

## *Covalent Bonding Work Answers Prentice Hall*

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that, Prentice hall chemistry chapter 8 covalent bonding answers gets packed inside the box it can be found in and obtains chucked right into the deep cob-webbed edges never to be viewed again. up until, human brain freeze strikes and also you cannot fairly make out what that little button on your glitzy remote does. we

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### Chemistry (12th Edition) Chapter 8 - Covalent Bonding - 8 ...

A diatomic molecule with a triple covalent bond is a. N<sub>2</sub>. c. H<sub>2</sub>. b. Br<sub>2</sub>. d. O<sub>2</sub>. \_\_\_\_ 15. A molecule of nitrous oxide, N<sub>2</sub>O, contains all of the following except a. a coordinate covalent bond. c. a double bond. b. a triple bond. d. nonbonding pairs of electrons. \_\_\_\_ 16. If a bonding pair of electrons is unequally shared between two atoms, the bond is a. ionic. c.

### Chapter Test A - Campbell County Schools

242 Chapter 8 • Covalent Bonding Single Covalent Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared electrons.

### Chapter 8: Covalent Bonding

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Atoms and Bonding Multiple Choice Write the letter of the correct answer on the line at the left. \_\_\_\_ 1. A(n) \_\_\_\_ is an atom or group of atoms that has an electric charge. a. compound b. ion c. crystal d. neutron \_\_\_\_ 2. A covalent bond is called \_\_\_\_ if the atoms equally share one or more pairs of

electrons. a. double b.

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