## **EXTRACTION OF ELEMENTAL SULPHUR**

- Sulphur from gasoline is further removed in the form of H<sub>2</sub>S
- CLAUS PROCESS: Process of getting elemental sulphur
- What we do in Claus process?
- First Step: Partial oxidation of H₂S
- H2S + O2 -> SO2+H2O + unburned H2S Exothermic Rex Unit Process : Burner
- H2S + SO2 <-> S2+2H2O Exothermic Rex
- First unit operation: Burner
- Since non condensable are released.
- H2S is absorbed in amine while other non condensable gases are vent out
- Tail gases contain gases like SO2, CS2 and COS which are treated through scot process or are incinerated.
- In the scot process, the SO2, CS2 and COS are reduced to H2S by reducing gases and passed back to the Claus process for further conversion.
- This ensures more than 99 percent conversion
- In order to ensure optimal conversion in the Claus plant, the H2S:SO2 ratio should be 2:1. The ratio is measured in the tail gas from the final condenser, and the air flow to the furnace is adjusted to achieve this ratio

## **Ammonia Production**

- Thermodynamic condition: High Pressure, Low temperature
- Requires Catalyst
- Source of Nitrogen : Air
- Source of Hydrogen: Natural Gas

- Steam reforming is used for getting H2
- CH4+H2O<-> CO+3H2 (endothermic)
- CO+H2O->CO2+H2 (exothermic)
- Major Steps involved:-
- Desulphurization of Natural Gas To protect downstream catalyst
- Next Step is reforming
- Primary reforming
- Clean Gas + h2o ->CO+H2+CH4+H2O
- This mixture goes to secondary reformer
- Primary reforming is done in a furnace
- Since temp is lower at the outlet, we partially combust h2 so that temperature increases again
- Therefore secondary reformer has two chambers : one for combustion and one for Reaction