

Directed A Grouping The Elements Answer Key

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Directed A Grouping The Elements

• Elements in a group often have similar properties because their atoms have the same number of electrons in their outer energy level. • Hydrogen is set off by itself in the periodic table because its properties do not match the properties of any one group. • Metals are in Groups 1–16. Metalloids are in Groups 13–16.

Section 2 Grouping the Elements - Midway Middle School Science

Directed A Grouping The Elements Gestalt principles, or gestalt laws, are rules of the organization of perceptual scenes. When we look at the world, we usually perceive complex scenes composed of many groups of objects on some background, with the objects themselves consisting of parts, which

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DIRECTED A GROUPING THE ELEMENTS ANSWER KEY

This section is about the ways in which elements are grouped together. Learn with flashcards, games, and more — for free.

Chapter 12, section 2 Grouping the Elements Flashcards ...

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Skills Worksheet Directed Reading A Section: Grouping the Elements Write the letter of the correct answer in the space provided. ____ 1. Which of the following gives elements in a family or group similar properties? a. the same atomic mass b. the same number of protons c. the same number of electrons in their outer energy level

Skills Worksheet Directed Reading A - RCSMSMrs. Botten

Directed Reading A SECTION: ARRANGING THE ELEMENTS 1. Answers will vary. Sample answer: Scientists might have been frustrated because the elements weren't organized and therefore their properties couldn't be predicted. 2. D 3. periodic 4. periodic table of the elements 5. Mendeleev was able to predict the properties of unknown elements by

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Skills Worksheet Directed Reading B Section: Grouping the Elements ____ 1. What gives elements in a family or group similar properties? a. the same atomic mass b. the same number of protons in their nuclei c. the same number of electrons in their outer energy level d. the same number of neutrons ____ 2. What makes elements reactive at the ...

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Directed Reading A continued GROUP 17: HALOGENS 27. The atoms of need to gain only one electron to have a complete outer level. 28. What important use do the halogens iodine and chlorine have in common? 29. Halogens combine with most metals to form , such as . 30. How does chlorinating water help protect people? GROUP 18: NOBLE GASES ____ 31.

Skills Worksheet Directed Reading A

Directed Reading A Section: Grouping the Elements 1. What gives elements in a family or group similar properties? a. the same atomic mass b. the same number of protons in their nuclei c. e

same number of electrons in their outer energy level d. the same number of total electrons GROUP 1: ALKALI METALS 2.

janellmcclure.typepad.com

Directed Reading B Section: Grouping the Elements Circle the letter of the best answer for each question. 1. What do atoms of elements in a group have that makes their properties similar? a. the same atomic mass b. the same number of protons c. the same number of outer level electrons d. the same total number of electrons 2.

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Section 2 Grouping the Elements Bellringer How do you know that a bird is a bird, that a kangaroo is a kangaroo, and that a shark is a shark? What characteristics of each animal help you to tell the animals apart? How can such an analysis of characteristics be applied to elements?

Chapter 12 Section 2 Grouping the Elements Chapter 12

Directed Reading A continued Atomic Number ____ 22. How many of the recently discovered elements follow the periodic law? a. half of them b. none of them c. every fifth element d. all of them ____ 23. What does the periodic law say about the properties of elements? a. They change periodically with the atomic numbers of the elements. b.

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The columns of the table represent groups, or families, of elements. The elements in a group often look and behave similarly, because they have the same number of electrons in their outermost ...

How Are Elements Grouped in the Periodic Table?

Rows of elements are called , and columns of elements are called or. 14. Who will approve the names of the newest elements? a. the scientist who discovered each element b. an international committee of scientists c. the chemists from a research institute 15. Silicon is a . (metal, nonmetal, or metalloid) Finding Your Way Around the Periodic ...

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