将到
$$5e^-$$

KClO₃ + 6HCl = KCl + 3Cl₂↑ + 3H₂O
失去 $5 \times e^-$

$$\frac{2e^{-}}{\sqrt[]{\text{CuO} + \text{H}_2}} = \text{Cu} + \text{H}_2\text{O}$$

$$\frac{e^{-}}{\text{Cl}_{2} + 2\text{NaOH}} = \text{NaCl} + \text{NaClO} + \text{H}_{2}\text{O}$$