

Chemistry

Richard "teslafreak" Burris

September 9, 2021

Contents

1	Preface	9
2	Warning	11
2.1	Legal	11
2.2	Safety	12
2.3	Reference materials and resources	13
2.3.1	YouTube channels	13
2.3.2	Websites	13
3	General Safety	15
4	Equipment	17
5	The Elements	19
5.1	1 - Hydrogen	19
5.1.1	Properties	19
5.2	2 - Helium	19
5.2.1	Properties	19
5.3	3 - Lithium	20
5.3.1	Properties	20
5.4	4 - Beryllium	20
5.4.1	Properties	20
5.5	5 - Boron	20
5.5.1	Properties	20
5.6	6 - Carbon	21
5.6.1	Properties	21
5.7	7 - Nitrogen	21
5.7.1	Properties	21
5.8	8 - Oxygen	21
5.8.1	Properties	21
5.9	9 - Fluorine	22
5.9.1	Properties	22
5.10	10 - Neon	22
5.10.1	Properties	22
5.11	11 - Sodium	23
5.11.1	Properties	23
5.12	12 - Magnesium	23
5.12.1	Properties	23
5.13	13 - Aluminium	23
5.13.1	Properties	23
5.14	14 - Silicon	24
5.14.1	Properties	24
5.15	15 - Phosphorus	24
5.15.1	Properties	24
5.16	16 - Sulfur	24
5.16.1	Properties	24
5.17	17 - Chlorine	25
5.17.1	Properties	25
5.18	18 - Argon	25

5.18.1	Properties	25
5.19	19 - Potassium	26
5.19.1	Properties	26
5.20	20 - Calcium	26
5.20.1	Properties	26
5.21	21 - Scandium	26
5.21.1	Properties	26
5.22	22 - Titanium	27
5.22.1	Properties	27
5.23	23 - Vanadium	27
5.23.1	Properties	27
5.24	24 - Chromium	27
5.24.1	Properties	27
5.25	25 - Manganese	28
5.25.1	Properties	28
5.26	26 - Iron	28
5.26.1	Properties	28
5.27	27 - Cobalt	29
5.27.1	Properties	29
5.28	28 - Nickel	29
5.28.1	Properties	29
5.29	29 - Copper	29
5.29.1	Properties	29
5.30	30 - Zinc	30
5.30.1	Properties	30
5.31	31 - Gallium	30
5.31.1	Properties	30
5.32	32 - Germanium	30
5.32.1	Properties	30
5.33	33 - Arsenic	31
5.33.1	Properties	31
5.34	34 - Selenium	31
5.34.1	Properties	31
5.35	35 - Bromine	32
5.35.1	Properties	32
5.36	36 - Krypton	32
5.36.1	Properties	32
5.37	37 - Rubidium	32
5.37.1	Properties	32
5.38	38 - Strontium	33
5.38.1	Properties	33
5.39	39 - Yttrium	33
5.39.1	Properties	33
5.40	40 - Zirconium	33
5.40.1	Properties	33
5.41	41 - Niobium	34
5.41.1	Properties	34
5.42	42 - Molybdenum	34
5.42.1	Properties	34
5.43	43 - Technetium	35
5.43.1	Properties	35
5.44	44 - Ruthenium	35
5.44.1	Properties	35
5.45	45 - Rhodium	35
5.45.1	Properties	35
5.46	46 - Palladium	36
5.46.1	Properties	36
5.47	47 - Silver	36
5.47.1	Properties	36
5.48	48 - Cadmium	36

5.48.1 Properties	36
5.49 49 - Indium	37
5.49.1 Properties	37
5.50 50 - Tin	37
5.50.1 Properties	37
5.51 51 - Antimony	38
5.51.1 Properties	38
5.52 52 - Tellurium	38
5.52.1 Properties	38
5.53 53 - Iodine	38
5.53.1 Properties	38
5.54 54 - Xenon	39
5.54.1 Properties	39
5.55 55 - Caesium	39
5.55.1 Properties	39
5.56 56 - Barium	39
5.56.1 Properties	39
5.57 57 - Lanthanum	40
5.57.1 Properties	40
5.58 58 - Cerium	40
5.58.1 Properties	40
5.59 59 - Praseodymium	41
5.59.1 Properties	41
5.60 60 - Neodymium	41
5.60.1 Properties	41
5.61 61 - Promethium	41
5.61.1 Properties	41
5.62 62 - Samarium	42
5.62.1 Properties	42
5.63 63 - Europium	42
5.63.1 Properties	42
5.64 64 - Gadolinium	42
5.64.1 Properties	42
5.65 65 - Terbium	43
5.65.1 Properties	43
5.66 66 - Dysprosium	43
5.66.1 Properties	43
5.67 67 - Holmium	44
5.67.1 Properties	44
5.68 68 - Erbium	44
5.68.1 Properties	44
5.69 69 - Thulium	44
5.69.1 Properties	44
5.70 70 - Ytterbium	45
5.70.1 Properties	45
5.71 71 - Lutetium	45
5.71.1 Properties	45
5.72 72 - Hafnium	45
5.72.1 Properties	45
5.73 73 - Tantalum	46
5.73.1 Properties	46
5.74 74 - Tungsten	46
5.74.1 Properties	46
5.75 75 - Rhenium	47
5.75.1 Properties	47
5.76 76 - Osmium	47
5.76.1 Properties	47
5.77 77 - Iridium	47
5.77.1 Properties	47
5.78 78 - Platinum	48

5.78.1 Properties	48
5.79 79 - Gold	48
5.79.1 Properties	48
5.80 80 - Mercury	48
5.80.1 Properties	48
5.81 81 - Thallium	49
5.81.1 Properties	49
5.82 82 - Lead	49
5.82.1 Properties	49
5.83 83 - Bismuth	50
5.83.1 Properties	50
5.84 84 - Polonium	50
5.84.1 Properties	50
5.85 85 - Astatine	50
5.85.1 Properties	50
5.86 86 - Radon	51
5.86.1 Properties	51
5.87 87 - Francium	51
5.87.1 Properties	51
5.88 88 - Radium	51
5.88.1 Properties	51
5.89 89 - Actinium	52
5.89.1 Properties	52
5.90 90 - Thorium	52
5.90.1 Properties	52
5.91 91 - Protactinium	53
5.91.1 Properties	53
5.92 92 - Uranium	53
5.92.1 Properties	53
5.93 93 - Neptunium	53
5.93.1 Properties	53
5.94 94 - Plutonium	54
5.94.1 Properties	54
5.95 95 - Americium	54
5.95.1 Properties	54
5.96 96 - Curium	54
5.96.1 Properties	54
5.97 97 - Berkelium	55
5.97.1 Properties	55
5.98 98 - Californium	55
5.98.1 Properties	55
5.99 99 - Einsteinium	56
5.99.1 Properties	56
5.100100 - Fermium	56
5.100.1 Properties	56
5.101101 - Mendelevium	56
5.101.1 Properties	56
5.102102 - Nobelium	57
5.102.1 Properties	57
5.103103 - Lawrencium	57
5.103.1 Properties	57
5.104104 - Rutherfordium	57
5.104.1 Properties	57
5.105105 - Dubnium	58
5.105.1 Properties	58
5.106106 - Seaborgium	58
5.106.1 Properties	58
5.107107 - Bohrium	59
5.107.1 Properties	59
5.108108 - Hassium	59

5.108.1 Properties	59
5.109109 - Meitnerium	59
5.109.1 Properties	59
5.110110 - Darmstadtium	60
5.110.1 Properties	60
5.111111 - Roentgenium	60
5.111.1 Properties	60
5.112112 - Copernicium	60
5.112.1 Properties	60
5.113113 - Nihonium	61
5.113.1 Properties	61
5.114114 - Flerovium	61
5.114.1 Properties	61
5.115115 - Moscovium	62
5.115.1 Properties	62
5.116116 - Livermorium	62
5.116.1 Properties	62
5.117117 - Tennessine	62
5.117.1 Properties	62
5.118118 - Oganesson	63
5.118.1 Properties	63
6 Chemicals	65
7 Composition reference	67
8 CAS reference	69
9 IUPAC reference	71
10 Glossary	73

Chapter 1

Preface

This is an attempt at gathering information from many disparate sources, and gathering them into one handy source. I can try to source things where possible, but it may not always be. I have always been interested in Chemistry, and indeed many facets of science. However, I was not well enough invested learning these things when I was in public school. Luckily in the modern age of the internet, there are many fantastic resources, and you're never too old to learn new things.

That said, I do want to be clear that I am an Information Technology worker by trade. Not a chemist. Not a doctor. Not a physicist.

I will make every reasonable attempt to ensure that all data is accurate, and that any risks I am aware of are stated, but these fields have many inherent risks, and death is a pretty common one. Please see the warning chapter, the general risks chapter, and read any warnings on individual subjects. I can't be held responsible for anything bad that happens, even if you follow these instructions to the letter.

The content of this book is intended to be as wide as I can make it, so it probably won't be a fun read through as a novel (not to mention, I am a terrible writer). It's primary usage should be as a reference.

Where a specific company is named, rest assured that as of this writing, it is a reference only. I am not sponsored by anyone, and have paid my own money for all equipment herein.

This book is maintained as a collection of documents in LaTeX format, which is compiled on a regular basis into PDF form, but if you should find the code itself useful, or wish to check if there are newer versions; please see my [GitHub page](#). Reproduction in part or in whole is not only allowed, it's encouraged.

Lastly, let me just apologize in advance for my poor writing again. Poor sentence structure, grammar, and likely even spelling will probably pepper this document. Corrections are accepted and welcomed in any regard you may have. Even excepting the aforementioned issues, I've always felt I was a poor writer, producing mostly quite dull content. I apologize if things are overly dry.

Chapter 2

Warning

While science can be a great deal of fun, the ways in which you can run into trouble are **numerous**.

2.1 Legal

First and probably worth consideration before you really start in on this: is it legal?

Some countries/states/locations don't really allow you to do anything, and some don't really care as long as you don't directly break laws. The most common thing people worry about seems to be what they could legally make. That is a fair concern, and many places won't really have much concern as long as you don't make drugs, explosives, or meddle in radioactive substances. Some places will even work with you on some of that (notably explosives) as long as your local law enforcement is aware. A number of years back, I received permission to do small (less than a gram) experiments with black powder as long as it was out in the hills away from any homes and I called to let the police know when I was working on it so they would know what was going on if anyone called it in. Your mileage may vary. By a lot. You could have some very understanding local law enforcement that will work with you as long as you are upfront with them and can explain your safety measures. You could get law enforcement that not only says no, but now wishes to keep an eye on you.

Unfortunately, this is a situation that is really only likely to get worse because there is a strange perception that science in general (and things like chemistry and biology in particular) should strictly be the domain of institutions and businesses. If you want to learn this stuff as an individual, it's not uncommon for people to assume the only possible reason is to make illicit substances. This has made things even more convoluted by causing to more very important considerations: Can I even buy (a chemical)? Can I even buy (a piece of equipment)?

Some chemicals that have a massive number of very legitimate uses are restricted or banned (elemental Iodine comes to mind) purely under the concept that you *could* use them to make illicit substances. This is, I will say, an unabashedly stupid policy for much the same reason that you could use nearly any weighted solid object to illegally bludgeon someone to death. Things very much should be consider on the pure and strict basis of what they are used for, not what they could be used for. This is not the reality of the situation though, and as a result, you must be aware of your local laws. I can only speak for the US, as I have only ever lived here. The biggest consideration in this regard will be the DEA controlled chemicals list. You are encouraged to keep familiar with it. For most chemicals, it doesn't outright ban you from having them, but merely sets limits on how much of them you can have. Some states may have additional restrictions, so be aware of that. Probably the biggest problem with this list is that although it shouldn't matter to someone that wants (let's say) a couple grams of a substance on it, it does cause havoc for the chemistry store you may want to purchase from. When they need to order some of these chemicals in bulk, they are required to fill out and maintain paperwork detailing what they are ordering, how much, what for, etc. For many sources, they've decided they just don't want to bother. This means that although it would be okay for you to buy it, you may have a hard time sourcing it. The internet is going to be your friend here.

Additionally, some equipment is actually forbidden. I personally believe this is absurd as well, but I have heard there are places that don't allow jointed glassware. Yes seriously. Apparently the mere act of putting a nice mating joint on a piece of glass makes it malicious. I am not sure what other equipment would be, but I can reasonably guess that this isn't the only example.

So, if you have found that it's legal to have some chemicals, and some equipment, and you aren't planning to break any laws with it; are you good now? You probably won't be shocked to hear that you may still not be. Some (and really many) places have policies that can cause otherwise legal things to become additional charges or evidence should you become legally

entangled. In anything. For instance, where I live (and at this time) in Utah, it is legal to own lock picks. Many places consider lock picks to be burglary tools, and ban them outright, but not Utah. However, in the event that you were say kicked out of a property and the police found that you had some; even if you didn't use them; they are burglary devices now and can act as evidence that you had intent to break and enter. I have heard the same thing can happen where they can magically decide you have chemical pre-cursors or glassware that could be used for crimes, and now they are evidence that you surely must be involved in such crimes. The laws around using such "evidence" are murky and if in doubt, I would strongly suggest speaking to a lawyer. As mentioned in the preface, I am not one.

Legal considerations aside, the biggest suggestion I would make here is to avoid even the appearance of criminality. This may sound cliched, but really, people will be far more likely to trust you if there are no reasons not to. If you have a prior drug charge, and are caught with certain materials; you may well be hosed. It doesn't matter in that case if you are clean now, people will often take the dimmest view of a situation that they can. You should try not to have any real cause for them to do so.

2.2 Safety

I don't want to get too specific here, as I think the better place will be general risks and individual safety instructions. What I do want to point out is that the equipment, substances, and methods herein constitute everything from a completely safe situation with no perceptible risk, all the way up to situations that may have several decent ways to kill you and/or those in the area for a single experiment. It should really go without stating that you **must exercise proper safety procedures**. This can vary widely, and I will give you the best guidance I can. It also pays to read up on your own.

The most general advise would be:

- Always ensure good ventilation. This isn't just for fumes or gasses, it's also not uncommon for dusts to get lofted in to the air. Better safe than sorry. Have you ever heard stories of toxic homes because someone used to make meth there? Same idea. Chemicals can accumulate and become dangerous over time, and it may not even be obvious.
- Always ensure safe storage. Not only so no one gets into things they shouldn't, but also to ensure a leak doesn't involve a bad interaction. Some chemicals also don't store well at certain temperatures. Some store just fine, but will decay faster. Read up on proper storage procedures.
- Always ensure safe disposal. Many chemicals are not safe to put in the garbage or down the drain. The risks can range from just putting something there that shouldn't be, to severely poisoning anything downstream (or groundwater). It also may be illegal, but really that should be a secondary concern to just being a good person.
- Always wear proper attire and PPE (personal protective equipment). At a bare minimum, this is safety glasses. In most circumstances it will also include a lab coat and gloves, maybe a respirator. Some people think the lab coat is just to look cool (and boy does it!) but it's actually there to protect you from spills. Get one, and wear it.
- Always follow all safety procedures. You never know when something will go wrong, and you will be happy that you mitigated it as much as possible.
- Never skip safety "this time". It's very easy to be complacent. It's very easy to decide you just don't want to deal with it "this time" because it's "not a concern". This mentality **kills**. If you don't have time to do it right, don't do it.
- Never leave a project un-supervised. If it's boiling now, it could boil over. If it's un-covered, it could get spilled. Act appropriately.
- Be cautious of anyone else involved. This is everything from making sure people in the area or who may help directly are appropriately safe and knowledgeable; to making sure that if you provide a chemical or equipment to someone, you know them and trust they will use it appropriately. If they don't, you can be legally liable in some circumstances.
- Let someone know what you are doing. In the event that you become incapacitated, this could save your life. I am not saying it's likely (it's quite unlikely if you are being safe) but again, hedging your bets towards safety is good policy.
- Lastly, have some common sense. This is not a hobby to practice while drunk/high/otherwise in an altered state. It's not safe to do risky things when tired either. I won't tell you how to live your life, but being "clean" in this regard will also mitigate your chances of getting involved in a call to law enforcement, which again, is prudent.

Many chemicals will state hazards they could present such as poisonous, flammable, explosive, corrosive, etc. It's important to understand how these properties relate to their concentrations. For instance, Hydrogen Peroxide in it's common

store bought form at 3% is something many people would not hesitate to dump all over an open wound (indeed, that is what it's commonly sold for). If you bring that up to 30%, it will cause chemical burns. Even more concentrated and those burns can be quite severe. Some acids in dilute concentrations present very minor threat to skin when washed off quickly, but will put you in a hospital in concentrated forms (or kill you outright). Some chemicals have no safe limit whatsoever. If you get so much as a drop of methyl-mercury on your skin, you're not long for this world (mercury poisoning will set in, and is fairly universally fatal). Worse still, methyl-mercury goes through gloves, so that won't even protect you. Thankfully methylmercury is something you are unlikely to ever come in contact with, and isn't really a useful chemical to have around either. There are plenty of chemicals that are nasty though, and which you have a good chance of coming across. Some present interesting safety considerations by the way they interact with the environment. You probably know that you shouldn't breath chloroform, but did you know that in regular atmosphere it can turn into phosgene? Phosgene is so utterly nasty that it was used as a chemical weapon. One is kinda bad for you, the other is extremely toxic.

Possibly my favorite example is Nitrogen Trichloride, a great video is [here](#) (this is a video by Tom who runs the YouTube channels [Explosions&Fire](#) and [Extractions&Ire](#), I can't recommend his content highly enough, it's a decent contender for my favorite YouTube channel). For those that can't or don't wish to watch it, the premise is that it's very easy to have it form by accident. When you are in a pool, and something is irritating your eyes, it's probably this (people often suspect it to be chlorine itself, but that is not the case). In this terrifically low concentration, it's merely unpleasant to your eyes. When concentrated, not only is it very poisonous, it's an extraordinarily unstable explosive! To my mind, it's these very sorts of things that add to the fascination of chemistry, but it also serves to remind you that risks are inherent and not always obvious.

2.3 Reference materials and resources

2.3.1 YouTube channels

[ChemicalForce](#)
[Doug's Lab](#)
[Explosions&Fire](#)
[Extractions&Ire](#)
[NileBlue](#)
[NileRed](#)
[Periodic Videos](#)
[Thoisoi2](#)

2.3.2 Websites

[International Union of Pure and Applied Chemistry](#)
[Royal Society of Chemistry](#) - A fantastic resource for information on elements

Chapter 3

General Safety

Chapter 4

Equipment

Chapter 5

The Elements

This chapter will be by atomic number rather than alphabetic.

5.1 1 - Hydrogen

5.1.1 Properties

Melting point -259.16°C, -434.49°F, 13.99 K

Boiling point -252.879°C, -423.182°F, 20.271 K

Density 0.000082 g cm⁻³

Relative atomic mass 1.008

Electron configuration 1s¹

Elemental group 1

Elemental period 1

Elemental block s

Key isotopes ¹H, ²H

CAS number 133-74-0

Atomic radius, non-bonded 1.10 Å

Covalent radius 0.32 Å

Electronegativity (Pauling scale) 2.20

Electron affinity 72.769 (kJ mol⁻¹)

5.2 2 - Helium

5.2.1 Properties

Melting point Unknown

Boiling point -268.928°C, -452.07°F, 4.222 K

Density 0.000164 g cm⁻³

Relative atomic mass 4.003

Electron configuration 1s²

Elemental group 18

Elemental period 1

Elemental block s

Key isotopes ⁴He

CAS number 7440-59-7

Atomic radius, non-bonded 1.400 Å

Covalent radius 0.37 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Not stable

5.3 3 - Lithium

5.3.1 Properties

Melting point 180.50°C, 356.90°F, 453.65 K **Boiling point** 1342°C, 2448°F, 1615 K

Density 0.534 g cm⁻³

Relative atomic mass 6.94

Electron configuration [He] 2s¹

Elemental group 1

Elemental period 2

Elemental block s

Key isotopes ⁷Li

CAS number 7439-93-2

Atomic radius, non-bonded 1.82 Å

Covalent radius 1.30 Å

Electronegativity (Pauling scale) 0.98

Electron affinity 59.633 (kJ mol⁻¹)

5.4 4 - Beryllium

5.4.1 Properties

Melting point 1287°C, 2349°F, 1560 K

Boiling point 2468°C, 4474°F, 2741 K

Density 1.85 g cm⁻³

Relative atomic mass 9.012

Electron configuration [He] 2s²

Elemental group 2

Elemental period 2

Elemental block s

Key isotopes ⁹Be

CAS number 7440-41-7

Atomic radius, non-bonded 1.53 Å

Covalent radius 0.99 Å

Electronegativity (Pauling scale) 1.57

Electron affinity Not stable

5.5 5 - Boron

5.5.1 Properties

Melting point 2077°C, 3771°F, 2350 K

Boiling point 4000°C, 7232°F, 4273 K

Density 2.34 g cm⁻³

Relative atomic mass 10.81

Electron configuration [He] 2s²2p¹

Elemental group 13

Elemental period 2

Elemental block p

Key isotopes ¹¹B

CAS number 7440-42-8

Atomic radius, non-bonded 1.92 Å

Covalent radius 0.84 Å
Electronegativity (Pauling scale) 2.04
Electron affinity 26.989 (kJ mol⁻¹)

5.6 6 - Carbon

5.6.1 Properties

Melting point Sublimes at 3825°C, 6917°F, 4098 K
Boiling point Sublimes at 3825°C, 6917°F, 4098 K

Density 3.513 (diamond) g cm⁻³, 2.2 (graphite) g cm⁻³
Relative atomic mass 12.011
Electron configuration [He] 2s²2p²
Elemental group 14
Elemental period 2
Elemental block p
Key isotopes ¹²C, ¹³C, ¹⁴C
CAS number 7440-44-0

Atomic radius, non-bonded 1.70 Å
Covalent radius 0.75 Å
Electronegativity (Pauling scale) 2.55
Electron affinity 121.776 (kJ mol⁻¹)

5.7 7 - Nitrogen

5.7.1 Properties

Melting point -210.0°C, -346.0°F, 63.2 K
Boiling point -195.795°C, -320.431°F, 77.355 K

Density 0.001145 g cm⁻³
Relative atomic mass 14.007
Electron configuration [He] 2s²2p³
Elemental group 15
Elemental period 2
Elemental block p
Key isotopes ¹⁴N
CAS number 7727-37-9

Atomic radius, non-bonded 1.55 Å
Covalent radius 0.71 Å
Electronegativity (Pauling scale) 3.04
Electron affinity Not stable

5.8 8 - Oxygen

5.8.1 Properties

Melting point -218.79°C, -361.82°F, 54.36 K
Boiling point -182.962°C, -297.332°F, 90.188 K

Density 0.001308 g cm⁻³
Relative atomic mass 15.999
Electron configuration [He] 2s²2p⁴

Elemental group 16
Elemental period 2
Elemental block p
Key isotopes ^{16}O
CAS number 7782-44-7

Atomic radius, non-bonded 1.52 Å
Covalent radius 0.64 Å
Electronegativity (Pauling scale) 3.44
Electron affinity 140.976 (kJ mol⁻¹)

5.9 9 - Fluorine

5.9.1 Properties

Melting point -219.67°C, -363.41°F, 53.48 K
Boiling point -188.11°C, -306.6°F, 85.04 K

Density 0.001553 g cm⁻³
Relative atomic mass 18.998
Electron configuration [He] 2s²2p⁵
Elemental group 17
Elemental period 2
Elemental block p
Key isotopes ^{19}F
CAS number 7782-41-4

Atomic radius, non-bonded 1.47 Å
Covalent radius 0.60 Å
Electronegativity (Pauling scale) 3.98
Electron affinity 328.165 (kJ mol⁻¹)

5.10 10 - Neon

5.10.1 Properties

Melting point -248.59°C, -415.46°F, 24.56 K
Boiling point -246.046°C, -410.883°F, 27.104 K

Density 0.000825 g cm⁻³
Relative atomic mass 20.180
Electron configuration [He] 2s²2p⁶
Elemental group 18
Elemental period 2
Elemental block p
Key isotopes ^{20}Ne
CAS number 7440-01-9

Atomic radius, non-bonded 1.54 Å
Covalent radius 0.62 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Not stable

5.11 11 - Sodium

5.11.1 Properties

Melting point 97.794°C, 208.029°F, 370.944 K

Boiling point 882.940°C, 1621.292°F, 1156.090 K

Density 0.97 g cm⁻³

Relative atomic mass 22.990

Electron configuration [Ne] 3s¹

Elemental group 1

Elemental period 3

Elemental block s

Key isotopes ²³Na

CAS number 7440-23-5

Atomic radius, non-bonded 2.27 Å

Covalent radius 1.60 Å

Electronegativity (Pauling scale) 0.93

Electron affinity 52.867 (kJ mol⁻¹)

5.12 12 - Magnesium

5.12.1 Properties

Melting point 650°C, 1202°F, 923 K

Boiling point 1090°C, 1994°F, 1363 K

Density 1.74 g cm⁻³

Relative atomic mass 24.305

Electron configuration [Ne] 3s²

Elemental group 2

Elemental period 3

Elemental block s

Key isotopes ²⁴Mg

CAS number 7439-95-4

Atomic radius, non-bonded 1.73 Å

Covalent radius 1.40 Å

Electronegativity (Pauling scale) 1.31

Electron affinity Not stable

5.13 13 - Aluminium

Though commonly spelled Aluminum (without the second "i") in America, that is actually not accurate to the officially recognized spelling.

5.13.1 Properties

Melting point 660.323°C, 1220.581°F, 933.473 K

Boiling point 2519°C, 4566°F, 2792 K

Density 2.70 g cm⁻³

Relative atomic mass 26.982

Electron configuration [Ne] 3s²3p¹

Elemental group 13

Elemental period 3

Elemental block p

Key isotopes ^{27}Al
CAS number 7429-90-5

Atomic radius, non-bonded 1.84 Å
Covalent radius 1.24 Å
Electronegativity (Pauling scale) 1.61
Electron affinity 41.762 (kJ mol⁻¹)

5.14 14 - Silicon

5.14.1 Properties

Melting point 1414°C, 2577°F, 1687 K
Boiling point 3265°C, 5909°F, 3538 K

Density 2.3296 g cm⁻³
Relative atomic mass 28.085
Electron configuration [Ne] 3s²3p²
Elemental group 14
Elemental period 3
Elemental block p
Key isotopes ^{28}Si , ^{30}Si
CAS number 7440-21-3

Atomic radius, non-bonded 2.10 Å
Covalent radius 1.14 Å
Electronegativity (Pauling scale) 1.90
Electron affinity 134.068 (kJ mol⁻¹)

5.15 15 - Phosphorus

5.15.1 Properties

Melting point 44.15°C, 111.47°F, 317.3 K
Boiling point 280.5°C, 536.9°F, 553.7 K

Density 1.823 g cm⁻³ (white)
Relative atomic mass 30.974
Electron configuration [Ne] 3s²3p³
Elemental group 15
Elemental period 3
Elemental block p
Key isotopes ^{31}P
CAS number 7723-14-0

Atomic radius, non-bonded 1.80 Å
Covalent radius 1.09 Å
Electronegativity (Pauling scale) 2.19
Electron affinity 72.037 (kJ mol⁻¹)

5.16 16 - Sulfur

5.16.1 Properties

Melting point 115.21°C, 239.38°F, 388.36 K
Boiling point 444.61°C, 832.3°F, 717.76 K

Density 2.07 g cm⁻³
Relative atomic mass 32.06
Electron configuration [Ne] 3s²3p⁴
Elemental group 16
Elemental period 3
Elemental block p
Key isotopes ³²S
CAS number 7704-34-9

Atomic radius, non-bonded 1.80 Å
Covalent radius 1.04 Å
Electronegativity (Pauling scale) 2.58
Electron affinity 200.41 (kJ mol⁻¹)

5.17 17 - Chlorine

5.17.1 Properties

Melting point -101.5°C, -150.7°F, 171.7 K
Boiling point -34.04°C, -29.27°F, 239.11 K

Density 0.002898 g cm⁻³
Relative atomic mass 35.45
Electron configuration [Ne] 3s²3p⁵
Elemental group 17
Elemental period 3
Elemental block p
Key isotopes ³⁵Cl, ³⁷Cl
CAS number 7782-50-5

Atomic radius, non-bonded 1.75 Å
Covalent radius 1.00 Å
Electronegativity (Pauling scale) 3.16
Electron affinity 348.575 (kJ mol⁻¹)

5.18 18 - Argon

5.18.1 Properties

Melting point -189.34°C, -308.81°F, 83.81 K
Boiling point -185.848°C, -302.526°F, 87.302 K

Density 0.001633 g cm⁻³
Relative atomic mass 39.95
Electron configuration [Ne] 3s²3p⁶
Elemental group 18
Elemental period 3
Elemental block p
Key isotopes ⁴⁰Ar
CAS number 7440-37-1

Atomic radius, non-bonded 1.88 Å
Covalent radius 1.01 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Not stable

5.19 19 - Potassium

5.19.1 Properties

Melting point 63.5°C, 146.3°F, 336.7 K

Boiling point 759°C, 1398°F, 1032 K

Density 0.89 g cm⁻³

Relative atomic mass 39.098

Electron configuration [Ar] 4s¹

Elemental group 1

Elemental period 4

Elemental block s

Key isotopes ³⁹K

CAS number 7440-09-7

Atomic radius, non-bonded 2.75 Å

Covalent radius 2.00 Å

Electronegativity (Pauling scale) 0.82

Electron affinity 48.385 (kJ mol⁻¹)

5.20 20 - Calcium

5.20.1 Properties

Melting point 842°C, 1548°F, 1115 K

Boiling point 1484°C, 2703°F, 1757 K

Density 1.54 g cm⁻³

Relative atomic mass 40.078

Electron configuration [Ar] 4s²

Elemental group 2

Elemental period 4

Elemental block s

Key isotopes ⁴⁰Ca

CAS number 7440-70-2

Atomic radius, non-bonded 2.31 Å

Covalent radius 1.74 Å

Electronegativity (Pauling scale) 1.00

Electron affinity 2.369 (kJ mol⁻¹)

5.21 21 - Scandium

5.21.1 Properties

Melting point 1541°C, 2806°F, 1814 K

Boiling point 2836°C, 5137°F, 3109 K

Density 2.99 g cm⁻³

Relative atomic mass 44.956

Electron configuration [Ar] 3d¹4s²

Elemental group 3

Elemental period 4

Elemental block d

Key isotopes ⁴⁵Sc

CAS number 7440-20-2

Atomic radius, non-bonded 2.15 Å
Covalent radius 1.59 Å
Electronegativity (Pauling scale) 1.36
Electron affinity 18.139 (kJ mol⁻¹)

5.22 22 - Titanium

5.22.1 Properties

Melting point 1670°C, 3038°F, 1943 K
Boiling point 3287°C, 5949°F, 3560 K

Density 4.506 g cm⁻³
Relative atomic mass 47.867
Electron configuration [Ar] 3d²4s²
Elemental group 4
Elemental period 4
Elemental block d
Key isotopes ⁴⁸Ti
CAS number 7440-32-6

Atomic radius, non-bonded 2.11 Å
Covalent radius 1.48 Å
Electronegativity (Pauling scale) 1.54
Electron affinity 7.622 (kJ mol⁻¹)

5.23 23 - Vanadium

5.23.1 Properties

Melting point 1910°C, 3470°F, 2183 K
Boiling point 3407°C, 6165°F, 3680 K

Density 6.0 g cm⁻³
Relative atomic mass 50.942
Electron configuration [Ar] 3d³4s²
Elemental group 5
Elemental period 4
Elemental block d
Key isotopes ⁵¹V
CAS number 7440-62-2

Atomic radius, non-bonded 2.07 Å
Covalent radius 1.44 Å
Electronegativity (Pauling scale) 1.63
Electron affinity 50.655 (kJ mol⁻¹)

5.24 24 - Chromium

5.24.1 Properties

Melting point 1907°C, 3465°F, 2180 K
Boiling point 2671°C, 4840°F, 2944 K

Density 7.15 g cm⁻³
Relative atomic mass 51.996

Electron configuration [Ar] 3d⁵4s¹
Elemental group 6
Elemental period 4
Elemental block d
Key isotopes ⁵²Cr
CAS number 7440-47-3

Atomic radius, non-bonded 2.06 Å
Covalent radius 1.30 Å
Electronegativity (Pauling scale) 1.66
Electron affinity 64.259 (kJ mol⁻¹)

5.25 25 - Manganese

5.25.1 Properties

Melting point 1246°C, 2275°F, 1519 K
Boiling point 2061°C, 3742°F, 2334 K

Density 7.3 g cm⁻³
Relative atomic mass 54.938
Electron configuration [Ar] 3d⁵4s²
Elemental group 7
Elemental period 4
Elemental block d
Key isotopes ⁵⁵Mn
CAS number 7439-96-5

Atomic radius, non-bonded 2.05 Å
Covalent radius 1.29 Å
Electronegativity (Pauling scale) 1.55
Electron affinity Not stable

5.26 26 - Iron

5.26.1 Properties

Melting point 1538°C, 2800°F, 1811 K
Boiling point 2861°C, 5182°F, 3134 K

Density 7.87 g cm⁻³
Relative atomic mass 55.845
Electron configuration [Ar] 3d⁶4s²
Elemental group 8
Elemental period 4
Elemental block d
Key isotopes ⁵⁶Fe
CAS number 7439-89-6

Atomic radius, non-bonded 2.04 Å
Covalent radius 1.24 Å
Electronegativity (Pauling scale) 1.83
Electron affinity 14.569 (kJ mol⁻¹)

5.27 27 - Cobalt

5.27.1 Properties

Melting point 1495°C, 2723°F, 1768 K

Boiling point 2927°C, 5301°F, 3200 K

Density 8.86 g cm⁻³

Relative atomic mass 58.933

Electron configuration [Ar] 3d⁷4s²

Elemental group 9

Elemental period 4

Elemental block d

Key isotopes ⁵⁹Co

CAS number 7440-48-4

Atomic radius, non-bonded 2.00 Å

Covalent radius 1.18 Å

Electronegativity (Pauling scale) 1.88

Electron affinity 63.873 (kJ mol⁻¹)

5.28 28 - Nickel

5.28.1 Properties

Melting point 1455°C, 2651°F, 1728 K

Boiling point 2913°C, 5275°F, 3186 K

Density 8.90 g cm⁻³

Relative atomic mass 58.693

Electron configuration [Ar] 3d⁸4s²

Elemental group 10

Elemental period 4

Elemental block d

Key isotopes ⁵⁸Ni

CAS number 7440-02-0

Atomic radius, non-bonded 1.97 Å

Covalent radius 1.17 Å

Electronegativity (Pauling scale) 1.91

Electron affinity 111.537 (kJ mol⁻¹)

5.29 29 - Copper

5.29.1 Properties

Melting point 1084.62°C, 1984.32°F, 1357.77 K

Boiling point 2560°C, 4640°F, 2833 K

Density 8.96 g cm⁻³

Relative atomic mass 63.546

Electron configuration [Ar] 3d¹⁰4s¹

Elemental group 11

Elemental period 4

Elemental block d

Key isotopes ⁶³Cu

CAS number 7440-50-8

Atomic radius, non-bonded 1.96 Å
Covalent radius 1.22 Å
Electronegativity (Pauling scale) 1.90
Electron affinity 119.159 (kJ mol⁻¹)

5.30 30 - Zinc

5.30.1 Properties

Melting point 419.527°C, 787.149°F, 692.677 K
Boiling point 907°C, 1665°F, 1180 K

Density 7.134 g cm⁻³
Relative atomic mass 65.38
Electron configuration [Ar] 3d¹⁰4s²
Elemental group 12
Elemental period 4
Elemental block d
Key isotopes ⁶⁴Zn
CAS number 7440-66-6

Atomic radius, non-bonded 2.01 Å
Covalent radius 1.20 Å
Electronegativity (Pauling scale) 1.65
Electron affinity Not stable

5.31 31 - Gallium

Much like water, this is one of the few materials that actually expands when it freezes rather than contracting. For this reason, it is not recommended to store it in glass containers.

5.31.1 Properties

Melting point 29.7646°C, 85.5763°F, 302.9146 K
Boiling point 2229°C, 4044°F, 2502 K

Density 5.91 g cm⁻³
Relative atomic mass 69.723
Electron configuration [Ar] 3d¹⁰4s²4p¹
Elemental group 13
Elemental period 4
Elemental block p
Key isotopes ⁶⁹Ga
CAS number 7440-55-3

Atomic radius, non-bonded 1.87 Å
Covalent radius 1.23 Å
Electronegativity (Pauling scale) 1.81
Electron affinity 41.49 (kJ mol⁻¹)

5.32 32 - Germanium

5.32.1 Properties

Melting point 938.25°C, 1720.85°F, 1211.4 K
Boiling point 2833°C, 5131°F, 3106 K

Density 5.3234 g cm⁻³
Relative atomic mass 72.630
Electron configuration [Ar] 3d¹⁰4s²4p²
Elemental group 14
Elemental period 4
Elemental block p
Key isotopes ⁷³Ge, ⁷⁴Ge
CAS number 7440-56-4

Atomic radius, non-bonded 2.11 Å
Covalent radius 1.20 Å
Electronegativity (Pauling scale) 2.01
Electron affinity 118.939 (kJ mol⁻¹)

5.33 33 - Arsenic

5.33.1 Properties

Melting point Sublimes at 616°C, 1141°F, 889 K
Boiling point Sublimes at 616°C, 1141°F, 889 K

Density 5.75 g cm⁻³
Relative atomic mass 74.922
Electron configuration [Ar] 3d¹⁰4s²4p³
Elemental group 15
Elemental period 4
Elemental block p
Key isotopes ⁷⁵As
CAS number 7440-38-2

Atomic radius, non-bonded 1.85 Å
Covalent radius 1.20 Å
Electronegativity (Pauling scale) 2.18
Electron affinity 77.574 (kJ mol⁻¹)

5.34 34 - Selenium

5.34.1 Properties

Melting point 220.8°C, 429.4°F, 494 K
Boiling point 685°C, 1265°F, 958 K

Density 4.809 g cm⁻³
Relative atomic mass 78.971
Electron configuration [Ar] 3d¹⁰4s²4p⁴
Elemental group 16
Elemental period 4
Elemental block p
Key isotopes ⁸⁰Se
CAS number 7782-49-2

Atomic radius, non-bonded 1.90 Å
Covalent radius 1.18 Å
Electronegativity (Pauling scale) 2.55
Electron affinity 194.965 (kJ mol⁻¹)

5.35 35 - Bromine

5.35.1 Properties

Melting point -7.2°C, 19°F, 266 K

Boiling point 58.8°C, 137.8°F, 332 K

Density 3.1028 g cm⁻³

Relative atomic mass 79.904

Electron configuration [Ar] 3d¹⁰4s²4p⁵

Elemental group 17

Elemental period 4

Elemental block p

Key isotopes ⁷⁹Br

CAS number 7726-95-6

Atomic radius, non-bonded 1.85 Å

Covalent radius 1.17 Å

Electronegativity (Pauling scale) 2.96

Electron affinity 324.537 (kJ mol⁻¹)

5.36 36 - Krypton

5.36.1 Properties

Melting point -157.37°C, -251.27°F, 115.78 K

Boiling point -153.415°C, -244.147°F, 119.735 K

Density 0.003425 g cm⁻³

Relative atomic mass 83.798

Electron configuration [Ar] 3d¹⁰4s²4p⁶

Elemental group 18

Elemental period 4

Elemental block p

Key isotopes ⁸⁴Kr

CAS number 7439-90-9

Atomic radius, non-bonded 2.02 Å

Covalent radius 1.16 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Not stable

5.37 37 - Rubidium

5.37.1 Properties

Melting point 39.30°C, 102.74°F, 312.45 K

Boiling point 688°C, 1270°F, 961 K

Density 1.53 g cm⁻³

Relative atomic mass 85.468

Electron configuration [Kr] 5s¹

Elemental group 1

Elemental period 5

Elemental block s

Key isotopes ⁸⁵Rb, ⁸⁷Rb

CAS number 7440-17-7

Atomic radius, non-bonded 3.03 Å
Covalent radius 2.15 Å
Electronegativity (Pauling scale) 0.82
Electron affinity 46.884 (kJ mol⁻¹)

5.38 38 - Strontium

5.38.1 Properties

Melting point 777°C, 1431°F, 1050 K
Boiling point 1377°C, 2511°F, 1650 K

Density 2.64 g cm⁻³
Relative atomic mass 87.62
Electron configuration [Kr] 5s²
Elemental group 2
Elemental period 5
Elemental block s
Key isotopes ⁸⁶Sr, ⁸⁷Sr, ⁸⁸Sr
CAS number 7440-24-6

Atomic radius, non-bonded 2.49 Å
Covalent radius 1.90 Å
Electronegativity (Pauling scale) 0.95
Electron affinity 4.631 (kJ mol⁻¹)

5.39 39 - Yttrium

5.39.1 Properties

Melting point 1522°C, 2772°F, 1795 K
Boiling point 3345°C, 6053°F, 3618 K

Density 4.47 g cm⁻³
Relative atomic mass 88.906
Electron configuration [Kr] 4d¹5s²
Elemental group 3
Elemental period 5
Elemental block d
Key isotopes ⁸⁹Y
CAS number 7440-65-5

Atomic radius, non-bonded 2.32 Å
Covalent radius 1.76 Å
Electronegativity (Pauling scale) 1.22
Electron affinity 29.621 (kJ mol⁻¹)

5.40 40 - Zirconium

5.40.1 Properties

Melting point 1854°C, 3369°F, 2127 K
Boiling point 4406°C, 7963°F, 4679 K

Density 6.52 g cm⁻³
Relative atomic mass 91.224

Electron configuration [Kr] 4d²5s²
Elemental group 4
Elemental period 5
Elemental block d
Key isotopes ⁹⁰Zr, ⁹²Zr, ⁹⁴Zr
CAS number 7440-67-7

Atomic radius, non-bonded 2.23 Å
Covalent radius 1.64 Å
Electronegativity (Pauling scale) 1.33
Electron affinity 41.103 (kJ mol⁻¹)

5.41 41 - Niobium

5.41.1 Properties

Melting point 2477°C, 4491°F, 2750 K
Boiling point 4741°C, 8566°F, 5014 K

Density 8.57 g cm⁻³
Relative atomic mass 92.906
Electron configuration [Kr] 4d⁴5s¹
Elemental group 5
Elemental period 5
Elemental block d
Key isotopes ⁹³Nb
CAS number 7440-03-1

Atomic radius, non-bonded 2.18 Å
Covalent radius 1.56 Å
Electronegativity (Pauling scale) 1.6
Electron affinity 88.381 (kJ mol⁻¹)

5.42 42 - Molybdenum

5.42.1 Properties

Melting point 2622°C, 4752°F, 2895 K
Boiling point 4639°C, 8382°F, 4912 K

Density 10.2 g cm⁻³
Relative atomic mass 95.95
Electron configuration [Kr] 4d⁵5s¹
Elemental group 6
Elemental period 5
Elemental block d
Key isotopes ⁹⁵Mo, ⁹⁶Mo, ⁹⁸Mo
CAS number 7439-98-7

Atomic radius, non-bonded 2.17 Å
Covalent radius 1.46 Å
Electronegativity (Pauling scale) 2.16
Electron affinity 72.171 (kJ mol⁻¹)

5.43 43 - Technetium

5.43.1 Properties

Melting point 2157°C, 3915°F, 2430 K

Boiling point 4262°C, 7704°F, 4535 K

Density 11 g cm⁻³

Relative atomic mass [98]

Electron configuration [Kr] 4d⁵5s²

Elemental group 7

Elemental period 5

Elemental block d

Key isotopes Unknown

CAS number 7440-26-8

Atomic radius, non-bonded 2.16 Å

Covalent radius 1.38 Å

Electronegativity (Pauling scale) 2.10

Electron affinity 53.07 (kJ mol⁻¹)

5.44 44 - Ruthenium

5.44.1 Properties

Melting point 2333°C, 4231°F, 2606 K

Boiling point 4147°C, 7497°F, 4420 K

Density 12.1 g cm⁻³

Relative atomic mass 101.07

Electron configuration [Kr] 4d⁷5s¹

Elemental group 8

Elemental period 5

Elemental block d

Key isotopes ¹⁰¹Ru, ¹⁰²Ru, ¹⁰⁴Ru

CAS number 7440-18-8

Atomic radius, non-bonded 2.13 Å

Covalent radius 1.36 Å

Electronegativity (Pauling scale) 2.2

Electron affinity 101.31 (kJ mol⁻¹)

5.45 45 - Rhodium

5.45.1 Properties

Melting point 1963°C, 3565°F, 2236 K

Boiling point 3695°C, 6683°F, 3968 K

Density 12.4 g cm⁻³

Relative atomic mass 102.906

Electron configuration [Kr] 4d⁸5s¹

Elemental group 9

Elemental period 5

Elemental block d

Key isotopes ¹⁰³Rh

CAS number 7440-16-6

Atomic radius, non-bonded 2.10 Å
Covalent radius 1.34 Å
Electronegativity (Pauling scale) 2.28
Electron affinity 109.704 (kJ mol⁻¹)

5.46 46 - Palladium

5.46.1 Properties

Melting point 1554.8°C, 2830.6°F, 1828 K
Boiling point 2963°C, 5365°F, 3236 K

Density 12.0 g cm⁻³
Relative atomic mass 106.42
Electron configuration [Kr] 4d¹⁰
Elemental group 10
Elemental period 5
Elemental block d
Key isotopes ¹⁰⁶Pd
CAS number 7440-05-3

Atomic radius, non-bonded 2.10 Å
Covalent radius 1.30 Å
Electronegativity (Pauling scale) 2.20
Electron affinity 54.225 (kJ mol⁻¹)

5.47 47 - Silver

5.47.1 Properties

Melting point 961.78°C, 1763.2°F, 1234.93 K
Boiling point 2162°C, 3924°F, 2435 K

Density 10.5 g cm⁻³
Relative atomic mass 107.868
Electron configuration [Kr] 4d¹⁰5s¹
Elemental group 11
Elemental period 5
Elemental block d
Key isotopes ¹⁰⁷Ag
CAS number 7440-22-4

Atomic radius, non-bonded 2.11 Å
Covalent radius 1.36 Å
Electronegativity (Pauling scale) 1.93
Electron affinity 125.624 (kJ mol⁻¹)

5.48 48 - Cadmium

5.48.1 Properties

Melting point 321.069°C, 609.924°F, 594.219 K
Boiling point 767°C, 1413°F, 1040 K

Density 8.69 g cm⁻³
Relative atomic mass 112.414

Electron configuration [Kr] 4d¹⁰5s²
Elemental group 12
Elemental period 5
Elemental block d
Key isotopes ¹¹⁴Cd
CAS number 7440-43-9

Atomic radius, non-bonded 2.18 Å
Covalent radius 1.40 Å
Electronegativity (Pauling scale) 1.69
Electron affinity Not stable

5.49 49 - Indium

5.49.1 Properties

Melting point 156.60°C, 313.88°F, 429.75 K
Boiling point 2027°C, 3681°F, 2300 K

Density 7.31 g cm⁻³
Relative atomic mass 114.818
Electron configuration [Kr] 4d¹⁰5s²5p¹
Elemental group 13
Elemental period 5
Elemental block p
Key isotopes ¹¹⁵In
CAS number 7440-74-6

Atomic radius, non-bonded 1.93 Å
Covalent radius 1.42 Å
Electronegativity (Pauling scale) 1.78
Electron affinity 28.9 (kJ mol⁻¹)

5.50 50 - Tin

5.50.1 Properties

Melting point 231.928°C, 449.47°F, 505.078 K
Boiling point 2586°C, 4687°F, 2859 K

Density 7.287 g cm⁻³
Relative atomic mass 118.710
Electron configuration [Kr] 4d¹⁰5s²5p²
Elemental group 14
Elemental period 5
Elemental block p
Key isotopes ¹²⁰Sn
CAS number 7440-31-5

Atomic radius, non-bonded 2.17 Å
Covalent radius 1.40 Å
Electronegativity (Pauling scale) 1.96
Electron affinity 107.298 (kJ mol⁻¹)

5.51 51 - Antimony

5.51.1 Properties

Melting point 630.628°C, 1167.13°F, 903.778 K

Boiling point 1587°C, 2889°F, 1860 K

Density 6.68 g cm⁻³

Relative atomic mass 121.760

Electron configuration [Kr] 4d¹⁰5s²5p³

Elemental group 15

Elemental period 5

Elemental block p

Key isotopes ¹²¹Sb

CAS number 7440-36-0

Atomic radius, non-bonded 2.06 Å

Covalent radius 1.40 Å

Electronegativity (Pauling scale) 2.05

Electron affinity 100.924 (kJ mol⁻¹)

5.52 52 - Tellurium

5.52.1 Properties

Melting point 449.51°C, 841.12°F, 722.66 K

Boiling point 988°C, 1810°F, 1261 K

Density 6.232 g cm⁻³

Relative atomic mass 127.60

Electron configuration [Kr] 4d¹⁰5s²5p⁴

Elemental group 16

Elemental period 5

Elemental block p

Key isotopes ¹³⁰Te

CAS number 13494-80-9

Atomic radius, non-bonded 2.06 Å

Covalent radius 1.37 Å

Electronegativity (Pauling scale) 2.1

Electron affinity 190.161 (kJ mol⁻¹)

5.53 53 - Iodine

5.53.1 Properties

Melting point 113.7°C, 236.7°F, 386.9 K

Boiling point 184.4°C, 363.9°F, 457.6 K

Density 4.933 g cm⁻³

Relative atomic mass 126.904

Electron configuration [Kr] 4d¹⁰5s²5p⁵

Elemental group 17

Elemental period 5

Elemental block p

Key isotopes ¹²⁷I

CAS number 7553-56-2

Atomic radius, non-bonded 1.98 Å
Covalent radius 1.36 Å
Electronegativity (Pauling scale) 1.36
Electron affinity 295.152 (kJ mol⁻¹)

5.54 54 - Xenon

5.54.1 Properties

Melting point -111.75°C, -169.15°F, 161.4 K
Boiling point -108.099°C, -162.578°F, 165.051 K

Density 0.005366 g cm⁻³
Relative atomic mass 131.293
Electron configuration [Kr] 4d¹⁰5s²5p⁶
Elemental group 18
Elemental period 5
Elemental block p
Key isotopes ¹³²Xe
CAS number 7440-63-3

Atomic radius, non-bonded 2.16 Å
Covalent radius 1.36 Å
Electronegativity (Pauling scale) 2.60
Electron affinity Not stable

5.55 55 - Caesium

5.55.1 Properties

Melting point 28.5°C, 83.3°F, 301.7 K
Boiling point 671°C, 1240°F, 944 K

Density 1.873 g cm⁻³
Relative atomic mass 132.905
Electron configuration [Xe] 6s¹
Elemental group 1
Elemental period 6
Elemental block s
Key isotopes ¹³³Cs
CAS number 7440-46-2

Atomic radius, non-bonded 3.43 Å
Covalent radius 2.38 Å
Electronegativity (Pauling scale) 0.79
Electron affinity 45.505 (kJ mol⁻¹)

5.56 56 - Barium

5.56.1 Properties

Melting point 727°C, 1341°F, 1000 K
Boiling point 1845°C, 3353°F, 2118 K

Density 3.62 g cm⁻³
Relative atomic mass 137.327

Electron configuration [Xe] 6s²
Elemental group 2
Elemental period 6
Elemental block s
Key isotopes ¹³⁸Ba
CAS number 7440-39-3

Atomic radius, non-bonded 2.68 Å
Covalent radius 2.06 Å
Electronegativity (Pauling scale) 0.89
Electron affinity 13.954 (kJ mol⁻¹)

5.57 57 - Lanthanum

5.57.1 Properties

Melting point 920°C, 1688°F, 1193 K
Boiling point 3464°C, 6267°F, 3737 K

Density 6.15 g cm⁻³
Relative atomic mass 138.905
Electron configuration [Xe] 5d¹6s²
Elemental group Lanthanides
Elemental period 6
Elemental block d
Key isotopes ¹³⁹La
CAS number 7439-91-0

Atomic radius, non-bonded 2.43 Å
Covalent radius 1.94 Å
Electronegativity (Pauling scale) 1.10
Electron affinity 45.35 (kJ mol⁻¹)

5.58 58 - Cerium

5.58.1 Properties

Melting point 799°C, 1470°F, 1072 K
Boiling point 3443°C, 6229°F, 3716 K

Density 6.77 g cm⁻³
Relative atomic mass 140.116
Electron configuration [Xe] 4f¹5d¹6s²
Elemental group Lanthanides
Elemental period 6
Elemental block f
Key isotopes ¹⁴⁰Ce
CAS number 7440-45-1

Atomic radius, non-bonded 2.42 Å
Covalent radius 1.84 Å
Electronegativity (Pauling scale) 1.12
Electron affinity 62.72 (kJ mol⁻¹)

5.59 59 - Praseodymium

5.59.1 Properties

Melting point 931°C, 1708°F, 1204 K

Boiling point 3520°C, 6368°F, 3793 K

Density 6.77 g cm⁻³

Relative atomic mass 140.908

Electron configuration [Xe] 4f³6s²

Elemental group Lanthanides

Elemental period 6

Elemental block f

Key isotopes ¹⁴¹Pr

CAS number 7440-10-0

Atomic radius, non-bonded 2.40 Å

Covalent radius 1.90 Å

Electronegativity (Pauling scale) 1.13

Electron affinity 92.819 (kJ mol⁻¹)

5.60 60 - Neodymium

5.60.1 Properties

Melting point 1016°C, 1861°F, 1289 K

Boiling point 3074°C, 5565°F, 3347 K

Density 7.01 g cm⁻³

Relative atomic mass 144.242

Electron configuration [Xe] 4f⁴6s²

Elemental group Lanthanides

Elemental period 6

Elemental block f

Key isotopes ¹⁴²Nd

CAS number 7440-00-8

Atomic radius, non-bonded 2.39 Å

Covalent radius 1.88 Å

Electronegativity (Pauling scale) 1.14

Electron affinity Unknown

5.61 61 - Promethium

5.61.1 Properties

Melting point 1042°C, 1908°F, 1315 K

Boiling point 3000°C, 5432°F, 3273 K

Density 7.26 g cm⁻³

Relative atomic mass [145]

Electron configuration [Xe] 4f⁵6s²

Elemental group Lanthanides

Elemental period 6

Elemental block f

Key isotopes ¹⁴⁵Pm, ¹⁴⁷Pm

CAS number 7440-12-2

Atomic radius, non-bonded 2.38 Å
Covalent radius 1.86 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Unknown

5.62 62 - Samarium

5.62.1 Properties

Melting point 1072°C, 1962°F, 1345 K
Boiling point 1794°C, 3261°F, 2067 K

Density 7.52 g cm⁻³
Relative atomic mass 150.36
Electron configuration [Xe] 4f⁶6s²
Elemental group Lanthanides
Elemental period 6
Elemental block f
Key isotopes ¹⁵²Sm
CAS number 7440-19-9

Atomic radius, non-bonded 2.36 Å
Covalent radius 1.85 Å
Electronegativity (Pauling scale) 1.17
Electron affinity Unknown

5.63 63 - Europium

5.63.1 Properties

Melting point 822°C, 1512°F, 1095 K
Boiling point 1529°C, 2784°F, 1802 K

Density 5.24 g cm⁻³
Relative atomic mass 151.964
Electron configuration [Xe] 4f⁷6s²
Elemental group Lanthanides
Elemental period 6
Elemental block f
Key isotopes ¹⁵³Eu
CAS number 7440-53-1

Atomic radius, non-bonded 2.35 Å
Covalent radius 1.83 Å
Electronegativity (Pauling scale) Unknown
Electron affinity 83.363 (kJ mol⁻¹)

5.64 64 - Gadolinium

5.64.1 Properties

Melting point 1313°C, 2395°F, 1586 K
Boiling point 3273°C, 5923°F, 3546 K

Density 7.90 g cm⁻³
Relative atomic mass 157.25

Electron configuration [Xe] 4f⁷5d¹6s²
Elemental group Lanthanides
Elemental period 6
Elemental block f
Key isotopes ¹⁵⁸Gd
CAS number 7440-54-2

Atomic radius, non-bonded 2.34 Å
Covalent radius 1.82 Å
Electronegativity (Pauling scale) 1.20
Electron affinity Unknown

5.65 65 - Terbium

5.65.1 Properties

Melting point 1359°C, 2478°F, 1632 K
Boiling point 3230°C, 5846°F, 3503 K

Density 8.23 g cm⁻³
Relative atomic mass 158.925
Electron configuration [Xe] 4f⁹6s²
Elemental group Lanthanides
Elemental period 6
Elemental block f
Key isotopes ¹⁵⁹Tb
CAS number 7440-27-9

Atomic radius, non-bonded 2.33 Å
Covalent radius 1.81 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Unknown

5.66 66 - Dysprosium

5.66.1 Properties

Melting point 1412°C, 2574°F, 1685 K
Boiling point 2567°C, 4653°F, 2840 K

Density 8.55 g cm⁻³
Relative atomic mass 162.500
Electron configuration [Xe] 4f¹⁰6s²
Elemental group Lanthanides
Elemental period 6
Elemental block f
Key isotopes ¹⁶⁴Dy
CAS number 7429-91-6

Atomic radius, non-bonded 2.31 Å
Covalent radius 1.80 Å
Electronegativity (Pauling scale) 1.22
Electron affinity Unknown

5.67 67 - Holmium

5.67.1 Properties

Melting point 1472°C, 2682°F, 1745 K

Boiling point 2700°C, 4892°F, 2973 K

Density 8.80 g cm⁻³

Relative atomic mass 164.930

Electron configuration [Xe] 4f¹¹6s²

Elemental group Lanthanides

Elemental period 6

Elemental block f

Key isotopes ¹⁶⁵Ho

CAS number 7440-60-0

Atomic radius, non-bonded 2.30 Å

Covalent radius 1.79 Å

Electronegativity (Pauling scale) 1.23

Electron affinity Unknown

5.68 68 - Erbium

5.68.1 Properties

Melting point 1529°C, 2784°F, 1802 K

Boiling point 2868°C, 5194°F, 3141 K

Density 9.07 g cm⁻³

Relative atomic mass 167.259

Electron configuration [Xe] 4f¹²6s²

Elemental group Lanthanides

Elemental period 6

Elemental block f

Key isotopes ¹⁶⁶Er

CAS number 7440-52-0

Atomic radius, non-bonded 2.29 Å

Covalent radius 1.77 Å

Electronegativity (Pauling scale) 1.24

Electron affinity Unknown (kJ mol⁻¹)

5.69 69 - Thulium

5.69.1 Properties

Melting point 1545°C, 2813°F, 1818 K

Boiling point 1950°C, 3542°F, 2223 K

Density 9.32 g cm⁻³

Relative atomic mass 168.934

Electron configuration [Xe] 4f¹³6s²

Elemental group Lanthanides

Elemental period 6

Elemental block f

Key isotopes ¹⁶⁹Tm

CAS number 7440-30-4

Atomic radius, non-bonded 2.27 Å
Covalent radius 1.77 Å
Electronegativity (Pauling scale) 1.25
Electron affinity 99.283 (kJ mol⁻¹)

5.70 70 - Ytterbium

5.70.1 Properties

Melting point 824°C, 1515°F, 1097 K
Boiling point 1196°C, 2185°F, 1469 K

Density 6.90 g cm⁻³
Relative atomic mass 173.045
Electron configuration [Xe] 4f¹⁴6s²
Elemental group Lanthanides
Elemental period 6
Elemental block f
Key isotopes ¹⁷²Yb, ¹⁷³Yb, ¹⁷⁴Yb
CAS number 7440-64-4

Atomic radius, non-bonded 2.26 Å
Covalent radius 1.78 Å
Electronegativity (Pauling scale) Unknown
Electron affinity -1.93 (kJ mol⁻¹)

5.71 71 - Lutetium

5.71.1 Properties

Melting point 1663°C, 3025°F, 1936 K
Boiling point 3402°C, 6156°F, 3675 K

Density 9.84 g cm⁻³
Relative atomic mass 174.967
Electron configuration [Xe] 4f¹⁴5d¹6s²
Elemental group Lanthanides
Elemental period 6
Elemental block f
Key isotopes ¹⁷⁵Lu
CAS number 7439-94-3

Atomic radius, non-bonded 2.24 Å
Covalent radius 1.74 Å
Electronegativity (Pauling scale) 1.0
Electron affinity 32.81 (kJ mol⁻¹)

5.72 72 - Hafnium

5.72.1 Properties

Melting point 2233°C, 4051°F, 2506 K
Boiling point 4600°C, 8312°F, 4873 K

Density 13.3 g cm⁻³
Relative atomic mass 178.486

Electron configuration [Xe] 4f¹⁴5d²6s²
Elemental group 4
Elemental period 6
Elemental block d
Key isotopes ¹⁷⁷Hf, ¹⁷⁸Hf, ¹⁸⁰Hf
CAS number 7440-58-6

Atomic radius, non-bonded 2.23 Å
Covalent radius 1.64 Å
Electronegativity (Pauling scale) 1.3
Electron affinity 1.351 (kJ mol⁻¹)

5.73 73 - Tantalum

5.73.1 Properties

Melting point 3017°C, 5463°F, 3290 K
Boiling point 5455°C, 9851°F, 5728 K

Density 16.4 g cm⁻³
Relative atomic mass 180.948
Electron configuration [Xe] 4f¹⁴5d³6s²
Elemental group 5
Elemental period 6
Elemental block d
Key isotopes ¹⁸⁰Ta, ¹⁸¹Ta
CAS number 7440-25-7

Atomic radius, non-bonded 2.22 Å
Covalent radius 1.58 Å
Electronegativity (Pauling scale) 1.5
Electron affinity 31.068 (kJ mol⁻¹)

5.74 74 - Tungsten

5.74.1 Properties

Melting point 3414°C, 6177°F, 3687 K
Boiling point 5555°C, 10031°F, 5828 K

Density 19.3 g cm⁻³
Relative atomic mass 183.84
Electron configuration [Xe] 4f¹⁴5d⁴6s²
Elemental group 6
Elemental period 6
Elemental block d
Key isotopes ¹⁸²W, ¹⁸⁴W, ¹⁸⁶W
CAS number 7440-33-7

Atomic radius, non-bonded 2.18 Å
Covalent radius 1.50 Å
Electronegativity (Pauling scale) 1.7
Electron affinity 78.757 (kJ mol⁻¹)

5.75 75 - Rhenium

5.75.1 Properties

Melting point 3185°C, 5765°F, 3458 K

Boiling point 5590°C, 10094°F, 5863 K

Density 20.8 g cm⁻³

Relative atomic mass 186.207

Electron configuration [Xe] 4f¹⁴5d⁵6s²

Elemental group 7

Elemental period 6

Elemental block d

Key isotopes ¹⁸⁷Re

CAS number 7440-15-5

Atomic radius, non-bonded 2.16 Å

Covalent radius 1.41 Å

Electronegativity (Pauling scale) 1.9

Electron affinity 14.47 (kJ mol⁻¹)

5.76 76 - Osmium

5.76.1 Properties

Melting point 3033°C, 5491°F, 3306 K

Boiling point 5008°C, 9046°F, 5281 K

Density 22.5872 g cm⁻³

Relative atomic mass 190.23

Electron configuration [Xe] 4f¹⁴5d⁶6s²

Elemental group 8

Elemental period 6

Elemental block d

Key isotopes ¹⁹²Os

CAS number 7440-04-2

Atomic radius, non-bonded 2.16 Å

Covalent radius 1.36 Å

Electronegativity (Pauling scale) 2.2

Electron affinity 106.1 (kJ mol⁻¹)

5.77 77 - Iridium

5.77.1 Properties

Melting point 2446°C, 4435°F, 2719 K

Boiling point 4428°C, 8002°F, 4701 K

Density 22.5622 g cm⁻³

Relative atomic mass 192.217

Electron configuration [Xe] 4f¹⁴5d⁷6s²

Elemental group 9

Elemental period 6

Elemental block d

Key isotopes ¹⁹³Ir

CAS number 7439-88-5

Atomic radius, non-bonded 2.13 Å
Covalent radius 1.32 Å
Electronegativity (Pauling scale) 2.2
Electron affinity 150.884 (kJ mol⁻¹)

5.78 78 - Platinum

5.78.1 Properties

Melting point 1768.2°C, 3214.8°F, 2041.4 K
Boiling point 3825°C, 6917°F, 4098 K

Density 21.5 g cm⁻³
Relative atomic mass 195.084
Electron configuration [Xe] 4f¹⁴5d⁹6s¹
Elemental group 10
Elemental period 6
Elemental block d
Key isotopes ¹⁹⁵Pt
CAS number 7440-06-4

Atomic radius, non-bonded 2.13 Å
Covalent radius 1.30 Å
Electronegativity (Pauling scale) 2.2
Electron affinity 205.321 (kJ mol⁻¹)

5.79 79 - Gold

5.79.1 Properties

Melting point 1064.18°C, 1947.52°F, 1337.33 K
Boiling point 2836°C, 5137°F, 3109 K

Density 19.3 g cm⁻³
Relative atomic mass 196.967
Electron configuration [Xe] 4f¹⁴5d¹⁰6s¹
Elemental group 11
Elemental period 6
Elemental block d
Key isotopes ¹⁹⁷Au
CAS number 7440-57-5

Atomic radius, non-bonded 2.14 Å
Covalent radius 1.30 Å
Electronegativity (Pauling scale) 2.4
Electron affinity 222.749 (kJ mol⁻¹)

5.80 80 - Mercury

5.80.1 Properties

Melting point -38.829°C, -37.892°F, 234.321 K
Boiling point 356.619°C, 673.914°F, 629.769 K

Density 13.5336 g cm⁻³
Relative atomic mass 200.592

Electron configuration [Xe] 4f¹⁴5d¹⁰6s²
Elemental group 12
Elemental period 6
Elemental block d
Key isotopes ²⁰²Hg
CAS number 7439-97-6

Atomic radius, non-bonded 2.23 Å
Covalent radius 1.32 Å
Electronegativity (Pauling scale) 1.9
Electron affinity Not stable

5.81 81 - Thallium

5.81.1 Properties

Melting point 304°C, 579°F, 577 K
Boiling point 1473°C, 2683°F, 1746 K

Density 11.8 g cm⁻³
Relative atomic mass 204.38
Electron configuration [Xe] 4f¹⁴5d¹⁰6s²6p¹
Elemental group 13
Elemental period 6
Elemental block p
Key isotopes ²⁰⁵Tl
CAS number 7440-28-0

Atomic radius, non-bonded 1.96 Å
Covalent radius 1.44 Å
Electronegativity (Pauling scale) 1.8
Electron affinity 36.375 (kJ mol⁻¹)

5.82 82 - Lead

5.82.1 Properties

Melting point 327.462°C, 621.432°F, 600.612 K
Boiling point 1749°C, 3180°F, 2022 K

Density 11.3 g cm⁻³
Relative atomic mass 207.2
Electron configuration [Xe] 4f¹⁴5d¹⁰6s²6p²
Elemental group 14
Elemental period 6
Elemental block p
Key isotopes ²⁰⁸Pb
CAS number 7439-92-1

Atomic radius, non-bonded 2.02 Å
Covalent radius 1.45 Å
Electronegativity (Pauling scale) 1.8
Electron affinity 35.121 (kJ mol⁻¹)

5.83 83 - Bismuth

5.83.1 Properties

Melting point 271.406°C, 520.531°F, 544.556 K

Boiling point 1564°C, 2847°F, 1837 K

Density 9.79 g cm⁻³

Relative atomic mass 208.980

Electron configuration [Xe] 4f¹⁴5d¹⁰6s²6p³

Elemental group 15

Elemental period 6

Elemental block p

Key isotopes ²⁰⁹Bi

CAS number 7440-69-9

Atomic radius, non-bonded 2.07 Å

Covalent radius 1.50 Å

Electronegativity (Pauling scale) 1.9

Electron affinity 90.924 (kJ mol⁻¹)

5.84 84 - Polonium

5.84.1 Properties

Melting point 254°C, 489°F, 527 K

Boiling point 962°C, 1764°F, 1235 K

Density 9.20 g cm⁻³

Relative atomic mass [209]

Electron configuration [Xe] 4f¹⁴5d¹⁰6s²6p⁴

Elemental group 16

Elemental period 6

Elemental block p

Key isotopes ²⁰⁹Po, ²¹⁰Po

CAS number 7440-08-6

Atomic radius, non-bonded 1.97 Å

Covalent radius 1.42 Å

Electronegativity (Pauling scale) 2.0

Electron affinity 183.3 (kJ mol⁻¹)

5.85 85 - Astatine

5.85.1 Properties

Melting point 300°C, 572°F, 573 K

Boiling point 350°C, 662°F, 623 K

Density Unknown

Relative atomic mass [210]

Electron configuration [Xe] 4f¹⁴5d¹⁰6s²6p⁵

Elemental group 17

Elemental period 6

Elemental block p

Key isotopes ²¹⁰At, ²¹¹At

CAS number 7440-68-8

Atomic radius, non-bonded 2.02 Å
Covalent radius 1.48 Å
Electronegativity (Pauling scale) 2.2
Electron affinity 270.2 (kJ mol⁻¹)

5.86 86 - Radon

5.86.1 Properties

Melting point -71°C, -96°F, 202 K
Boiling point -61.7°C, -79.1°F, 211.5 K

Density 0.009074 g cm⁻³
Relative atomic mass [222]
Electron configuration [Xe] 4f¹⁴5d¹⁰6s²6p⁶
Elemental group 18
Elemental period 6
Elemental block p
Key isotopes ²¹¹Rn, ²²⁰Rn, ²²²Rn
CAS number 10043-92-2

Atomic radius, non-bonded 2.20 Å
Covalent radius 1.46 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Not stable

5.87 87 - Francium

5.87.1 Properties

Melting point 21°C, 70°F, 294 K
Boiling point 650°C, 1202°F, 923 K

Density Unknown
Relative atomic mass [223]
Electron configuration [Rn] 7s¹
Elemental group 1
Elemental period 7
Elemental block s
Key isotopes ²²³Fr
CAS number 7440-73-5

Atomic radius, non-bonded 3.48 Å
Covalent radius 2.42 Å
Electronegativity (Pauling scale) 0.7
Electron affinity 44.38 (kJ mol⁻¹)

5.88 88 - Radium

5.88.1 Properties

Melting point 696°C, 1285°F, 969 K
Boiling point 1500°C, 2732°F, 1773 K

Density 5 g cm⁻³
Relative atomic mass [226]

Electron configuration [Rn] 7s²
Elemental group 2
Elemental period 7
Elemental block s
Key isotopes ²²⁶Ra
CAS number 7440-14-4

Atomic radius, non-bonded 2.83 Å
Covalent radius 2.11 Å
Electronegativity (Pauling scale) 0.9
Electron affinity 9.65 (kJ mol⁻¹)

5.89 89 - Actinium

5.89.1 Properties

Melting point 1050°C, 1922°F, 1323 K
Boiling point 3200°C, 5792°F, 3473 K

Density 10 g cm⁻³
Relative atomic mass [227]
Electron configuration [Rn] 6d¹7s²
Elemental group Actinides
Elemental period 7
Elemental block d
Key isotopes ²²⁷Ac
CAS number 7440-34-8

Atomic radius, non-bonded 2.47 Å
Covalent radius 2.01 Å
Electronegativity (Pauling scale) 1.1
Electron affinity 33.77 (kJ mol⁻¹)

5.90 90 - Thorium

5.90.1 Properties

Melting point 1750°C, 3182°F, 2023 K
Boiling point 4785°C, 8645°F, 5058 K

Density 11.7 g cm⁻³
Relative atomic mass 232.038
Electron configuration [Rn] 6d²7s²
Elemental group Actinides
Elemental period 7
Elemental block f
Key isotopes ²³⁰Th, ²³²Th
CAS number 7440-29-1

Atomic radius, non-bonded 2.45 Å
Covalent radius 1.90 Å
Electronegativity (Pauling scale) 1.3
Electron affinity Unknown

5.91 91 - Protactinium

5.91.1 Properties

Melting point 1572°C, 2862°F, 1845 K

Boiling point 4000°C, 7232°F, 4273 K

Density 15.4 g cm⁻³

Relative atomic mass 231.036

Electron configuration [Rn] 5f²6d¹7s²

Elemental group Actinides

Elemental period 7

Elemental block f

Key isotopes ²³¹Pa

CAS number 7440-13-3

Atomic radius, non-bonded 2.43 Å

Covalent radius 1.84 Å

Electronegativity (Pauling scale) 1.5

Electron affinity Unknown

5.92 92 - Uranium

5.92.1 Properties

Melting point 1135°C, 2075°F, 1408 K

Boiling point 4131°C, 7468°F, 4404 K

Density 19.1 g cm⁻³

Relative atomic mass 238.029

Electron configuration [Rn] 5f³6d¹7s²

Elemental group Actinides

Elemental period 7

Elemental block f

Key isotopes ²³⁴U, ²³⁵U, ²³⁸U

CAS number 7440-61-1

Atomic radius, non-bonded 2.41 Å

Covalent radius 1.83 Å

Electronegativity (Pauling scale) 1.7

Electron affinity Unknown

5.93 93 - Neptunium

5.93.1 Properties

Melting point 644°C, 1191°F, 917 K

Boiling point 3902°C, 7056°F, 4175 K

Density 20.2 g cm⁻³

Relative atomic mass [237]

Electron configuration [Rn] 5f⁴6d¹7s²

Elemental group Actinides

Elemental period 7

Elemental block f

Key isotopes ²³⁷Np

CAS number 7439-99-8

Atomic radius, non-bonded 2.39 Å
Covalent radius 1.80 Å
Electronegativity (Pauling scale) 1.3
Electron affinity Unknown

5.94 94 - Plutonium

5.94.1 Properties

Melting point 640°C, 1184°F, 913 K
Boiling point 3228°C, 5842°F, 3501 K

Density 19.7 g cm⁻³
Relative atomic mass [244]
Electron configuration [Rn] 5f⁶7s²
Elemental group Actinides
Elemental period 7
Elemental block f
Key isotopes ²³⁸Pu, ²³⁹Pu, ²⁴⁰Pu
CAS number 7440-07-5

Atomic radius, non-bonded 2.43 Å
Covalent radius 1.80 Å
Electronegativity (Pauling scale) 1.3
Electron affinity Unknown

5.95 95 - Americium

5.95.1 Properties

Melting point 1176°C, 2149°F, 1449 K
Boiling point 2011°C, 3652°F, 2284 K

Density 12 g cm⁻³
Relative atomic mass [243]
Electron configuration [Rn] 5f⁷7s²
Elemental group Actinides
Elemental period 7
Elemental block f
Key isotopes ²⁴¹Am, ²⁴³Am
CAS number 7440-35-9

Atomic radius, non-bonded 2.44 Å
Covalent radius 1.73 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Unknown

5.96 96 - Curium

5.96.1 Properties

Melting point 1345°C, 2453°F, 1618 K
Boiling point Unknown

Density 13.51 g cm⁻³
Relative atomic mass [247]

Electron configuration [Rn] 5f⁷6d¹7s²

Elemental group Actinides

Elemental period 7

Elemental block f

Key isotopes ²⁴³Cm, ²⁴⁸Cm

CAS number 7440-51-9

Atomic radius, non-bonded 2.450 Å

Covalent radius 1.68 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.97 97 - Berkelium

5.97.1 Properties

Melting point 986°C, 1807°F, 1259 K

Boiling point Unknown

Density 14.78 g cm⁻³

Relative atomic mass [247]

Electron configuration [Rn] 5f⁹7s²

Elemental group Actinides

Elemental period 7

Elemental block f

Key isotopes ²⁴⁷Bk, ²⁴⁹Bk

CAS number 7440-40-6

Atomic radius, non-bonded 2.44 Å

Covalent radius 1.68 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.98 98 - Californium

5.98.1 Properties

Melting point 900°C, 1652°F, 1173 K

Boiling point Unknown

Density 15.1 g cm⁻³

Relative atomic mass [251]

Electron configuration [Rn] 5f¹⁰7s²

Elemental group Actinides

Elemental period 7

Elemental block f

Key isotopes ²⁴⁹Cf, ²⁵²Cf

CAS number 7440-71-3

Atomic radius, non-bonded 2.45 Å

Covalent radius 1.68 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.99 99 - Einsteinium

5.99.1 Properties

Melting point 860°C, 1580°F, 1133 K

Boiling point Unknown

Density Unknown

Relative atomic mass [252]

Electron configuration [Rn] 5f¹¹7s²

Elemental group Actinides

Elemental period 7

Elemental block f

Key isotopes ²⁵²Es

CAS number 7429-92-7

Atomic radius, non-bonded 2.45 Å

Covalent radius 1.65 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.100 100 - Fermium

5.100.1 Properties

Melting point 1527°C, 2781°F, 1800 K

Boiling point Unknown

Density Unknown

Relative atomic mass [257]

Electron configuration [Rn] 5f¹²7s²

Elemental group Actinides

Elemental period 7

Elemental block f

Key isotopes ²⁵⁷Fm

CAS number 7440-72-4

Atomic radius, non-bonded 2.45 Å

Covalent radius 1.67 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.101 101 - Mendelevium

5.101.1 Properties

Melting point 827°C, 1521°F, 1100 K

Boiling point Unknown

Density Unknown

Relative atomic mass [258]

Electron configuration [Rn] 5f¹³7s²

Elemental group Actinides

Elemental period 7

Elemental block f

Key isotopes ²⁵⁸Md, ²⁶⁰Md

CAS number 7440-11-1

Atomic radius, non-bonded 2.46 Å
Covalent radius 1.73 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Unknown

5.102 102 - Nobelium

5.102.1 Properties

Melting point 827°C, 1521°F, 1100 K
Boiling point Unknown

Density Unknown
Relative atomic mass [259]
Electron configuration [Rn] 5f¹⁴7s²
Elemental group Actinides
Elemental period 7
Elemental block f
Key isotopes ²⁵⁹No
CAS number 10028-14-5

Atomic radius, non-bonded 2.46 Å
Covalent radius 1.76 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Unknown

5.103 103 - Lawrencium

5.103.1 Properties

Melting point 1627°C, 2961°F, 1900 K
Boiling point Unknown

Density Unknown
Relative atomic mass [262]
Electron configuration [Rn] 5f¹⁴7s²7p¹
Elemental group Actinides
Elemental period 7
Elemental block f
Key isotopes ²⁶²Lr
CAS number 22537-19-5

Atomic radius, non-bonded 2.46 Å
Covalent radius 1.61 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Unknown

5.104 104 - Rutherfordium

5.104.1 Properties

Melting point Unknown
Boiling point Unknown

Density Unknown
Relative atomic mass [267]

Electron configuration [Rn] 5f¹⁴6d²7s²

Elemental group 4

Elemental period 7

Elemental block d

Key isotopes ²⁶⁵Rf

CAS number 53850-36-5

Atomic radius, non-bonded Unknown

Covalent radius 1.57 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.105 105 - Dubnium

5.105.1 Properties

Melting point Unknown

Boiling point Unknown

Density Unknown

Relative atomic mass [268]

Electron configuration [Rn] 5f¹⁴6d³7s²

Elemental group 5

Elemental period 7

Elemental block d

Key isotopes ²⁶⁸Db

CAS number 53850-35-4

Atomic radius, non-bonded Unknown

Covalent radius 1.49 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.106 106 - Seaborgium

5.106.1 Properties

Melting point Unknown

Boiling point Unknown

Density Unknown

Relative atomic mass [269]

Electron configuration [Rn] 5f¹⁴6d⁴7s²

Elemental group 6

Elemental period 7

Elemental block d

Key isotopes ²⁷¹Sg

CAS number 54038-81-2

Atomic radius, non-bonded Unknown

Covalent radius 1.43 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.107 107 - Bohrium

5.107.1 Properties

Melting point Unknown

Boiling point Unknown

Density Unknown

Relative atomic mass [270]

Electron configuration [Rn] 5f¹⁴6d⁵7s²

Elemental group 7

Elemental period 7

Elemental block d

Key isotopes ²⁷²Bh

CAS number 54037-14-8

Atomic radius, non-bonded Unknown

Covalent radius 1.41 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.108 108 - Hassium

5.108.1 Properties

Melting point Unknown

Boiling point Unknown

Density Unknown

Relative atomic mass [269]

Electron configuration [Rn] 5f¹⁴6d⁶7s²

Elemental group 8

Elemental period 7

Elemental block d

Key isotopes ²⁷⁰Hs

CAS number 54037-57-9

Atomic radius, non-bonded Unknown

Covalent radius 1.34 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.109 109 - Meitnerium

5.109.1 Properties

Melting point Unknown

Boiling point Unknown

Density Unknown

Relative atomic mass [278]

Electron configuration [Rn] 5f¹⁴6d⁷7s²

Elemental group 9

Elemental period 7

Elemental block d

Key isotopes ²⁷⁶Mt

CAS number 54038-01-6

Atomic radius, non-bonded Unknown
Covalent radius 1.29 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Unknown

5.110 110 - Darmstadtium

5.110.1 Properties

Melting point Unknown
Boiling point Unknown

Density Unknown
Relative atomic mass [281]
Electron configuration [Rn] 5f¹⁴6d⁹7s¹
Elemental group 10
Elemental period 7
Elemental block d
Key isotopes ²⁸¹Ds
CAS number 54083-77-1

Atomic radius, non-bonded Unknown
Covalent radius 1.28 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Unknown

5.111 111 - Roentgenium

5.111.1 Properties

Melting point Unknown
Boiling point Unknown

Density Unknown
Relative atomic mass [280]
Electron configuration [Rn] 5f¹⁴6d¹⁰7s¹
Elemental group 11
Elemental period 7
Elemental block d
Key isotopes ²⁸⁰Rg
CAS number 54386-24-2

Atomic radius, non-bonded Unknown
Covalent radius 1.21 Å
Electronegativity (Pauling scale) Unknown
Electron affinity Unknown

5.112 112 - Copernicium

5.112.1 Properties

Melting point Unknown
Boiling point Unknown

Density Unknown
Relative atomic mass [285]

Electron configuration [Rn] 5f¹⁴6d¹⁰7s²**Elemental group** 12**Elemental period** 7**Elemental block** d**Key isotopes** ²⁸⁵Cn**CAS number** 54084-26-3**Atomic radius, non-bonded** Unknown**Covalent radius** 1.22 Å**Electronegativity (Pauling scale)** Unknown**Electron affinity** Unknown

5.113 113 - Nihonium

5.113.1 Properties

Melting point Unknown**Boiling point** Unknown**Density** Unknown**Relative atomic mass** [286]**Electron configuration** [Rn] 5f¹⁴6d¹⁰7s²7p¹**Elemental group** 13**Elemental period** 7**Elemental block** p**Key isotopes** ²⁸⁶Nh**CAS number** 54084-70-7**Atomic radius, non-bonded** Unknown**Covalent radius** 1.36 Å**Electronegativity (Pauling scale)** Unknown**Electron affinity** Unknown

5.114 114 - Flerovium

5.114.1 Properties

Melting point Unknown**Boiling point** Unknown**Density** Unknown**Relative atomic mass** [289]**Electron configuration** [Rn] 5f¹⁴6d¹⁰7s²7p²**Elemental group** 14**Elemental period** 7**Elemental block** p**Key isotopes** ²⁸⁹Fl**CAS number** 54085-16-4**Atomic radius, non-bonded** Unknown**Covalent radius** 1.43 Å**Electronegativity (Pauling scale)** Unknown**Electron affinity** j0 (kJ mol⁻¹)

5.115 115 - Moscovium

5.115.1 Properties

Melting point Unknown

Boiling point Unknown

Density Unknown

Relative atomic mass [289]

Electron configuration [Rn] 5f¹⁴6d¹⁰7s²7p³

Elemental group 15

Elemental period 7

Elemental block p

Key isotopes ²⁸⁹Mc

CAS number 54085-64-2

Atomic radius, non-bonded Unknown

Covalent radius 1.62 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.116 116 - Livermorium

5.116.1 Properties

Melting point Unknown

Boiling point Unknown

Density Unknown

Relative atomic mass [293]

Electron configuration [Rn] 5f¹⁴6d¹⁰7s²7p⁴

Elemental group 16

Elemental period 7

Elemental block p

Key isotopes ²⁹³Lv

CAS number 54100-71-9

Atomic radius, non-bonded Unknown

Covalent radius 1.75 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.117 117 - Tennessine

5.117.1 Properties

Melting point Unknown

Boiling point Unknown

Density Unknown

Relative atomic mass [294]

Electron configuration [Rn] 5f¹⁴6d¹⁰7s²7p⁵

Elemental group 17

Elemental period 7

Elemental block p

Key isotopes ²⁹⁴Ts

CAS number 87658-56-8

Atomic radius, non-bonded Unknown

Covalent radius 1.65 Å

Electronegativity (Pauling scale) Unknown

Electron affinity Unknown

5.118 118 - Oganesson

5.118.1 Properties

Melting point Unknown

Boiling point Unknown

Density Unknown

Relative atomic mass [294]

Electron configuration [Rn] 5f¹⁴6d¹⁰7s²7p⁶

Elemental group 18

Elemental period 7

Elemental block p

Key isotopes ²⁹⁴Og

CAS number 54144-19-3

Atomic radius, non-bonded Unknown

Covalent radius 1.57 Å

Electronegativity (Pauling scale) Unknown

Electron affinity 5.403 (kJ mol⁻¹)

Chapter 6

Chemicals

Chapter 7

Composition reference

Ac	Ge	Po
Ag	H	Pr
Al	He	Pt
Am	Hf	Pu
Ar	Hg	Ra
As	Ho	Rb
At	Hs	Re
Au	I	Rf
B	In	Rg
Ba	Ir	Rh
Be	K	Rn
Bh	Kr	Ru
Bi	La	S
Bk	Li	Sb
Br	Lr	Sc
C	Lu	Se
Ca	Lv	Sg
Cd	Mc	Si
Ce	Md	Sm
Cf	Mg	Sn
Cl	Mn	Sr
Cm	Mo	Ta
Cn	Mt	Tb
Co	N	Tc
Cr	Na	Te
Cs	Nb	Th
Cu	Nd	Ti
Db	Ne	Tl
Ds	Nh	Tm
Dy	Ni	Ts
Er	No	U
Es	Np	V
Eu	O	W
F	Og	Xe
Fe	Os	Y
Fl	P	Yb
Fm	Pa	Zn
Fr	Pb	Zr
Ga	Pd	
Gd	Pm	

Chapter 8

CAS reference

CAS references will vary in structure as to how many digits are separated by commas. I believe they are always a group of three numbers (so two hyphens), but in the interest of figuring out a sort order, I have chosen to interpret the number as though it was a solid number with no separations. I'm not sure if this is the preferred method, but it's the one you will see here.

133-74-0 - Hydrogen	7440-19-9 - Samarium
7704-34-9 - Sulfur	7440-20-2 - Scandium
7723-14-0 - Phosphorus	7440-21-3 - Silicon
7727-37-9 - Nitrogen	7440-22-4 - Silver
7429-90-5 - Aluminium	7440-23-5 - Sodium
7429-91-6 - Dysprosium	7440-24-6 - Strontium
7429-92-7 - Einsteinium	7440-25-7 - Tantalum
7439-88-5 - Iridium	7440-26-8 - Technetium
7439-89-6 - Iron	7440-27-9 - Terbium
7439-90-9 - Krypton	7440-28-0 - Thallium
7439-91-0 - Lanthanum	7440-29-1 - Thorium
7439-92-1 - Lead	7440-30-4 - Thulium
7439-93-2 - Lithium	7440-31-5 - Tin
7439-94-3 - Lutetium	7440-32-6 - Titanium
7439-95-4 - Magnesium	7440-33-7 - Tungsten
7439-96-5 - Manganese	7440-34-8 - Actinium
7439-97-6 - Mercury	7440-35-9 - Americium
7439-98-7 - Molybdenum	7440-36-0 - Antimony
7439-99-8 - Neptunium	7440-37-1 - Argon
7440-00-8 - Neodymium	7440-38-2 - Arsenic
7440-01-9 - Neon	7440-39-3 - Barium
7440-02-0 - Nickel	7440-40-6 - Berkelium
7440-03-1 - Niobium	7440-41-7 - Beryllium
7440-04-2 - Osmium	7440-42-8 - Boron
7440-05-3 - Palladium	7440-43-9 - Cadmium
7440-06-4 - Platinum	7440-44-0 - Carbon, atomic
7440-07-5 - Plutonium	7440-45-1 - Cerium
7440-08-6 - Polonium	7440-46-2 - Caesium
7440-09-7 - Potassium	7440-47-3 - Chromium
7440-10-0 - Praseodymium	7440-48-4 - Cobalt
7440-11-1 - Mendelevium	7440-50-8 - Copper
7440-12-2 - Promethium	7440-51-9 - Curium
7440-13-3 - Protactinium	7440-52-0 - Erbium
7440-14-4 - Radium	7440-53-1 - Europium
7440-15-5 - Rhenium	7440-54-2 - Gadolinium
7440-16-6 - Rhodium	7440-55-3 - Gallium
7440-17-7 - Rubidium	7440-56-4 - Germanium
7440-18-8 - Ruthenium	7440-57-5 - Gold

7440-58-6 - Hafnium
7440-59-7 - Helium
7440-60-0 - Holmium
7440-61-1 - Uranium
7440-62-2 - Vanadium
7440-63-3 - Xenon
7440-64-4 - Ytterbium
7440-65-5 - Yttrium
7440-66-6 - Zinc
7440-67-7 - Zirconium
7440-68-8 - Astatine
7440-69-9 - Bismuth
7440-70-2 - Calcium
7440-71-3 - Californium
7440-72-4 - Fermium
7440-73-5 - Francium
7440-74-6 - Indium
7553-56-2 - Iodine
7726-95-6 - Bromine
7782-41-4 - Fluorine
7782-44-7 - Oxygen
7782-49-2 - Selenium

7782-50-5 - Chlorine
10028-14-5 - Nobelium
10043-92-2 - Radon
13494-80-9 - Tellurium
22537-19-5 - Lawrencium
53850-35-4 - Dubnium
53850-36-5 - Rutherfordium
54037-14-8 - Bohrium
54037-57-9 - Hassium
54038-01-6 - Meitnerium
54038-81-2 - Seaborgium
54083-77-1 - Darmstadtium
54084-26-3 - Copernicium
54084-70-7 - Nihonium
54085-16-4 - Flerovium
54085-64-2 - Moscovium
54100-71-9 - Livermorium
54144-19-3 - Oganesson
54386-24-2 - Roentgenium
87658-56-8 - Tennessine

Chapter 9

IUPAC reference

Chapter 10

Glossary

Ion - A particle with an electrical charge (can be positive or negative, see anion and cation).

Valence - A measurement of how readily a given atom will bond to another.