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b. a solution containing the maximum amount of. solute that can be dissolved at a given temperature. c. the number of moles of solute dissolved in. 1 L of solution. d. describes liquids that are insoluble in each other. e. ... Chapter 16 Test REVIEW ...

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Chapter 16

Modern Chemistry 129 Chapter Test Chapter: Acids and Bases PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. Which of the following is an oxyacid? a. HCl b. H₂O c. H₂S d. H₂SO₄ ____ 2. Which acid is produced in the stomach? a. HNO₃ b. CH₃COOH c ...

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Chapter 16 Solutions 169 15. Describe the two diagrams of a bottled carbonated beverage below as greater pressure or lower pressure, and then as greater solubility or lower solubility.

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Chapter 16 Sample Test Instructions: You may use a "clean" periodic table. Sample test questions for the optional sections on molality, normality, and colligative properties follow this test. 1) A solution that can dissolve more of a substance than it currently contains is said to be ____.

Chapter 16

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