## Mass and Energy balances for a steady-flow process

$$\sum_{in} \dot{m} = \sum_{out} \dot{m} \qquad (kg/s)$$

$$\dot{m}_1 = \dot{m}_2$$

$$\rho_1 V_1 A_1 = \rho_2 V_2 A_2$$

Mass balance

$$\underline{\dot{E}_{\text{in}} - \dot{E}_{\text{out}}} = \underbrace{dE_{\text{system}}/dt}^{0 \text{ (steady)}} = 0$$
Rate of net energy transfer by heat, work, and mass potential, etc., energies



 $\dot{m}_{2} = \dot{m}_{1}$ 

Hot water

out

A water heater in steady operation.

(Hot-water tank)

## (kW) Rate of net energy transfer out

Rate of net energy transfer in by heat, work, and mass

by heat, work, and mass

$$\dot{Q}_{\rm in} + \dot{W}_{\rm in} + \sum_{\rm in} \dot{m}\theta = \dot{Q}_{\rm out} + \dot{W}_{\rm out} + \sum_{\rm out} \dot{m}\theta$$

$$\dot{Q}_{\rm in} + \dot{W}_{\rm in} + \sum_{\rm in} \dot{m} \left( h + \frac{V^2}{2} + gz \right) = \dot{Q}_{\rm out} + \dot{W}_{\rm out} + \sum_{\rm out} \dot{m} \left( h + \frac{V^2}{2} + gz \right)$$
for each inlet

Energy balance Electric heating

element

 $W_{\rm in}$ 

 $m_1$ 

Cold water in

## Energy balance relations with sign conventions (i.e., heat input and work output are positive)

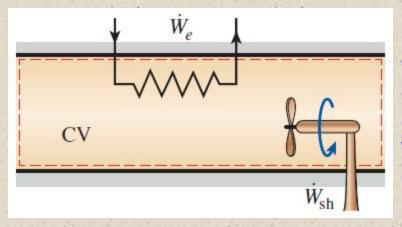
$$\dot{Q} - \dot{W} = \sum_{\text{out}} \dot{m} \left( h + \frac{V^2}{2} + gz \right) - \sum_{\text{in}} \dot{m} \left( h + \frac{V^2}{2} + gz \right)$$
for each exit
for each inlet

$$\dot{Q} - \dot{W} = \dot{m} \left[ h_2 - h_1 + \frac{V_2^2 - V_1^2}{2} + g(z_2 - z_1) \right]$$

$$q - w = h_2 - h_1 + \frac{V_2^2 - V_1^2}{2} + g(z_2 - z_1)$$

$$q - w = h_2 - h_1 \quad q = \dot{Q}/\dot{m} \quad w = \dot{W}/\dot{m}$$

when kinetic and potential energy changes are negligible



Under steady operation, shaft work and electrical work are the only forms of work a simple compressible system may involve.

$$\frac{J}{kg} \equiv \frac{N \cdot m}{kg} \equiv \left(kg \frac{m}{s^2}\right) \frac{m}{kg} \equiv \frac{m^2}{s^2}$$

$$\left(\text{Also, } \frac{\text{Btu}}{\text{1bm}} = 25,037 \frac{\text{ft}^2}{\text{s}^2}\right)$$

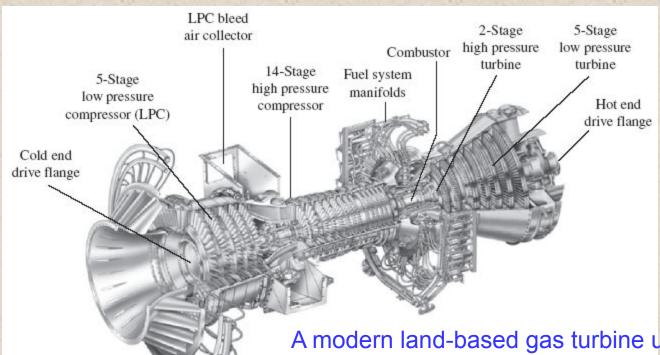
The units m<sup>2</sup>/s<sup>2</sup> and J/kg are equivalent.

$V_1$	$V_2$	Δke
m/s	m/s	kJ/kg
0	45	1
50	67	1
100	110	1
200	205	1
500	502	1

At very high velocities, even small changes in velocities can cause significant changes in the kinetic energy of the fluid.

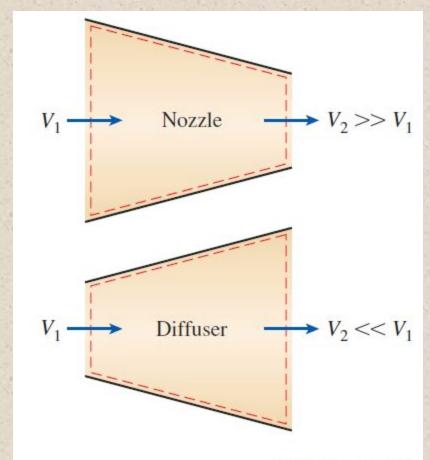
## SOME STEADY-FLOW ENGINEERING DEVICES

Many engineering devices operate essentially under the same conditions for long periods of time. The components of a steam power plant (turbines, compressors, heat exchangers, and pumps), for example, operate nonstop for months before the system is shut down for maintenance. Therefore, these devices can be conveniently analyzed as steady-flow devices.



A modern land-based gas turbine used for electric power production. This is a General Electric LM5000 turbine. It has a length of 6.2 m, it weighs 12.5 tons, and produces 55.2 MW at 3600 rpm with steam injection.

## **Nozzles and Diffusers**



## FIGURE 5-26

Nozzles and diffusers are shaped so that they cause large changes in fluid velocities and thus kinetic energies. Nozzles and diffusers are commonly utilized in jet engines, rockets, spacecraft, and even garden hoses.

A **nozzle** is a device that *increases the velocity of a fluid* at the expense of pressure.

A **diffuser** is a device that *increases* the pressure of a fluid by slowing it down.

The cross-sectional area of a nozzle decreases in the flow direction for subsonic flows and increases for supersonic flows. The reverse is true for diffusers.

Energy balance for a nozzle or diffuser:

$$\dot{E}_{\text{in}} = \dot{E}_{\text{out}}$$

$$\dot{m}\left(h_1 + \frac{V_1^2}{2}\right) = \dot{m}\left(h_2 + \frac{V_2^2}{2}\right)$$

(since  $\dot{Q} \cong 0$ ,  $\dot{W} = 0$ , and  $\Delta pe \cong 0$ )

# Deceleration of Air in a Diffuser

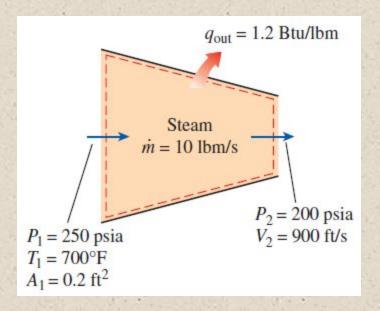


$$\frac{\dot{E}_{\rm in} - \dot{E}_{\rm out}}{\dot{E}_{\rm in} - \dot{E}_{\rm out}} = \underbrace{dE_{\rm system}/dt} = 0$$
Rate of net energy transfer by heat, work, and mass
$$\dot{E}_{\rm in} = \dot{E}_{\rm out}$$

$$\dot{m} \left(h_1 + \frac{V_1^2}{2}\right) = \dot{m} \left(h_2 + \frac{V_2^2}{2}\right) \quad \text{(since } \dot{Q} \cong 0, \, \dot{W} = 0, \, \text{and } \Delta \text{pe} \cong 0)$$

$$h_2 = h_1 - \frac{V_2^2 - V_1^2}{2}$$

# Acceleration of Steam in a Nozzle



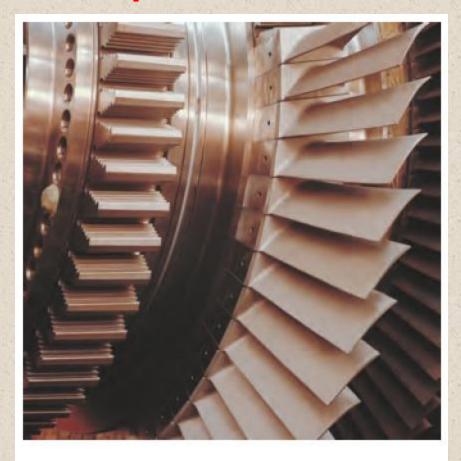
$$\frac{\dot{E}_{\rm in} - \dot{E}_{\rm out}}{\dot{E}_{\rm in} - \dot{E}_{\rm out}} = \underbrace{dE_{\rm system}/dt}_{\rm Rate \ of \ net \ energy \ transfer \ by \ heat, \ work, \ and \ mass}_{\rm Rate \ of \ change \ in \ internal, \ kinetic, \ potential, \ etc., \ energies}$$

$$\dot{E}_{\rm in} = \dot{E}_{\rm out}$$

$$\dot{m}\left(h_1 + \frac{V_1^2}{2}\right) = \dot{Q}_{\rm out} + \dot{m}\left(h_2 + \frac{V_2^2}{2}\right) \quad \text{(since } \dot{W} = 0, \ \text{and } \Delta \text{pe} \cong 0\text{)}$$

$$h_2 = h_1 - q_{\rm out} - \frac{V_2^2 - V_1^2}{2}$$

## **Turbines and Compressors**



### FIGURE 5-29

Turbine blades attached to the turbine shaft.

Turbine drives the electric generator In steam, gas, or hydroelectric power plants.

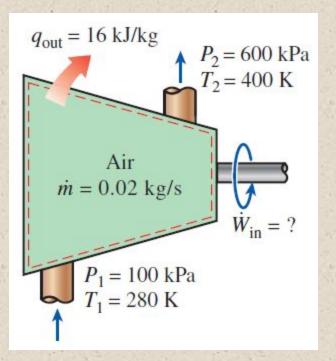
As the fluid passes through the turbine, work is done against the blades, which are attached to the shaft. As a result, the shaft rotates, and the turbine produces work.

Compressors, as well as pumps and fans, are devices used to increase the pressure of a fluid. Work is supplied to these devices from an external source through a rotating shaft.

A *fan* increases the pressure of a gas slightly and is mainly used to mobilize a gas.

A *compressor* is capable of compressing the gas to very high pressures.

**Pumps** work very much like compressors except that they handle liquids instead of gases.



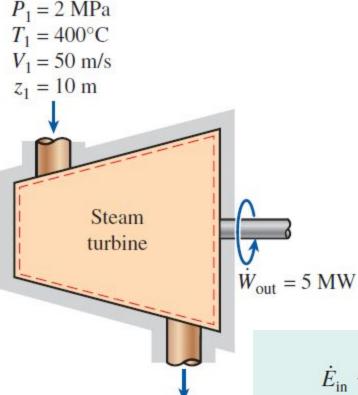
## **Compressing Air**by a Compressor

$$\dot{E}_{\rm in} - \dot{E}_{\rm out} = \underbrace{dE_{\rm system}/dt}^{0 \, (\rm steady)} = 0$$
Rate of net energy transfer by heat, work, and mass
$$E_{\rm in} - \dot{E}_{\rm out} = \underbrace{dE_{\rm system}/dt}^{0 \, (\rm steady)} = 0$$
Rate of change in internal, kinetic, potential, etc., energies

 $\dot{W}_{\rm in} = \dot{m}q_{\rm out} + \dot{m}(h_2 - h_1)$ 

 $\dot{W}_{\rm in} + \dot{m}h_1 = \dot{Q}_{\rm out} + \dot{m}h_2$  (since  $\Delta ke = \Delta pe \approx 0$ )

 $\dot{E}_{\rm in} = \dot{E}_{\rm out}$ 



 $P_2 = 15 \text{ kPa}$ 

 $V_2 = 180 \text{ m/s}$ 

 $x_2 = 0.90$ 

 $z_2 = 6 \text{ m}$ 

# Power Generation by a Steam Turbine

$$\dot{E}_{\rm in} - \dot{E}_{\rm out}$$

Rate of net energy transfer by heat, work, and mass

$$dE_{\text{system}}/dt = 0$$

Rate of change in internal, kinetic, potential, etc., energies

$$\dot{E}_{\rm in} = \dot{E}_{\rm out}$$

$$\dot{m}\left(h_1 + \frac{V_1^2}{2} + gz_1\right) = \dot{W}_{\text{out}} + \dot{m}\left(h_2 + \frac{V_2^2}{2} + gz_2\right) \quad \text{(since } \dot{Q} = 0\text{)}$$

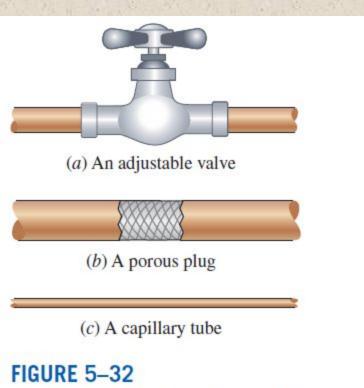
$$w_{\text{out}} = -\left[ (h_2 - h_1) + \frac{V_2^2 - V_1^2}{2} + g(z_2 - z_1) \right] = -(\Delta h + \Delta \text{ke} + \Delta \text{pe})$$

## Throttling valves

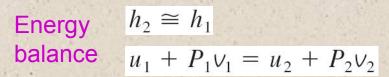
Throttling valves are any kind of flow-restricting devices that cause a significant pressure drop in the fluid.

What is the difference between a turbine and a throttling valve?

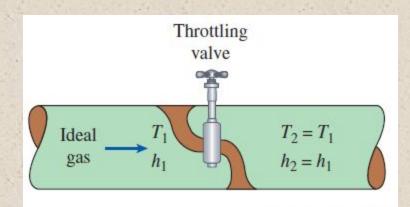
The pressure drop in the fluid is often accompanied by a *large drop in temperature*, and for that reason throttling devices are commonly used in refrigeration and air-conditioning applications.



Throttling valves are devices that cause large pressure drops in the fluid.

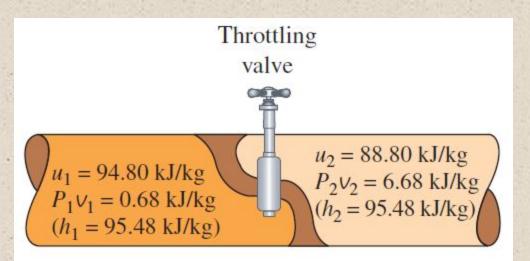


Internal energy + Flow energy = Constant



#### FIGURE 5-33

The temperature of an ideal gas does not change during a throttling (h = constant) process since h = h(T).



# Expansion of Refrigerant-134a in a Refrigerator

#### FIGURE 5-34

During a throttling process, the enthalpy (flow energy + internal energy) of a fluid remains constant. But internal and flow energies may be converted to each other.

$$h_2 \cong h_1$$
 (kJ/kg)  
 $u_1 + P_1 v_1 = u_2 + P_2 v_2$ 

 $Internal\ energy\ +\ Flow\ energy\ =\ Constant$ 

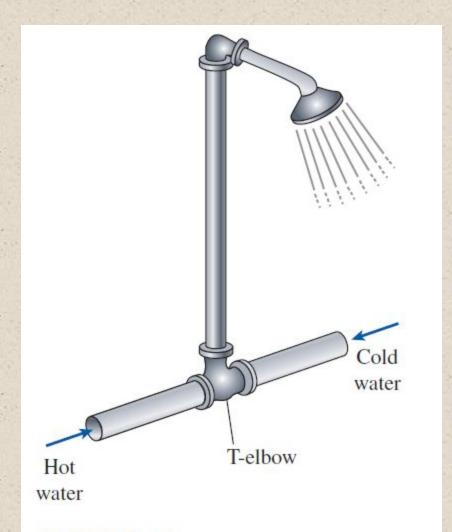
## Mixing chambers

In engineering applications, the section where the mixing process takes place is commonly referred to as a **mixing chamber**.

The mixing chamber does not have to be a distinct "chamber." An ordinary T-elbow or a Y-elbow in a shower, for example, serves as the mixing chamber for the cold- and hot-water streams.

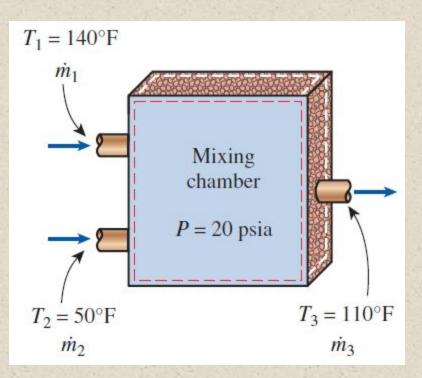
The conservation of mass principle for a mixing chamber requires that the sum of the incoming mass flow rates equal the mass flow rate of the outgoing mixture.

The conservation of energy equation is analogous to the conservation of mass equation.



#### FIGURE 5-35

The T-elbow of an ordinary shower serves as the mixing chamber for the hot- and the cold-water streams.



## Mixing of Hot and Cold Waters in a Shower

$$\dot{m}_1 + \dot{m}_2 = \dot{m}_3$$

$$\dot{E}_{\rm in} - \dot{E}_{\rm out} = \underbrace{dE_{\rm system}/dt}^{0 \text{ (steady)}}_{= 0}$$
Rate of net energy transfer by heat, work, and mass potential, etc., energies 
$$\dot{E}_{\rm in} = \dot{E}_{\rm out}$$

$$\dot{m}_1 h_1 + \dot{m}_2 h_2 = \dot{m}_3 h_3 \text{ (since } \dot{Q} \cong 0, \, \dot{W} = 0, \, \text{ke} \cong \text{pe} \cong 0)$$

$$\dot{m}_1 h_1 + \dot{m}_2 h_2 = (\dot{m}_1 + \dot{m}_2) h_3$$

# Fluid B 70°C Heat Fluid A 20°C

## **Heat exchangers**

**Heat exchangers** are devices where two moving fluid streams exchange heat without mixing.

Heat exchangers are widely used in various industries, and they come in various designs.

#### FIGURE 5-38

A heat exchanger can be as simple as two concentric pipes.

The heat transfer associated with a heat exchanger may be zero or nonzero depending on how the control volume is selected.

