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Name Class Date Concept Review continued 5. What is the empirical formula for a compound that is 29.44% calcium, 23.55% sulfur, and 47.01% oxygen? This compound is a common ingredient in plaster. 6. The empirical formula of a compound is CH. The experimental molar mass is 78.12 g/mol. Determine the molecular formula of the compound. 7.

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Concept Review Compound Names And Formulas Answer

Concept Review Section: Compound Names and Formulas 1. Explain the difference between iron(II) nitrate and iron(III) nitrate. What is the significance of the Roman numerals? _____ 2. Name the following ionic compounds, keeping in mind that a transition metal cation must include its charge. ...

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Holt Chemistry 7 Covalent Compounds Name Class Date Concept Review continued Use the system of prefixes and the rule for suffixes to name the following compounds. 15. PbCl 2 16. KCl 17. LiO 2 18. As 2O 3 19. PBr 3 20. SF 4 21. N 2O 5 Write the formulas for the following compounds. 22. nitrogen monoxide 23. carbon dioxide 24. carbon ...

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The name begins with the name of the first element—carbon. The second element, chlorine, becomes chloride, and we attach the correct numerical prefix ("tetra-") to indicate that the molecule contains four chlorine atoms. Putting these pieces together gives the name carbon tetrachloride for this compound.

5.2: Covalent Compounds: Formulas and Names - Chemistry ...

Nomenclature CHEM 10 Review Worksheet ... Which of the following non-metals would you write first in a covalent compound's name or formula? oxygen or nitrogen iodine or chlorine silicon or sulfur 2. Write the names or formulas as required for each of the following covalent compounds. ...

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Skills Worksheet Concept Names And Formulas Of Ionic ...

The name begins with the name of the first element—carbon. The second element, chlorine, becomes chloride, and we attach the correct numerical prefix ("tetra-") to indicate that the molecule contains four chlorine atoms. Putting these pieces together gives the name carbon tetrachloride for this compound.

4.3: Covalent Compounds: Formulas and Names - Chemistry ...

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