

Understanding Ozone Pollution:

A Comparison of Chemical Mechanisms

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Motivation

- ► Importance of O₃ production chemistry representation future emission scenarios.
- ► Compare different O₃ chemistry representations used in chemical transport models.
- ▶ Determine effects on O₃ production by comparing treatment of Volatile Organic Compounds (VOCs) degradation products.

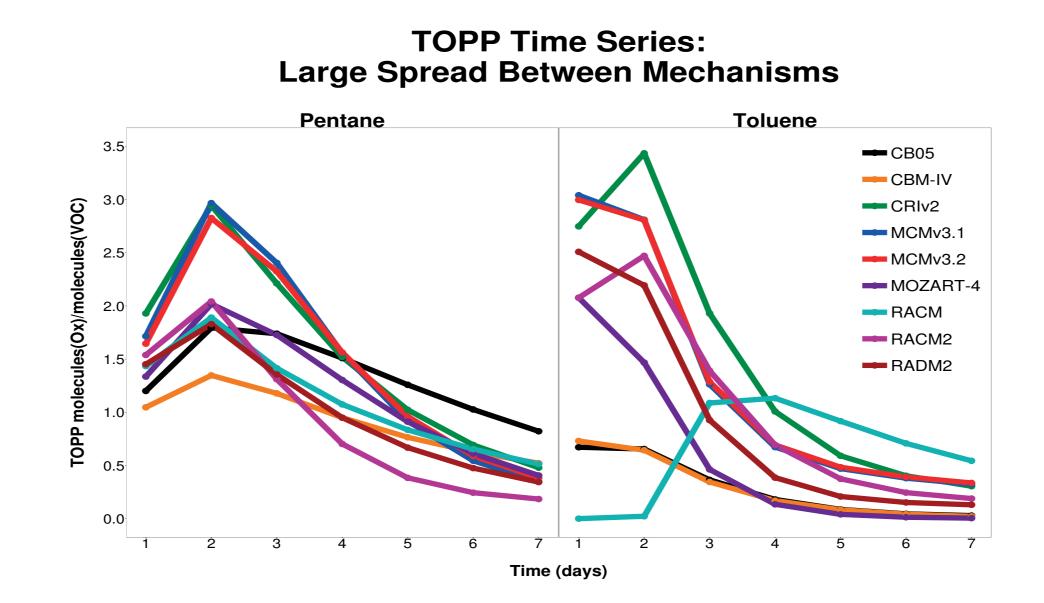
Approach

- ► Tagged Ozone Production Potentials (TOPPs) [1] calculated over 7 days for VOCs common to urban environments.
- ▶ Following mechanisms are compared to MCM v3.2.

MCM v3.1 CRI v2 CBM-IV CB05 RADM2 RACM RACM2 MOZART-4

▶ O_x (= O_3 + NO_2) production allocated to the emitted VOC by 'tagging' organic degradation products.

Results



Conclusions

- ▶ More explicit mechanisms have larger O_x production than less explicit mechanisms.
- ▶ VOCs break down into smaller fragments quicker in less-explicit mechanisms resulting in less O₃ production.
- ▶ The first day O₃ production similar between many mechanisms, larger differences over time.
- ▶ Differences in VOC degradation products can impact on O₃ production RACM aromatic chemistry.

References

[1] T. M. Butler, M. G. Lawrence, D. Taraborrelli, and J. Lelieveld. Multi-day ozone production potential of volatile organic compounds calculated with a tagging approach. Atmospheric Environment, 45(24):4082–4090, 2011.









