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#### **Chapter 16 Properties Of Solutions**

Chapter 16: Properties of Solutions. STUDY. PLAY. This type of solution contains the maximum amount of solute for a given amount of solvent at a constant temperature. This is the amount of substance that dissolves in a given quantity of a solvent at a given temperature to produce a saturated solution.

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Chapter 16: Colligative Properties of Solutions 45 16-4. The mole fraction of (NH 4) 2SO 4(aq) is given by x (NH 4) 2SO 4 = n (NH 4) 2SO 4 n (NH 4) 2SO 4 + n H 2O Because (NH 4) 2SO 4(aq) is a strong electrolyte, it dissociates completely into NH + 4 (aq) and SO2 – 4 (aq) ions. Assume a one kilogram solution. The number of moles of ions in one ...

### **CHAPTER 16. Colligative Properties of Solutions**

Chapter 16 Solutions 167. SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471–477) This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves. Solution Formation (pages 471–472) Look at Figure 16.1 on page 471 to help you answer Questions 1 and 2.

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Chapter 16 - Solutions - 16.1 Properties of Solutions - 16.1 Lesson Check - Page 524: 5 Answer The question is asking you to identify / recall factors that contribute to solubility, which is the mass of solute that dissolve in a given mass of solvent.

# Chapter 16 - Solutions - 16.1 Properties of Solutions - 16 ...

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