

Solution: CSTR with Cooling Coil

Given Data

Reaction: $A \rightarrow C$ (exothermic, first-order)

$$V = 50 \text{ dm}^3, \quad v_0 = 5 \text{ dm}^3/\text{min}, \quad C_{A0} = 2.0 \text{ mol/dm}^3,$$

$$T_f = 300 \text{ K}, \quad \Delta H_R = -50,000 \text{ cal/mol},$$

$$T_c = 270 \text{ K}, \quad \dot{m}_c = 1500 \text{ g/min}, \quad C_{p,c} = 1 \text{ cal/(g} \cdot \text{K)},$$

$$U = 50 \text{ cal/(\text{min} \cdot \text{m}^2 \cdot \text{K})},$$

$$k(300 \text{ K}) = 0.20 \text{ dm}^3/(\text{mol} \cdot \text{min}), \quad E = 20,000 \text{ cal/mol},$$

$$R = 1.987 \text{ cal/(\text{mol} \cdot \text{K})}.$$

(a) Minimum Conversion without Cooling Limitations

Mass Balance: For a steady-state CSTR:

$$F_{A0} - F_A = -r_A V$$

with

$$F_A = C_A v_0 = C_{A0}(1 - X)v_0, \quad -r_A = k(T)C_A.$$

Substitute:

$$C_{A0}Xv_0 = k(T)C_{A0}(1 - X)V.$$

Cancelling C_{A0} :

$$X = \frac{k(T)V/v_0}{1 + k(T)V/v_0}.$$

Temperature Dependence:

$$k(T) = k(300) \exp \left[\frac{E}{R} \left(\frac{1}{300} - \frac{1}{T} \right) \right].$$

Energy Balance (no coil): Only sensible heat removal by outlet stream:

$$\rho C_p v_0 (T_f - T) + (-\Delta H_R) v_0 C_{A0} X = 0.$$

Hence:

$$T = T_f + \frac{(-\Delta H_R)C_{A0}X}{\rho C_p}.$$

Coupling Equations: The above mass and energy balance equations are solved simultaneously for X and T .

(b) Cooling Coil Surface Area to Maintain Conversion from (a)

With a coil, the steady-state energy balance is:

$$\rho C_p v_0 (T_f - T) + UA(T_c - T) + (-\Delta H_R) v_0 C_{A0} X = 0.$$

Rearranging for A :

$$A = \frac{(-\Delta H_R) v_0 C_{A0} X - \rho C_p v_0 (T - T_f)}{U(T - T_c)}.$$

Here, X and T are taken from part (a) to maintain the same conversion.

(c) Minimum Conversion in the System

The minimum conversion corresponds to the thermal runaway limit, obtained by simultaneously solving:

$$\frac{dT}{dX} \text{ from energy balance} \quad \text{and} \quad X(T) \text{ from mass balance.}$$

This requires checking the intersection points of the X - T curves.