

1. A light source has a power output of 75 W.
 - a) What is the frequency of the light if the entire energy of the source is emitted as light of wavelength $\lambda = 600 \text{ nm}$?
 - b) How many photons are emitted by the light source per second?
2. How many photons per second are emitted by a 7.5 mW carbon dioxide laser if its wavelength is $10.6 \mu\text{m}$?
3. A laser pulse of energy 20 J has a duration of $5 \times 10^{-7} \text{ s}$. The light, of wavelength $\lambda = 580 \text{ nm}$, strikes a metal surface.
 - (a) What is the power of the laser?
 - (b) How many photons strike the metal surface?
 - (c) What is the maximum velocity of the electrons ejected from the metal by the light, if the work function is $3 \times 10^{-19} \text{ J}$? (Electron mass: $m_e = 9.11 \times 10^{-31} \text{ kg}$)
4. What is the energy of a photon of X-ray radiation with a wavelength of 10^{-10} m , expressed in joules and in electronvolts? What are the photon's momentum and mass?
5. A metal surface is illuminated with light of wavelength $\lambda_1 = 492 \text{ nm}$. The maximum kinetic energy of the emitted electrons is $W_{\text{kin1}} = 7.9 \times 10^{-19} \text{ J}$. When the wavelength of the light is changed to $\lambda_2 = 579 \text{ nm}$, the kinetic energy decreases to $W_{\text{kin2}} = 3.8 \times 10^{-20} \text{ J}$. Based on these data, determine the Planck constant and the work function characteristic of the metal!
6. When light of a certain frequency is incident on a metal surface, the measured stopping potential of the emitted electrons is $U_{f1} = 3.19 \text{ V}$. For light of half that frequency, the stopping potential is $U_{f2} = 0.625 \text{ V}$. Determine the work function and the wavelength of the light!

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