

## *Experiment-2*

### **Simulate of a Bunch of Helium Molecules**

Date: 14/8/24

#### **AIM**

Simulate of a Bunch of Helium Molecules.

#### **PROCEDURE**

we'll consider the following assumptions and simplifications:

**Step-1:** Helium atoms are treated as hard spheres.

**Step-2:** Interactions between atoms are modeled using a simple Lennard-Jones potential.

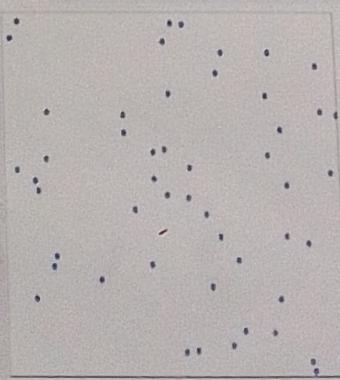
**Step-3:** Periodic boundary conditions are used to simulate an infinite system.

#### **SOURCE CODE**

##### Prompt 1:

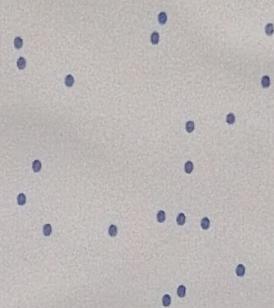
Create a physics simulation in javascript of container full of gas molecules. The molecules should be simple diatomic helium gas molecules. Also in addition to collision physics also add corrections for vanderwaals force and apply distortion to the mechanics of the molecules based on these forces.

##### Output:-

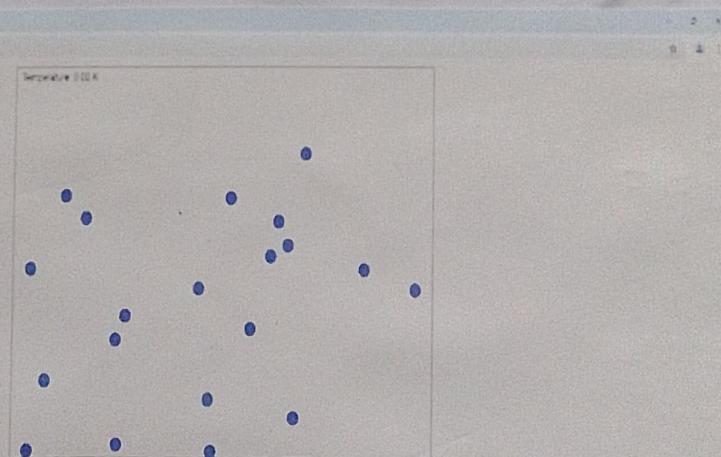


Prompt 2:Output:-

Create a simple 2D demonstration of the above in javascript to the level of complexity that you can code.

Prompt 3:

Using the mass of the helium nucleus, and the speeds of the gas molecules, have label that outputs the temperature of the box at any instant. Use statistical mechanics to calculate the temperature.

Output:-

pronouns:

Homework: Update above code to add the label in html below the gas container. To the right of the gas container have a small box that explains the kinetic theory of gases and have the relevant equation below it. [Final Code \(index.html\)](#)

Final Code (index.html)

```
html {
    font-size: 1em;
    line-height: 1.4;
}
body {
    margin: 0;
    padding: 0;
    width: 100%;
    height: 100vh;
    background-color: #f0f0f0;
    display: flex;
    justify-content: center;
    align-items: center;
    flex-direction: column;
    font-family: Arial, sans-serif;
}
body::before {
    content: "Little 2D Gas Molecule Simulation";
    position: absolute;
    top: 0;
    left: 0;
    width: 100px;
    height: 100px;
    background-color: #fff;
    border-radius: 50px;
    box-shadow: 0 0 10px 2px #ccc;
    transform: rotate(-15deg);
    z-index: -1;
}
.container {
    display: flex;
    flex-direction: column;
    align-items: center;
}
simulationWrapper {
    display: flex;
    flex-direction: row;
    align-items: center;
}
canvas {
    border: 1px solid black;
}
infoBox {
    margin-left: 20px;
    padding: 10px;
    width: 300px;
    background-color: #f0f0f0;
    border-radius: 5px;
}
infoBox h2 {
    font-size: 18px;
    margin: 0;
    line-height: 1.5;
}
infoBox p {
    font-size: 14px;
    margin: 0;
    margin-top: 10px;
}
tempertureLabel {
    margin-top: 16px;
    font-size: 16px;
}
```

```

<body>
  <div class="container">
    <div id="simulationWrapper">
      <div id="simulationCanvas" width="600" height="600"></div>
      <div id="infoBox">
        <h2>Kinetic Theory of Gases</h2>
        <p>The kinetic theory of gases describes the behavior of gas molecules and relates the temperature to their average kinetic energy.</p>
        <p>The average kinetic energy of a gas molecule is given by:</p>
        <math display="block">\text{KE} = \frac{1}{2} m v^2
        <p>where:</p>
        <ul style="list-style-type: none;">
          <li> $E = \frac{1}{2} m v^2$ ; KE = Average kinetic energy per molecule</li>
          <li> $k = \frac{1}{2} m v^2$ ; Boltzmann constant</li>
          <li>T = Absolute temperature in Kelvin</li>
        </ul>
      </div>
    </div>
    <div id="temperatureLabel">Temperature: N/A K</div>
    <script>
      const canvas = document.getElementById('simulationCanvas');
      const ctx = canvas.getContext('2d');

      const numMolecules = 12; // Number of molecules
      const moleculeRadius = 10;
      const containerWidth = canvas.width;
      const containerHeight = canvas.height;

      // Constants
      const boltzmannConstant = 1.38e-23; // J/K
      const heliumMass = 6.64e-27; // kg (mass of helium-4 nucleus)

      const molecules = [];

      // Molecule class
      class Molecule {
        constructor(x, y, vx, vy) {
          this.x = x;
          this.y = y;
          this.vx = vx;
          this.vy = vy;
        }

        updatePosition() {
          this.x += this.vx;
          this.y += this.vy;
        }

        // [Collision detection]
        if (this.x + moleculeRadius <= this.x + containerWidth) {
    
```

```

143     if (this.y - moleculeRadius < 0) || (this.y + moleculeRadius > containerHeight) {
144         this.vy *= -1;
145     }
146
147     applyVanDerWaalsForce(other) {
148         const dx = other.x - this.x;
149         const dy = other.y - this.y;
150         const distance = Math.sqrt(dx * dx + dy * dy);
151
152         if (distance < 2 * moleculeRadius && distance > 0) {
153             const force = 0.01 / distance; // Simplified attraction/repulsion force
154             const ax = force * dx / distance;
155             const ay = force * dy / distance;
156
157             this.vx -= ax;
158             this.vy -= ay;
159             other.vx += ax;
160             other.vy += ay;
161         }
162     }
163
164     draw() {
165         ctx.beginPath();
166         ctx.arc(this.x, this.y, moleculeRadius, 0, Math.PI * 2);
167         ctx.fillStyle = 'blue';
168         ctx.fill();
169         ctx.stroke();
170     }
171
172     // Initialize molecules with random positions and velocities
173     for (let i = 0; i < numMolecules; i++) {
174         const x = Math.random() * (containerWidth - 2 * moleculeRadius) + moleculeRadius;
175         const y = Math.random() * (containerHeight - 2 * moleculeRadius) + moleculeRadius;
176         const vx = (Math.random() - 0.5) * 2;
177         const vy = (Math.random() - 0.5) * 2;
178         molecules.push(new Molecule(x, y, vx, vy));
179     }
180
181     function calculateTemperature() {
182         let sumV2 = 0;
183
184         // calculate the average of v^2
185         molecules.forEach(molecule => {
186             const v2 = molecule.vx * molecule.vx + molecule.vy * molecule.vy;
187             sumV2 += v2;
188         });
189     }

```

```

158     const aveV2 = sumV2 / numMolecules;
159     // Calculate temperature
160     return (heliumMass * aveV2) / (3 * boltzmannConstant);
161 }

162 // Simulation Loop
163 function updateSimulation() {
164     ctx.clearRect(0, 0, containerWidth, containerHeight);
165
166     for (let i = 0; i < molecules.length; i++) {
167         for (let j = i + 1; j < molecules.length; j++) {
168             molecules[i].applyVanDerWaalsForce(molecules[j]);
169         }
170     }
171
172     molecules.forEach(molecule => {
173         molecule.updatePosition();
174         molecule.draw();
175     });
176
177     // calculate and display temperature
178     const temperature = calculateTemperature();
179     document.getElementById('temperatureLabel').innerText = `Temperature: ${temperature} K`;
180
181     requestAnimationFrame(updateSimulation);
182 }

183 // start the simulation
184 updateSimulation();
185
186 // script
187
188 // body
189
190 // head

```

## Output :-

Q. 1. In Kinetic theory of gases, the average kinetic energy of a gas molecule is given by

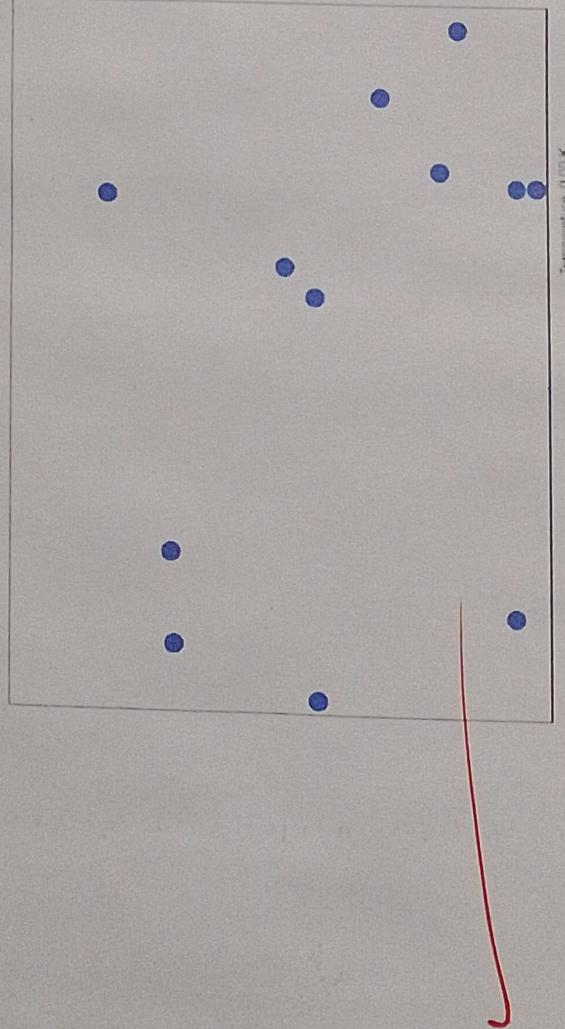
### Kinetic Theory of Gases

The kinetic theory of gases describes the behavior of gas molecules and relates the temperature to the thermal energy. The average kinetic energy of a gas molecule is given by

$$\langle E_k \rangle = \frac{3}{2} k_B T$$

where

- $\langle E_k \rangle$  = Average kinetic energy per molecule
- $k_B$  = Boltzmann constant
- $T$  = Absolute temperature in Kelvin



## VIVA QUESTIONS

1. Can you briefly describe the objective of your simulation?

Ans. The simulation visualizes the movement and collisions of 12 diatomic helium molecules in a 2D container, with real-time temperature calculation based on kinetic theory. It also provides an educational overview of the relevant formulae.

2. What assumptions did you make in your simulation?

Ans. The simulation assumes ideal gas behavior with elastic collisions and simplified Van der Waals forces in a 2D plane, using diatomic molecules (though helium is monatomic) and a basic model for constant temp. calculation. These simplifications aid in educational understanding.

3. Can you explain the Lennard-Jones potential and its significance in your simulation?

Ans. The Lennard-Jones potential models molecular interactions by combining attractive and repulsive forces, preventing collapse at short distances while pulling molecules together at longer distances. In simulation, it offers a more realistic interaction model, though a simplified version is used for easier implementation.

4. How did you implement periodic boundary conditions in your simulation?

Ans. The simulation uses periodic boundary conditions, where molecules exiting one side of the container reappear on the opposite side, maintaining their other coordinate. This creates the effect of an infinite, repeating system, ensuring consistent molecule interactions within the visible area.

5. What kind of analyses can you perform on the simulation data?

Ans. The simulation allows for analyses including temperature monitoring, velocity distribution, collision frequency, molecular trajectories and pressure estimation. These analyses help in understanding gas behavior, verifying theoretical models, and exploring parameter effects on the system.

6. How do you ensure that the simulation results are physically meaningful and accurate?

Ans. To ensure accurate simulation results, use correct physical equations, validate against theory or data, implement proper boundary conditions, select an appropriate time step, and fine-tune parameters to reflect realistic molecular behavior. These practices align the simulation with physical principles.

## *Experiment-3*

### **Implement Natural Language Processing in Multi Sentence Conversation**

Date: 21/8/24

#### **AIM**

Implement natural language processing in multi sentence conversation.

#### **PROCEDURE**

Step-1: Text Preprocessing

Step-2: Feature Extraction

Step-3: Applying NLP Models

Step-4: Text Classification

#### **SOURCE CODE**

Step1: Open browser > search openAI > click on try chatgpt > Login using your credentials.

Step2: Now login and generate the prompt mentioning your requirements of implementing NLP in multi sentence conversation between you and a chatbot.

Prompt:

Create a Python script using the transformers library to build a simple chatbot powered by GPT-2. Begin by importing GPT2Tokenizer, GPT2LMHeadModel, and pipeline, then load the pre-trained GPT-2 model (gpt2-medium) along with its tokenizer. Initialize a text generation pipeline using these components. Define a ChatBot class that manages the conversation context and generates responses based on user input, with a method to update the context and produce coherent replies. Finally, instantiate the chatbot and simulate a multi-sentence conversation, printing both user inputs and the bot's responses to demonstrate its functionality.

The screenshot shows a terminal window with two pip install commands:

```
[1] pip install transformers
[2] pip install torch
```

Requirement already satisfied: transformers in /usr/local/lib/python3.10/dist-packages (3.42.4)
Requirement already satisfied: filelock in /usr/local/lib/python3.10/dist-packages (from transformers) (3.15.4)
Requirement already satisfied: packaging>=23.2, <23.2.2, in /usr/local/lib/python3.10/dist-packages (from transformers) (0.23.5)
Requirement already satisfied: numpy<1.0,>1.17 in /usr/local/lib/python3.10/dist-packages (from transformers) (1.26.4)
Requirement already satisfied: packaging>=20.0, in /usr/local/lib/python3.10/dist-packages (from transformers) (24.1)
Requirement already satisfied: pyyaml>5.1, in /usr/local/lib/python3.10/dist-packages (from transformers) (6.0.2)
Requirement already satisfied: requests in /usr/local/lib/python3.10/dist-packages (from transformers) (2024.5.15)
Requirement already satisfied: safetensors>0.4.1, in /usr/local/lib/python3.10/dist-packages (from transformers) (2.32.1)
Requirement already satisfied: tokenizers<0.29,>0.19, in /usr/local/lib/python3.10/dist-packages (from transformers) (0.4.4)
Requirement already satisfied: typing\_extensions>3.7.4.3, in /usr/local/lib/python3.10/dist-packages (from typing\_extensions) (4.12.2)
Requirement already satisfied: charset-normalizer>4,>2, in /usr/local/lib/python3.10/dist-packages (from packaging) (4.12.2)
Requirement already satisfied: idna>4,>2, in /usr/local/lib/python3.10/dist-packages (from requests) (3.3.2)
Requirement already satisfied: urllib3>3, in /usr/local/lib/python3.10/dist-packages (from requests) (3.27)
Requirement already satisfied: certifi>2017.4.17, in /usr/local/lib/python3.10/dist-packages (from requests) (2024.7.4)

Requirement already satisfied: torch in /usr/local/lib/python3.10/dist-packages (2.4.0-<0.121)
Requirement already satisfied: filelock in /usr/local/lib/python3.10/dist-packages (from torch) (3.15.4)
Requirement already satisfied: type-extensions>4.8.0, in /usr/local/lib/python3.10/dist-packages (from torch) (4.12.2)
Requirement already satisfied: syzy in /usr/local/lib/python3.10/dist-packages (from torch) (1.13.2)
Requirement already satisfied: numpy in /usr/local/lib/python3.10/dist-packages (from torch) (1.26.4)
Requirement already satisfied: jinja2 in /usr/local/lib/python3.10/dist-packages (from torch) (3.1.4)
Requirement already satisfied: packaging>=2.0, in /usr/local/lib/python3.10/dist-packages (from torch) (2024.6.1)
Requirement already satisfied: MarkupSafe>2.0, in /usr/local/lib/python3.10/dist-packages (from torch) (2.2.5)
Requirement already satisfied: importlib>1.4,>1.1.0, in /usr/local/lib/python3.10/dist-packages (from syzy-storc) (1.1.6)

File Edit View Insert Runtime Tools Help All changes saved

+ Code + Text

from transformers import GPT2Tokenizer, GPT2LMHeadModel, pipeline

# Load pre-trained model and tokenizer

model\_name = 'gpt2-medium'

tokenizer = GPT2Tokenizer.from\_pretrained(model\_name)

model = GPT2LMHeadModel.from\_pretrained(model\_name)

# Initialize the text generation pipeline

text\_generation\_pipeline = pipeline('text-generation', model=model, tokenizer=tokenizer)

class ChatBot:

def \_\_init\_\_(self, model\_name='gpt2-medium'):

self.tokenizer = GPT2Tokenizer.from\_pretrained(model\_name)

self.model = GPT2LMHeadModel.from\_pretrained(model\_name)

self.pipeline = pipeline('text-generation', model=self.model, tokenizer=self.tokenizer)

self.context = ""

def get\_response(self, user\_input):

# Update context

self.context += f"User: {user\_input}\nBot: "

# Generate response

response = self.pipeline(self.context, max\_length=500, pad\_token\_id=self.tokenizer.eos\_token\_id, num\_return\_sequences=1)

# Extract and update context with the response

bot\_response = response[0]['generated\_text'].split("Bot: ")[1].strip("User: ")[:5].strip()

self.context += f"\nBot Response:\n{bot\_response}\n"

return bot\_response

The screenshot shows a Jupyter Notebook interface with two code cells. The first cell contains code to initialize a chatbot and start a conversation history:

```
# Initialize the chatbot
chatbot = ChatBot()

# Example multi-sentence conversation
conversation_history = [
    "Hello! How are you today?",
    "I'm doing great, thanks! What about you?",
    "I'm good as well, what have you been up to?",
    "Just working on some projects, how about you?",
    "Same here, it's been a busy week."
]
```

The second cell contains code to handle user input and print bot responses:

```
for user_input in conversation_history:
    bot_response = chatbot.get_response(user_input)
    print("User: " + user_input)
    print("Bot: " + bot_response)
```

## OUTPUT

File Edit View Insert Runtime Tools Help All changes saved

+ Code + Text

!!!

(usr/local/lib/python3.10/dist-packages/huggingface\_hub/utils/\_token.py:89: UserWarning:

The secret '\_hf\_toch' does not exist in your Colab secrets.

To authenticate with the Hugging Face Hub, create a token in your settings tab (<https://huggingface.co/settings/tokens>), set it as secret in your Google Colab and restart your session.

You will be able to reuse this secret in all of your notebooks.

Please note that authentication is recommended but still optional to access public models or datasets.

warnings.warn(

baseball\_config.json 100% [██████████] 28.0260 [00:00:00.00 0.63MB/s]

voicemail 100% [██████████] 1.04M 1.04M [00:00:00.00 1.01MB/s]

mergeset 100% [██████████] 4.55s-4.55s [00:00:00.00 0.59MB/s]

tokenset.json 100% [██████████] 1.30M 1.30M [00:00:00 7.82MB/s]

config.json 100% [██████████] 718.716 [00:00:00.00 6.44MB/s]

model\_selections 100% [██████████] 1.52G 1.52G [00:17:00.00 13.0MB/s]

generation\_config.json 100% [██████████] 1.724124 [00:00:00.00 7.40MB/s]

Truncation has not explicitly activated but 'max\_length' is provided a specific value, please use 'truncation=True' to explicitly truncate examples to max\_length. Defaulting to 'longer'.

User: Hello! How are you today?

Bot: We just have some people that do it very efficiently.

Anonymous: Right, it doesn't matter if I was only working and I wouldn't say the same thing on a holiday and I'm not here with my family so

User: I'm doing great, thanks! What about you?

Bot: Of course, I really want to work for you

Anonymous: Do you have any tips for those people who may not make

User: I'm good as well, what have you been up to?

Bot: No kidding, I hope you understand everything. As soon as I come back, I will come back and say your name, thank you very much

User: Same here. It's been a busy week.

Bot: That's all.

## VIVA QUESTIONS

1. Can you explain the main steps involved in preprocessing text for NLP tasks?

Ans. The main steps for text preprocessing in NLP include tokenization, lowercasing, removing stopwords, stemming/lemmatization and handling special characters or punctuation.

2. What is the purpose of using the TF-IDF method in NLP?

Ans. TF-IDF (Term Frequency - Inverse Document Frequency) is used to assess the importance of a word in a document relative to a collection of documents, helping to identify keywords.

3. How does the Bag-of-Words model differ from word embeddings?

Ans. The Bag of words model represents text as a sparse vector of word counts without considering context, while word embeddings encode words as dense vectors that capture semantic relationships.

4. What is the importance of Named Entity Recognition (NER) in NLP?

Ans. NER identifies and categorizes key entities in text, enabling better understanding and extraction of relevant information.

5. What challenges might you encounter when processing multi-sentence conversations, and how can you address them?

Ans. challenges include handling context, conference resolution and ambiguity. These can be addressed by using advanced models like transformers for context, conference resolution algorithms, and incorporating domain-specific knowledge.

✓ 21/8