Mathematical modelling in biology exercises

Giacomo Fantoni

telegram: @GiacomoFantoni

 $Github:\ https://github.com/giacThePhantom/mathematical-modelling-in-biology$

June 4, 2023

Contents

1	\mathbf{Pro}	Problems of week 1		
	1.1	Proble	em 1	
	1.2	Proble	em 2	
	1.3	Proble	em 3	
		1.3.1	Solve the equation for $\mathbf{U}(\mathbf{t})$	
			How can the quantities of ${\bf a}$ and ${\bf b}$ can be interpreted? From the data provided	
			can we infer the half-live of Uranium-234 and Thorium-230?	
		1.3.3	Calculate $\mathbf{T}(\mathbf{t})$, solution of the second differential equation $\ldots \ldots \ldots$	
		1.3.4	Compute $\lim_{\mathbf{t}\to\infty} \frac{\mathbf{T}(\mathbf{t})}{\mathbf{U}(\mathbf{t})}$	
		1.3.5	Explain why it is possible to estimate the rock age from the knowledge of $\frac{\mathbf{T}(\mathbf{t})}{\mathbf{U}(\mathbf{t})}$ at current time	

Chapter 1

Problems of week 1

- 1.1 Problem 1
- 1.2 Problem 2
- 1.3 Problem 3

The thorium-uranium method for dating rocks is based on the fact that Uranium234 decays into Thorium-230 which in turn decays into other elements. Set t=0 the rock formation time and denoting U(t) [T(t)] the amount of Uranium-234 [Thorium-230] in the rock at time t (measured in years), the following differential equation system is written:

$$\begin{cases} \frac{dU(t)}{dt} = -aU(t) \\ \frac{dT(t)}{dt} = aU(t) - bT(t) \\ U(0) = U_0 \\ T(0) = 0 \end{cases}$$

Where:

•
$$a \approx 5.9 \cdot 10^{-6} \frac{1}{years}$$

- $b \approx 1.9 \cdot 10^{-5} \frac{1}{years}$
- U_0 is the initial amount of Uranium-234

Note that, based on geological principles, it is believed that there was no thorium at the time of rock formation.

1.3.1 Solve the equation for U(t)

$$\frac{dU(t)}{dt} = -aU(t)$$

$$\frac{dU(t)}{dU(t)} = -adt$$

$$\int \frac{q}{dU(t)} dt \int -adt$$

$$\log(U(t)) = -at + C$$

$$U(t) = e^{-at+C}$$

$$U(t) = C_1 e^{-at}$$

Considering $U(0) = U_0$:

$$C_1 e^{-a0} = U_0$$
$$C_1 = U_0$$

So, in conclusion:

$$U(t) = U_0 e^{-at}$$

1.3.2 How can the quantities of a and b can be interpreted? From the data provided can we infer the half-live of Uranium-234 and Thorium-230?

- a is the rate at which Uranium-234 decays into Thorium-230.
- b is the rate at which Thorium-230 decays into other elements.

To compute the half-life of Uranium-234:

$$U(t_1) = \frac{1}{2}U(t_2)$$

$$V_0e^{-at_1} = \frac{1}{2}V_0e^{-at_2}$$

$$2 = \frac{e^{-at_2}}{e^{-at_1}}$$

$$e^{-at_2+at_1} = 2$$

$$e^{a(t_1-t_2)} = 2$$

$$a(t_1-t_2) = \log(2)$$

$$t_1-t_2 = \frac{\log(2)}{a}$$

$$t_1-t_2 = \frac{\log(2)}{5.9 \cdot 10^{-6}} = 117482.6 \text{ years}$$

1.3.3 Calculate T(t), solution of the second differential equation

$$\frac{dT(t)}{dt} = aU_0e^{-at} - bT(t)$$

Let:

$$w(t) = \frac{T(t)}{e^{-bt}}$$
$$w(t) = T(t)e^{bt}$$

Taking its derivative:

$$\frac{dw(t)}{dt} = bT(t)e^{bt} + [aU_0e^{-at} - bT(t)]e^{bt}$$

$$\frac{dw(t)}{dt} = bT(t)e^{bt} + aU_0e^{-at+bt} - bT(t)e^{bt}$$

$$\frac{dw(t)}{dt} = bT(t)e^{bt} + aU_0e^{-at+bt} - bT(t)e^{bt}$$

$$\frac{dw(t)}{dt} = aU_0e^{-at+bt}$$

Now taking the integral:

$$\int \frac{dw(t)}{dt}dt = \int aU_0e^{-at+bt}dt$$

$$w(t) = \int aU_0 \int e^{(b-a)t}dt$$

$$w(t) = \frac{aU_0}{b-a}e^{b-at} + C_2$$

Now, considering that:

$$T(t) = w(t)e^{-bt}$$

$$T(t) = \frac{aU_0}{b-a}e^{(b-a)t}e^{-bt} + C_2e^{-bt}$$

$$T(t) = \frac{aU_0}{b-a}e^{bt}e^{-at}e^{-bt} + C_2e^{-bt}$$

$$= \frac{aU_0}{b-a}e^{-at} + C_2e^{-bt}$$

Considering T(0) = 0:

$$\frac{aU_0}{b-a}e^{-at} + C_2e^{-bt} = 0$$
$$\frac{aU_0}{b-a} + C_2 = 0$$
$$C_2 = \frac{-aU_0}{b-a}$$

So, in the end:

$$T(t) = \frac{aU_0}{b - a}e^{-at} - \frac{aU_0}{b - a}e^{-bt}$$

$\textbf{1.3.4} \quad Compute \ \lim_{t \to \infty} \frac{\mathbf{T(t)}}{\mathbf{U(t)}}$

Computing the ratio:

$$\begin{split} \frac{T(t)}{U(t)} &= \frac{\frac{aU_0}{b-a}e^{-at} - \frac{aU_0}{b-a}e^{-bt}}{U_0e^{-at}} \\ &= \frac{aU_0}{b-a}e^{-at}\frac{1}{U_0e^{at}} - \frac{aU_0}{b-a}e^{-bt}\frac{1}{U_0e^{at}} \\ &= \frac{aU_0}{b-a}e^{-at}\frac{1}{U_0e^{at}} - \frac{aU_0}{b-a}e^{-bt}\frac{1}{U_0e^{-at}} \\ &= \frac{a}{b-a} - \frac{a}{b-a}e^{(a-b)t} \end{split}$$

$$\lim_{t \to \infty} \frac{T(t)}{U(t)} = \lim_{t \to \infty} \frac{a}{b-a} - \frac{a}{b-a} e^{(a-b)t} = \frac{a}{b-a} - \frac{a}{b-a} e^{(a-b)\infty}$$

$$= \frac{a}{b-a} - \frac{a}{b-a} 0$$

$$= \frac{a}{b-a}$$

1.3.5 Explain why it is possible to estimate the rock age from the knowledge of $\frac{T(t)}{U(t)}$ at current time

Hint: write down the function $\frac{T(t)}{U(t)}$ and evaluate its behaviour.

Consider:

$$\frac{T(t)}{U(t)} = \frac{a}{b-a} - \frac{a}{b-a}e^{(a-b)}$$

Now, this function does not depend on the initial concentration of Uranium and can be inverted for t:

$$\frac{T(t)}{U(t)} = \frac{a}{b-a} - \frac{a}{b-a}e^{(a-b)t}$$

$$\frac{a}{b-a}e^{(a-b)t} = \frac{a}{b-a} - \frac{T(t)}{U(t)}$$

$$e^{(a-b)t} = \frac{a}{b-a} - \frac{b-a}{a} \frac{T(t)}{U(t)}$$

$$e^{(a-b)t} = 1 - \frac{b-a}{a} \frac{T(t)}{U(t)}$$

$$(a-b)t = \log\left(1 - \frac{b-a}{a} \frac{T(t)}{U(t)}\right)$$

$$t\left(\frac{T(t)}{U(t)}\right) = \frac{\log\left(1 - \frac{b-a}{a} \frac{T(t)}{U(t)}\right)}{a-b}$$

So now from the ratio $\frac{T(t)}{U(t)}$ the rock age can be estimated.