

Review of Central Potential - always covered in undergrad. Q.M.

$V(\vec{r}) = V(r) \leftarrow$  spherically symmetric

$H = \frac{\vec{p}^2}{2m} + V(r)$

$\hookrightarrow \vec{p} = -i\hbar \vec{\nabla} \quad \nwarrow \quad \frac{1}{r^2} \frac{\partial}{\partial r} \left( r^2 \frac{\partial}{\partial r} \right) = \frac{\vec{L}^2}{\hbar^2 r^2} \quad \leftarrow$  spherical coordinates

Orbital angular momentum is conserved

$$\begin{aligned} [L_z, \vec{p}^2] &= [x p_y - y p_x, p_x^2 + p_y^2 + p_z^2] \\ &= [x p_y, p_x^2] + [x p_y, p_y^2] + [x p_y, p_z^2] \\ &\quad - [y p_x, p_x^2] - [y p_x, p_y^2] - [y p_x, p_z^2] \\ &= x p_y p_x p_x - p_x p_x x p_y - y p_x p_y p_y + p_y p_y y p_x \end{aligned}$$

$$\begin{aligned} & \quad \quad \quad x p_x + \underbrace{[p_x, x]}_{-i\hbar} \quad \quad \quad y p_y + \underbrace{[p_y, y]}_{-i\hbar} \end{aligned}$$

$$\begin{aligned} &= x p_y p_x p_x - p_x (x p_x - i\hbar) p_y - y p_x p_y p_y + p_y (y p_y - i\hbar) p_x \\ &= x p_y p_x p_x - p_x x p_x p_y + i\hbar p_x p_y - y p_x p_y p_y + p_y y p_y p_x - i\hbar p_y p_x \end{aligned}$$

$$\begin{aligned} & \quad \quad \quad x p_x + \underbrace{[p_x, x]}_{-i\hbar} \quad \quad \quad y p_y + \underbrace{[p_y, y]}_{-i\hbar} \end{aligned}$$

$$\begin{aligned} &= x p_y p_x p_x - [x p_x - i\hbar] p_x p_y + i\hbar p_x p_y - y p_x p_y p_y \\ &= 0 \quad \quad \quad + y p_y p_y p_x - i\hbar p_y p_x - i\hbar p_y p_x \end{aligned}$$

$[L_z, V(r)] = [-i\hbar \frac{\partial}{\partial \phi}, V(r)] = 0$

$\therefore$  similarly  $[L_x, V(r)] = [L_y, V(r)] = 0$   
 $\therefore [L^2, V(r)] = 0$

$\therefore \boxed{[L_z, H] = 0}$   $\rightarrow$  orbital ang. mom. is conserved  
 $\boxed{[L^2, H] = 0}$

$\Rightarrow$  Can make simultaneous eigenkets of  $H, L_z, L^2$

(2)

# Schrodinger Equation for Central Potential

$$\text{S.E. } H \Psi_E(r, \theta, \phi) = E \Psi_E(r, \theta, \phi)$$

$$R(r) Y_{lm}(\theta, \phi)$$

$$\therefore \left[ \frac{\partial}{\partial r} \left( r^2 \frac{\partial R}{\partial r} \right) - \frac{2mr^2}{\hbar^2} \left[ V(r) + \frac{\hbar^2 l(l+1)}{2mr^2} - E \right] \right] R = 0$$

$V_{\text{eff}}(r)$   $\nwarrow$  ang. mom. barrier

- Need  $V(r)$  to solve
- $R = R(E, l, r)$
- No  $m$  dependence

Transform  $R(r) = \frac{u(r)}{r}$   $\leftarrow$  need  $u = rR \rightarrow 0$  as  $r \rightarrow 0$  for well behaved  $R$

$$-\frac{\hbar^2}{2m} \frac{d^2 u}{dr^2} + \left[ V(r) + \frac{\hbar^2 l(l+1)}{2mr^2} \right] u = E u$$

Looks like 1-D S.E. but  $V_{\text{eff}}(r) = V(r) + \frac{\hbar^2 l(l+1)}{2mr^2}$

- dif. eq. for each  $l$
  - $r \geq 0$
  - $u = rR \rightarrow 0$  as  $r \rightarrow 0$
  - depends on  $l$  but not  $m$
- $\therefore$  same  $E_l$  for  $m = -l, \dots, l$

Solve for specific  $V(r)$

① Free particles  $V(r) = 0$

② Infinite square well

$$V(r) = \begin{cases} 0, & r \leq a \\ \infty, & r > a \end{cases}$$

③ Hydrogen "atom"  $V(r) = -\frac{1}{4\pi\epsilon_0} \frac{Ze^2}{r}$   
(ion for  $z > 1$ )  $\nearrow$

(3)

Hydrogen Atom

$$V(r) = -\frac{1}{4\pi\epsilon_0} \frac{Ze^2}{r}$$

Radial equation:  $-\frac{\hbar^2}{2m} \frac{d^2 u}{dr^2} + \left[ V(r) + \frac{\hbar^2 l(l+1)}{2mr^2} \right] u = Eu$

$u = rR$  so that  $u \rightarrow 0$  as  $r \rightarrow 0$   
for well behaved  $R$

negative  
for bound  
states

Clean up notation:

$$K = \sqrt{\frac{-2mE}{\hbar^2}}$$

$$\rho = \frac{mZe^2}{2\pi\epsilon_0 \hbar^2 K}$$

$$\rho = Kr$$

$$\frac{d^2 u}{d\rho^2} = \left[ 1 - \frac{\rho_0}{\rho} + \frac{l(l+1)}{\rho^2} \right] u$$

Asymptotic forms

large  $\rho$ :  $\frac{d^2 u}{d\rho^2} = u \leadsto u = Ae^{-\rho} + Be^{\rho}$

need  $B=0$   
to keep  $u$  finite  
when  $\rho \rightarrow \infty$

small  $\rho$ :  $\frac{d^2 u}{d\rho^2} = \frac{l(l+1)}{\rho^2} u \leadsto u = Cp^{l+1} + D\rho^{-l}$

need  $D=0$   
to keep  $u$  finite  
when  $\rho \rightarrow 0$

(Note:  $l=0 \rightarrow u \sim D$   
 $rR \sim 0$   
cannot have  $\uparrow$   
 $u \rightarrow 0$  as  $r \rightarrow 0$ )

Put in asymptotic behavior

$$u = e^{-\rho} \rho^{l+1} v(\rho)$$

hope  $v$  is simpler

$$\rho \frac{d^2 v}{d\rho^2} + 2(l+1-\rho) \frac{dv}{d\rho} + [\rho_0 - 2(l+1)]v = 0$$

Seek power series solution:  $v = \sum_{j=0}^{\infty} c_j \rho^j$

$$\Rightarrow c_{j+1} = \frac{2(j+l+1) - \rho_0}{(j+1)(j+2l+2)} c_j$$

Need  $j_{\max}$  to have power series converge as  $\rho \rightarrow \infty$

-  $e^{-\rho}$  will make  $R$  well behaved at large  $\rho$   
if series is finite

termination:

$$c_{j_{\max}} \neq 0$$

$$c_{j_{\max}+1} = 0$$

$$2(j_{\max} + l + 1) - \rho_0 = 0$$

$$\equiv n$$

$\longleftrightarrow$  integer

$$\therefore \underline{\underline{\rho_0 = 2n}}$$

$$j_{\max} \geq 0$$

quantum numbers:

$$l = 0, 1, 2, \dots$$

$$j_{\max} = 0, 1, 2, \dots$$

$$\left. \begin{array}{l} l = 0, 1, 2, \dots \\ j_{\max} = 0, 1, 2, \dots \end{array} \right\} n = 1, 2, \dots$$

$$1 = n = j_{\max} + l + 1 \rightarrow \begin{array}{c} l = -j_{\max} \\ \geq 0 \end{array} \rightarrow \begin{array}{c} l = 0 \\ \geq 0 \end{array}$$

$$2 = n = j_{\max} + l + 1 \rightarrow \begin{array}{c} l = 1 - j_{\max} \\ \geq 0 \end{array} \rightarrow \begin{array}{c} l = 0, 1 \\ \geq 0 \end{array}$$

etc.

$$\underline{\underline{l = 0, 1, \dots, n-1}}$$

$$n = 1, 2, \dots$$

$$l = 0, 1, \dots, n-1$$

$$m = -l, -l+1, \dots, l$$

(5)

# Natural scaling of $r$

Recall  $k = \sqrt{\frac{-2mE}{\hbar^2}}$ .

$$\rho_0 = \frac{mZe^2}{2\pi\epsilon_0\hbar^2 k}$$

$$\rho = 2n$$

$$\rho = kr = \frac{r}{\left(\frac{1}{k}\right)}$$

$$k = \frac{mZe^2}{2\pi\epsilon_0\hbar^2 \rho_0}$$

$$= \frac{mZe^2}{2\pi\epsilon_0\hbar^2 2n}$$

$$= \underbrace{\frac{me^2}{4\pi\epsilon_0\hbar^2}}_{\frac{1}{a_B}} \frac{Z}{n}$$

$$a_B = \frac{4\pi\epsilon_0\hbar^2}{me^2}$$

$$= 0.0529 \text{ nm}$$

$$\frac{1}{k} = a_B \frac{n}{Z}$$

← bigger for bigger  $n$

← smaller for bigger  $Z$

## Energy eigenvalues

$$E = -\frac{\hbar^2 k^2}{2m} = -\frac{\hbar^2}{2m} \frac{Z^2}{a_B^2 n^2} = \frac{a_B^2}{(4\pi\epsilon_0)^2 \hbar^4} \frac{m^2 e^4}{n^2}$$

$$E_n = -\frac{me^4}{2(4\pi\epsilon_0)^2 \hbar^2} \frac{Z^2}{n^2}$$

6

$$\psi_3^{(0)}(r) = \eta_1^{(0)\dagger} \eta_2^{(0)\dagger} r^2 e^{-r/3a_0} \propto \left(1 - \frac{2}{3} \frac{r}{a_0} + \frac{2}{27} \frac{r^2}{a_0^2}\right) e^{-r/3a_0} \quad (64)$$

$$\psi_1^{(1)}(r) = r e^{-r/2a_0} \quad (65)$$

$$\psi_2^{(1)}(r) = \eta_1^{(1)\dagger} r^2 e^{-r/3a_0} \propto r \left(1 - \frac{r}{6a_0}\right) e^{-r/3a_0} \quad (66)$$

$$\psi_1^{(2)}(r) = r^2 e^{-r/3a_0} \quad (67)$$

Exercise 8. Obtain the results stated in Eqs. (66) and (68).

The customary notation for these radial wavefunctions is  $R_{nl}$ , where  $n = l + j$ . With this notation, and with the appropriate normalization factors, Eqs. (62)–(67) become

$$R_{10}(r) = \frac{2}{a_0^{3/2}} e^{-r/a_0} \quad (68)$$

$$R_{20}(r) = \frac{2}{(2a_0)^{3/2}} \left(1 - \frac{r}{2a_0}\right) e^{-r/2a_0} \quad (69)$$

$$R_{30}(r) = \frac{2}{(3a_0)^{3/2}} \left(1 - \frac{2}{3} \frac{r}{a_0} + \frac{2}{27} \frac{r^2}{a_0^2}\right) e^{-r/3a_0} \quad (70)$$

$$R_{21}(r) = \frac{1}{(2a_0)^{3/2}} \frac{1}{\sqrt{3}} \frac{r}{a_0} e^{-r/2a_0} \quad (71)$$

$$R_{31}(r) = \frac{1}{(3a_0)^{3/2}} \frac{4\sqrt{2}}{3} \frac{r}{a_0} \left(1 - \frac{r}{6a_0}\right) e^{-r/3a_0} \quad (72)$$

$$R_{32}(r) = \frac{1}{(3a_0)^{3/2}} \frac{2\sqrt{2}}{27\sqrt{5}} \frac{r^2}{a_0^2} e^{-r/3a_0} \quad (73)$$

The normalization condition satisfied by these radial wavefunctions is similar to Eq. (34):

$$\int_0^\infty (R_{nl})^2 r^2 dr = 1$$

The normalization of the wavefunctions  $R_{nl}$  can be checked by evaluating this integral. Alternatively, we can use the results of Section 6.3 (see Problem 10).

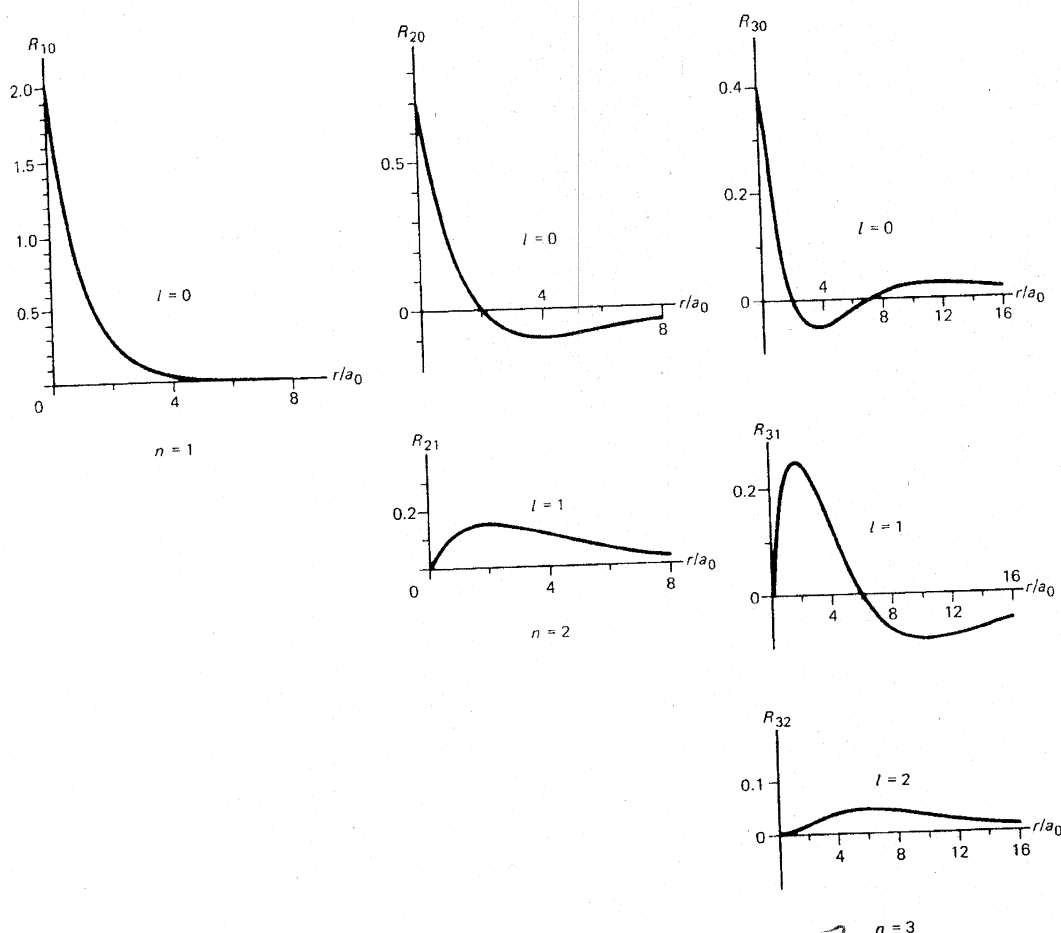


Fig. 8.3 Plots of the radial wavefunctions  $R_{10}$ ,  $R_{20}$ ,  $R_{30}$ ,  $R_{21}$ ,  $R_{31}$ , and  $R_{32}$ .

Figure 8.3 shows plots of the radial wavefunctions  $R_{nl}$ . All the wavefunctions with  $l = 0$  have a maximum at  $r = 0$ . The other wavefunctions all vanish at  $r = 0$ . This says that when the electron is in a state of zero angular momentum, the probability for finding the electron in a volume element  $dV = dx dy dz$  is largest at the nucleus.<sup>1</sup> Note, however, that the large probability per unit volume at  $r = 0$  does not necessarily mean that this is the most proba-

<sup>1</sup> The electron does not interact with the nuclear material except by the Coulomb interaction; thus, the electron can move through the nucleus without hindrance.

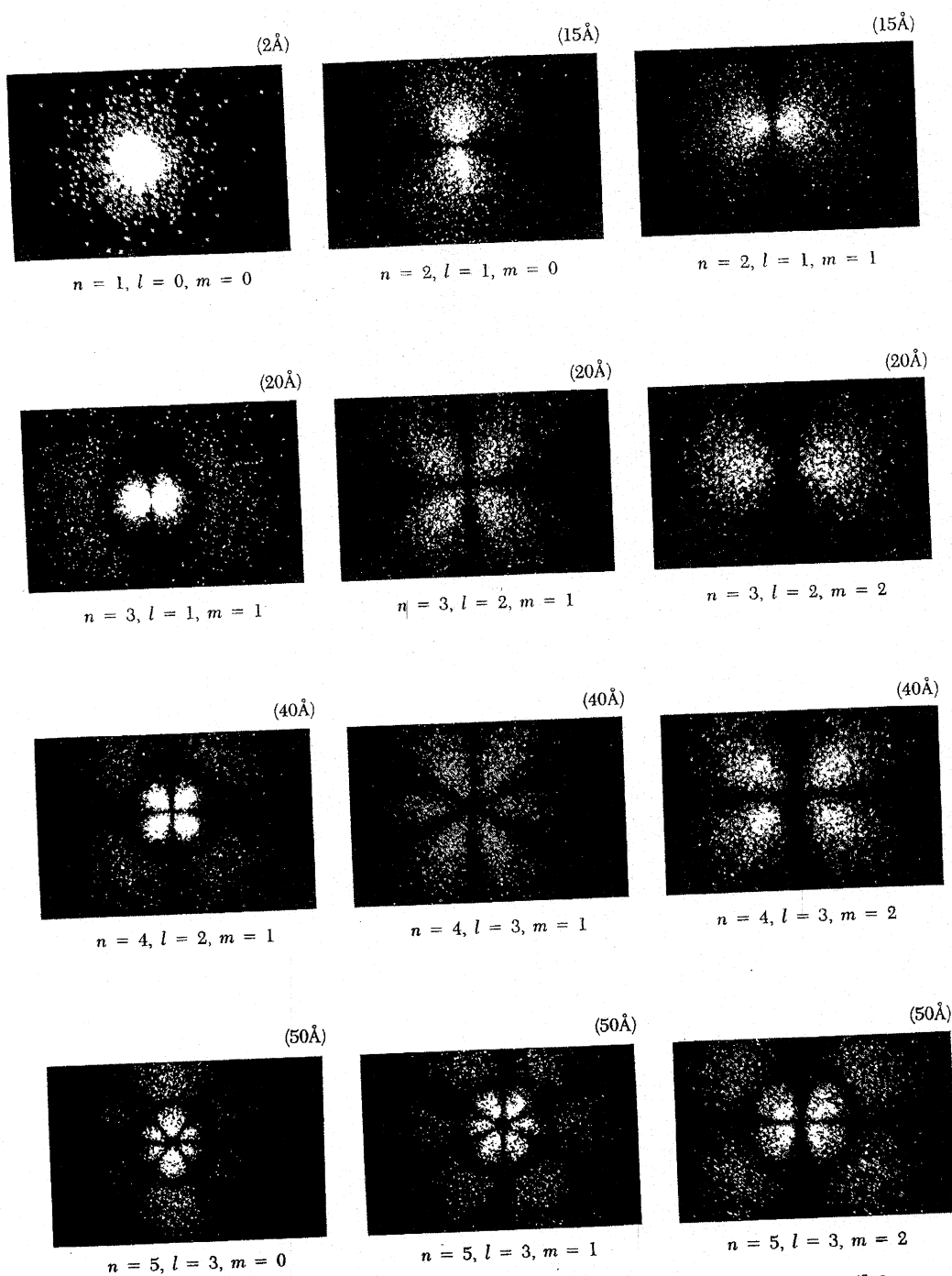


Fig. 8.5 Pictures of the probability distributions  $|\psi_{nlm}(r, \theta, \phi)|^2$  for some states of the hydrogen atom. The density of dots in these pictures is proportional to the probability. These pictures were generated by a Monte-Carlo computer program that selected points  $r, \theta, \phi$  at random, and then decided to plot them or not plot them according to the value of  $|\psi_{nlm}(r, \theta, \phi)|^2$ . (Courtesy A. F. Burr and A. Fisher, New Mexico State University.)



(9)

# Free Particle - Spherical Coordinates

Summary only

$$\begin{aligned}
 H &= \frac{\vec{p}^2}{2m} \\
 &= -\frac{\hbar^2}{2m} \nabla^2 \\
 &= -\frac{\hbar^2}{2m} \left[ \frac{1}{r^2} \frac{\partial}{\partial r} \left( r^2 \frac{\partial}{\partial r} \right) - \frac{\vec{L}^2}{\hbar^2 r^2} \right] \\
 &= -\frac{\hbar^2}{2m} \frac{1}{r^2} \frac{\partial}{\partial r} \left( r^2 \frac{\partial}{\partial r} \right) + \underbrace{\frac{\vec{L}^2}{2mr^2}}_{\text{all angular dep.}}
 \end{aligned}$$

Look for separable solution:  $R(r) Y_{lm}(\theta, \phi)$

now any  $>0$  energy is possible

$$E R Y_{lm} = \left[ -\frac{\hbar^2}{2m} \frac{1}{r^2} \frac{\partial}{\partial r} \left( r^2 \frac{\partial}{\partial r} \right) + \frac{\hbar^2 l(l+1)}{2mr^2} \right] R Y_{lm}$$

Let  $E = \frac{\hbar^2 k^2}{2m}$   $\Leftarrow$  defines  $k$

$$\begin{aligned}
 0 &= \left[ \frac{\hbar^2 k^2}{2m} + \frac{\hbar^2}{2m} \frac{1}{r^2} \frac{\partial}{\partial r} \left( r^2 \frac{\partial}{\partial r} \right) - \frac{\hbar^2}{2m} \frac{l(l+1)}{r^2} \right] R(r) \\
 &= \left[ 1 + \frac{1}{(kr)^2} \frac{\partial}{\partial (kr)} \left( (kr)^2 \frac{\partial}{\partial (kr)} \right) - \frac{l(l+1)}{(kr)^2} \right] j_l(kr)
 \end{aligned}$$

$\Rightarrow$  Well behaved solutions

spherical Bessel functions  $\left\{ \begin{aligned} j_l(kr) &= (kr)^l \left( -\frac{1}{kr} \frac{d}{d(kr)} \right)^l \frac{\sin(kr)}{kr} \end{aligned} \right.$

e.g.  $j_0(\rho) = \frac{\sin \rho}{\rho}$

$$j_1(\rho) = \frac{\sin \rho}{\rho^2} - \frac{\cos \rho}{\rho}$$

$$j_2(\rho) = \left( \frac{3}{\rho^3} - \frac{1}{\rho} \right) \sin \rho - \frac{3}{\rho^2} \cos \rho$$

Expand functions in complete set

$$e^{ikz} = \sqrt{4\pi} \sum_{l=0}^{\infty} \sqrt{2l+1} i^l Y_{l0}(\theta, \phi) j_l(kr)$$

↑

plane wave — important because  
used in scattering theory