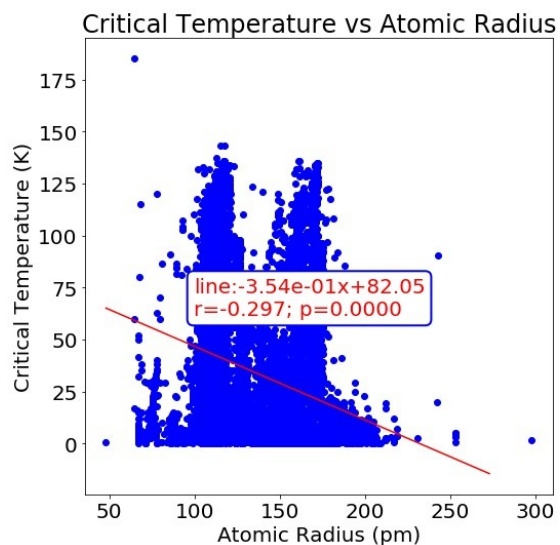
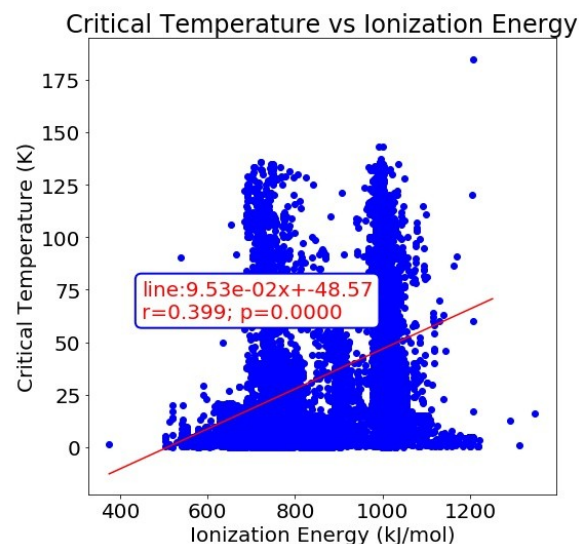
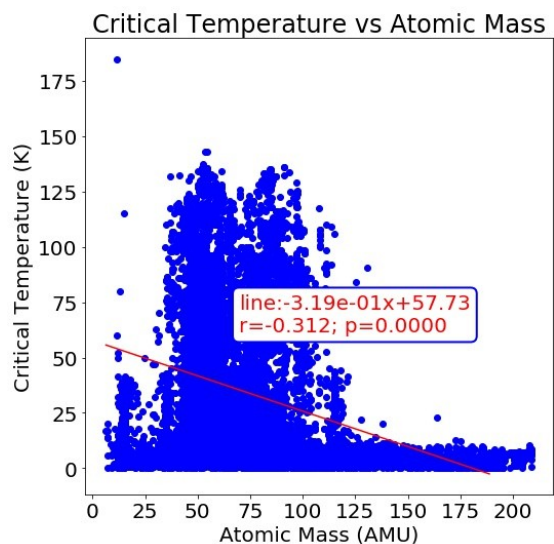
Critical temperature distribution

- The distributions suggest that there are two populations of superconductors in the data set.
- Possibly type I and type II superconductors.
- Type I are significantly lower temperature than type II

Distribution of Critical Temperatures



Scatter Plots of the Weighted Mean



Scatter Plots of the Weighted Mean

