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chemical reaction. It uses symbols to show the relationship between the reactants and the products. 9. product 10. reactant 11. produces or yields, reversible reaction, , symbols for solid, liquid, gas,, substance dissolved in water Concept Review: Balancing Chemical Equations 1. The law of conservation of mass states that in ordinary chemical ...

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Holt Science Spectrum 2 Chemical Reactions Section: Reaction Types 1. Name the compound that is a reactant in all combustion reactions. 2. Explain how you can determine if a chemical reaction represents a single-replacement reaction or a double-replacement reaction. 3. Describe what happens during a reduction/oxidation reaction. 4.

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Section: Properties of Matter 1. Classify each of the following as a physical or chemical property of sulfur. ... Concept Review Skills Worksheet. Bellringer Transparencies SECTION: WHAT IS MATTER? 1. a. ... (chemical property). 4. 136.3 g silver 5. 2.3 g/cm3 6. Characteristic properties are proper-

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(Section 8.1) 2. Be able to show which reactant is losingelectrons and which is gaining electrons, and how ... Chapter Review 135. LEO the lion goes GER or OIL RIG. Ifcopper is losing electrons, some other substance in this reaction must gain electrons. ... 8 Classifying Chemical Reactions ...

#### **CHAPTER 8: CLASSIFYING CHEMICAL REACTIONS**

Section: Limiting Reactants and Percentage Yield ... reactant is used up first and thus controls the quantity of that can be formed in a chemical reaction. 3. The reactant that runs out first is the reactant. 4. ... Concept Review: Stoichiometry and Cars 1. c 2. b 3. d 4. b 5. d 6. If there is too much oxygen and not

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9.2 Classifying Chemical Reactions MAIN Idea There are four types of chemical reactions: synthesis, ... 282 Chapter 9 • Chemical Reactions Section 99.1.1 ... Objectives Recognize evidence of chemical change. Represent chemical reactions with equations. Balance chemical equations. Review Vocabulary chemical change: a process involving one or ...

# **Chapter 9: Chemical Reactions**

Section 8-3: Classifying Chemical Reactions Reaction Type General Format Characteristics of Reactants Characteristics of Products Example (s) Decomposition AB A + B only one reactant two or more simpler products CuCO 3 (s) CuO + CO 2 (g) 2 H 2 O (l) 2 H

# **Section 8-3: Classifying Chemical Reactions**

Concept Review: Energy 1. energy 2. physical 3. chemical 4. endothermic 5. exothermic 6. kinetic 7. transferred 8. In any chemical or physical change, the total quantity of energy remains constant. Energy cannot be created or destroyed. 9. Heat is the enegy transferred between objects that are at different tempera-tures. 10. Temperature is a ...

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Holt Chemistry 13 Chemical Equations and Reactions Name Class Date Quiz continued \_\_\_\_\_\_ 6. Which statement is true for the balanced formula equation? a. The coefficients are the same for each substance. b. There are more reactant elements than product elements. c. The number of atoms of an element that makes up the reactants is

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Chemistry classifying chemical reactions. this is just stuff that I found in the concept review. STUDY. PLAY. Chemical reaction. The process by which one or more substances change to produce one or more different substances. Give three examples of a chemical change that has occurred in your home.

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