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•covalent bond: a bond formed by the sharing of electrons between atoms •molecule: a neutral group of atoms joined by covalent bonds •diatomic molecule: a molecule consisting of two atoms  
Glossary Terms

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single covalent bond. two atoms held together by sharing one pair of electrons. structural formula. represents the covalent bonds as dashes and shows the arrangement of covalently bonded atoms. unshared pair. a pair of valence electrons that is not shared between atoms, also known as a lone pair or non-bonding pair.

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Chapter 8 Covalent Bonding and Molecular Structure 8-3 There are two types of repulsive forces between the two atoms. First, the nuclei repel because they are both positively charged. Second, the electrons repel because they are both negatively charged. The attractive forces between the two atoms result from the

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