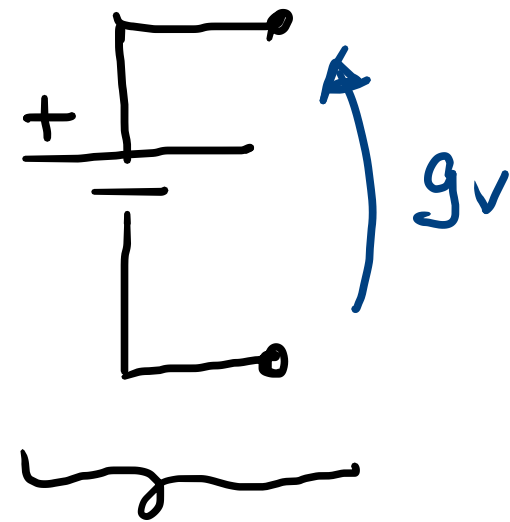
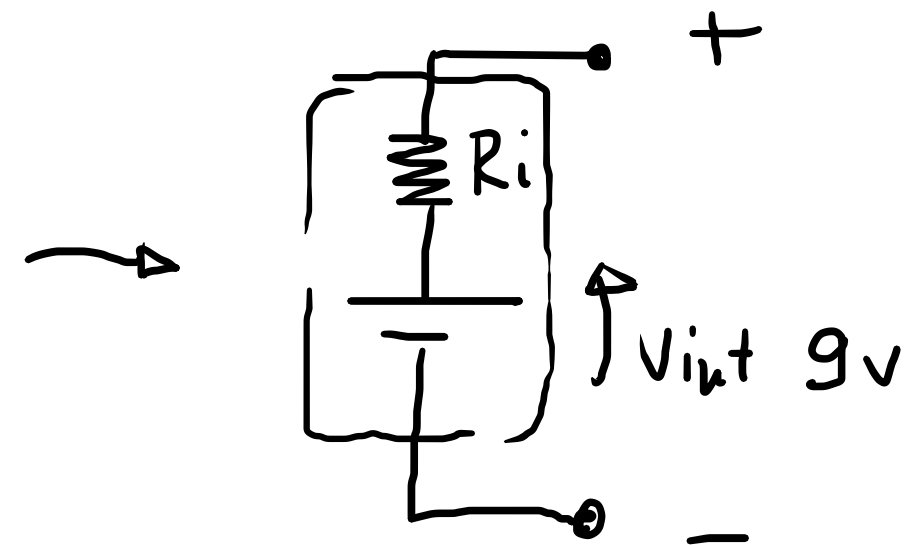


275

BATTERIE



CELLA

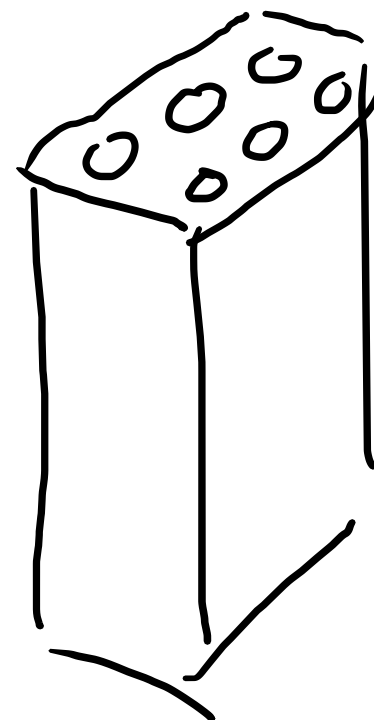


1,5 V

AA

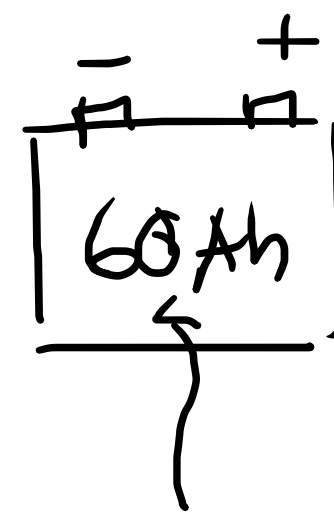


AAA ... AAAA



mAh
Ah

$\underline{\underline{C}}$
 $\underline{\underline{1C}}$



1200 mAh

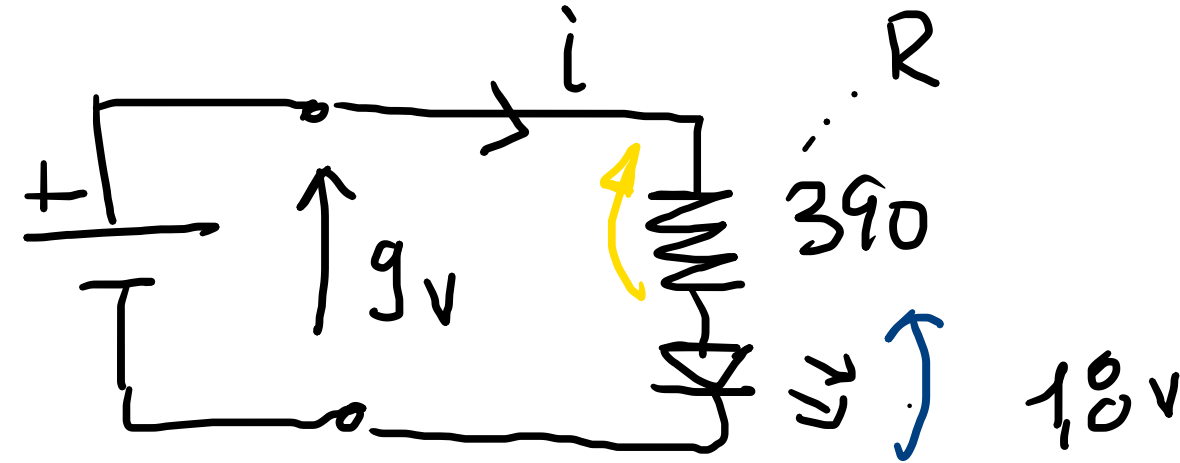
C/10



120 mA

Ah
mAh

9V Alkaline 600 mAh



$$t = \frac{600 \text{ mAh}}{19 \text{ mA}} \approx \underline{31,6 \text{ h}}$$

$$9V = iR + 1,8V$$

$$iR = 9 - 1,8$$

$$i = \frac{9 - 1,8}{R} = \frac{7,2}{390} = 0,019 \text{ A}$$

$$\underline{19 \text{ mA}}$$

