The Carbon Cycle: Mineral Weathering

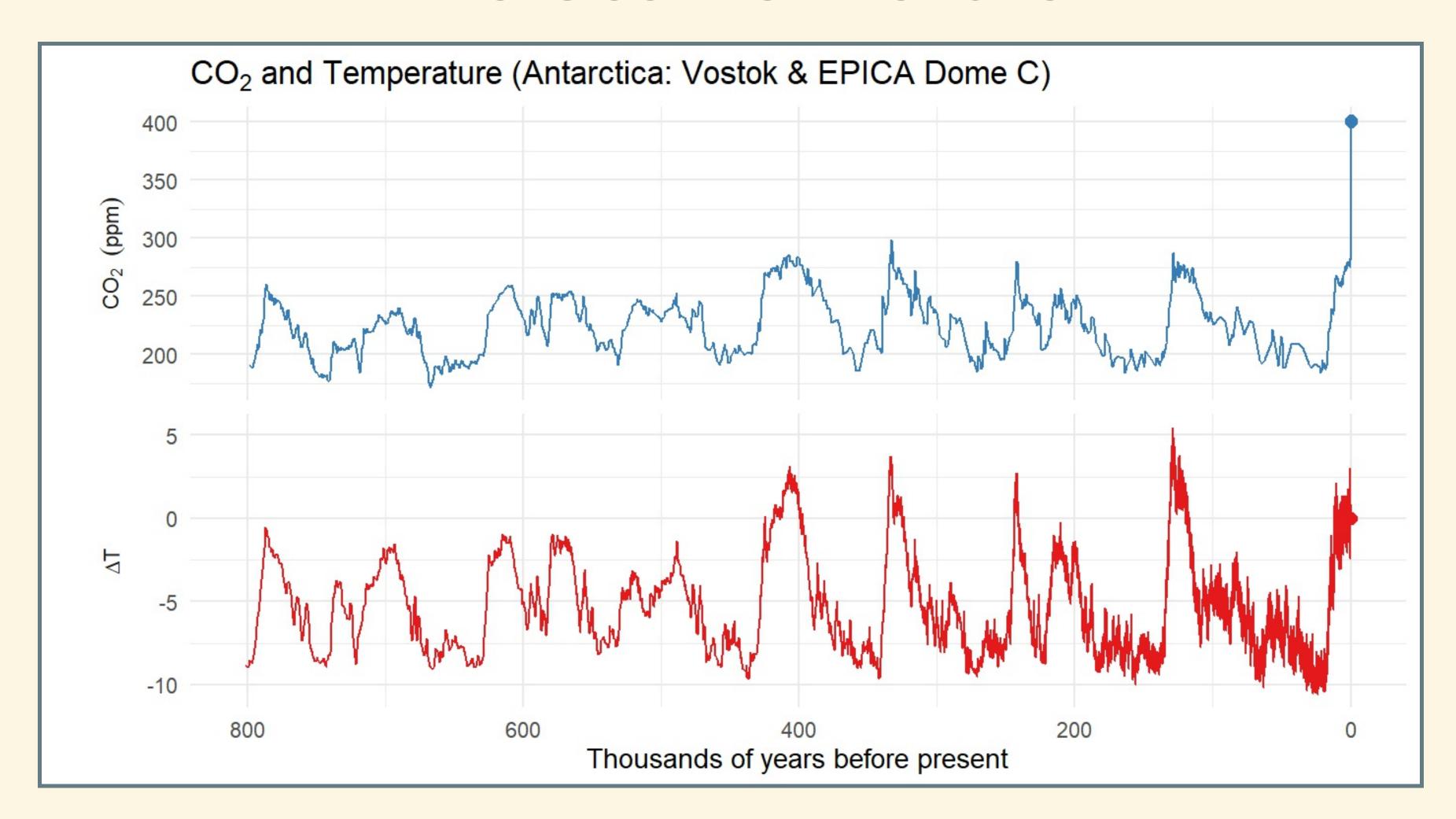
EES 3310/5310
Global Climate Change
Jonathan Gilligan

Class #11: Friday Sept. 14 2018

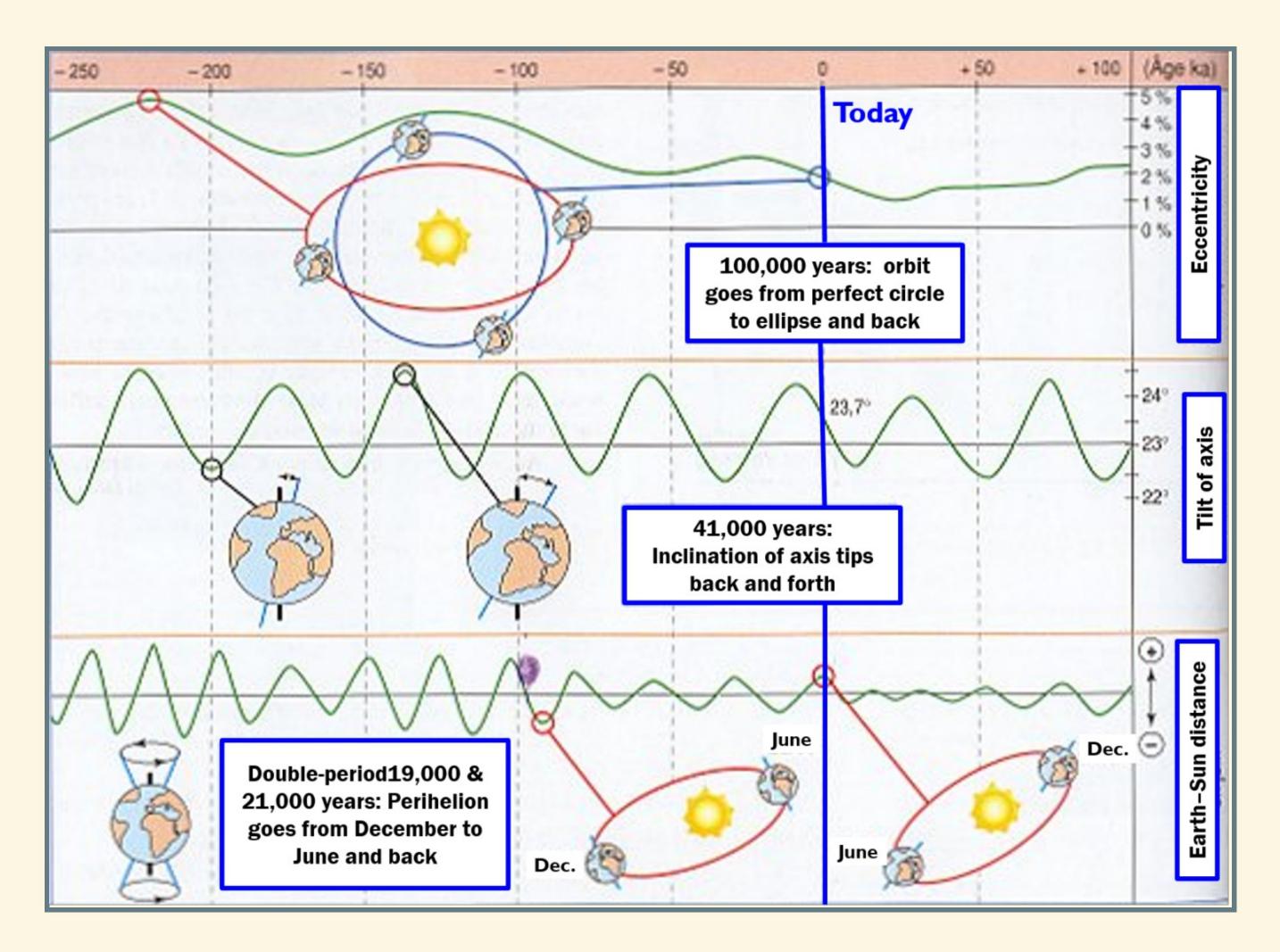


The Oceans Breathe: Ice Ages

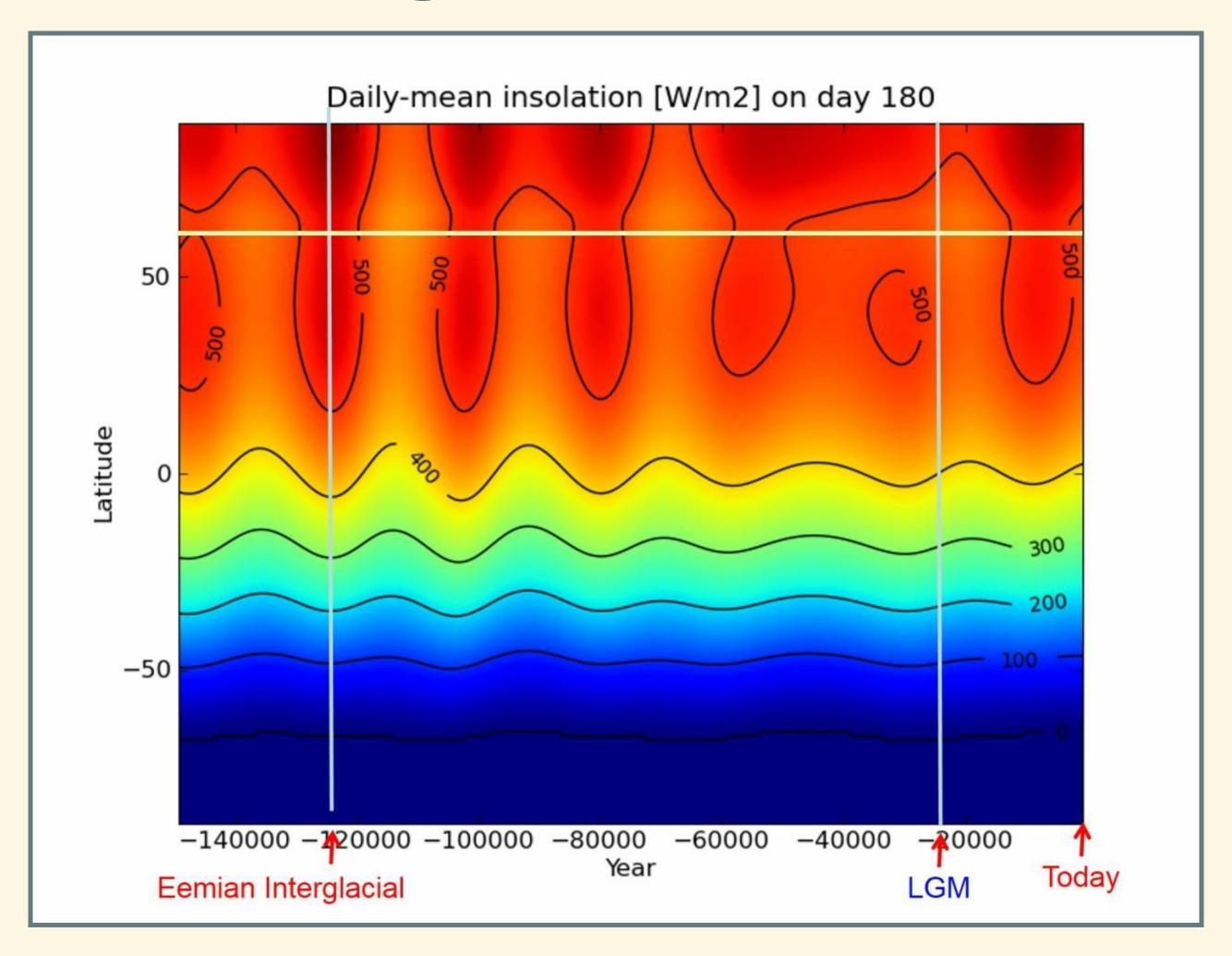
The Oceans Breathe



Causes



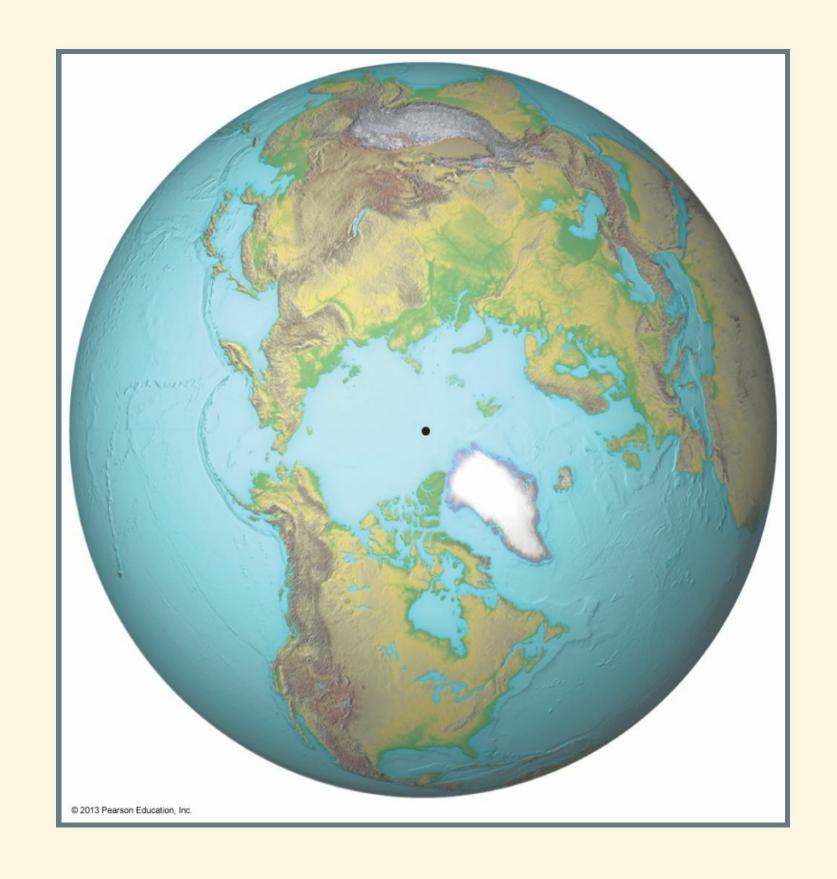
Sunlight on June 29th

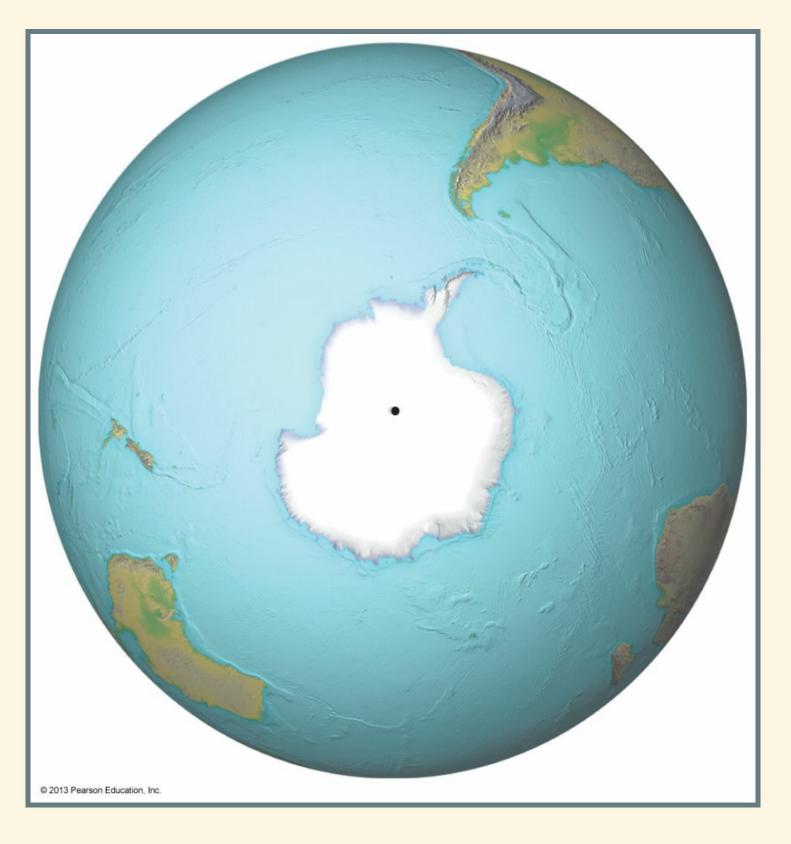


Question

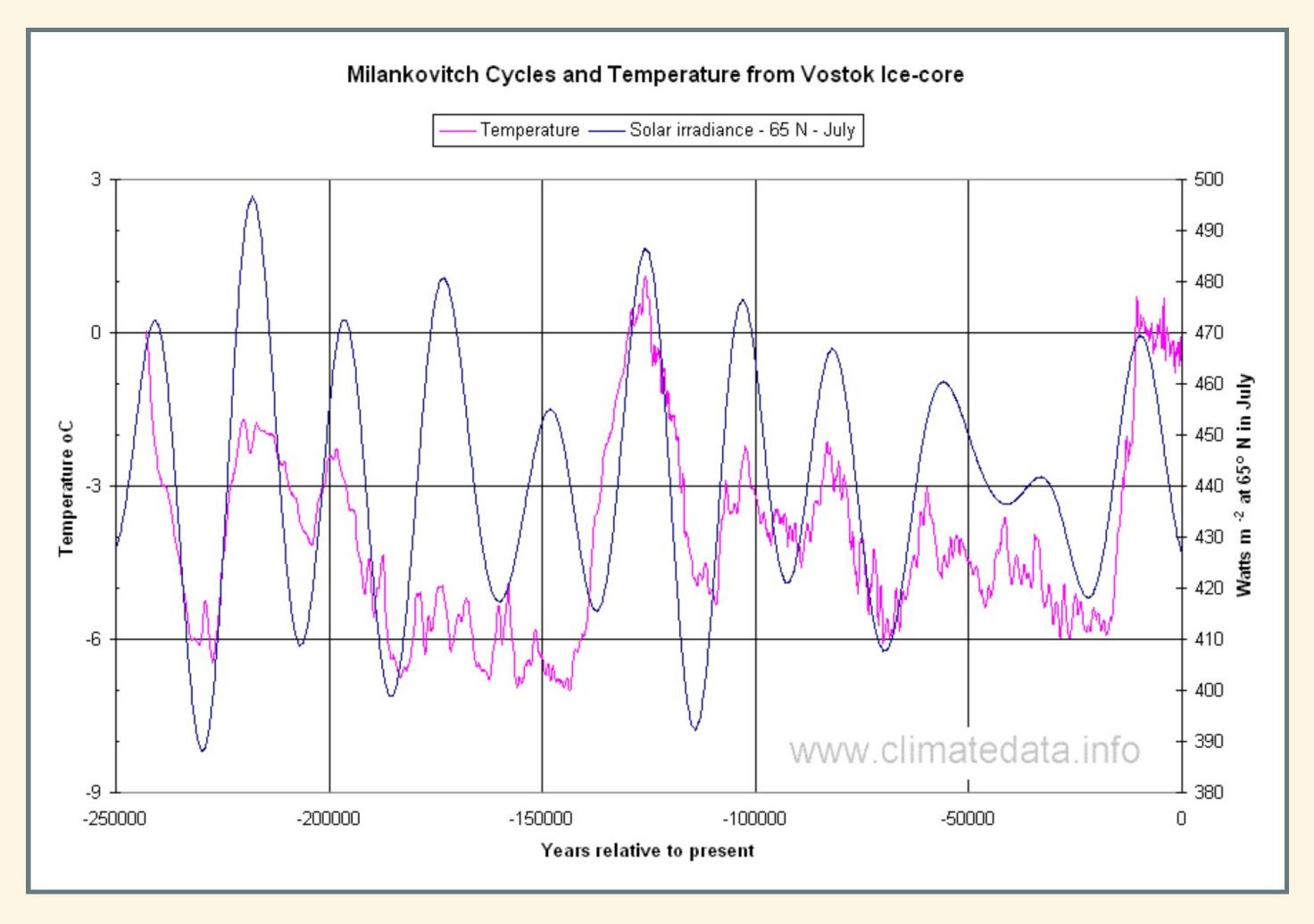
Why would the summer sunlight in the far northern hemisphere be so important?

Northern vs. Southern Hemisphere





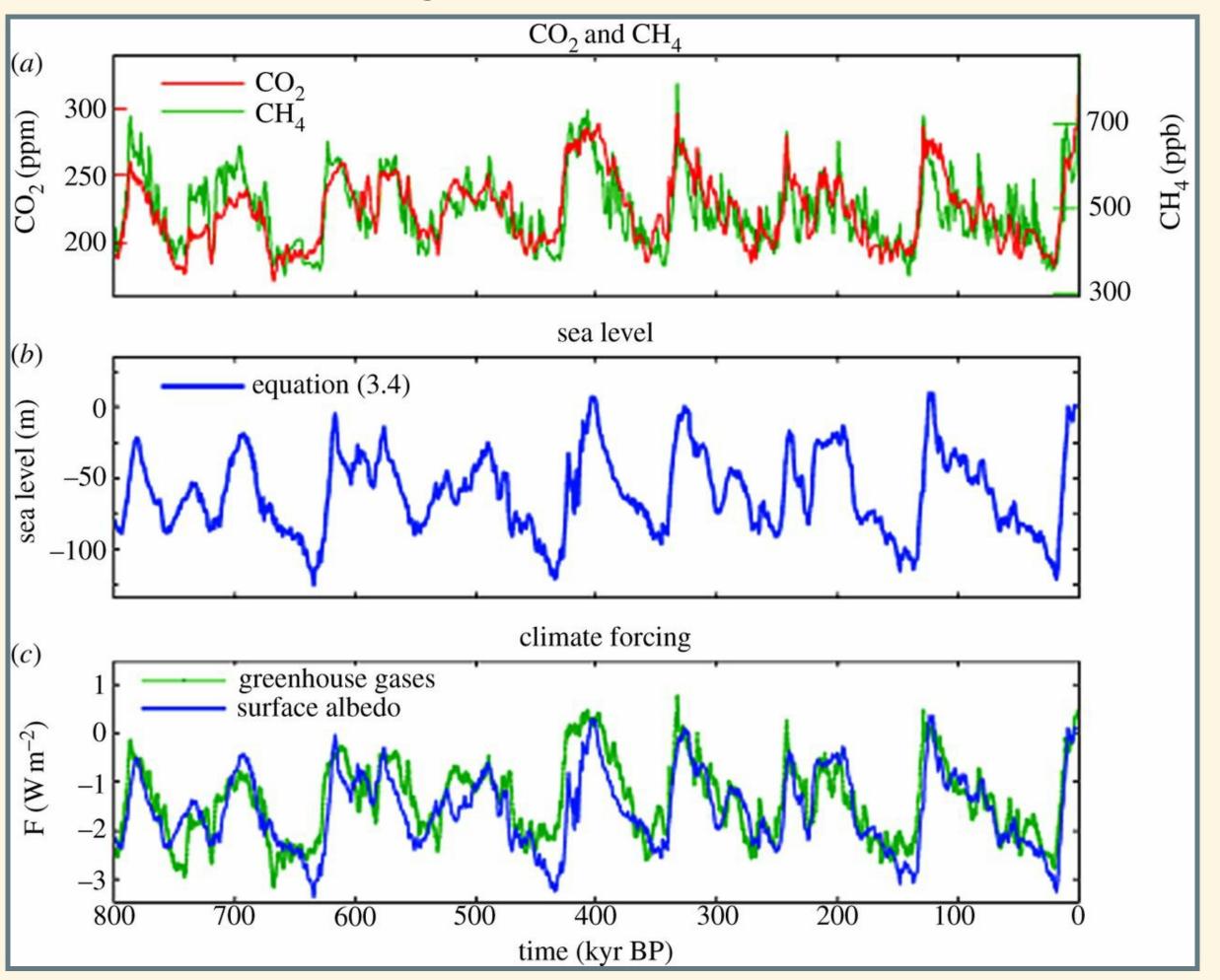
Timing of Ice Ages



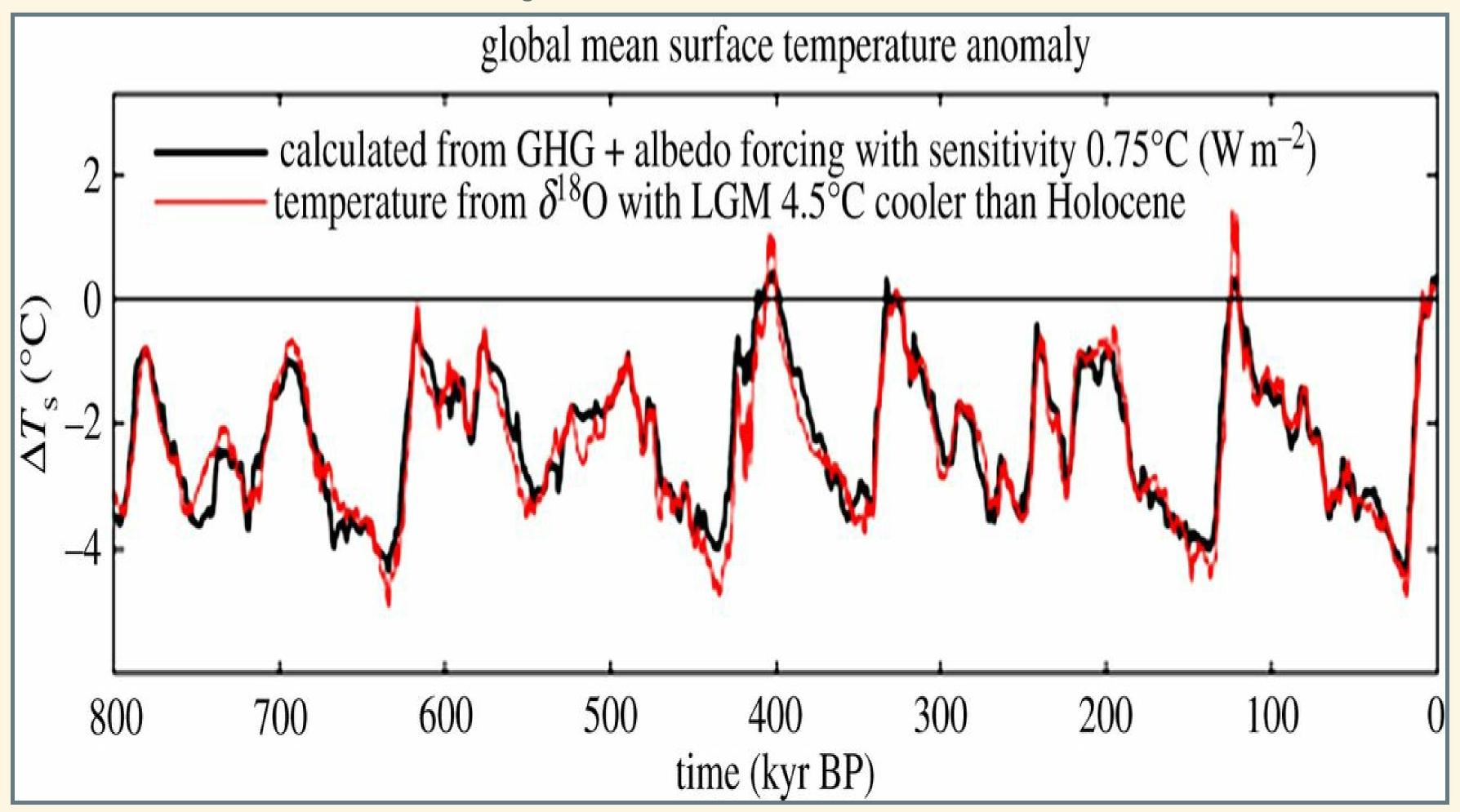
Ice Age Feedbacks

- Orbital cycles match timing of ice ages
- Changes in sunlight are too small to explain temperature changes
- There must be positive feedbacks to amplify them

Theory of Feedbacks



Theory vs. Observations



Ice-Age Feedbacks:

- Temperature starts to fall
 - Glaciers grow → higher albedo
 - co₂ drops → weaker greenhouse
 - Colder
- Temperature starts to rise
 - Glaciers retreat → higher albedo
 - co₂ rises → stronger greenhouse
 - Warmer
- Without co₂ and ice-albedo feedbacks, ice-ages couldn't happen
- Ice ages can't happen with today's co₂ levels.

The Carbon Dioxide Theory of Climatic Change

By GILBERT N. PLASS

The Johns Hopkins University, Baltimore, Md.¹ (Manuscript received August 9 1955)

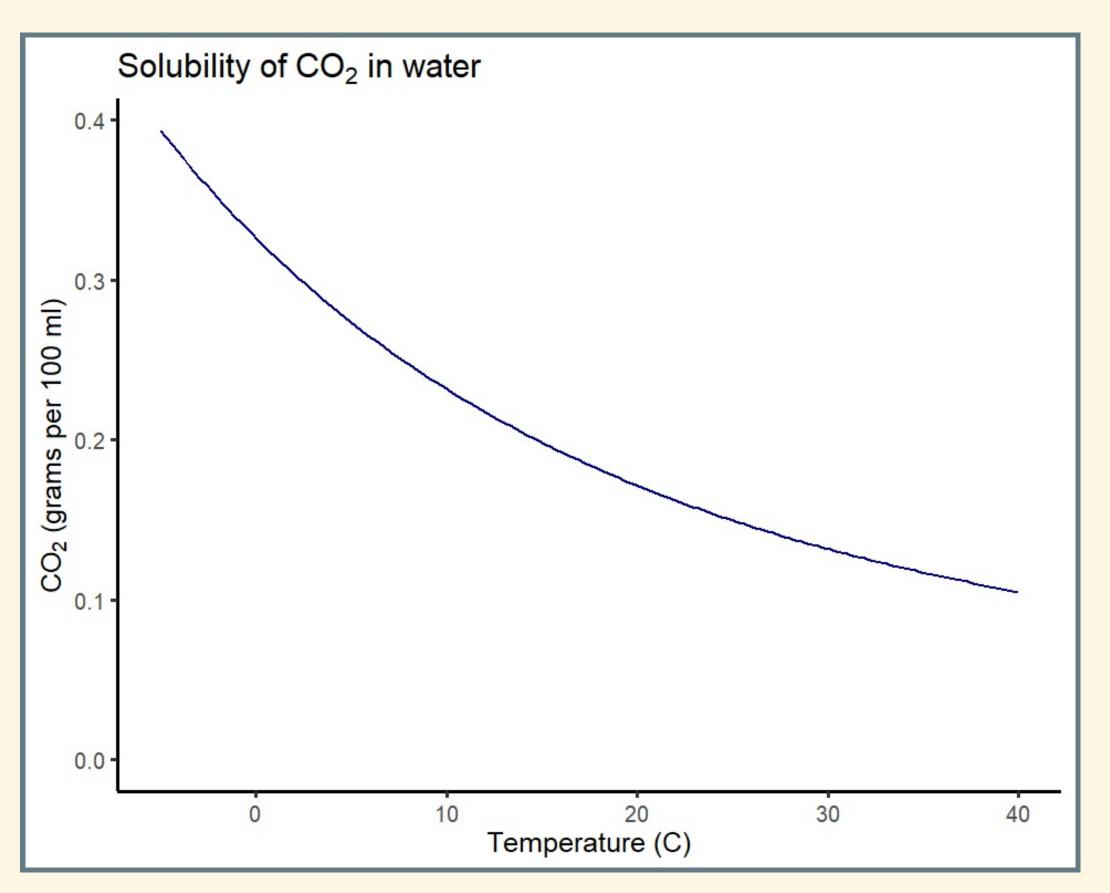
Abstract

The most recent calculations of the infra-red flux in the region of the 15 micron CO₂ band show that the average surface temperature of the earth increases 3.6° C if the CO₂ concentration in the atmosphere is doubled and decreases 3.8° C if the CO₂ amount is halved, provided that no other factors change which influence the radiation balance. Variations in CO₂ amount of this magnitude must have occurred during geological history; the resulting temperature changes were sufficiently large to influence the climate. The CO₂ balance is discussed. The CO₂ equilib-

predicted by the CO₂ theory. When the total CO₂ is reduced below a critical value, it is found that the climate continuously oscillates between a glacial and an inter-glacial stage with a period of tens of thousands of years; there is no possible stable state for the climate. Simple explanations are provided by the CO₂ theory for the increased precipitation at the onset of a glacial period, the time lag of millions of years between periods of mountain building and the ensuing glaciation, and the severe glaciation at the end of the Carboniferous. The extra CO₂ released into the atmosphere by industrial processes and other human activities may have caused the temperature rise during the present century. In contrast with other theories of climate, the CO₂ theory predicts that this warming trend will continue, at least for several centuries.

The Oceans Breathe

The Oceans Breathe

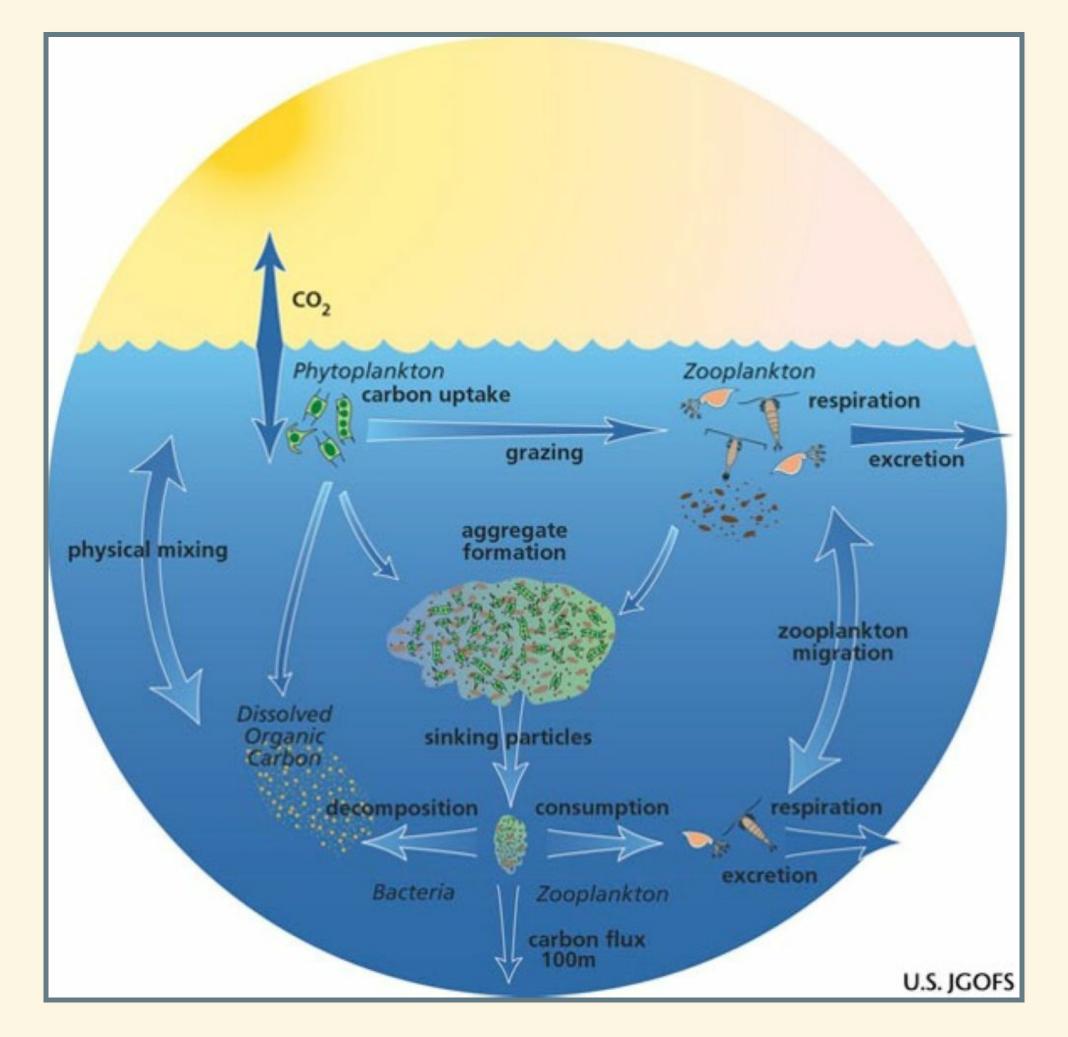


Solubility pump:

- Temperature rises:
 - CO₂ moves from ocean to atmosphere.
- Temperature falls:
 - CO₂ moves from atmospher to ocean.

Positive feedback

Biological Pump



Structure of the ocean

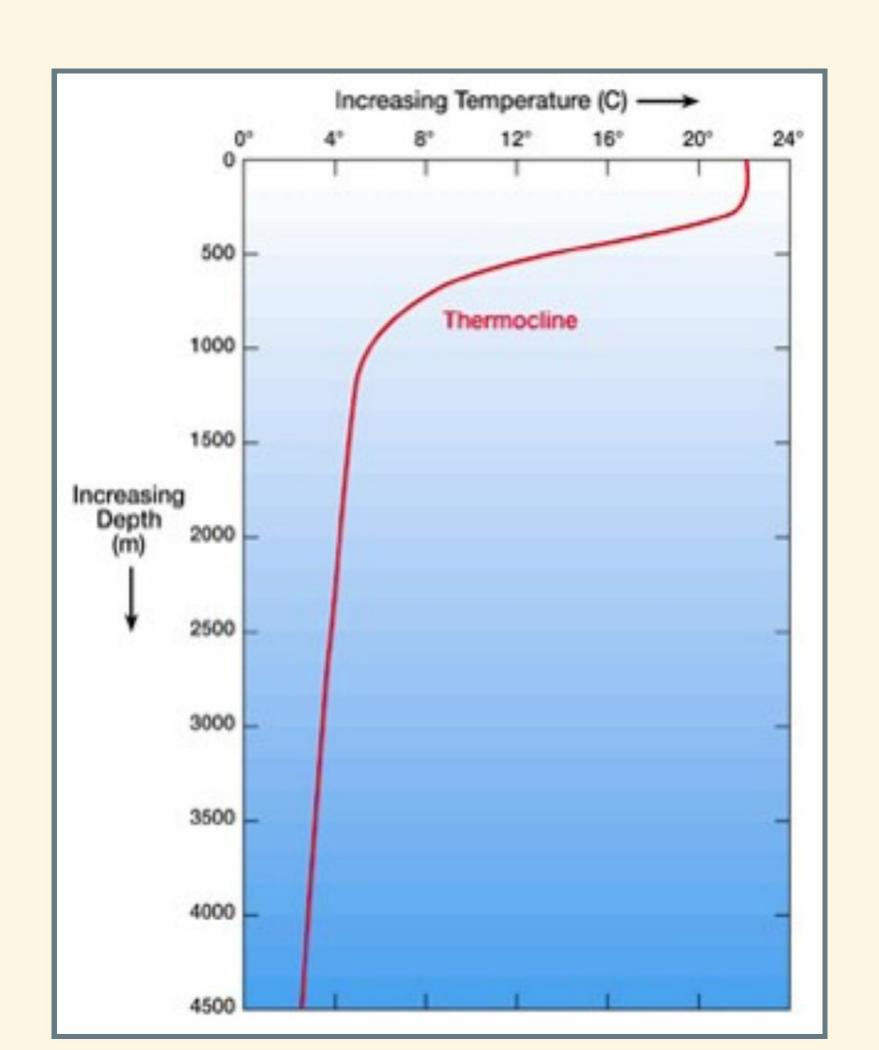
Structure of the ocean

Troposphere:

- Heated from bottom
 - (sunlight absorbed at surface)
- Warmer at bottom
- Unstable → well-mixed

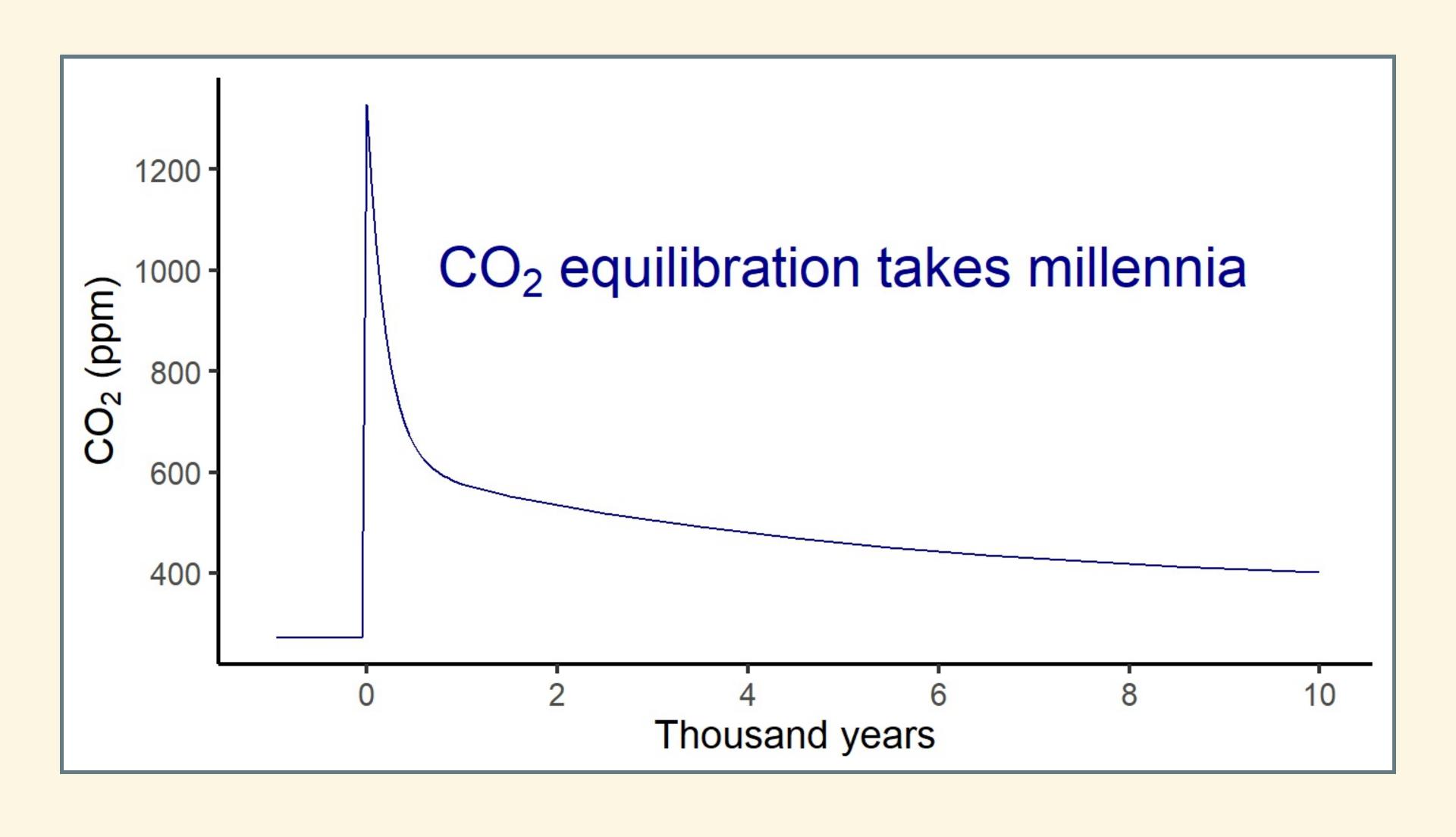
Ocean:

- Heated from top
 - (sunlight absorbed near surface)
- Warmer at top
- Thermocline as barrier to mixing
- Surface layer mixed by wind
- Deep ocean poorly mixed



Ocean Carbon Cycle Numbers:

- Air
 ⇔ Upper ocean:
 - 1000 GT carbon in upper ocean
 - Very fast: 92 GT/year from atmosphere
- - 38,000 GT carbon in deep ocean
 - Slow: 6 GT/year from upper ocean



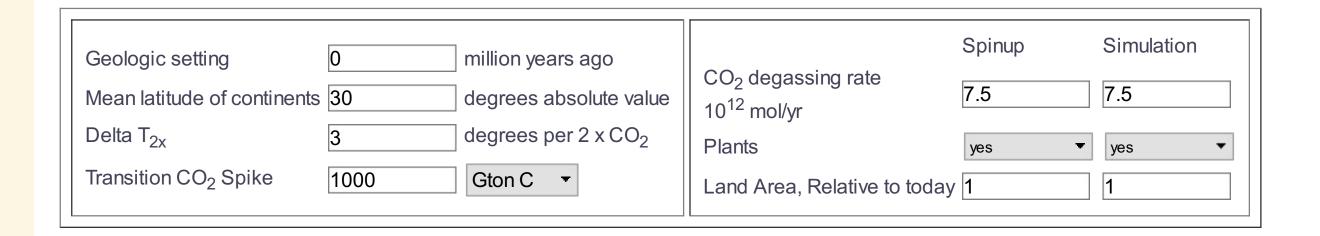
GEOCARB Model

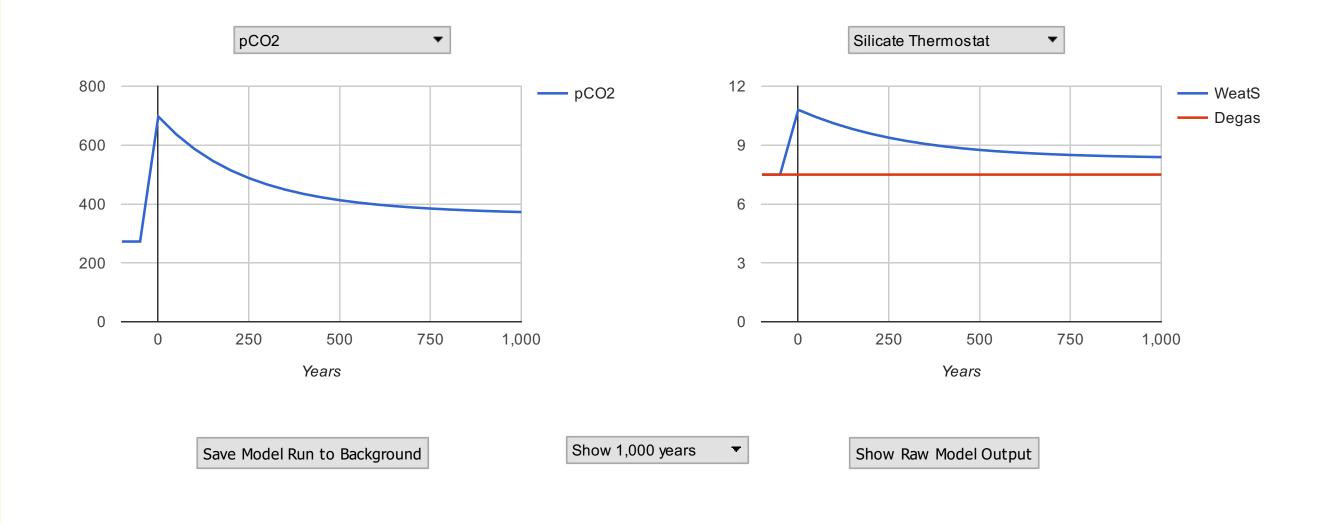
GEOCARB Model

GEOCARB Geologic Carbon Cycle

About this model

Other Models





GEOCARB Model

- 1. Set Transition CO₂ spike to 0
- 2. Monitor pCO₂ and Temperature
- 3. Set Transition CO₂ spike to 2000 GTon C
- 4. Look at different time scales
- 5. After 1000 years, how do pCO₂ and T compare to original values?
- 6. After 10,000 years?

The Rocks Breathe

The Rocks Breathe

- Carbonate vs. Silicate minerals
- Urey Reaction:

$$\mathsf{CaSiO_3} + \mathsf{CO_2} \Leftrightarrow \mathsf{CaCO_3} + \mathsf{SiO_2}$$

- weathering (reactions near surface)
- metamorphism (high temp./pressure deep beneath surface)
- Silicate rocks formed at high temperature (igneous)
- Carbonate rocks formed at low temperature (sedimentary)

Why this is important

- Rain falls on silicate rocks
 - CO₂ dissolves into rainwater
 - Dissolved CO₂ makes rainwater is acidic
- Water dissolves silicate rocks
 - Dissolved ions (ca+2, sio₃-2, etc.)
- In oceans, plankton convert dissolved CO₂ and ions to calcite (calcium carbonate)
- Calcite ends up as limestone on sea floor
- Bottom line:
 - Weathering silicate rocks transforms atmospheric CO₂ to rocks on sea floor.
 - Detailed chemistry on Monday

Weathering as Thermostat

CO₂ is balance of volcanic outgassing and chemical weathering

• Higher temperatures:

- More rain, faster chemical reactions
- Faster weathering
- Atmospheric CO₂ falls

Lower temperatures

- Less rain, slower chemical reactions
- Slower weathering
- Atmospheric CO₂ rises

Temperature of Earth

- As long as outgassing is constant, weathering acts as thermostat.
- Earth's temperature has been remarkably stable over time.
- Change of volcanic outgassing changes "setting" of thermostat.

Temperature of Mars and Venus

- Mars used to be warm.
- Now it is frozen.
- Why?
 - Volcanic outgassing stopped.
 - All CO₂ converted to rocks.
 - No new CO₂ from volcanoes.
- Venus is scorching hot
- Why?
 - Runaway greenhouse:
 - All water evaporated
 - Chemical weathering stopped
 - Outgassing/metamorphism converted all carbonate rocks to CO₂ gas.