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The vibrational Stark effect

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Expressions for the frequency and intensity of the fundamental-vibrational transitions of a polyatomic molecule in the presence of a uniform or nonuniform electric field are reported. They have been developed on the basis of perturbation theory and take into account, individually, first-order electrical and mechanical anharmonicity. They are more general than previously-used expressions, since terms which are quadratic in the field strength are included. As an example of their use and of their interpretative capability, they are applied to CO.

I. INTRODUCTION

The Stark effect¹ is the generic name for the effect of an electric field on optical spectra. An electric field (F) perturbs the vibration of a molecule and consequently alters its infrared spectrum and this is known as the vibrational Stark effect (VSE). The changes caused by the field to the frequency (ω_n) and intensity (A_n) of a given fundamental vibrational transition $(n \leftarrow 0)$ are experimentally represented by the vibrational Stark tuning rate $(\delta_{vE} = d\omega_n/dF)$, where $\hbar\omega_n$ is the *n*th fundamental-vibrational transition energy in the presence of the field) and the fractional change in the infrared cross section with respect to the field $[\delta_{SE} = A_n^{-1}(\partial A_n/\partial F)]$.

The source of the perturbation may simply be that of an externally-applied field, or it may be a field induced by collisions, or a field induced by surrounding the molecule by point charges, or one emanating from chemiabsorption of the molecule on a metal surface, or it may lie in the fact that the molecule is trapped in a chemical cage. This variety of situations permits the vibrational Stark effect to be a rich field for exploration from both experimental and theoretical standpoints. In some cases, the effect is quite dramatic as with the appearance of formerly forbidden vibrational transitions; this was first noted by Condon,² and later reemphasized by Woodward,³ and seen in the early experimental work by Crawford and co-workers⁴ on the vibrational rotation absorption of H2. Since then, there have been many more experiments on all kinds of species and in all kinds of environments. Martí et al. have drawn attention to some of the more recent ones; these include measurements at electrodes, in zeolites, and in special infrared cells.

Theoretical studies are important not only in their own right, as is the case with the calculation of any molecular property, but also because, when linked to the experimental results, they allow us to make inferences concerning the electric field of the molecular surroundings: for example, a transition-metal surface or a zeolite cage. One such attempt for the latter is the work of Goulay et al.⁶

For the calculation of the change in a fundamentalvibrational transition frequency $(\Delta\omega)_n$ by an electric field, there are three quantum mechanical techniques which can be applied in the context of VSE. These are (a) the calculation of the potential energy with a Hamiltonian which includes the electric field with a fixed value (F). The vibrational Schrödinger equation is then solved for the nuclear motion and different vibrational energies are obtained for different values of F. In the context of the present study, Bauschlicher⁷ has used this approach to consider the effects of an electric field on ω_n for a free CO molecule; he used a finite difference method to solve the vibrational Schrödinger equation; Caffarel et al.⁸ and Andzelm et al.,⁹ in their work on CO absorbed on Pd clusters, have used a quantum Monte Carlo method to solve the vibrational Schrödinger equation. In all cases, the results are a series of changes for a set of chosen field strengths. Alternatively, (b) the derivatives with respect to the normal coordinates (Q_n) of the field-perturbed energies may be found analytically and directly without finding the perturbed vibrational energies. For self-consistent-field (SCF) calculations, evaluation of such derivatives is now rather routine^{10,11} and coupled with the FG method of Wilson, Decius, and Cross¹² (i.e., diagonalization of the massweighted Cartesian second derivatives), the ω_n are readily obtained. The electric field (with different fixed finite values) is incorporated in the Fock equation. Duran and coworkers have used this method for CH4, 13 H2O, NH3, HCHO, and C₂H₄,¹⁴ and CO¹⁵ and, by replacing the complete polyelectronic Hamiltonian by a semiempirical one, for a number of larger species.⁵ Andzelm et al.⁹ have taken the same route with a density functional technique in place of the SCF one, and used it in a study of CO on Pd clusters.

The final method (c) at our disposal is that of perturbation theory. Since this is the choice of the present study and the details of its application will follow in the next section, suffice it to say, at this stage, that the change to ω_n by the field can be developed to different orders of the field strength and the results expressed in terms of the derivatives of various molecular electrical properties (dipole moments, polarizabilities, etc.) and spectroscopic constants (anharmonic force constants, etc.) of the unperturbed molecule. Bagus et al., ¹⁶ for systems of CO and CN absorbed on Cu, determined the perturbation terms linear in F for ω_n . The required derivatives were obtained from finite field calculations. Quadratic terms in F, requiring derivatives of the polarizability (α) were not considered. Pacchioni and Bagus¹⁷ dealt with the question of CO on

Cu and Pd clusters with fields stemming from point charges. Lambert, ^{18,19} again with the same first-order perturbation expression, investigated CO on a Ni cluster and in an aqueous double layer; however, his parameters were adapted so as to be appropriate for an absorbed rather than a free species. Cohen de Lara and co-workers have also used limited perturbation-theoretic expressions (anharmonicity was ignored) to study CH₄ (Ref. 20) and CO₂ (Ref. 6) when they are embedded in a zeolite cage. For CO₂, in particular, their results were in disaccord with experiment.

Calculations of changes to the fundamental transition intensities (A_n) in the presence of an electric field are much less common. Nearly all calculations begin with the approximation that expresses A_n in terms of the derivative of the electric transition dipole moment with respect to the normal coordinate Q_n . This derivative can then be obtained analytically from field-perturbed SCF equations. This was done by Duran and co-workers for CO, 15 H₂O, NH₃, HCHO, and C₂H₄ (Ref. 14) and certain larger species at the semi-empirical level.⁵ Results were given as values for particular field strengths. On the other hand, one can, as is explored in detail in this work, use perturbation theory. To date, this has been done only in an incomplete manner; for example, Cohen de Lara and co-workers in their study of N₂²¹ and CH₄,²² where no anharmonicity was taken into account, or by Lambert¹⁹ for CO, where the effects of change in dipole moment which relate to a change in polarizability were not considered.

The perturbation-theoretic treatment of VSE which is given in Sec. II follows, in spirit, the work of Bishop and Kirtman,²³ on vibrational polarizabilities and hyperpolarizabilities. There are several advantageous aspects to the approach taken. Foremost is its general applicability: polyatomic as well as diatomic molecules can be treated, expressions for the perturbed frequencies and intensities can be found as a power series in F, and the coefficients of the series can be directly related to experimental quantities $(\delta_{nE},\delta_{SE})$. Further, there is the possibility of greater understanding and clearer interpretation since the sources of the changes in ω_n and A_n by the field can be identified. From the point-of-view of producing numbers, the general availability of the required parameters from standard computer packages, 10 makes this a straight-forward matter. These parameters may, of course, pertain to the free molecule or be adapted to suit a particular chemical environment. Finally, though a question of taste, many find results which are in the form of a compact mathematical expression to be more aesthetic than a table of numbers corresponding to a set of chosen field strengths.

The perturbations-theoretic treatment, given in Sec. II, differs from previous ones in that it is more complete. That is to say, both electrical and mechanical anharmonicity are taken into account (each individually up to first order); expressions are developed which include quadratic as well as linear terms in F (this, e.g., allows δ_{vE} to be field dependent) and they can also be used where a nonuniform field is present. The common initial assumption that the intensity can be written in terms of $\partial \mu/\partial Q_n$ rather than the fundamental-vibrational transition dipole moment is not

made. Because there have been several previous theoretical VSE investigations of CO, this molecule has been chosen as an *example* of the use of the formulated methodology. It should not be construed as a definitive calculation.

As with all the work mentioned so far, field directions have been tacitly considered to be related to molecule-fixed axis system. That is, the molecule is non-rotating and, experimentally, this comes closest to those cases where a molecule is absorbed on a surface or, perhaps, trapped in a cage. The expressions developed, however, can be simply extended, if necessary, to take into account rotation by using the classical orientational-averaging technique. ²⁴ For H₂, purely quantum-mechanical methods must be used and the theory for doing this has been given by Dows and Buckingham²⁵ and in the work of Poll and co-workers²⁶ on the perturbation of the rotational-vibrational levels of H₂ and its isotopes by point charges.

Reviews which contain certain useful background material for this article have been written by Ashley and Pons²⁷ (experimental), Hirschfelder *et al.*²⁸ (theoretical), Amos¹⁰ (theoretical), Bishop²⁴ (theoretical), and Buckingham²⁹ (theoretical and experimental).

When, in this paper, atomic units are used, they are defined bv^{30}

length
$$(a_0) = 4\pi\epsilon_0 \hbar^2/m_e e^2 \approx 0.529 \ 177 \ \text{Å},$$

energy $(E_h) = \hbar^2/m_e a_0^2 \approx 4.359 \ 75 \times 10^{-18} \ \text{J},$
mass $(m_e) \approx 5.485 \ 80 \times 10^{-4} \ \text{u},$
electric field $= e/4\pi\epsilon_0 a_0^2 \approx 5.142 \ 21 \times 10^{11} \ \text{Vm}^{-1},$
dipole moment $= ea_0 \approx 8.478 \ 36 \times 10^{-30} \ \text{Cm}$
 $\approx 2.541 \ 75 \ \text{D}.$

Circular frequencies (ω) are defined such that $\hbar\omega$ is in units of E_h and the equivalence $E_h{\cong}219$ 474.6 cm⁻¹ can be made. Intensities are usually given in km mol⁻¹ with

$$A_n = (4\pi\epsilon_0)^{-1} (2\pi N_A/3\hbar c^2) g_n \mu_n^2 \omega_n, \tag{1}$$

where μ_n is the *n*th fundamental-vibrational transition dipole moment and g_n the degeneracy of the *n*th fundamental vibrational mode; we can also write

$$g_n \mu_n^2 = (\mu_x)_n^2 + (\mu_y)_n^2 + (\mu_z)_n^2$$

If μ_n and ω_n are in atomic units then

$$A_n = 3.5540 \times 10^6 \ g_n \mu_n^2 \omega_n \tag{2}$$

for A_n to be in km mol⁻¹. In the harmonic oscillator approximation, Eqs. (1) and (2) become

$$A_n = (4\pi\epsilon_0)^{-1} (\pi N_A/3c^2) g_n (\partial \mu/\partial Q_n)^2$$

$$\approx 1.1770 \times 10^6 g_n (\partial \mu/\partial Q_n)^2. \tag{3}$$

II. THEORY

First of all we will consider the effect of a non-uniform electric field on the *frequency* of a fundamental-vibrational

transition of a polyatomic molecule. The perturbation to the vibrational Hamiltonian due to the field(s) F may be written as

$$H' = -\mu_{\alpha} F_{\alpha} - \frac{1}{2} \alpha_{\alpha\beta} F_{\alpha} F_{\beta} - \frac{1}{3} \Theta_{\alpha\beta} F_{\alpha\beta} - \frac{1}{6} \beta_{\alpha\beta\gamma} F_{\alpha} F_{\beta} F_{\gamma}$$

$$- \frac{1}{3} A_{\alpha\beta\gamma} F_{\alpha\beta} F_{\beta\gamma} + \cdots$$

$$(4)$$

where μ_{α} and $\Theta_{\alpha\beta}$ are the dipole moment and quadrupole moment functions; $\alpha_{\alpha\beta}$, $\beta_{\alpha\beta\gamma}$, and $A_{\alpha,\beta\gamma}$ are the polarizability, first hyperpolarizability, and dipole–quadrupole polarizability, respectively. The greek subscripts α , β , ... denote vector or tensor components and can be equal to x,y,z (a molecular-based axis system); a repeated subscript denotes a summation over all three Cartesian components; F_{α} and $F_{\alpha\beta}$ are the electric field and field gradient.

The change in fundamental vibrational frequency is related to the change in the energy difference between the nth fundamental-vibrational level and the ground state and according to perturbation theory this is

$$\hbar(\Delta\omega)_n = \Delta(E_n - E_0) = \langle n | H' | n^F \rangle - \langle 0 | H' | 0^F \rangle, \tag{5}$$

where $|n^F\rangle$ and $|0^F\rangle$ are the field-perturbed vibrational wave functions for the fundamental (n) and ground vibrational states, respectively.

To evaluate the quantities in Eq. (5), we express the field-perturbed wave functions $|n^F\rangle$ and $|0^F\rangle$ in terms of the unperturbed ones. Since we require only terms in $(\Delta\omega)_n$ up to F^2 , we need only use first-order perturbation theory. The first-order wave functions introduce the vibrational polarizabilities 23,24 $\alpha^{\nu}_{\alpha\beta}(n)$ and $\alpha^{\nu}_{\alpha\beta}(0)$ and the complete expression is

$$\begin{split} \tilde{n}(\Delta\omega)_{n} &= -(\Delta\mu_{\alpha})F_{\alpha} - \frac{1}{2}(\Delta\alpha_{\alpha\beta})F_{\alpha}F_{\beta} - \frac{1}{3}(\Delta\Theta_{\alpha\beta})F_{\alpha\beta}) \\ &\quad - \frac{1}{3}(\Delta A_{\alpha,\beta\gamma})F_{\alpha}F_{\beta\gamma} - \frac{1}{2}[\alpha^{\nu}_{\alpha\beta}(n) - \alpha^{\nu}_{\alpha\beta}(0)]F_{\alpha}F_{\beta} \end{split} \tag{6}$$

where for the electric properties (P)

$$(\Delta P) = \langle n|P|n\rangle - \langle 0|P|0\rangle. \tag{7}$$

To determine these expectation values, we initially use the harmonic-oscillator approximation and then augment it by incorporating, individually, first-order electrical and mechanical anharmonicity, The former means writing any electric property as

$$P = P^{0} + \sum_{a} (\partial P/\partial Q_{a}) Q_{a} + \frac{1}{2} \sum_{a,b} (\partial^{2} P/\partial Q_{a} \partial Q_{b}) Q_{a} Q_{b},$$
(8)

where Q_a are the normal coordinates and denoting the second derivatives as the first-order terms. The latter means expressing the potential energy as

$$V = V^0 + \frac{1}{2} \sum_{\alpha} \lambda_{\alpha} Q_{\alpha}^2 + \frac{1}{6} \sum_{a,b,c} f_{abc} Q_{\alpha} Q_b Q_c$$

$$\tag{9}$$

where f_{abc} are the anharmonic force constants; these cubic terms will be taken as the first-order ones. A simplified notation will be convenient for the derivatives of the electrical properties $(\partial P/\partial Q_a)$, etc., namely (P/a), (P/ab), etc., or, more specifically, for $P=\mu_{\alpha}$ and $P=\alpha_{\alpha\beta}$ we will write (α/a) , (α/ab) , and $(\alpha\beta/a)$, $(\alpha\beta/ab)$.

In the harmonic oscillator approximation $\alpha_{\alpha\beta}^{\nu}(n)$ and $\alpha_{\alpha\beta}^{\nu}(0)$ are the same and the first-order electrical and mechanical anharmonic corrections are zero,²³ so these terms, to this order, can now be dropped, the remaining (ΔP) terms may be evaluated from the integrals in Appendix III of Ref. 12. The final result is

$$(\Delta P) = (\hbar/2\omega_n) \left[(P/nn) - \Sigma_a f_{ann} (P/n) \omega_n^{-2} \right]$$
 (10)

where the first term in the square brackets results from the electrical anharmonicity and the second one from the mechanical anharmonicity. In this context, and later, the vibrational frequencies ω_n and ω_a are the harmonic ones. Equations (6) and (10), together, define the change in frequency of a fundamental-vibrational transition when a molecule is placed in an electric field. For a diatomic molecule, Eq. (10) becomes simply

$$(\Delta P) = (\hbar/2\omega_n)[(P/nn) - f_{nnn}(P/n)\omega_n^{-2}]. \tag{11}$$

To approximate, via perturbation theory, the change in the intensity A_n , of the *n*th fundamental-vibrational transition, we begin with Eq. (2). This means that we must determine the shift in $\mu_n^2 \omega_n$. We have already considered the shift in ω_n and are therefore left with finding $\langle 0^F | \mu_\alpha^F | n^F \rangle$ and $\langle 0 | \mu_\alpha | n \rangle$ to first order in the anharmonicities. μ_α^F is defined by

$$\mu_{\alpha}^{F} = \mu_{\alpha} + \alpha_{\alpha\beta} F_{\beta} + \frac{1}{2} \beta_{\alpha\beta\gamma} F_{\beta} F_{\gamma} + \frac{1}{3} A_{\alpha,\beta\gamma} F_{\beta\gamma}. \tag{12}$$

For determination of the term in $\langle 0^F | \mu_{\alpha}^F | n^F \rangle$ which is *linear* in F, we follow in the same fashion as for $\Delta \omega_n$, but for the higher-order terms in F we restrict ourselves to the harmonic oscillator approximation only.

The unperturbed transition dipole integral, to first order in both anharmonicities, separately, is found to be

$$\langle 0 | \mu_{\alpha} | n \rangle = (\hbar/2\omega_n)^{1/2} (\alpha/n). \tag{13}$$

The equivalent field-perturbed integral is more complex. First, $|0^F\rangle$ and $|n^F\rangle$ must be determined to second order in the perturbation $-\mu_{\alpha}F_{\alpha}$ (assuming, for the moment, a uniform field); the rest of the perturbation in Eq. (4), if we consistently limit ourselves to the electrical and mechanical harmonic oscillator approximation for the nonlinear terms in F in $\langle 0^F | \mu_{\alpha}^F | n^F \rangle$, will appear only in $\langle 0 | \mu_{\alpha}^F | n \rangle$. Once $|0^F\rangle$ and $|n^F\rangle$ have been found, using standard perturbation-theoretic formulas, ²⁸ the anharmonic corrections to the harmonic oscillator approximation can be made and an expression for the complete integral found. This requires great care in identifying the intermediate vibrational states which occur in the $|0^F\rangle$ and $|n^F\rangle$ expansions. That done, the actual integration over the harmonic oscillator wave functions is trivial. The final result is

$$\langle 0^{F} | \mu_{\alpha}^{F} | n^{F} \rangle = (\hbar/2\omega_{n})^{1/2} (D_{1} + D_{2}F_{\beta} + \frac{1}{2}D_{3}F_{\beta}F_{\gamma} + \frac{1}{3}D_{4}F_{\beta\gamma}), \tag{14}$$

where

$$D_1 = (\alpha/n), \tag{15}$$

$$D_{2} = (\alpha \beta/n) + (\alpha/n) (\beta/nn) (2\omega_{n})^{-2} + \Sigma_{a}(\alpha/an)$$

$$\times (\beta/a) \omega_{a}^{-2} + \Sigma_{a \neq n} (\alpha/a) (\omega_{a}^{2} - \omega_{n}^{2})^{-1}$$

$$\times [(\beta/an) - \Sigma_{b} f_{abn}(\beta/b) \omega_{b}^{-2}], \tag{16}$$

$$D_3 = (\alpha \beta \gamma / n) - \hbar^{-1}(\alpha / n) \Sigma_a(\beta / a) (\gamma / a) \omega_a^{-3}, \quad (17)$$

$$D_4 = (\alpha, \beta \gamma / n). \tag{18}$$

It is informative to identify the source of the contributions to these D_i terms. In all cases, the initial term in D_i originates from μ_{α}^{F} and is the same as that which would be given in the harmonic oscillator approximation since the first-order anharmonic corrections are zero. D_3 and D_4 have only been found at the level of the harmonic oscillator approximation; the second term of D_3 stems from the second order terms in the expansion of $|0^F\rangle$ and $|n^F\rangle$ —this term would not appear if the conventional intensity approximation, Eq. (3), were used; it is, however, not negligible. The second and third terms and the first part of the fourth term in D_2 arise from the first-order expansion of $|0^F\rangle$ and $|n^F\rangle$ coupled with the first-order electrical anharmonicity in $(\partial^2 P/\partial Q_a \partial Q_b)$; the second part of the fourth term in D_2 is similar to the first part but instead accounts for the mechanical anharmonicity.

In summary, Eqs. (6), (11), (14)–(18) provide a set of working equations, at a consistent level of approximation, which allow changes to ω_n and $\langle 0|\mu_\alpha|n\rangle$ due to an electric field to be evaluated. Combined together they provide the means of determining the changes to the intensity of a fundamental-vibrational transition. The form of the expressions is such that the sources of the changes (anharmonicities, etc.) can be identified.

For diatomic molecules, with a uniform field along the internuclear axis (z), the frequency shift will be

$$(\Delta\omega)_{n} = (2\omega_{n})^{-1} [(z/nn) - f_{nnn}(z/n)\omega_{n}^{-2}] (-F_{z})$$

$$+ (2\omega_{n})^{-1} [(zz/nn) - f_{nnn}(zz/n)\omega_{n}^{-2}]$$

$$\times (-\frac{1}{2}F_{z}^{2}).$$
(19)

Again, the shift $(\Delta\omega)_n$ is to the fundamental frequency and the ω_n on the left-hand side is the harmonic frequency. The linear term is expressed in $(-F_z)$, since the field direction is usually defined with respect to the molecular dipole moment in such a way that a positive field increases $(\Delta\omega)_n$. For the change in the vibrational transition integral, we have for a diatomic molecule:

$$\langle 0^{F} | \mu_{\alpha}^{F} | n^{F} \rangle = (\hbar/2\omega_{n})^{1/2} [(z/n) - \{(zz/n) + (5/4)(z/n) \\ \times (z/nn)\omega_{n}^{-2}\} (-F_{z}) + \{(zzz/n) - \hbar^{-1}(z/n)^{3}\omega_{n}^{-3}\}(\frac{1}{2}F_{z}^{2})].$$
(20)

The form of Eq. (19) follows, transparently, the well-known expression²⁴ for zero-point-vibrational averaging. In comparison with previous work, over and above the use of Eq. (3) rather than Eq. (2), there are certain other points which may now be made. Lambert, ¹⁹ Bagus and co-workers ^{16,17} used only the first (linear) term in Eq.

TABLE I. Molecular properties for CO in atomic units.

	Set I ^a	Set II ^c
ω_n	9.9638×10 ⁻³	1.043×10^{-2}
f_{nnn}	-3.206×10^{-6} b	-3.54×10^{-6}
$\partial \mu_z / \partial Q_n$	5.660×10^{-3}	9.326×10^{-3}
$\partial^2 \mu_z / \partial Q_n^2$	-1.92×10^{-6}	3.049×10^{-5}
$\partial \alpha_{zz}/\partial Q_n$	8.502×10^{-2} c	8.502×10^{-2}
$\partial^2 \alpha_{zz} / \partial Q_n^2$	5.457×10^{-4} °	5.457×10^{-4}
$\partial \beta_{zzz} / \partial \widetilde{Q}_n$		8.209×10^{-2}

^aWith the exceptions noted, these values are taken from Ref. 7. ^bDerived from the spectroscopic constants α_e , B_e and ω_e given in Ref. 7. ^cBased on a SCF calculation using a (10s 6p 4d) basis set for both C and O contracted to [5s 3p 2d]. This basis set was taken from Ref. 31.

(19), Cohen de Lara and co-workers^{6,20} neglected the effects of mechanical anharmonicity and this may well explain the disaccord with experiment which they found.⁶ As for the intensity formula, Eq. (20), Lambert's expression¹⁹ corresponds to just the third term in D_2 in Eq. (17) and Cohen de Lara and co-workers's treatment^{21,22} was limited to simply the first term in D_2 . Finally, the form of the expressions given above, in terms of powers of F, allows for easy association of the coefficients with the experimental parameters (δ_{vE} , δ_{SE}). It may be noted that for a diatomic molecule, at this level of approximation, there are no mechanical-anharmonic contributions to the field-perturbed transition dipole moment in Eq. (20).

III. AN EXAMPLE: CO

As an example of the application of the theory given in the preceding section, the vibrational Stark effect for CO had been chosen. For this molecule there are other treatments^{7,15} available with which comparison can be made. The intention is not to provide definitive values for the frequency and intensity shifts for CO but rather to assess the contributions to the shifts and identify their source.

First, we assemble the required parameters. Initially we take a set, augmented by values of $\partial \alpha_{zz}/\partial Q$ and $\partial^2 \alpha_{zz}/\partial Q^2$, from the work of Bauschlicher.⁷ We do this in order to make a comparison with his finite-field results for $(\Delta \omega)_n$. These parameters are called Set I in Table I. The normal coordinate (using atomic units of mass) for CO is given by

$$Q = \mu^{1/2}(r - r_e) = 111.79(r - r_e)$$

with r being the internuclear separation and r_e =2.132 a_0 . Equation (19) then reads, in atomic units, as

$$(\Delta\omega)_n = 50.182(-1.92 \times 10^{-6} + 1.824 \times 10^{-4})(-F_z) + 50.182(5.46 \times 10^{-4} + 2.740 \times 10^{-3}) \times (-\frac{1}{2}F_z^2)$$
(21)

and, in cm^{-1} , as

$$(\Delta\omega)_n = 1.99 \times 10^3 (-F_z) - 1.81 \times 10^4 (F_z^2).$$
 (22)

Values of $(\Delta \omega)_n$ using the latter equation are given for three different field strengths in Table II, along with the

TABLE II. Values of $(\Delta\omega)_n$ in cm⁻¹ for CO, using Eq. (19) and parameter Set I; these are compared with those of Bauschlicher (Ref. 7).

$-F_z$ (a.u.)	This work	Bauschlicher (Ref. 7)	
0.005	9.50	9.62	-
0.01	18.09	18.11	
0.015	25.78	25.36	

corresponding values found by Bauschlicher. The agreement is very good. Of interest is the fact that it is the mechanical-anharmonic term which dominates both the linear and quadratic terms in F_z and that the quadratic contribution, though small, is significant and will lead to a field dependence in the Stark tuning rate δ_{vE} .

To make a comparison of the values of the frequency and intensity shifts given by the formulas in this work with those values determined by Andrés *et al.*, ¹⁵ the latter's values have been fitted, for the two basis sets which they used (DZ) and DZP), to

$$(\Delta\omega)_n = a(-F_z) + bF_z^2 \tag{23}$$

and

$$(\Delta A)_n = A(-F_z) + BF_z^2. \tag{24}$$

Values of a,b,A, and B are given in Table III and compared with those found directly from the perturbation-theoretic formulas. Also in Table III, the unperturbed ω and A values are shown. Bearing in mind that the basis set³¹ used in this work is different, the agreement is gratifying, the only exception being the coefficient B, but this term is both small and very sensitive to both basis set choice and the values of the derivative parameters. For $(\Delta \omega)_n$, in both the linear and quadratic terms in F_z it is again the mechanical-anharmonic contribution which dominates (by an order of magnitude).

Using the parameters of Set II of this work (Table I), the perturbed transition dipole moment is

$$\langle 0^F | \mu_{\alpha}^F | n^F \rangle = (\hbar/2\omega_n)^{1/2} (9.326 \times 10^{-3} + 8.829 \times 10^{-2} F_z$$

$$-3.164 \times 10^{-1} F_z^2) \tag{25}$$

TABLE III. Coefficients for Eqs. (23) and (24) for CO, with frequency and intensity shifts in cm⁻¹ and km mol⁻¹, respectively, and field strengths in atomic units, together with the unperturbed values of ω_n and A_n .

	Ref. 15 DZ basis	Ref. 15 DZP basis	This work parameter Set II
a	3.7×10^{3}	3.3×10^{3}	3.514×10^{3}
b	-2×10^{4}	-2×10^{4}	-1.743×10^4
A	-3.2×10^{3}	-2.9×10^{3}	-2.69×10^{3}
\boldsymbol{B}	1×10^{4}	1×10^4	-2.30×10^{3}
ω_n	2265	2424	2289
A_n	175	164	154.6

and the coefficient of the linear term is largely dominated by the polarizability derivative $(\alpha \beta/n)$. The perturbed fundamental-vibrational frequency (in cm⁻¹) is similarly given by

$$\omega_n^F = 2289 - 3.514 \times 10^3 F_z - 1.743 \times 10^4 F_z^2.$$
 (26)

This leads to the fact that the linear term (in F_z) in A_n is dominated (by an order of magnitude) by a product of the constant and the linear term in the perturbed transition dipole moment and thereby, by implication, by the polarizability derivative. The quadratic term in A_n is a rather delicate balance of contributions; the two most important ones are of similar size but of opposite sign.

IV. SUMMARY

As an alternative to finite-field calculations of the vibrational Stark effect, perturbation-theoretic formulas have been developed at a consistent level of approximation. These formulas require knowledge of the derivatives of the dipole moment and polarizability functions with respect to the normal coordinates; quantities which are now readily available from standard quantum-chemical computer packages. The formulas have been applied to CO and it is seen that the major contribution to the shifts in fundamental-vibrational frequencies and intensities caused by an electric field may be identified. It has also been shown that the formulas lead to results which are compatible with previously-published finite-field calculations for CO.

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