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Concept Review: Calculating Quantities in Reactions 1. mole 2. balance 3. relative 4. coefficients 5. molar mass 6. liquids 7. density 8. Avogadro's number

**3 2 3 2 Answer Key 3 2 3 2 - morganparkcps.org**

SECTION: REACTION TYPES 1. oxygen 2. A single-replacement reaction will have one compound and one element as reactants and one compound and one element as products, while a double-replacement reaction has two compounds as reactants and two compounds as products. 3. Electrons are transferred between elements during the reaction. The sub-

**Concept Review - Manchester High School**

Holt Chemistry 48 Stoichiometry Section: Calculating Quantities in Reactions In the space provided, write the letter of the term or phrase that best answers the question. \_\_\_\_ 1. The chemical equation  $P_4O_{10} + 6H_2O \rightarrow 4H_3PO_4$  can be correctly interpreted as a. 1 molecule of  $P_4O_{10}$  and 6 molecules of  $H_2O$  form 4 molecules of  $H_3PO_4$ . b. 1 mol  $P_4O_{10}$  ...

## Concept Review Section Stoichiometry Cars Answers

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