

# Bohm's Quantum Potential

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The **Bohm** or **Bohmian interpretation** of [quantum mechanics](#), which Bohm called the **causal**, or later, the **ontological interpretation**, is an [interpretation](#) postulated by [David Bohm](#) in 1952 as an alternative to the standard [Copenhagen interpretation](#). The Bohm interpretation grew out of the search for an alternative model based on the assumption of [hidden variables](#). Its basic formalism corresponds in the main to [Louis de Broglie's pilot-wave theory](#) of 1927. Consequently it is sometimes called the **de Broglie-Bohm theory**.

The Bohm interpretation is [causal](#) but not [local](#) and is non-[relativistic](#). It has not been disproven, but there are other schemes (such as the [Copenhagen interpretation](#)) that give the same theoretical predictions, so are equally confirmed by the experimental results.

The Bohm interpretation is based on these principles:

- **Every particle travels in a definite path**
- **We do not know what that path is**
- **The state of N particles is affected by a 3N dimensional field, which guides the motion of the particles**

De Broglie called this the **pilot wave**; Bohm called it the  **$\psi$ -field**. This field has a piloting influence on the motion of the particles. The *quantum potential* is derived from the  **$\psi$ -field**.

- **This 3N dimensional field satisfies the [Schrödinger equation](#)**

Mathematically, the field corresponds to the [wavefunction](#) of conventional quantum mechanics, and evolves according to the [Schrödinger equation](#). The positions of the particles do not affect the wave function.

- **Each particle's momentum  $\mathbf{p}$  is  $\nabla S(\mathbf{x}, t)$**
- **The particles form a statistical ensemble, with probability density  $\rho(\mathbf{x}, t) = |\psi(\mathbf{x}, t)|^2$**

## Reformulating the Schrödinger equation

Some of Bohm's insights are based on a reformulation of the [Schrödinger equation](#); instead of using the wavefunction  $\psi(\mathbf{x}, t)$ , he defines the wavefunction as

$$\psi(\mathbf{x}, t) = R(\mathbf{x}, t)e^{iS(\mathbf{x}, t)/\hbar}$$

and solves it for the (real) magnitude function  $R(\mathbf{x}, t)$  and (real) phase function  $S(\mathbf{x}, t)$ .

The [Schrödinger equation](#) for one particle of mass  $m$  is

$$i\hbar \frac{\partial \psi(\mathbf{x}, t)}{\partial t} = \frac{-\hbar^2}{2m} \nabla^2 \psi(\mathbf{x}, t) + V(\mathbf{x}) \psi(\mathbf{x}, t),$$

where the [wavefunction](#)  $\psi(\mathbf{x}, t)$  is a complex function of the spatial coordinate  $\mathbf{x}$  and time  $t$ . It can then be split into two coupled equations by expressing it in terms of  $R$  and  $S$ :

$$\begin{aligned} \frac{\partial R(\mathbf{x}, t)}{\partial t} &= \frac{-1}{2m} [R(\mathbf{x}, t) \nabla^2 S(\mathbf{x}, t) + 2 \nabla R(\mathbf{x}, t) \cdot \nabla S(\mathbf{x}, t)] \\ \frac{\partial S(\mathbf{x}, t)}{\partial t} &= - \left[ V + \frac{1}{2m} (\nabla S(\mathbf{x}, t))^2 - \frac{\hbar^2}{2m} \frac{\nabla^2 R(\mathbf{x}, t)}{R(\mathbf{x}, t)} \right]. \end{aligned}$$

The probability density  $\rho(\mathbf{x}, t)$  is a real function defined as the magnitude of the wave function:

$$\rho(\mathbf{x}, t) = |\psi(\mathbf{x}, t)|^2 = R^2(\mathbf{x}, t)$$

Thus

$$\psi(\mathbf{x}, t) = \sqrt{\rho(\mathbf{x}, t)} e^{iS(\mathbf{x}, t)/\hbar}.$$

Therefore we can substitute  $\rho(\mathbf{x}, t)$  for  $R^2(\mathbf{x}, t)$  and get:

$$-\frac{\partial \rho(\mathbf{x}, t)}{\partial t} = \nabla \cdot \left( \rho(\mathbf{x}, t) \frac{\nabla S(\mathbf{x}, t)}{m} \right) \quad (1)$$

$$-\frac{\partial S(\mathbf{x}, t)}{\partial t} = V(\mathbf{x}) + Q(\mathbf{x}, t) + \frac{1}{2m} (\nabla S(\mathbf{x}, t))^2 \quad (2)$$

where

$$Q(\mathbf{x}, t) = -\frac{\hbar^2}{2m} \frac{\nabla^2 R(\mathbf{x}, t)}{R(\mathbf{x}, t)} = -\frac{\hbar^2}{2m} \frac{\nabla^2 \sqrt{\rho(\mathbf{x}, t)}}{\sqrt{\rho(\mathbf{x}, t)}}$$

Bohm called the function  $Q(\mathbf{x}, t)$  the *quantum potential*.

We can use the same argument on the many-particle [Schrödinger equation](#):

$$i\hbar \frac{\partial \psi(\mathbf{x}_1, \mathbf{x}_2, \dots, t)}{\partial t} = \sum_i \frac{-\hbar^2}{2m_i} \nabla_i^2 \psi(\mathbf{x}_1, \mathbf{x}_2, \dots, t) + V(\mathbf{x}_1, \mathbf{x}_2, \dots) \psi(\mathbf{x}_1, \mathbf{x}_2, \dots, t),$$

where the  $i$ -th particle has mass  $m_i$  and position coordinate  $\mathbf{x}_i$  at time  $t$ . The [wavefunction](#)  $\psi(\mathbf{x}_1, \mathbf{x}_2, \dots, t)$  is a complex function of all the  $\mathbf{x}_i$  and time  $t$ .  $\nabla_i$  is the grad operator with respect to  $\mathbf{x}_i$ , i.e. of the  $i$ -th particle's position coordinate. As before the probability density  $\rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)$  is a real function defined by

$$\rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t) = |\psi(\mathbf{x}_1, \mathbf{x}_2, \dots, t)|^2.$$

As before, we can define a real function  $S(\mathbf{x}_1, \mathbf{x}_2, \dots, t)$  to be the [complex phase](#), so that we can define a similar relationship to the 1-particle example:

$$\psi(\mathbf{x}_1, \mathbf{x}_2, \dots, t) = \sqrt{\rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)} e^{iS(\mathbf{x}_1, \mathbf{x}_2, \dots, t)/\hbar}.$$

We can use the same argument to express the Schrödinger equation in terms of  $\rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)$  and  $S(\mathbf{x}_1, \mathbf{x}_2, \dots, t)$ :

$$-\frac{\partial \rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)}{\partial t} = \sum_i \nabla_i \cdot (\rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t) \frac{\nabla_i S(\mathbf{x}_1, \mathbf{x}_2, \dots, t)}{m_i}) \quad (3)$$

$$\begin{aligned} -\frac{\partial S(\mathbf{x}_1, \mathbf{x}_2, \dots, t)}{\partial t} &= V(\mathbf{x}_1, \mathbf{x}_2, \dots) + Q(\mathbf{x}_1, \mathbf{x}_2, \dots, t) \\ &+ \sum_i \frac{1}{2m_i} (\nabla_i S(\mathbf{x}_1, \mathbf{x}_2, \dots, t))^2 \end{aligned} \quad (4)$$

where

$$Q(\mathbf{x}_1, \mathbf{x}_2, \dots, t) = - \sum_i \frac{\hbar^2}{2m_i} \frac{\nabla_i^2 \rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)}{\rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)} = - \sum_i \frac{\hbar^2}{2m_i} \left( \frac{\nabla_i^2 \rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)}{2\rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)} - \left( \frac{\nabla_i \rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)}{2\rho(\mathbf{x}_1, \mathbf{x}_2, \dots, t)} \right)^2 \right).$$

## One-particle formalism

In his 1952 paper, Bohm starts from the reformulated Schrödinger equation. He points out that in equation (2):

$$-\frac{\partial S(\mathbf{x}, t)}{\partial t} = V(\mathbf{x}) + Q(\mathbf{x}, t) + \frac{1}{2m} (\nabla S(\mathbf{x}, t))^2,$$

if one represents the world of classical physics by setting  $\hbar$  to zero (which results in  $Q$  becoming zero) then  $S(\mathbf{x}, t)$  is the solution to the [Hamilton-Jacobi equation](#). He

quotes a theorem that says that if an ensemble of particles (which follow the equations of motion) have trajectories that are normal to a surface of constant  $S$ , then they are normal to all surfaces of constant  $S$ , and that  $\nabla S(\mathbf{x}, t)/m$  is the velocity of any particle passing point  $\mathbf{x}$  at time  $t$ .

Therefore, we can express equation (1) as:

$$-\frac{\partial \rho(\mathbf{x}, t)}{\partial t} = \nabla \cdot (\rho(\mathbf{x}, t)\mathbf{v}).$$

This equation shows that it is consistent to express  $\rho(\mathbf{x}, t)$  as the probability density because  $\rho(\mathbf{x}, t)\mathbf{v}$  is then the mean current of particles, and the equation expresses the conservation of probability.

Of course,  $\hbar$  is non-zero. Bohm suggests that we still treat the particle velocity as  $\nabla S(\mathbf{x}, t)/m$ . The movement of a particle is described by equation (2):

$$-\frac{\partial S(\mathbf{x}, t)}{\partial t} = V(\mathbf{x}) + Q(\mathbf{x}, t) + \frac{1}{2m}(\nabla S(\mathbf{x}, t))^2$$

where

$$Q(\mathbf{x}, t) = -\frac{\hbar^2}{2m} \left( \frac{\nabla^2 \rho(\mathbf{x}, t)}{2\rho(\mathbf{x}, t)} - \left( \frac{\nabla \rho(\mathbf{x}, t)}{2\rho(\mathbf{x}, t)} \right)^2 \right).$$

$V$  is the classical potential, which influences the particle's movement in the ways described by the classical laws of motion.  $Q$  also has the form of a potential; it is known as the *quantum potential*. It influences particles in ways that are specific to quantum theory. **Thus the particle is moving under the influence of a quantum potential  $Q$  as well as the classical potential  $V$ .**

## Many-particle formalism

The momentum of Bohm's  $i$ -th particle's "hidden variable" is defined by

$$\mathbf{p}_i = m_i \mathbf{v}_i = \nabla_i S \quad (3)$$

and the particles' total energy as  $E = -\partial S/\partial t$ ; equation (1) is the [continuity equation](#) for probability with

$$\mathbf{j}_i = \rho \mathbf{v}_i = \rho \frac{\mathbf{p}_i}{m_i} = \rho \frac{\nabla_i S}{m_i},$$

and **equation (4) is a statement that total energy is the sum of the potential energy, quantum potential and the kinetic energies.**