

( website :- arjun00.com.np )  
Council for Technical Education and Vocational Training  
**Office of the Controller of Examinations**  
Sanothimi, Bhaktapur  
**Regular/Back Exam-2078, Bhadra**

<b>Program:</b>	<b>Diploma in Engineering All</b>	<b>Full Marks: 60</b>
<b>Year/Part:</b>	<b>I/I (New + Old)</b>	<b>Pass Marks: 24</b>
<b>Subject:</b>	<b>Engineering Chemistry I</b>	<b>Time: 3 hrs</b>

*Candidates are required to give their answers in their own words as far as practicable.  
The figures in the margin indicate full marks.*

**Attempt All questions.**

1. a) What do you mean by Eq. wt. of element? Prove that ; [1+4]  
Molecular wt. = 2 x Vapour density.  
b) How is Dalton's atomic theory modified in the light of [5]  
Modern Knowledge? ( website :- arjun00.com.np )
2. a) State and explain Faraday's First law of electrolysis. [2+3]  
Calculate the mass of copper deposited by electrolysis  
on passing 2.5A current for 45 minutes through the  
solution of  $\text{CuSO}_4$ . (At. wt. of Copper = 63.5)  
b) State drawback of Rutherford's atomic model. What are [2+3]  
the basic postulates of Bohr's atomic Model?
3. a) State Dulong's and Petit's law. 0.444 gram of Metal when [1+4]  
dissolved in dilute HCl gave 177 ml of dry hydrogen at  
 $10^\circ\text{C}$  and 750 mm Hg pressure, the specific heat of the  
metal is 0.107. Calculate exact atomic wt. of metal.  
b) State Mendeleev's periodic law? Explain Mendeleev's [1+2+2]  
periodic table in brief. Also mention it's anomalies.
4. a) What is redox reaction? Balance the following chemical [1+4]  
Equation by oxidation number method.  
$$\text{Cu} + \text{HNO}_3 \rightarrow \text{Cu}(\text{NO}_3)_2 + \text{NO} + \text{H}_2\text{O}$$
  
b) What do you mean by acid and base according to [3+2]  
Arrhenious concept? Also mention it's limitations.
5. a) What are the significance of given chemical Equation? [3+2]  
$$\text{CaCO}_3 + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2 \uparrow$$
 Also, Mention  
the limitations of chemical Equations.

**Cont.....**

b) How can you determine the Equivalent weight by indirect oxide formation method. [5]

6. Write short notes on : (Any Five) [5x2=10]

a) Radical

b) Covalent bond

c) Hund's rule

d) Titration

e) Normality

f) Primary standard substances

**Good Luck!**

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