

## *Cstephenmurray Answers Key Acids And Bases*

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J. \_\_\_\_ When acids in your stomach break down your food into nutrients your body can absorb. K. \_\_\_\_ When enzymes in your saliva pre-digest and soften your food in your mouth before you swallow. L. \_\_\_\_ The complete act of digestion (all of the above). 2. Salt is put into water. The water is stirred until the salt disappears. A.

### **Day 14—Classification of Matter Objective 4**

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Acids and BasesAcids and Bases Solutions Manual Chemistry: Matter and Change • Chapter 18 357 Section 18.1 Introduction to Acids and Bases pages 634–643 Practice Problems pages 635–640 1. Write balanced equations for reactions between the following. a. aluminum and sulfuric acid  $2\text{Al(s)} + 3\text{H}_2\text{SO}_4\text{(aq)} \rightarrow \text{Al}_2\text{(SO}_4)_3\text{(s)} + 3\text{H}_2\text{(g)}$

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### **Cstephenmurray Answer Key Thermodynamics**

C. Acid D. Neutral \_\_\_\_ pH 7, when there is equal amount of acid and base (distilled water is also this). \_\_\_\_ A chemical that adds  $\text{H}^+$  ions when mixed in water. \_\_\_\_ Scale used to measure acids and bases. \_\_\_\_ A chemical that adds  $\text{OH}^-$  ions to a water solution. A 30 milliliter object rock is 15 grams. Find its density.

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Brønsted-Lowry Acid/Base Pairs Label the following as acid, base, conjugate acid (CA) and conjugate base (CB). 1)  $\text{H}_2\text{SO}_4 + \text{OH}^- \rightarrow \text{HSO}_4^- + \text{H}_2\text{O}$  acid Base Titration Practice 3)  $\text{HSO}_4^- + \text{H}_2\text{O} \rightarrow \text{SO}_4^{2-} + \text{H}^+$  30 Ease CB 4)  $\text{OH}^- + \text{H}^+ \rightarrow \text{H}_2\text{O}$  Base 1) A 25.0 ml- sample of HCl

was titrated to the endpoint with a 15.0 mL of 2.0 M NaOH.

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