



CY104S1 (Subjective)_E...



carefully and answer.

(Non-anonymous question (i)) *

(30 Points)

1. Explain the extrusion molding technique for polymer processing with application. [3 marks]
2. In a Beer-lambert law cell, the aqueous solution of a substance of known concentration absorbs 10 percent of the light. What fraction of the incident light will be absorbed by a five times as long? [3 marks]
3. The molar absorption coefficient of a substance dissolved in cyclohexane is $1710 \text{ L mol}^{-1} \text{ cm}^{-1}$ at 500 nm. The reduction in intensity of light of the same wavelength that passes through a cell of 1 mm path length containing a 2 mmol L^{-1} solution is _____%.? [2 marks]
4. Classify nanoparticles very briefly with one examples. [3 Marks]
5. Describe the electrochemical (anodic and cathodic) reactions for galvanic corrosion. Give one example of passivation and sacrificial anodic protection. [3+1 marks]
6. A sample of water on analysis gave the following results: [3 marks]

$\text{Ca}^{2+} = 30 \text{ mg/L}$; $\text{Mg}^{2+} = 24 \text{ mg/L}$; $\text{CO}_2 = 24 \text{ mg/L}$; $\text{HCl} = 50 \text{ mg/L}$; $\text{K}^+ = 10 \text{ mg/L}$

 - (i) Calculate the total hardness of water sample
 - (ii) Also calculate the quantities of lime and soda needed for softening one litre of this water sample
7. Explain the effect of the following in boiler feed water : (i) silica (ii) CO_2 [2 marks]
8. Why can caustic embrittlement be controlled by adding sodium sulphate to boiler feed water? [1 marks]
9. Write down the Nernst's equation and calculate the reduction potential for the reduction of O_2 at pH=7. Given partial pressure of O_2 [$p\text{O}_2$] = 0.20 bar and $E^\circ = 1.229 \text{ V}$ at 25°C .
The reduction reaction involved is given: $2\text{H}^+ + \frac{1}{2}\text{O}_2(\text{g}) + 2\text{e}^- \rightleftharpoons \text{H}_2\text{O}(\text{l})$ [2 marks]
10. Give the example of following dyes with structures
 - a) Indigo dye and
 - b) anthraquinone dye
11. What is the Pharmaceutical phase and Pharmacodynamics phase? [2 marks]
12. What is antagonist? [1 mark]
13. What is bioavailability? [1 mark]
14. What are the requirements of ideal drug? (write four criteria) [1 mark]

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