

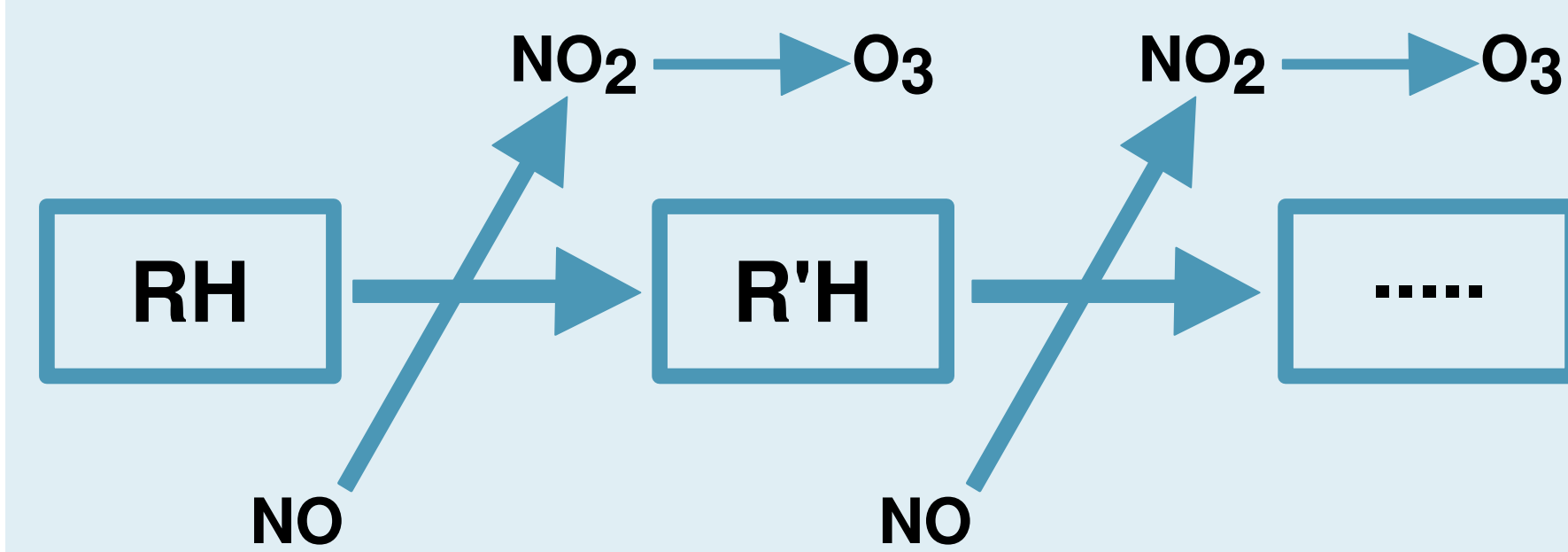
# The Influence of Atmospheric Conditions on the Production of Ozone during VOC Oxidation

Jane Coates and Tim Butler

## Background

- Temperature main meteorological driver of surface ozone in many areas.
- Temperature
  - increases isoprene emissions from vegetation,
  - increases reaction rates of chemical processes.

- VOC the “fuel” and NO<sub>x</sub> the “catalyst” of ozone production.



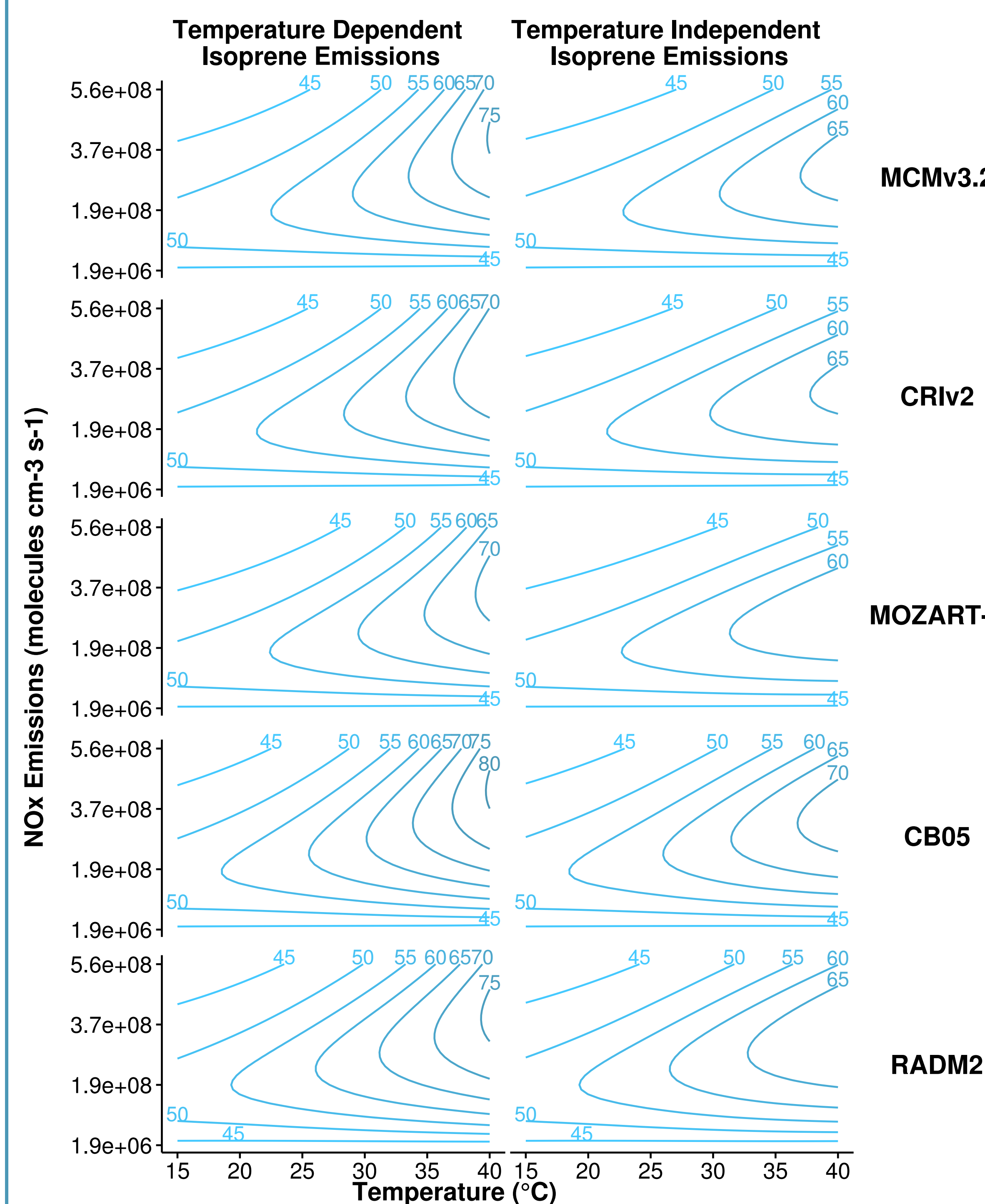
- What drives the ozone–temperature relationship? Increased isoprene emissions or chemistry?
- Can chemical mechanisms simulate the ozone–temperature relationship across NO<sub>x</sub> gradients?

## Approach

- Box model simulating Benelux region.
- Temperature dependent and independent isoprene emissions.
- NO<sub>x</sub> emissions systematically varied over 15 – 40 °C.
- Repeated with chemical mechanisms: MCMv3.2, CRIV2, MOZART-4, RADM2, CB05.

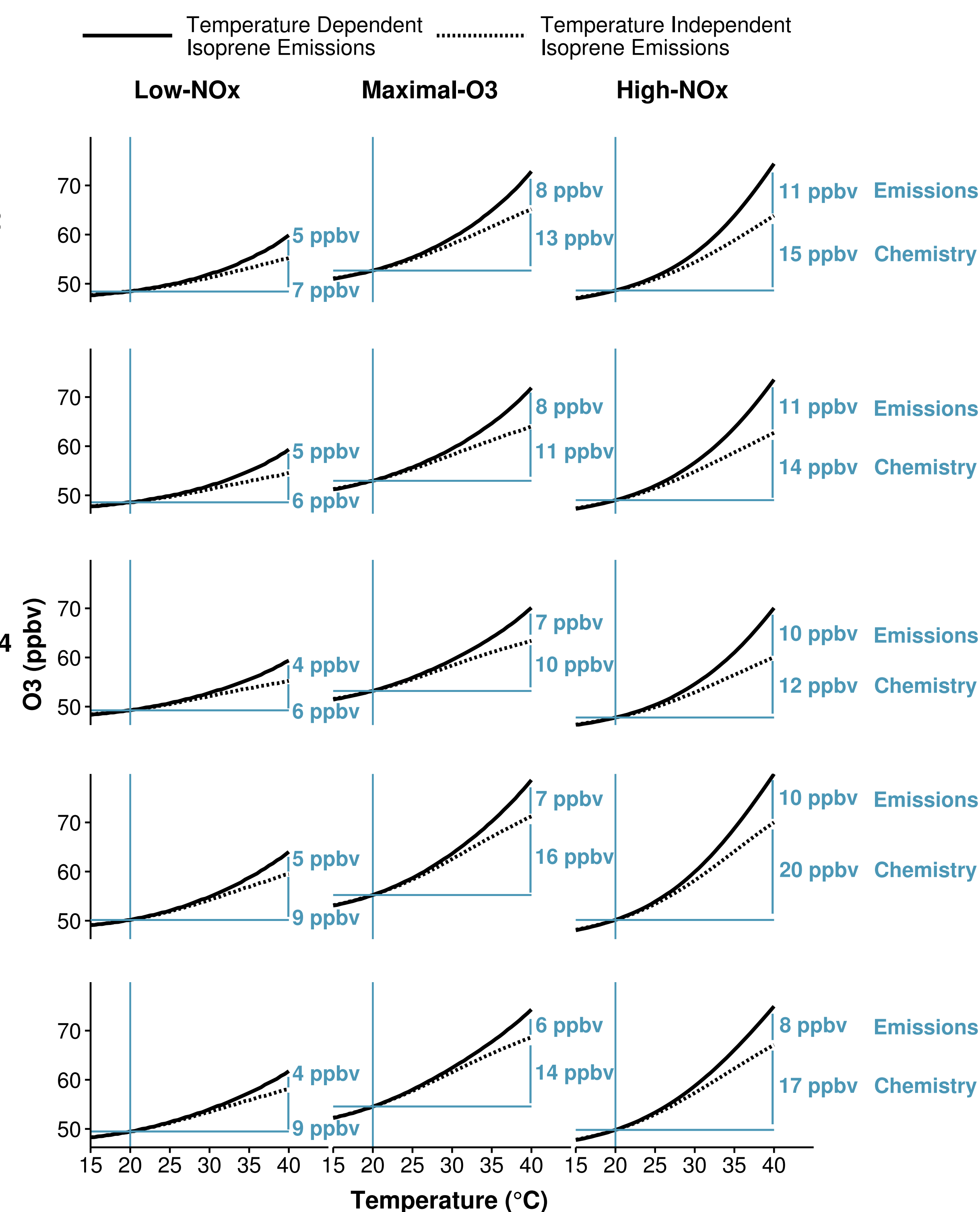
## Results

Ozone Mixing Ratios in ppbv as a Function of NO<sub>x</sub> and Temperature



- High-NO<sub>x</sub> and temperature-dependent isoprene ⇒ highest ozone levels.
- Low-NO<sub>x</sub> conditions has lowest ozone levels.
- Ozone levels vary non-linearly with NO<sub>x</sub> and temperature in each chemical mechanism.
- RADM2 and CB05 produce most ozone regardless of isoprene source.

Increase in O<sub>3</sub> mixing ratios from 20 °C due to Emissions and Chemistry



- Faster reaction rates cause larger increase in ozone than higher isoprene emissions.
- High-NO<sub>x</sub> conditions has largest increases in ozone from chemistry and emissions.
- Increased isoprene emissions in CB05 and RADM2 produces less ozone than other chemical mechanisms.

## Conclusions

- Temperature-dependent chemistry responsible for most of the increase in ozone with temperature.
- High-NO<sub>x</sub> conditions cause largest increases in ozone with temperature.
- Non-linear relationship of ozone with temperature and NO<sub>x</sub> reproduced by all chemical mechanisms.
- Representation of isoprene degradation in CB05 and RADM2 produces less ozone than other chemical mechanisms.

## Future Work

- Compare simulated ozone–temperature to observations.
- What are the most important temperature-dependent chemical processes? PAN decomposition? Increased radical production?
- Why do CB05 and RADM2 produce more ozone than other mechanisms?

The IASS is sponsored by