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Concept Review: Energy. 1. energy 2. physical 3. chemical 4. endothermic 5. exothermic 6. kinetic 7. transferred 8. In any chemical or physical change, the total quantity of energy remains constant. Energy cannot be created or destroyed. 9. Heat is the energy transferred between objects that are at different temperatures.

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Concept Review j j. 8. In the word equation hydrogen + oxygen + heat water, hydrogen is a {reactant, product}, and water is a [reactant product]. This is an example of a [physical, chemical] change. - 9. A [physical, chemical] reaction rearranges the atoms that make up the reactants.

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SECTION: RATES OF CHANGE 1. If a change is made to a system in chemical equilibrium, the equilibrium shifts to oppose the change until a new equilibrium is reached. 2. Answers may vary. The table salt can be ground up into small particles. The temperature of the water can be increased. 3. a. toward the left b. toward the right

Concept Review - Manchester Local School District

During a temperature inversion, a. sulfur oxides and nitrogen oxides combine with water in the

atmosphere. b. an influx of acidic water causes a rapid change in the pH of water. c. levels of ground-level ozone decrease.

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Holt Science Spectrum 2 Heat and Temperature Skills Worksheet Concept Review Section: Energy Transfer 1. Explain why a ceramic bowl will keep oatmeal hot longer than a stainless steel bowl. ____ ____ 2.

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