Nuclear Fuel Cycle

NUGN506 - LME1

By

GUILLAUME L'HER



Department of Nuclear Engineering COLORADO SCHOOL OF MINES

LME submitted for the Nuclear Fuel Cycle class at the Colorado School of Mines.

 $F_{\rm ALL} \ 2017$

TABLE OF CONTENTS

]	Pa	ge
1	Lea	rning l	Measuren	nent 1															1
	1.1	Extra	Credit		 	 								 					1
		1.1.1	Problem		 	 													1
		1.1.2	Solution		 	 							•			•			1
Bi	bliog	raphy																	3

CHAPTER

LEARNING MEASUREMENT 1

he extra-credit problems for the first learning measurement exercise is tackled in this document.

1.1 Extra Credit

1.1.1 Problem

Thermal neutron fission of U-235 produces nuclides of the A=132 isobar in 4.31% of all fissions. This isobar includes the radioactive doubly-magic nuclide Sn-132 with Z=50, N=82 and a half-life of 39.7 seconds, but Te-132 has the highest independent fission yield for A=132. Why aren't significant amounts of Cs-132 and Ba-132 (or Cs-131 or Ba-133 for the A=131 or A=133 isobars, respectively) observed in the cumulative fission yield when neighboring nuclides like Cs-133 and Ba-134 are? [Hint: It has nothing to do with magic numbers as Te-132 is the most probable fission product.]

1.1.2 Solution

The cumulative fission yield is different from the independent fission yield. The cumulativie fission yield also consider the decay of the precursors to form a nuclide of interest. In our case, we know that Te-132 has the highest independent fission yield for A = 132. According the the chart of the nuclides, it beta decay to I-132, I-132 then can beta decay to a stable Xe-132, or lose its excited state. Consequently, this decay chain does not produce any element with more protons than Xenon, and thus no Cs-133 or Ba-133. Similarly for A = 131 or A = 133, the decay chains stop at the elements Xe-131 and Cs-133 respectively.

Consequently, the cumulative fission yield for A132-isobars Cs-132 and Ba-132 are "only" their independent fission yield. In the meantime, Cs-133 and Ba-134 can also be created by the β - decay chains:

(1.1)
$$^{133}I \rightarrow ^{133}Xe \rightarrow ^{133}Cs$$

(1.2)
$$^{134}Cs \rightarrow ^{134}Ba$$

BIBLIOGRAPHY