

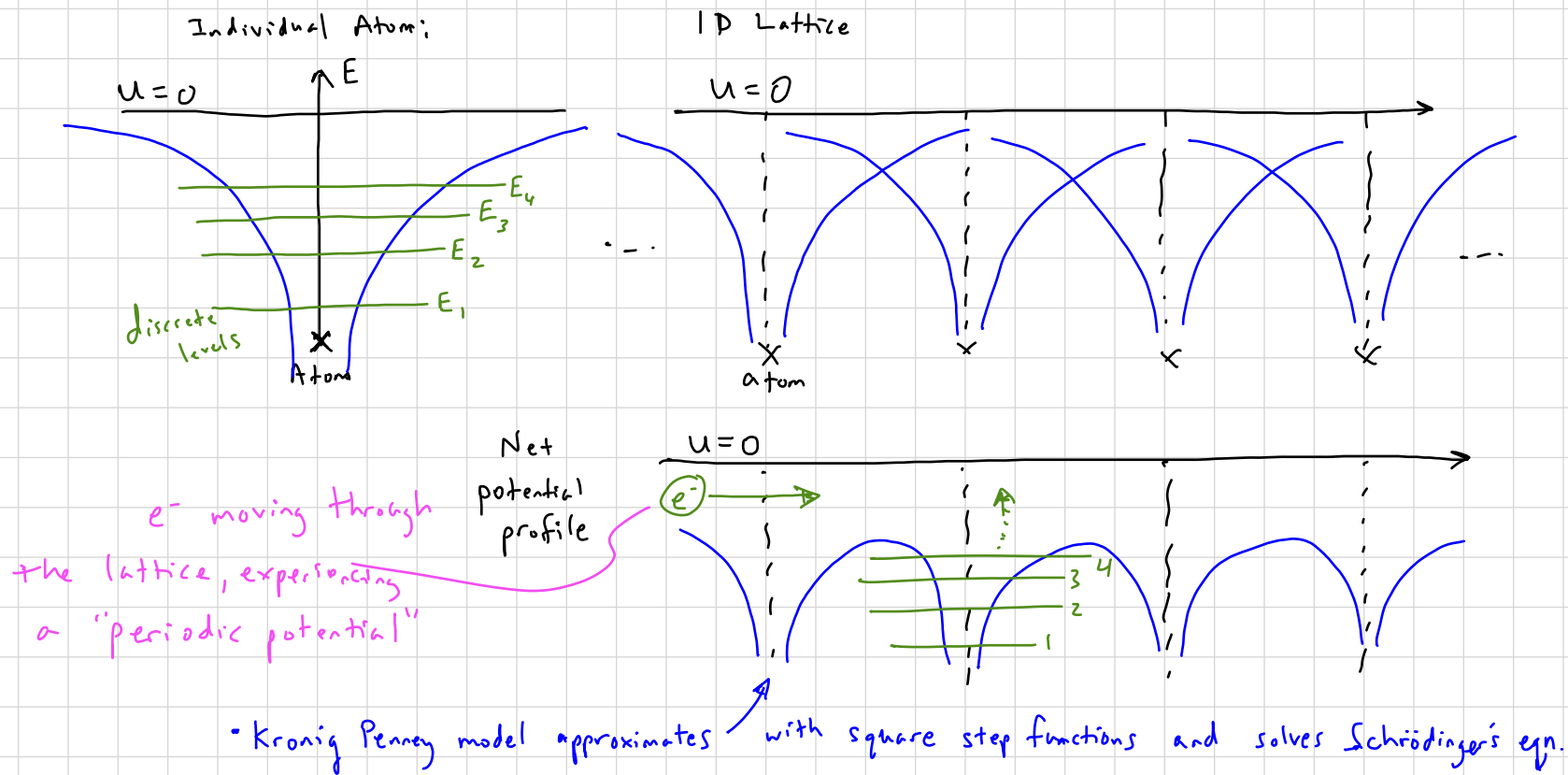
Lecture #4

Quantum Theory of Solids (cont.)

Kronig-Penney Model

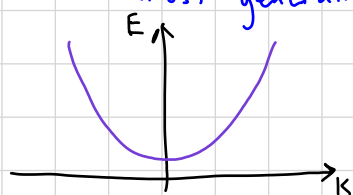
Energy bands tell us where e^- s are allowed to be, but what dictates how they move from one place to another?

→ most influential is their attraction to the atomic nuclei



What is the result?

- A solution for energy of an e^- with relation to K → "wave number" and the momentum (p) of the e^- is:
- This solution/relation between E and K is also known as a "dispersion relation"
- The most generalized solution gives this relation:

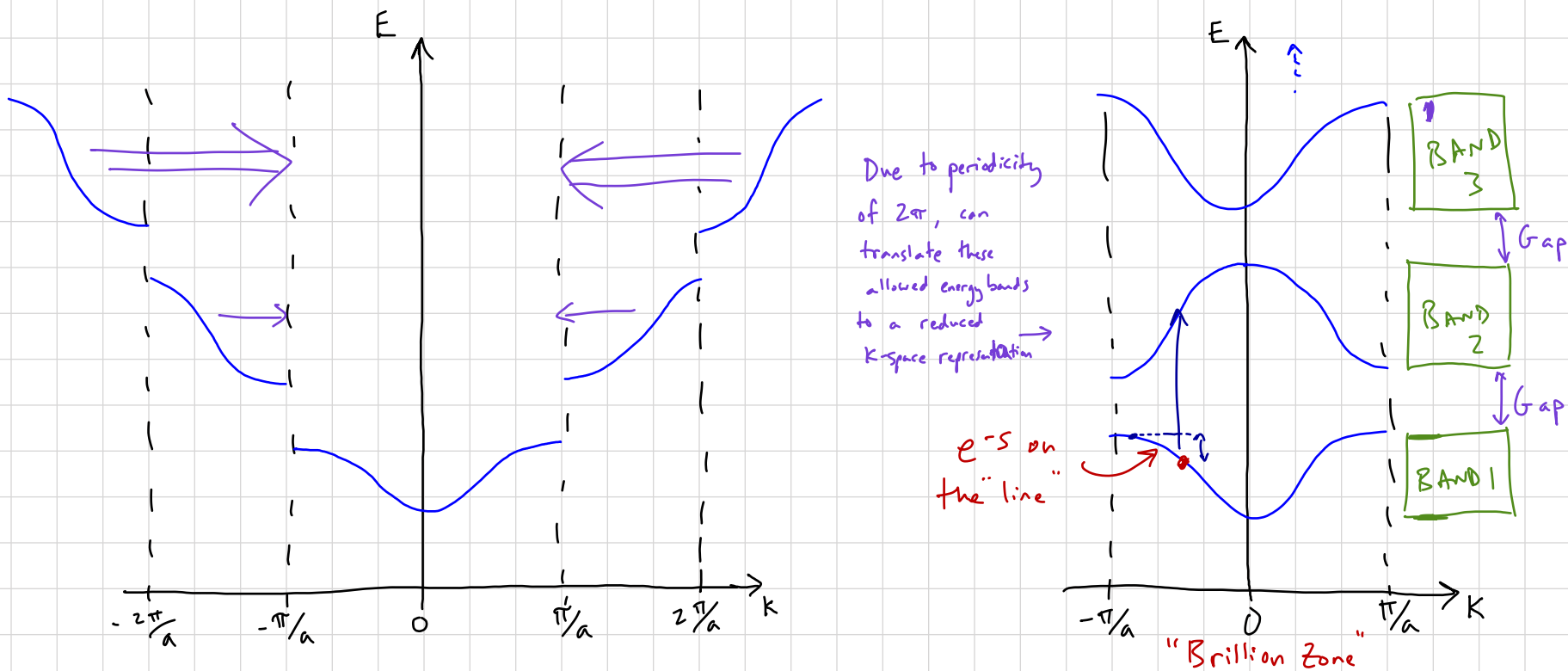


$$E = \frac{\hbar^2 K^2}{2m}$$

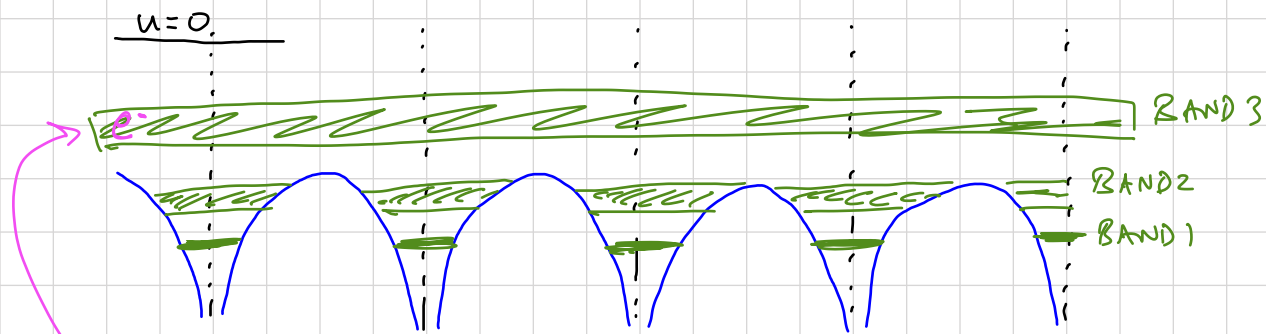
parabolic dispersion relation for "free particle" (NOT in a 1D periodic lattice!)

$$p = \hbar K, \quad E = \frac{p^2}{2m}$$

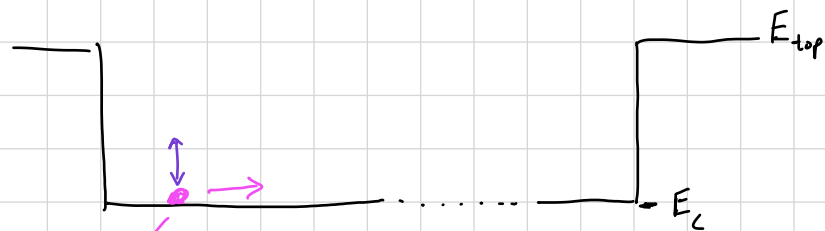
If the periodic conditions of the crystal are applied, then the "allowed" values of K for which the wave eqn. has a solution give you your $E-K$ diagram:



Visualizing these energy bands in a crystal:



Generally, an e^- will be in an "upper" or higher energy band and thus is effectively like a particle-in-a-box with:



effectively a "free particle" in a well with periodic B.C.'s

$$E = \frac{\hbar^2 k^2}{2m}, \quad k = \sqrt{\frac{2mE}{\hbar^2}}$$

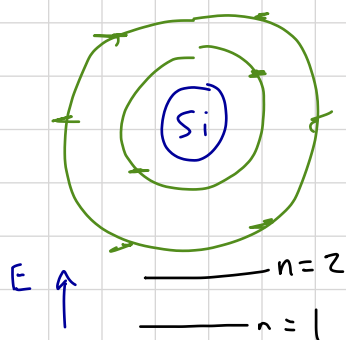
Comes from time-independent S.E.:

$$\frac{\partial^2 \psi}{\partial x^2} + \frac{2mE}{\hbar^2} \psi = 0$$

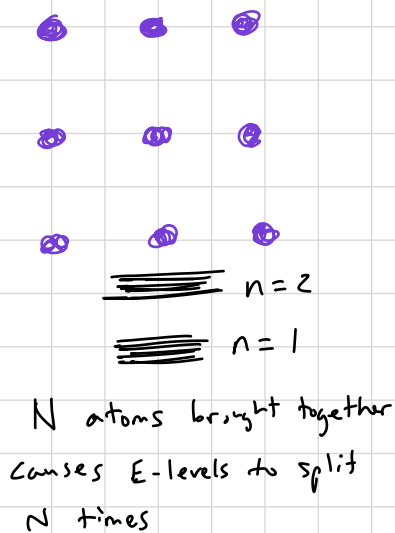
$\sim k^2$

→ Where are e^- s allowed to be in a semiconductor crystal?

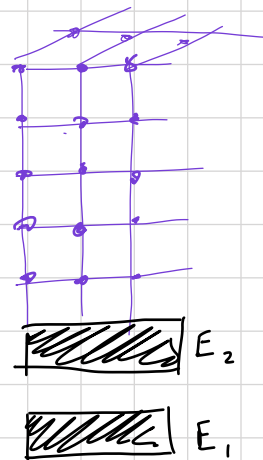
RECAP:



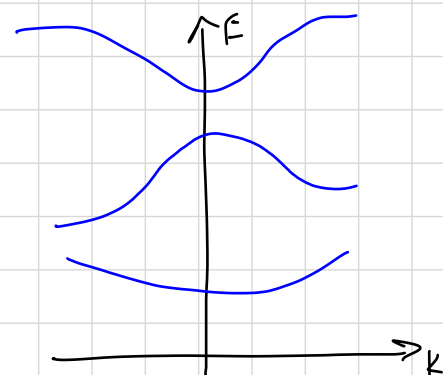
Atom: e^- s in orbitals



N atoms brought together causes E-levels to split N times



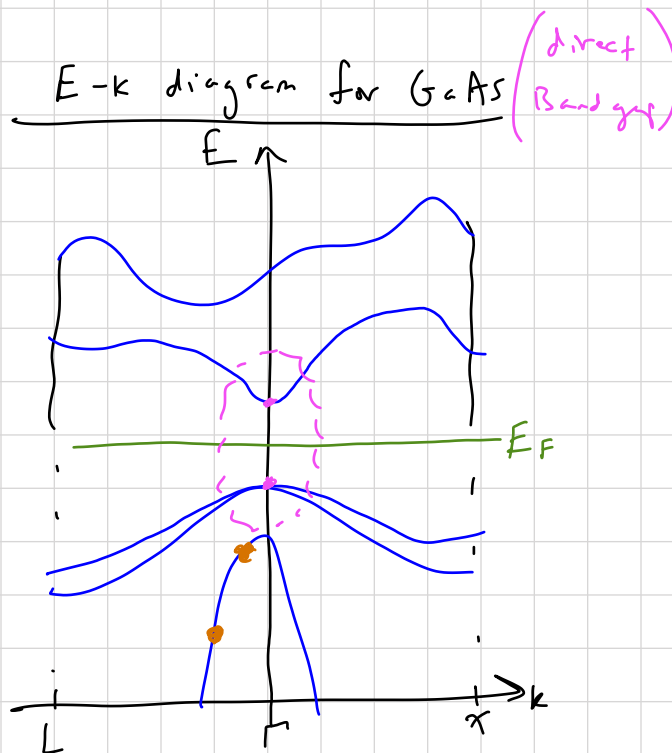
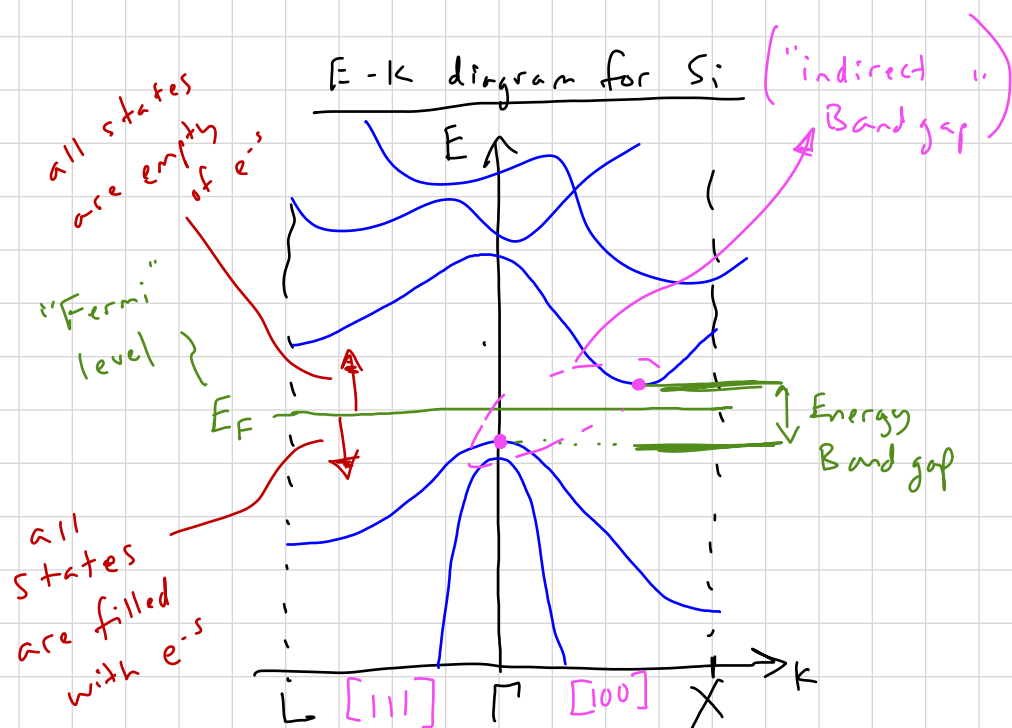
Energy bands are formed in atomic crystals



Solving wave eqn. in periodic crystal yields relationship between E-k

wave number
(crystallographic direction vector)

→ Thus far this was developed for 1D, how does it look in 3-D crystal?



→ What properties does an e^- have?

properties does an e^- have?

→ Effective mass (m^*) Consider parabolic $E-k$ relation: $E = \frac{\hbar^2 k^2}{2m}$

$$\frac{dE}{dk} = \frac{\hbar^2 k}{m} = \frac{\hbar p}{m}$$

relate p to velocity by $v = \frac{p}{m}$:

$$\frac{1}{\hbar} \frac{dE}{dk} = \frac{p}{m} = v$$

take 2nd derivative:

take 2nd derivative: $\frac{d^2E}{dk^2} = \frac{\hbar^2}{m} \rightarrow m^* = \hbar^2 \left(\frac{d^2E}{dk^2} \right)^{-1}$

★ For a free e^- mass is constant, but when bound in crystal with periodic potential it depends on inverse of $E-k$ curvature

m^* ties classical world to quantum: $F = ma = -eE$ (see textbook)