Q1.

(a) 
$$f(\varepsilon) = \frac{1}{1 + e^{(\varepsilon - \varepsilon f)/\kappa T}} - at equilibrium}$$

when  $\varepsilon = \varepsilon f$ 
 $f(\varepsilon) = \frac{1}{1 + e^{\circ}} = \frac{1}{2}$ 

Probability for an electron to occupy a state located at formi level is  $50\%$ .

b)  $\varepsilon_f = \varepsilon \varepsilon$ 
 $f(\varepsilon)$  at  $\varepsilon_c + \kappa_T$ 
 $= \frac{1}{1 + e^{\kappa_T/\kappa_T}} = \frac{1}{1 + e^{\kappa_T/\kappa_T}}$ 
 $= \frac{1}{1 + e^{\kappa_T/\kappa_T}} = \frac{1}{1 + e^{\kappa_T/\kappa_T}}$ 
 $= \frac{1}{1 + e^{\kappa_T/\kappa_T}} = \frac{1}{1 + e^{\kappa_T/\kappa_T}}$ 
 $= \frac{1}{1 + e^{\kappa_T/\kappa_T}} = \frac{1}{1 + e^{\kappa_T/\kappa_T}}$ 

c) 
$$\frac{\xi(\xi_{1})}{\xi(\xi_{1})} = 1 - \frac{1}{\xi(\xi_{2})}$$
 $\frac{1}{1 + e^{(\xi_{1} + \xi_{2})} + e^{(\xi_{2} + \xi_{2})} + e^{(\xi_{1} + \xi_{2})} + e^{(\xi_{2} + \xi_{2})} + e^{(\xi_$ 

f(E) = 1 + e(E-Eg)/KT if E = Eg 1 + e(ej-ep/kT 50%

b) 
$$f(\mathcal{E}c) = \frac{1}{1 + e^{(\mathcal{E}c - \mathcal{E}p)/kT}}$$

for finding an electron at  $\mathcal{E}c$ 

$$1 - f(\mathcal{E}v) = 1 - \frac{1}{1 + e^{(\mathcal{E}v - \mathcal{E}p)/kT}}$$

$$= \frac{(\mathcal{E}v - \mathcal{E}p)/kT}{1 + e^{(\mathcal{E}v - \mathcal{E}p)/kT}}$$

We know that
$$f(\mathcal{E}c) = 1 - f(\mathcal{E}v)$$

$$= \frac{1}{1 + e^{(\mathcal{E}c - \mathcal{E}p)/kT}}$$

1 = e(Ev-Eg)/kT (Ec+Ev-2Eg)/kT wring (i), can be written as 1+ e-(Ev-E4)/KT so it becomes. 1 + e(Ec-EP)/KT | + = (Ev-Eg)/KT)  $\frac{\varepsilon_{c}-\varepsilon_{f}}{\varepsilon_{g}}=\frac{-(\varepsilon_{v}-\varepsilon_{g})}{\varepsilon_{g}}$   $\frac{\varepsilon_{f}}{\varepsilon_{g}}=\frac{\varepsilon_{c}+\varepsilon_{v}}{\varepsilon_{g}}$ 

$$E = \mathcal{E}f + \Delta \mathcal{E}$$

$$J(\mathcal{E}) = \frac{1}{1 + e^{(\mathcal{E}g + \Delta \mathcal{E} - \mathcal{E}g)}KT}$$

$$= \frac{1}{1 + e^{\Delta \mathcal{E}/KT}}$$

$$1 - J(\mathcal{E}) = \frac{(\mathcal{E} - \mathcal{E}g)/KT}{1 + e^{(\mathcal{E} - \mathcal{E}g)/KT}}$$

$$1 + e^{(\mathcal{E} - \mathcal{E}g)/KT}$$

$$1 + e^{(\mathcal{E} - \mathcal{E}g)/KT}$$

$$1 + e^{\Delta \mathcal{E}/KT}$$

$$1$$

from i 4 iii, we can say f(Eg+AE) = 1-f(Eg-AE) (4) 150K 1.0 0.5 300K

At O'K Temperature. f(E) = probability. for E>Ep f(E)=0 for E<Ef f(E)=1 17 States means 14 electrons Since 7 states are under fermillere

QS)  $n(E) = g.(E) \cdot D_c(E)$ M N for Conduction band. = winy boltzman approximation.

f(E) = (E-Ef)/kT We know that carrier distribution for conduction band is directly proportional to VE-Ec 1 for valence band it is 1 EV- E f we know that corviere distribution is X J(E). VE-EC wing Dc(E) & Dv(E) 1 derivations.

So we can say.  $N(E) = K \cdot \sqrt{E - E_c} \cdot e^{-(E - E_g)/KT}$ (k is a constant) so for its maxima, we take derivative wort & d | k.VE-Ec. e (E-Ef)/KT)

dE | (E-Ef)/KT)  $= K \left[ \frac{1}{2\sqrt{\epsilon - \epsilon c}} - \frac{(\epsilon - \epsilon \rho)/kT}{\sqrt{\epsilon - \epsilon c \cdot e}} \right]$   $= K \left[ \frac{1}{2\sqrt{\epsilon - \epsilon c}} - \frac{(\epsilon - \epsilon \rho)/kT}{\sqrt{\epsilon - \epsilon c \cdot e}} \right]$ using product rule. + equating to  $\frac{1}{2\sqrt{E-Ec}} = \frac{\sqrt{E-Ec}}{\sqrt{KT}}$ E = Ec + KT

M Similarly for Valence band. P(E) = D(E). (1- f(E))  $\frac{d}{dE} = x \left[ \frac{-1}{e^{(\xi - \xi)/kT}} \frac{-(\xi - \xi)/kT}{\xi - \xi} \frac{-(\xi - \xi)/kT}{\xi - \xi} \right] \frac{-(\xi - \xi)/kT}{\xi - \xi}$  (x is a constant)equating . d(P(E)) = 0 we get,  $\frac{-1}{2\sqrt{\varepsilon v - \varepsilon}} + \sqrt{\varepsilon v - \varepsilon} = 0$ we get [E = EV- KT]

for a given 
$$T$$
 $N \cdot P = (n_1)^2$ 

at 300 k

 $Ni(S_1) = 1.5 \times 10^{10} \text{ cm}^3$ 
 $P = (1.5 \times 10^{10})^2$ 
 $10^5$ 
 $P = 2.25 \times 10^{15} \text{ cm}^3$ 

b) As  $Nd - Na > ni$ 

here on  $Nd - Na$ 

electron concentration is approximately  $Nd - Na$ .

So,  $P = (n_1)^2 = N_1^2$ 
 $Nd - Na$ 

eg-&; = 0.26eV Sample is n-type assuming & is located at midpoint

n = n; e (Eg-&i)/kT  $= 2.2 \times 10^{14} \, \text{cm}^3$ 

 $P = \frac{n^2}{n} = \frac{4.55 \times 10^5 \text{ cm}^3}{1}$ 

Calculate 'new' ni jor T=800K 7 ni = \( \Nc (800K) N\_V (800K).e^{-(Eg/2KT)} = 2.56 × 10 cm3 Hore at such high temperature, the semisconductor becomes intrinsic. 80 = ni mn mp x 2.56 x 10 6 cm3 As intrinsic semiconductor. EF is at middle of bandgap  $so = \begin{cases} E_F = E_C - E_V \\ 2 \end{cases}$ س 6

(27) we know ni=n=2p  $P = \frac{10^{13}}{\sqrt{2}} \approx 7.07 \times 10^{12} \text{ cm}^{-3}$ n= 2xp # 1.4 x 10 cm-3 So at Na = 0 Na = n = 1.4 x 10 13 cm3 QB) a) Boron acts as an acceptor (gives holes) So it's a Ptype Semiconductor. XEDT

b) At T=300K 5 Intrinsic so Dopant Concentration > E B ( avoisey density  $P = N_a = 4 \times 10^6 \text{ cm}^3$   $n = (n_i = 10^{10} \text{ cm}^3)^2/P$   $= 2.5 \times 10^3 \text{ cm}^3$ . 83 87 n; = \Nc(600) Nv(600) · e (Eg)/KT 87 At T=600K = 1.16 x 10 cm3 Simplifying of substituting constants So now we have some intrinsic carrier concentration. = 1  $P = Na + ni = (4 \times 10^{16} + 1.16 \times 16^{15})$ =  $[4.12 \times 10^{16} \text{ cm}^3]$ رسع ال n= (ni) = [3.27 x 10 3 cm3] 4 

b) At T = 300 K Intrinsic so Dopant Concentration > T ( aronieq density  $P = N_a = 4 \times 10^6 \text{ cm}^3$   $n = (n_i = 10^1 \text{ cm}^3)^2/p$   $= 2.5 \times 10^3 \text{ cm}^3$ At T=600K n== \Nc(600) Nv(600) · e (Eg)/KT  $= 1.16 \times 10^{15} \text{ cm}^3$ Simplifying of substituting constants So now me have some intrinsic carrier concentration. 1 2  $p = N_0 + n_i = (4 \times 10^{16} + 1.16 \times 16^{15})$ =  $[4.12 \times 10^{16} \text{ cm}^3]$ سع n= (ni)2 = [3.27 x103 cm3] 2 4 

c) At high temperatures, the number of intrinsic carriers increases due to more free electrons from solicon - silicon covalent bonds.

d) We have,  $\mathcal{E}_{g} - \mathcal{E}_{v} = \kappa T \ln \left[ N_{v}(T) / P(T) \right]$ 

= 0.34 eV at T= 600 k

F

so, Eg is located at 0.34 from Ev.

Ef-Er = 0.34 eV

(P9) we want, a) hi = 10 Nd. ni = \( Nc(T) Nv(T) - (Eg)/2kT = 1.71×10 (T 3/2-(Eg)/2KT K Calculating No f Nor f iteratively some for ni = 10 Nd for n-type get T# 778 K for p-type, similarly we get T = 633 K b) for n-type (arcsenic) n = Nd = 10'6 = 2.1×10 cm3 so = p=ni<sup>2</sup> for p-type (boron)  $p \approx Na \quad n = \frac{ni^2}{Na} = 2.1 \times 10^5 \text{ cm}^3$ 

$$E_{g} - E_{Y} = kT \ln \frac{NV}{P} = \frac{1.04 \times 10^{19}}{2.1 \times 10^{4}}$$

$$= \left[0.88 \text{ eV}\right]$$

$$E_{g} - E_{Y} = kT \ln \frac{NV}{P} = \frac{1.04 \times 10^{11}}{10^{15}}$$

$$= \left[0.24 \text{ eV}\right]$$

$$\frac{1}{2} = \frac{10^{15} \text{ cm}^{-3}}{10^{15}}$$

a) The mean free time bet " collisions Un = 9 tmm Mh Tom = Mama = 2.85 × 10 13 sec b) drift nelocity Vd = Un E = 5 x 104 cm/ d = Vd Tmn = 0.14 nm. approx thermal velocity  $Vm = \sqrt{\frac{3kT}{m}} = 2.29 \times 10^{7} \text{ cm/sec}$ Va = V4h = 2.29 × 106 cm/sec so't' for distance of 1 um t = 10 cm = 4.37 × 10" sec 2.29 × 10 cm/sec

mean free dime Un = 2 Tmn -> Tm = Unmn Tom = 2.1 × 10 300 t = 207.7 collision => 207 callisions Vd = - Un E = Vd = 2.29×106 cm/stc E = 1635.71 Vcm' V = E x width = 1635.71 Vcm x 10 cm JV=046V]

12) a) we know, 5 for N-type deped with 1×1016 cm² of phosphorous is b) P = 1 = 1 q. Nha Up 2 × 9 × 10 16 cm<sup>3</sup> × 250 cm<sup>2</sup>/Vsec = 0.28 Dun Here, acceptor density (Boron) is greater
though donor density (phosphorous) so

it's a 'p-type' semiconductor. E- Ey = KTln (Nc) =  $0.026V \ln \left( \frac{2.8 \times 10^{19} \text{ cm}^3}{10^{16} \text{ cm}^3} \right) = 0.21 \text{ eV}$ And for (a).

 $E_{g}-E_{v}=k + ln\left(\frac{N_{v}}{N_{net}}\right)$ = 0.026 V ln [ 1.01 × 10'9 cm<sup>3</sup> ] 0.12eV (a) 10.21eV BUE D ·Ev (b)

Q13) a) N-type, holes are minority  $p = n_i^2 = (10^{'0} \text{ cm}^3)^2 / 10^{17} \text{ cm}^3$ = [10<sup>2</sup> cm<sup>3</sup>] Sample 2 - p-type, electrons are minority carriers  $n = n_i^2 = (10^{10} \text{ cm}^{-3})^2$   $N_a = 15$   $10^{10} \text{ cm}^{-3}$   $10^{15} \text{ cm}^{-3}$ Sample 3, holes are minority carriers  $P = h_i^2 = (10^{10} \text{cm}^3)^2 / 9.9 \times 10^{17} \text{cm}^3)$ Nnet = /102 cm-3/

7 b) Sample 1 = NA = 1017 cm-3 len (Nd = 10 cm3) = 750 cm2/4 sec 0 = 9 Nalla = [12 II cm] Sample 2 = Na = 10 5 cm3 Up (Na = 10'5 cm 3) = 480 cm / Vsec 6 = q Nalla = [12 12 cm] Sample 3 = NT = Nd + Na = 1.01x10/tm3 Un (NT = 1.01 × 1017 cm3) = 750 cm2/ Vsec Net =  $N_q - N_a = 0.99 \times 10^{17} \text{ cm}^3$   $6 = 9 N_{\text{net}} U_n = 11.88 \cdot 9 \text{ cm}^3$ c) for sample 1 E- Eg = KTln (Nc) = 0.026 Vln (Nc) 2.8×10'9 cm3 = 0.15 eV Sample 2. Eg-Ev- KThe (NV) = 0.026 V ln (1.04 × 10 cm3)

[0.24eV] Sample 3 
Ec-Eg=kTln (Nc)

Nnet Net = Nd - Na i Substituting Nnet =  $0.026 \text{ V ln} \left( \frac{2.8 \times 10^{19} \text{ cm}^3}{9.9 \times 10^{16} \text{ cm}^3} \right)$ = 0.15eV Q(4)a)  $J = qnv = q D \left(\frac{dn}{dx}\right)$  $v = D(1)(\frac{dn}{dn}) = \left| -\frac{D}{\lambda} \right|$ b) J= quant = quo f v= la E So, & = - D = - (KT) c) &= -1000 V/cm 00 0000 &= -0.026/X 80 A = 0.25 Um

(215)  $\mathcal{E} = -\frac{dV}{dx} = \frac{1}{9} \frac{d\mathcal{E}_{v}}{dx}$ therefore the equation is E = 1