

Concept Review Compound Names And Formulas Answer

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Name Class Date Concept Review continued 5. What is the empirical formula for a compound that is 29.44% calcium, 23.55% sulfur, and 47.01% oxygen? This compound is a common ingredient in plaster. 6. The empirical formula of a compound is CH. The experimental molar mass is 78.12 g/mol. Determine the molecular formula of the compound. 7.

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Concept Review Section: Compound Names and Formulas 1. Explain the difference between iron(II) nitrate and iron(III) nitrate. What is the significance of the Roman numerals? _____ 2. Name the following ionic compounds, keeping in mind that a transition metal cation must include its charge. ...

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Holt Chemistry 7 Covalent Compounds Name Class Date Concept Review continued Use the system of prefixes and the rule for suffixes to name the following compounds. 15. PbCl₂ 16. KCl 17. LiO₂ 18. As₂O₃ 19. PBr₃ 20. SF₄ 21. N₂O₅ Write the formulas for the following compounds. 22. nitrogen monoxide 23. carbon dioxide 24. carbon ...

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The name begins with the name of the first element—carbon. The second element, chlorine, becomes chloride, and we attach the correct numerical prefix (“tetra-”) to indicate that the molecule contains four chlorine atoms. Putting these pieces together gives the name carbon tetrachloride for this compound.

5.2: Covalent Compounds: Formulas and Names - Chemistry ...

Nomenclature CHEM 10 Review Worksheet ... Which of the following non-metals would you write first in a covalent compound's name or formula? oxygen or nitrogen iodine or chlorine silicon or sulfur 2. Write the names or formulas as required for each of the following covalent compounds. ...

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