SECTION 7

CHEMISTRY

ÈKÓ ELÀ

INTRODUCTION / ORO AKOSO

Chemistry is the science of matter, especially	Èkộ Elà jệ ệkộ ìmộ-jinlệ ti ệdá, pápàá, ti
its properties, structure, composition, behavior,	ìṣesí***, ètò-inú, ohun-inú, ìṣèwàhù, àsè,
reactions, interactions and the changes it	ìbáraşepò àti àwọn ìyípadà tó má nní. A
undergoes. Chemistry is sometimes called "the	máa npè èkó-elà ní ìmò-jinlè t'àárín nítorí
central science" because it connects physics	òun ló şe àsopò àwọn èkọ-èdá pèlú àwọn
with other natural sciences such as astronomy,	ìmỳ-jinlệ míràn bí ệkộ (nípa) ìràwỳ, ệkộ
geology and biology.	ìṣè̞dále̞-ayé, àti e̞ko̞ e̞dá-oníyè.
Chemistry is the study of matter and energy and	Èkó elà jé èkó àwon èdá, agbára àti àwon
the interactions between them.	ìbáraşepò tó wà láarín wọn
Chemical reaction is a process that leads to the	Àsè ẹlà jệ ìlànà bí àwọn ìjọ ệdá kan ṣe
transformation of one set of chemical	nparadà sí àwọn ìjọ ệdá miran.
substances to another.	(àsè: reaction; elà: chemicals)
Chemical equation is the symbolic	Òmì (equation) ẹlà jệ àmì àpẹrẹ àsè ẹlà bí a
representation of a chemical reaction where the	ti nfi àwọn èsè-àsè (reactants) sí apá òsì tí a
reactant entities are given on the left hand side	sì fi àwọn ệsún-àsè (product of reaction) sí apá
and the product entities on the right hand side	òtún.
Chemical composition: the arrangement, type	Ètò Ohun-inú ẹlà: Ètò, oríși àti ìbùpín
and ratio of atoms in molecules of substances	(ratio) àwọn átòmù tó wà nínu mólékù
	àwọn ệdá
Chemical synthesis is purposeful execution of	Ìṣṇdá elà (Ìṣṇdá: bring to existence) jé
chemical reactions to get a product, or several	ìlànà tí a mộómò gbà se àsè láti ní èsún
products	tàbí àwọn ệsún tí a nfệ.

VOCABULARY / ÌTÚMÒ - ÒRÒ

VOCABULARI / HUMO - ORO				
English	Yoruba	English	Yoruba	
Acids and Bases	Èkan at'ègbo	Gas volume	Àye òyì	
Air	Afé	Inorganic chemistry	Èkó elà-aláìléedú	
Alkali	Aşoşe	Laboratory	Ilé ìmò-jinlè	
Alkali metals	Àlùro asose	Liquid	Aşàn	
Alloy	Àyópò	Matter	Èdá	
Amalgam	Àdàlú	Metal	Àlùrọ	
Analytical chemistry		Mixture	Àkópò	
Atom	Átòmù	Non-metal	Àdàrọ	
Atomic energy	Agbára átòmù	Organic chemistry	Èkó elà-eléedú	
Base	Ègbo	Physical chemistry	Èkó-elà t'ìṣeṣí-èdá	
Biochemistry	Èkó elà-ìyè	Physical science	Èkó ìmò-jìnlè ìṣeṣí-èdá	
Breeze	Aféré	Product of a reaction	Èsún-àsè	

ÌWÉ-ÌLÉWÓ ỆKÓ ÌMÒ-JÌNLỆ ÀTI ÌMÒ-ỆRÓ

Changes in Nature	Àyípadà nnú	Reactants	Èsè-àsè
-	àyànmó		
Chemical	ęlà	Reaction	Àsè
Chemical composition	Ètò-inú ẹlà	Shape	Ìrísí
Chemical equation	Òmì ẹlà	Soap Making	Oşę şişé
Chemical process		Solid	Adì
Chemist	Akéko elà	Solute	Gberefun; Àìpò
Chemistry	Èkó-elà	Solution	Àpòpò
Colour	Àwò	Solvent	Еро
Compound	Àsèpò	States of matter	Irú-ipò àwọn ệdá
Concentrate	Láti sọ dògidì	Substance	Èdá
Drug	Oògùn	Synthesis	Îşệdá
Dye	Aró	Wind	Aféfé
Element	Ìṣù-átòmù	Fire	Iná
Endothermic reactiom	Àsè aloná	Gas	Òyì
Energy	Agbára	Gas equation	Òmì òyì
Entropy	Ìdàrú	Gas Laws	Àwon Ofi-òyì
Exothermic reaction	Àsè afaná		

TABLE OF ELEMENTS / ÎTÉ ÀWON ÌŞÙ-ÁTÒMÙ

ELEMENT	ÌṢÙ-ÁTỌMÙ	ELEMENT	ÌṢÙ-ÁTỌMÙ	ELEMENT	ÌŞÙ-ÁTÒMÙ
Actinium	Actinium	Germanium	Gemania	Scandium	Scandia
Aluminum	Alumo	Gold	goolu; Wura	Selenium	Selenia
Americium	Amerisia	Hafnium	Hafnia	Silicon	Silika
Antimony		Helium	Helia	Silver	Fadaka
Argon		Holmium	Holmia	Phosphorus	
Arsenic		Indium	India	Platinum	
Barium	Báríà	Iodine	Iodo	Plutonium	Plutonia
Berkelium	Bekelia	Iridium	Iridia	Polonium	Polonia
Beryllium	Berilia	Iron	Irin	Potassium	
Bismuth		Krypton		Praseodymium	Praseodimia
Boron		Lanthanum		Promethium	Prometia
Bromine		Lawrencium	Laurensia	Protactinium	Purotatinia
Cadmium	Kadmia	Lead	Ojé	Sodium	Soda
Calcium	Kalsia	Lithium	Litia	Strontium	Strontia
Californium	Kalifonia	Lutetium	Lutetia	Sulfur	Imi-ọjó,
					Sulfuri
Carbon	Eedu	Magnesium	Magnesia	Tantalum	
Cerium	Ceria	Manganese		Technetium	Teknetia
Cesium	Cesia	Mendelevium	Mendelefia	Tellurium	Teluria
Chlorine	Òyì-iyò (Oyiyò)	Mercury		Terbium	Tebia
Chromium	Cromia	Molybdenum		Thallium	Talia
Cobalt		Neodymium	Neodimia	Thorium	Toria
Copper	Kopa, Kobo	Nickel	Nikeeli	Thulium	Tulia
Curium	Curia	Niobium	Niobia	Tin	Stania
Dysprosium	Disprosia	Nitrogen	Òyì-ilè,Oyile	Titanium	Titania
Einsteinium	Ainstenia	Nobelium	Nobelia	Tungsten	
Element 104	Ìṣù-átòmù 104	Osmium	Osmia	Uranium	Urania
Element 105	Ìşù 105	Oxygen	Òyì-iná	Vanadium	Fanadia

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Element 106	Ìşù 106	Palladium	Paladia	Xenon	Senia
Erbium	Ebia	Radium	Radia	Ytterbium	Yitabia
Europium	Europia	Radon	Redonu	Yttrium	Yitiria
Fermium	Femia	Rhenium	Renia	Zinc	
Fluorine		Rhodium	Rodia	Zirconium	Saconia
Francium	Fransia	Rubidium	Rubidia		
Gadolinium	Gadolinia	Ruthenium	Rutenia		
Gallium	Galia	Samarium	Samaria		

CHEMISTRY TERMS / ÀWON ÈNÒ ÈKÓ-ELÀ

English	Yoruba	Use	Ìlò
Reaction	Àsè	The reaction of A with	Àsè A pệlú B sun D
		B produces D	
React	Láti se (àwon èsè)	React A with B	Se A pệlú B
	, , ,		
Product	Èsún	The product of the	Èsún àsè A pệlú (or àti) B ni
		reaction of A with	D àti F
		(and) B are D and F	
Product of a reaction	Èsún àsè	What are the products	Kini àwọn ệsún àsè A àti (or
		of the reaction of A	pệlú) B
		and B	
D	+ + + + + + + + + + + + + + + + + + +	3371 · · · · · · · · · · · · · · · · · · ·	
Reactant	Èsè, Èsè-àsè	What are the reactants	Àwon èsè-àsè wo ló sun D or
C 1.4	À => > => = == <	that produced D	Kini àwọn èsè-àsè tó sun D
Complete reaction Incomplete reaction	Àsè-ìsèparí		
Reactor	Àsè-àìsèparí		
	Abó-àsè		
Equilibrium	Agbede		
Equilibrium Point	Ibi agbede		
Equilibrium reaction	Àsè adúrólágbede		
Equilibrium constant	Òòkà-àìyệ ibi-		
	agbede		
Equilibrium Point	Ibi agbede		
Endothermic	Àsè aloná		
Reaction	1 2 6 7		
Exothermic reaction	Àsè afaná		
Energy	Agbára		
Heat energy	Agbára Okun-iná	Heat = Okun-iná	
Hotness	Ìgbóná		
Free energy	Agbára wíwúlò		
Gibb's free energy	Agbára wíwúlò ti		
II-1114 £	Gíbìsì		
Helmholtz free	Agbára wíwúlò ti		
energy	Hélmhóòsì		
Temperature	Ìgbóná		
Temperature scale	Ìdíwòn ìgbóná		
Entropy	Ìdàrú		
Enthalpy	Agbára ***		
Enthalpy of a			
reaction		<u> </u>	