allylic + benzylic halogention stability of dienes S-cis vs S-trans conformations Kinetie vs thermodynamic control conjugate addition cydo addition Diels-Alder drawing It molecular orbitals Homo vs. Lumo Symmetry allowed us. Symmetry torbidden nomendature of avonatic molecules annulenes aromaticity - planar, cyclic, able to delocalize, thuckel Frost's circle (molecular orbital energy levels) I dentity as aromatic, non-aromatic, ant, - aromatic behaylic carbocation oxidation of side chain Birch reduction Substituent effects on reactivity rates. Test # 1 rework

Electrophilic aromatic substitution archium ion Halosenation (chloring & bromine) Halognation mechanism nitration mechanism sufonation mechanism reversibility of sulfanation Friedel crafts alkylation mechanism Friedel Cratts acylation mechanism clemmenson reduction of carbony! wolff - Kishner reduction of carbonyl Rancy Nickel reduction of carbonyl ortholpara activitors ortho Ipan deactivitors meta deadivators mo meta Friedel - crafts react ions Oxidation of side chains reduction at nitro groups ites I NH3 reduction of one nitro group activators beat deactivators ortho disubstituted bentenes order of reactions in synthesis partial rate factors substitution of naphthalene substitution of heterocycles

reactions Synthesis Electrophilic aromatic substitution mechanism Free question Spectral sheet (s) · resonance structures.

potential energy diagrams