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4 \*P46937A0424\* DO NOT WRITE IN THIS AREA DO NOT WRITE IN THIS AREA DO NOT WRITE IN THIS AREA 6 The table shows the mean bond enthalpies for some covalent bonds. Covalent bond C CCCBrBr C Br C H Mean bond enthalpy / kJ mol-1 347 612 193 290 413 What is the approximate enthalpy change, in kJ mol-1, for the reaction shown? H C C H

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Section Review 2-1 1. protons; neutrons 2. electrons 3. neutrons 4. electrons 5. ionic 6. The two main types of chemi-cal bonding are ionic and covalent bonding. Ionic bonds are formed when a transfer of electrons takes

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