## Organic II

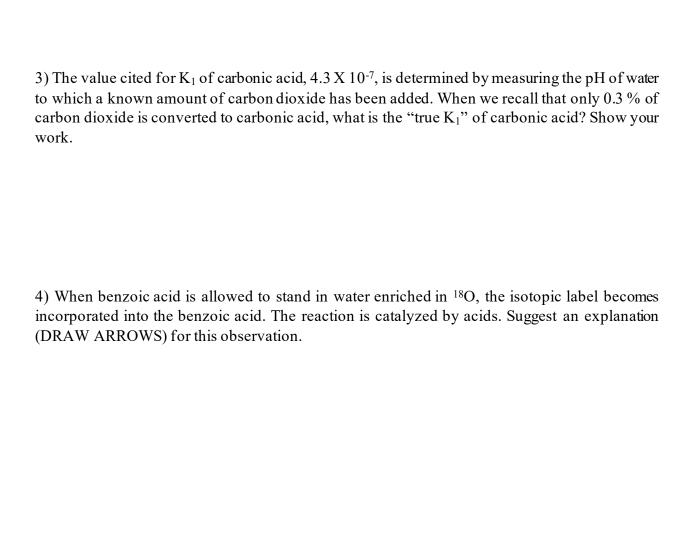
Assignment # 8 Spring 2021

- 1) Give the IUPAC names for the following molecules AND draw them OR if already given the drawing, give the IUPAC name.
- a) Methacrylic acid
- b) Crotonic acid
- c) Oxalic Acid
- d) p-Toluic acid

g) Palmitic acid

## h) Naphthenic acid (one is fine)

- k) What type of acid is the acid in part j using the omega nomenclature?
- 2) a) Lactic acid has a pKa of 3.9. What is the [lactate]/[lactic acid] ratio at the pH of blood (7.4)? Show your work.
- b) A 0.1 M solution of lactic acid in water has a pH of 2.5. What is the [lactate]/[lactic acid] ratio in this solution? Show your work.



5) The compound shown was subjected to the following series of reactions. Give the product of each reaction.

1) Lithium Aluminum Hydride
2) 
$$H_2O$$

A

SOCl<sub>2</sub>

D

1) Lithium Aluminum Hydride
2)  $H_3O^+$ 

C

DMSO

1) Lithium Aluminum Hydride
2)  $H_2O$ 

E  $(C_9H_9ClO_3)$ 

6) Circle the more acidic compound in each of the following pairs. Explain your choice.

7) a) Which stereoisomer of 4-hydroxycyclohexanecarboxylic acid (cis or trans) can form a lactone? Show this lactone.

b) What is the conformation of the cyclohexane ring in the starting hydroxy acid? Show this conformation.

 $c) \ What is the conformation of the cyclohexane ring in the lactone? \ Show this conformation.$ 

8) a) On standing in dilute aqueous acid, compound A is smoothly converted to mevalonolactone. Suggest (that means draw arrows) a reasonable mechanism for this reaction.

Compound A

- b) What other organic product is formed in the reaction above?
- 9) Give the products of the following reactions.

a) OH 
$$\xrightarrow{\text{Ethanol}}$$
 ?????

$$H_2SO_4$$

$$\xrightarrow{\text{1) LiAlD4}}$$
 ?????

1) Mg, diethyl ether

$$CF_3$$
 2)  $CO_2$ 
3)  $H_3O^+$ 

?????

C

 $H_2O$ , acetic acid

 $H_2SO_4$ , heat

OH

 $HBr$ 
 $Benzoyl peroxide$ 

?????

 $OH$ 
 $OH$ 

10) Compounds A and B are isomers having the molecular formula  $C_4H_8O_3$ . Identify A and B on the basis of their  $^1H$  NMR spectra.

Compound A: δ 1.3 (3H, triplet), 3.6 (2H, quartet), 4.1 (2H, singlet), 11.1 (1H, broad singlet)

Compound B: δ 2.6 (2H, triplet), 3.4 (3H, singlet), 3.7 (2H, triplet), 11.3 (1H, broad singlet)