

Energy and Environmental Engineering

Refrigeration and Air-conditioning



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2nd law of thermodynamics

- Kelvin Planck Statement: The Kelvin statement of the second law of thermodynamics: **It is impossible to convert the heat from a single source into work without any other effect.**
- Clausius Statement: The Clausius statement of the second law of thermodynamics states that **“it is impossible to construct a device that operates in a cycle and produces no effect other than the transfer of heat from a cooler body to a hotter body.”**

Introduction

Second law of thermodynamics

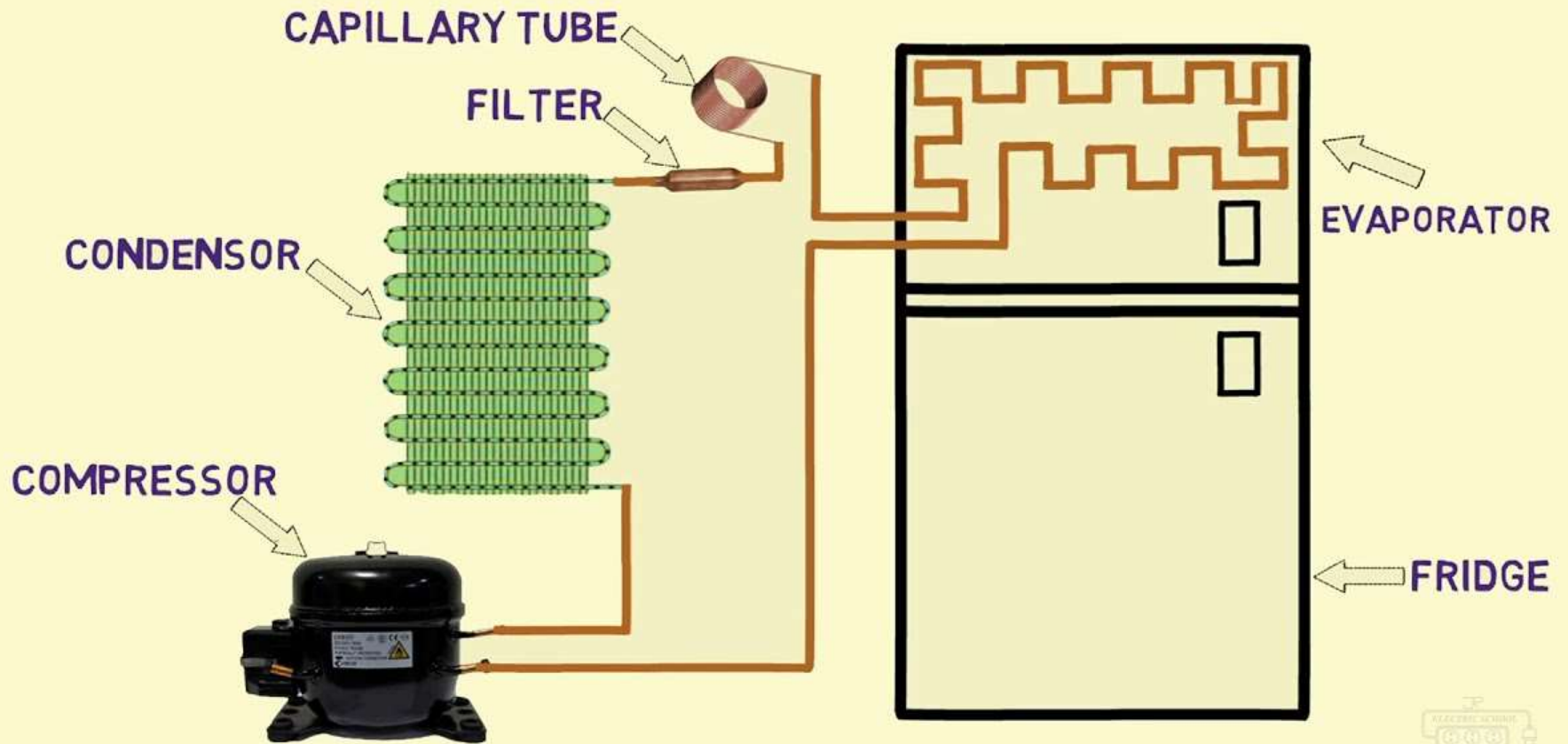


Domestic Refrigerator



Commercial Refrigerator

How Fridge Cooling System Work



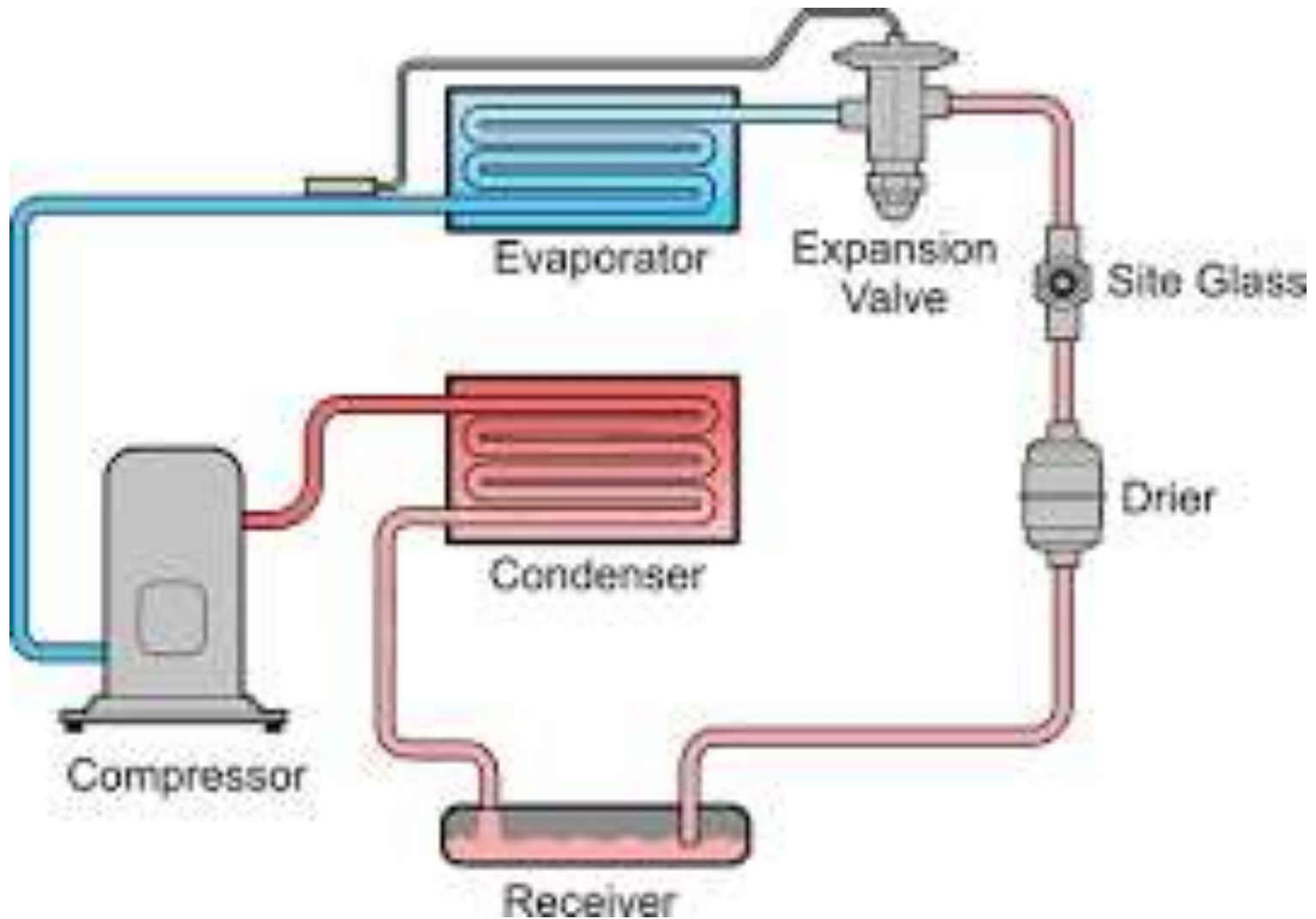
Refrigeration

- The term **refrigeration** means cooling a space, substance or system to lower and/or maintain its temperature below the ambient one (while the removed heat is rejected at a higher temperature).
- In other words, refrigeration is artificial (human-made) cooling.
- Energy in the form of heat is removed from a low-temperature reservoir and transferred to a high-temperature reservoir.

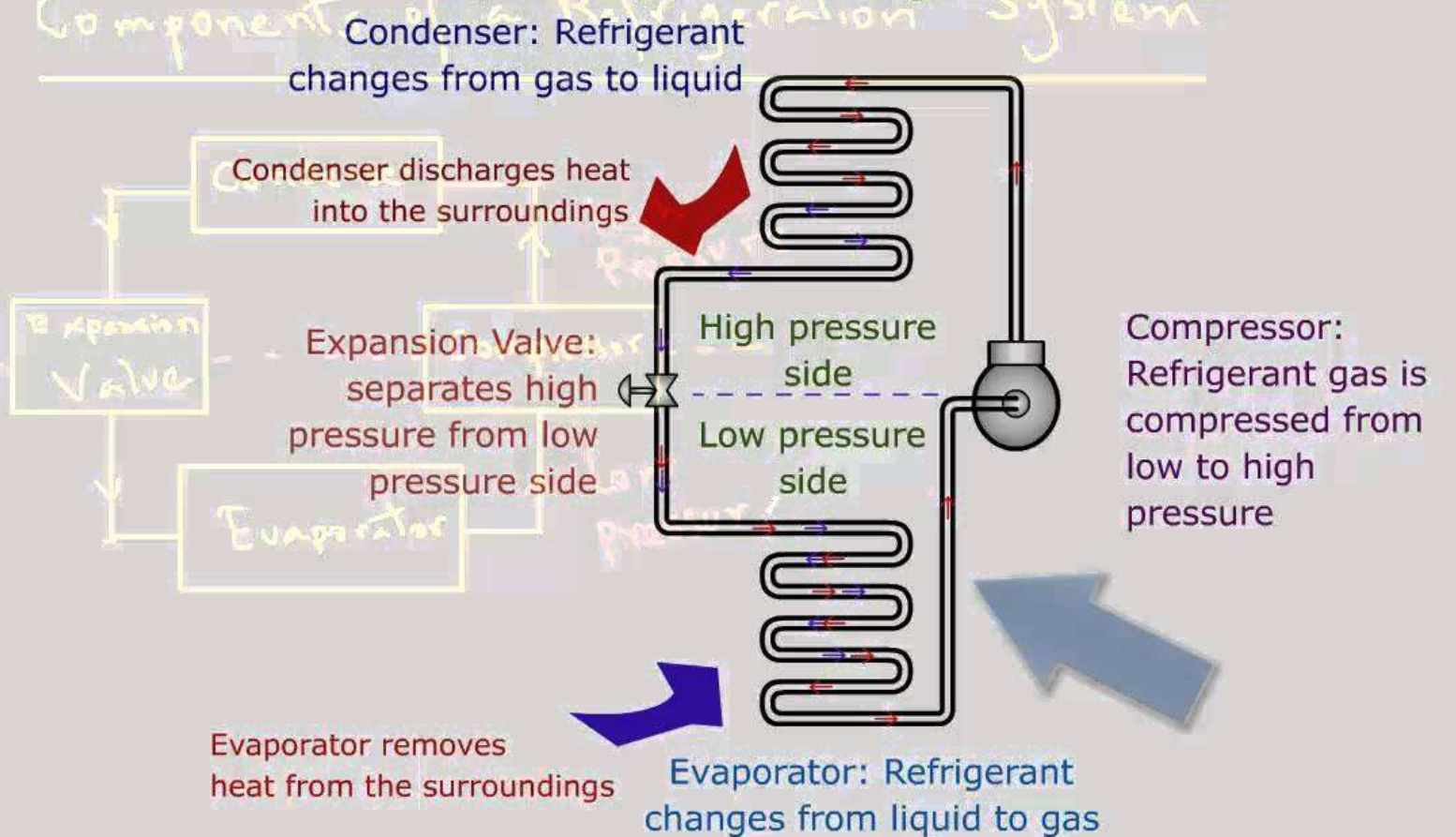
Refrigerant

- A **refrigerant** is a working fluid used in the refrigeration cycle of air conditioning systems and heat pumps where in most cases they undergo a repeated phase transition from a liquid to a gas and back again.
- Classification of refrigerants:
 - Primary
 - Secondary

Schematic diagram



Vapor Compression Refrigeration System

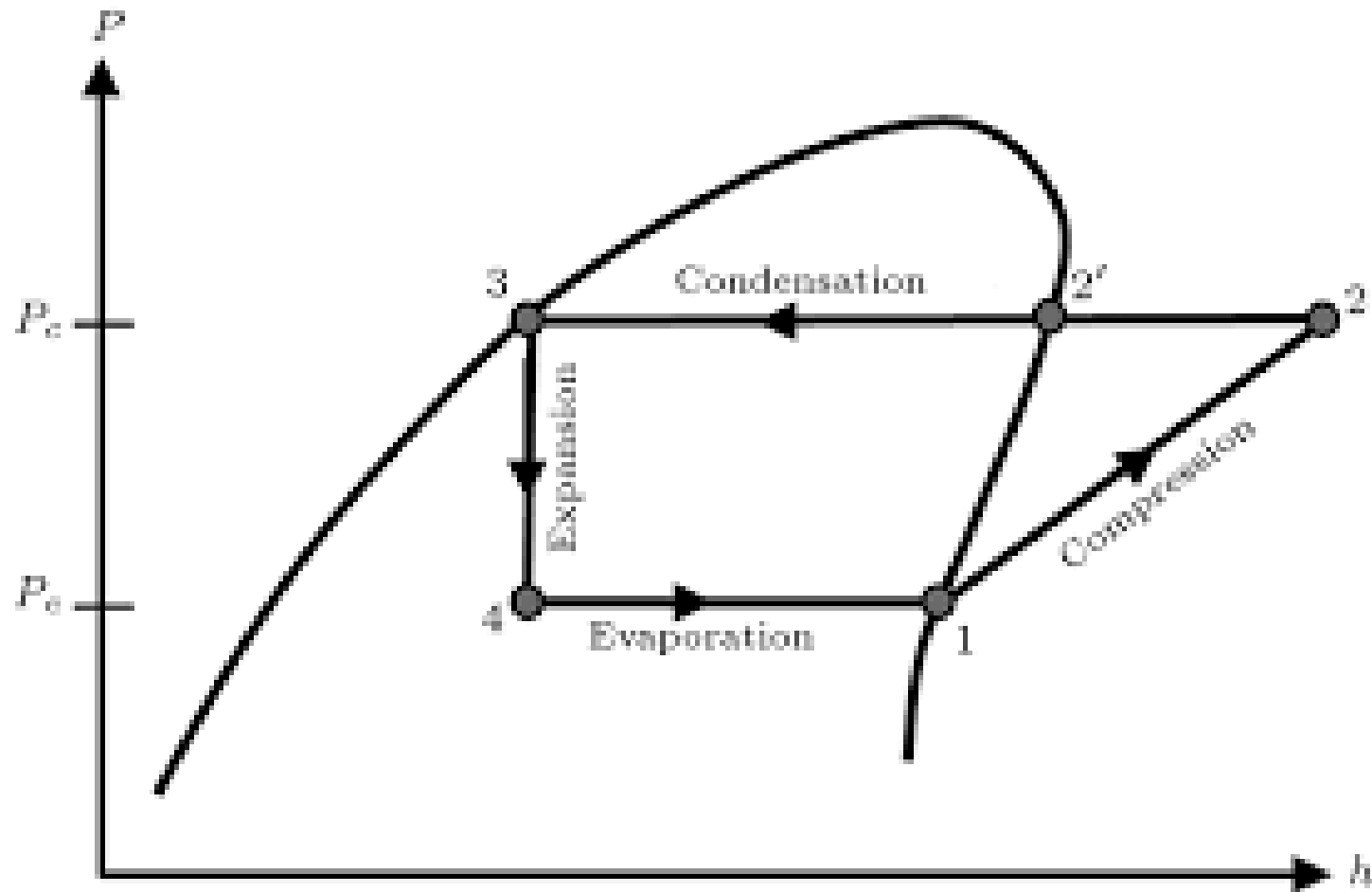


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Working animation

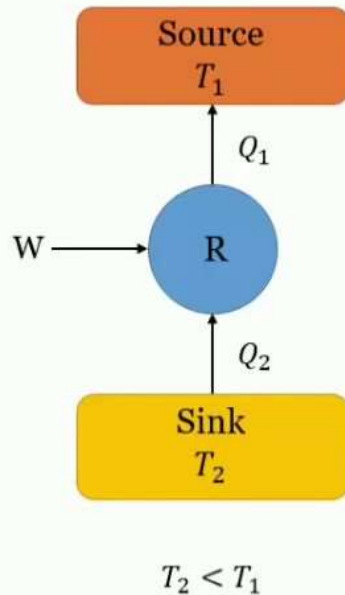
- <https://www.youtube.com/watch?v=h5wQoA15OnQ>

Refrigeration cycle



COP

Refrigerator



Where

Q_2 = Quantity of heat removed from a low temperature region

Q_1 = Quantity of heat supplied to a high temperature region

$$Q_1 = Q_2 + W$$

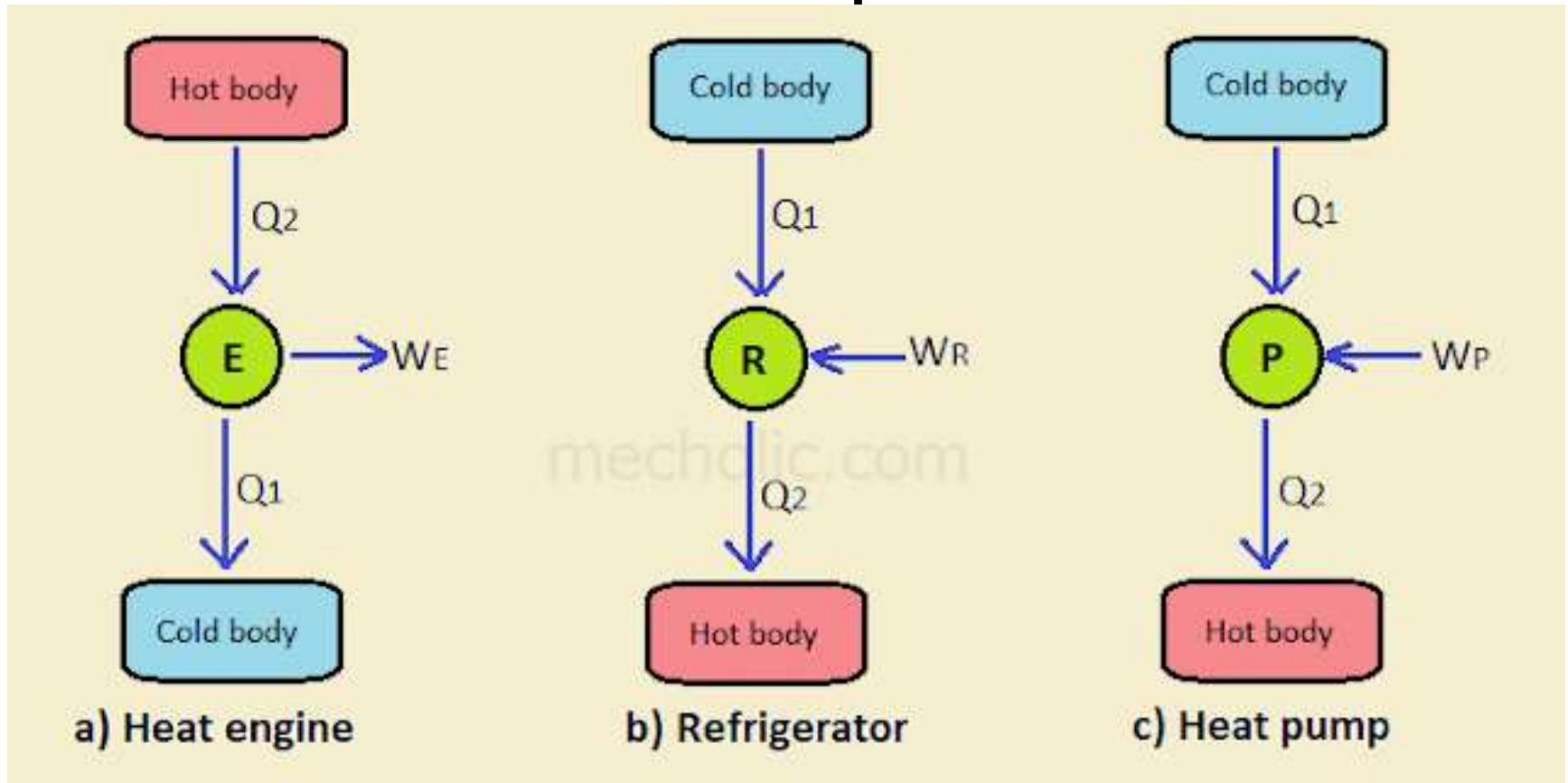
$$W = Q_1 - Q_2$$

$$(COP)_R = \frac{\text{Net Refrigeration Effect}}{\text{Work Supplied}}$$

$$(COP)_R = \frac{Q_2}{W} = \frac{Q_2}{Q_1 - Q_2} = \frac{T_2}{T_1 - T_2}$$

Difference between COP & Eff

Heat Engine, Refrigerator & Heat Pump



$$\eta_{\text{Ref}} = \frac{Q_c}{W} = \frac{Q_c}{Q_H - Q_c} = \text{COP}_{\text{Ref}}$$

$$\eta_{\text{HP}} = \frac{Q_H}{W} = \frac{Q_H}{Q_H - Q_c} = \text{COP}_{\text{HP}}$$

$$\text{COP}_{\text{HP}} \text{ or energy performance ratio} = \frac{Q_2}{W_p}$$

$$\text{COP}_{\text{HP}} = \frac{Q_2}{Q_2 - Q_1}$$

$$\text{COP}_{\text{HP}} = \frac{Q_1 + W_R}{Q_2 - Q_1} = \frac{Q_1}{Q_2 - Q_1} + \frac{W_R}{Q_2 - Q_1}$$

$$\text{COP}_{\text{HP}} = \frac{Q_1}{Q_2 - Q_1} + \frac{Q_2 - Q_1}{Q_2 - Q_1}$$

$$= \frac{Q_1}{Q_2 - Q_1} + 1$$

$$\text{COP}_{\text{HP}} = \text{COP}_R + 1$$

Unit of Refrigeration

- The unit of refrigeration is expressed in terms of **ton of refrigeration (TR)**.
- One ton of refrigeration is defined as the amount of refrigeration effect (heat transfer rate) produced during uniform melting of one ton (1000kg) of ice at 0°C to the water at the 0°C in 24 hours.

Application of Refrigeration

- Domestic
- Commercial
- Industrial
- Marine and transportation
- Comfort air conditioning
- Pharmaceutical industries

Type of Refrigeration

- VCR: Vapor Compression Refrigeration Cycle (mechanical energy)
- VAR: Vapor Absorption Refrigeration Cycle (use thermal energy)

Properties of Refrigerant

- Low Freezing Point. Refrigerants should have low freezing point than the normal operating conditions.
- Low Condensing Pressure. ...
- High Evaporator Pressure. ...
- High Critical Pressure. ...
- High Vapor Density. ...
- High Dielectric strength. ...
- High Latent Heat of Vaporization. ...
- High Heat Transfer Coefficient.

No.	Properties	R134a
1	Boiling Point	-14.9°F or -26.1°C
2	Auto-Ignition Temperature	1418°F or 770°C
3	Ozone Depletion Level	0
4	Solubility In Water	0.11% by weight at 77°F or 25°C
5	Critical Temperature	252°F or 122°C
6	Cylinder Color Code	Light Blue
7	Global Warming Potential (GWP)	1200

Refrigerant properties

Substance	R-Num	BP(°C)	ODP	GWP
Carbon dioxide	R744	−78.5	0	1
Ammonia	R717	−33.3	0	0
1,1,1,2-Tetrafluoro ethane	R134a	−26.3	0	1300
Isobutane	R600a	−11.6	0	20
Dichloro di-fluoro methane	R12	−29.8	0.84	10,600
Difluoromonochloromethane	R22	−40.8	0.055	1700

Vapour Compression Refrigeration System - Construction

- ❑ **Condenser** : It is a coil of tubes made of copper.
- ❑ **Receiver tank**: It is the reservoir of liquid refrigerant.
- ❑ **Expansion Valve**: This is a throttle valve. High pressure refrigerant is made to flow at a controlled rate through this valve.
- ❑ **Evaporator** : It is the actual cooler and kept in the space to be cooled. The evaporator is a coil of tubes made of copper

Introduction

- **Refrigeration**- process of removing heat from a substance under controlled conditions
- -continued extraction of heat from a body whose temperature is already below to its surroundings
- Substance used to remove heat is called as **refrigerant**

AC

- **Air conditioning**- branch of engineering which deals with study of conditioning of air
- Conditioning of air means supplying and maintaining internal atmospheric conditions for human comfort, industrial application, food processing, storage of food and other material

COP

- $\text{COP} = Q/W$
- $\text{Relative COP} = \text{act. COP} / \text{theoretical COP}$
- $Q = \text{amt. of heat extracted in refrigerator}$
- $W = \text{work done on refrigerant}$

IN USA

- $\text{EER} = \frac{\text{heat removed in BTU}}{\text{energy consumed in watts-hr}}$
- Amt. of heat removed from space in BTU in a HVAC system per Watts-hr of electricity consumed
- $\text{EER} = 3.412 \times \text{COP}$

- SEER= avg. performance of HVAC system
- SEER= seasonal energy efficiency ratio
- IPLV (integrated part load value)= RAC institute has fixed certain performance standards of chillers used in system. Rating of chillers is represented by term IPLV IN Kw/ton.
- Measures the efficiency under variable load and temperature for non-residential central air-conditioning plant

Unit

- tonne of refrigeration (TR)= amt. of refrigeration effect produced by the uniform melting of one tonne of ice from and at zero degree centigrade in 24 hrs
- $1 \text{ TR} = 1000 \text{ kg} \times 335 \text{ kJ in 24 hrs}$
 $= (1000 \times 335) / (24 \times 60 \times 60)$
 $= 3.87 \text{ kJ}$
 $1 \text{ tonne of refrigeration} = 14000 \text{ kJ/h}$
 $1 \text{ ton} = 0.9 \text{ tonne}).$

Domestic refrigerator

- Primary function-provide food storage space maintained at low temperature for the preservation of food
- Secondary function-formation of ice cubes
- Storage temperature- $273\text{ K} - 277\text{ K}$
- Freezers are provided at top portion (in some cases it may be in bottom position)
- May be single-door, double-door top freezer, double door bottom freezer and side by side door freezer

- Can work on VCRs or VARs
- VCRs is more compact and more efficient
- Power compression can be varied according to size (75 W, 92 W, 125 W, 180 W, 370 W)
- Condenser is tube and wire type
- In between condenser and capillary tube ,receiver and drier is used
- Evaporators:-
 - 1] Bare tube evaporators
 - 2] Plate surface evaporators
 - 3] Finned evaporators