

# Understanding Ozone Pollution:

# A Comparison of Chemical Mechanisms

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#### **Motivation**

- ► Importance of O<sub>3</sub> production chemistry representation future emission scenarios.
- ► Compare different O<sub>3</sub> chemistry representations used in chemical transport models.
- ► Determine effects on O<sub>3</sub> production by comparing treatment of Volatile Organic Compounds (VOCs) degradation products.

# Approach

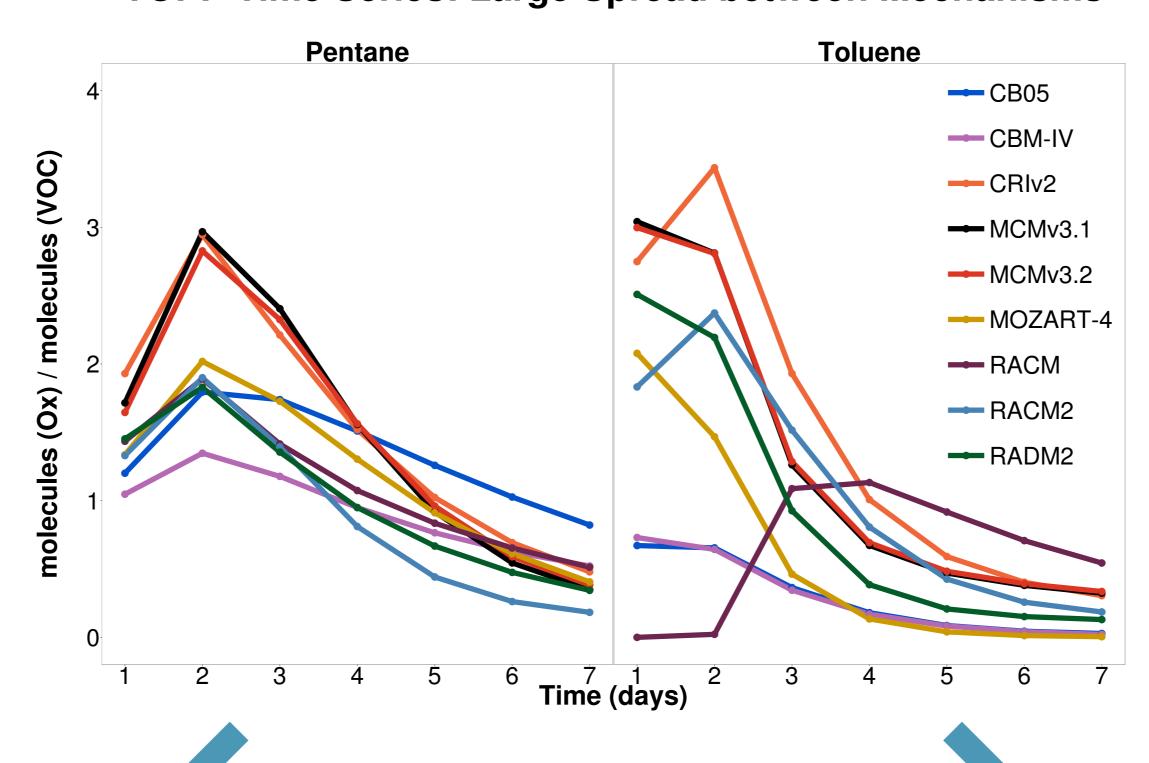
- ► Tagged Ozone Production Potentials (TOPPs) [1] calculated over 7 days for VOCs common to urban environments.
- ▶ Following mechanisms are compared to near-explicit MCM v3.2.

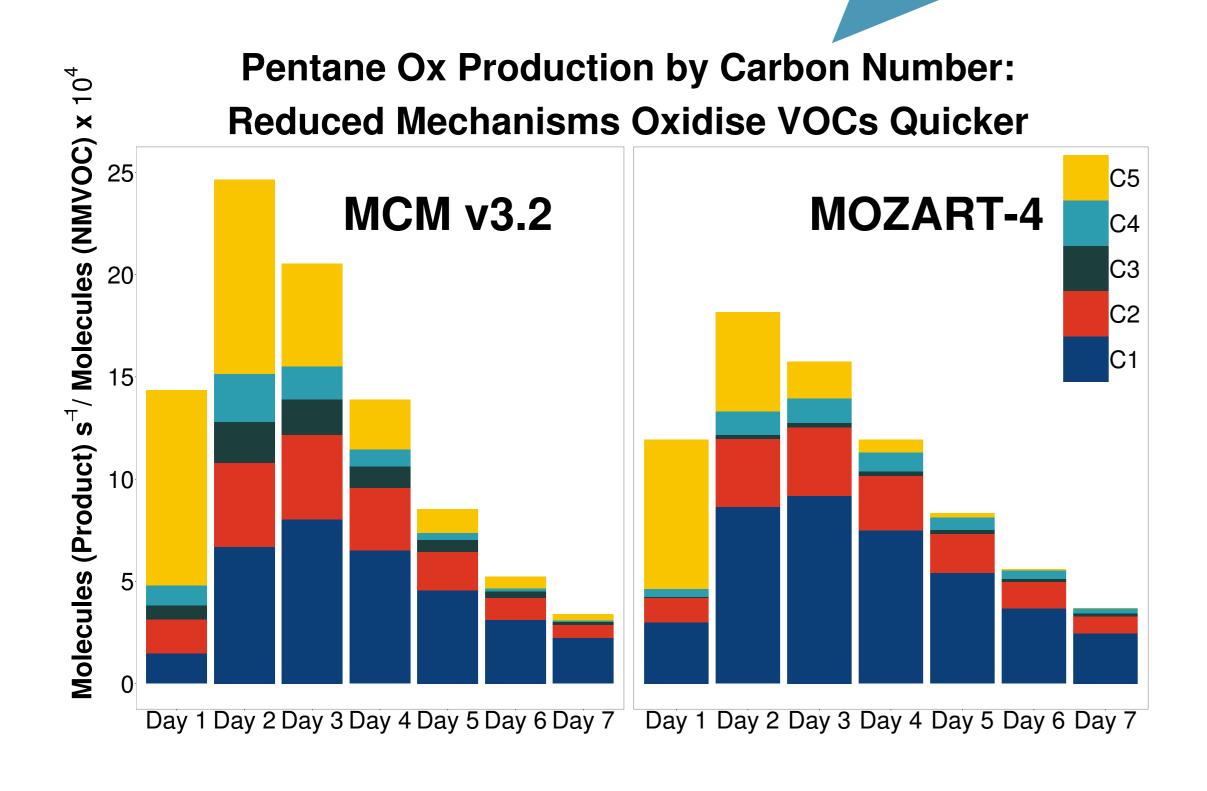
MCM v3.1 CRI v2 CBM-IV CB05 RADM2 RACM RACM2 MOZART-4

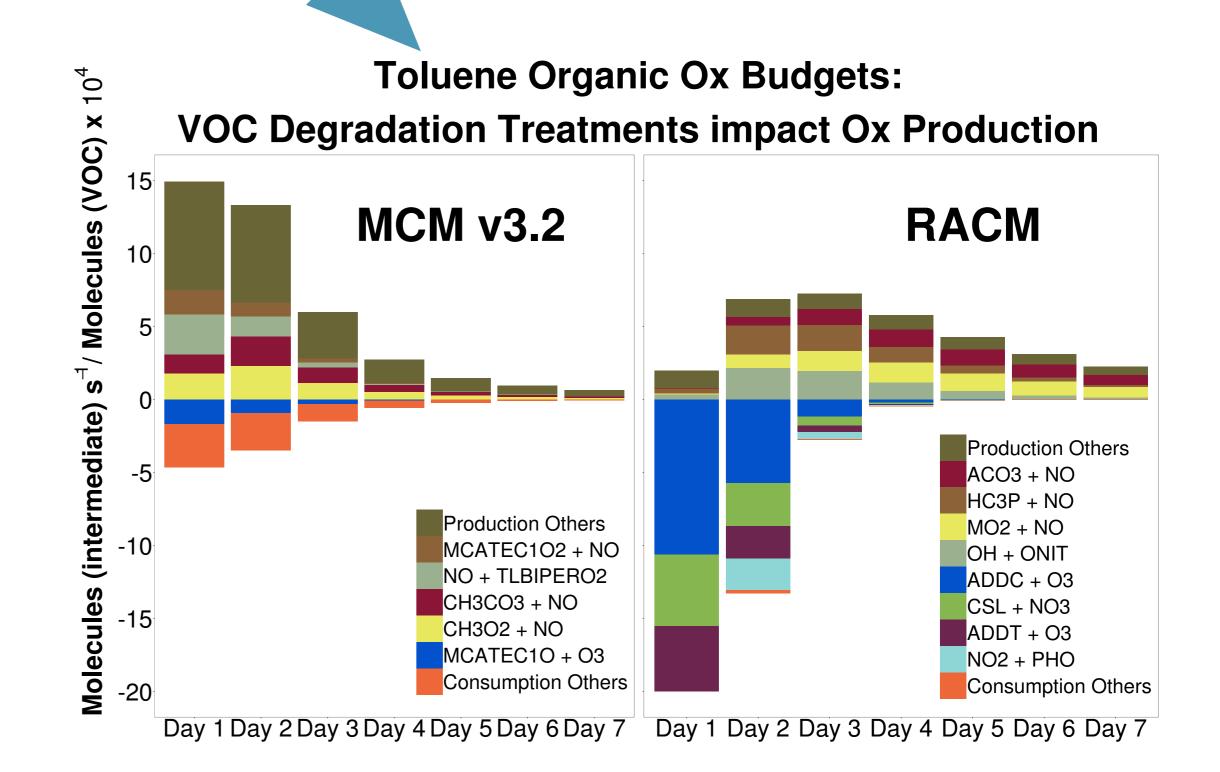
▶  $O_x$  (=  $O_3$  +  $NO_2$ ) production allocated to emitted VOC by 'tagging' its organic degradation products.

#### Results

# **TOPP Time Series: Large Spread between Mechanisms**







### **Conclusions**

- ▶ More explicit mechanisms produce more O<sub>x</sub> than less explicit mechanisms.
- $\triangleright$  VOCs broken down into smaller fragments quicker in less-explicit mechanisms resulting in less  $O_x$  production.
- ▶ First day O<sub>x</sub> production from VOCs similar between many mechanisms, larger differences over time.
- ▶ Differences in VOC degradation treatments impacts on O<sub>x</sub> production RACM aromatic chemistry.

# References

[1] T. M. Butler, M. G. Lawrence, D. Taraborrelli, and J. Lelieveld. Multi-day ozone production potential of volatile organic compounds calculated with a tagging approach. Atmospheric Environment, 45(24):4082–4090, 2011.



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