

Course: EE3013/ Semiconductor Devices and Processing
School: School of Electrical and Electronic Engineering
Part I - Highlights

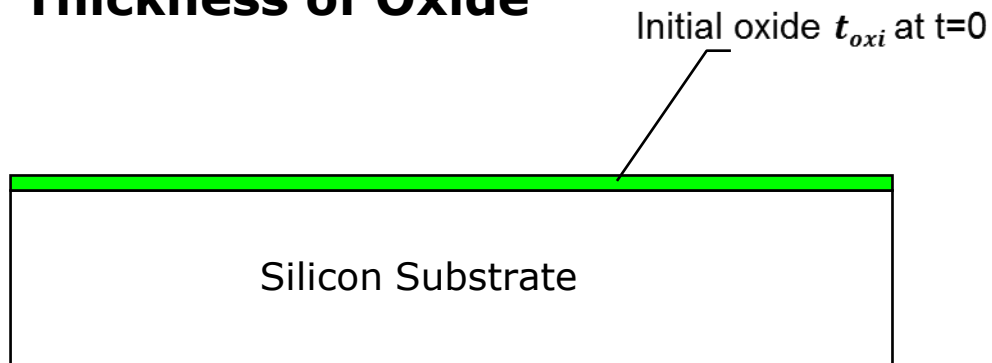
Week 7 - Thermal Oxidation

Thermal Oxidation

- **Applications:**

- Mask for implantation and diffusion,
- Isolation,
- Gate oxide of MOS structures for insulation, and
- Surface passivation.

- **Thickness of Oxide**



- There are two oxidation methods:

- Dry oxidation, and
- Wet oxidation.

$$t = \frac{t_{ox}^2}{B} + \frac{t_{ox}}{B/A} - \tau$$

or $t_{ox}^2 + At_{ox} = B(t + \tau)$ (Equation 7.34)

where $A = \frac{2D}{k_s}$, $B = \frac{2DN_0}{M}$ and $\tau = \frac{t_{oxi}^2}{B} + \frac{t_{oxi}}{B/A}$

t is time, t_{oxi} is the initial oxide thickness on the wafer before oxidation and τ is the time required to grow the initial oxide.

Thickness of Oxide

$$t_{ox} = \frac{-A + \sqrt{A^2 + 4B(t + \tau)}}{2} \quad (\text{Equation 7.35})$$

- For **short time** with $t + \tau \ll A^2/4B$:

$$t_{ox} = \frac{B}{A}(t + \tau) \quad (\text{Equation 7.36})$$

- For **long time** with $t + \tau \gg A^2/4B$, $t > \tau$:

$$t_{ox} = \sqrt{Bt} \quad (\text{Equation 7.37})$$

Note:

- B/A is called **linear rate** coefficient. (Growth rate is limited by the reaction at the silicon interface; it depends on Si-Si bond strength and not by diffusivity.)
- B is called **parabolic rate** coefficient. (Oxidation rate is diffusion limited.)

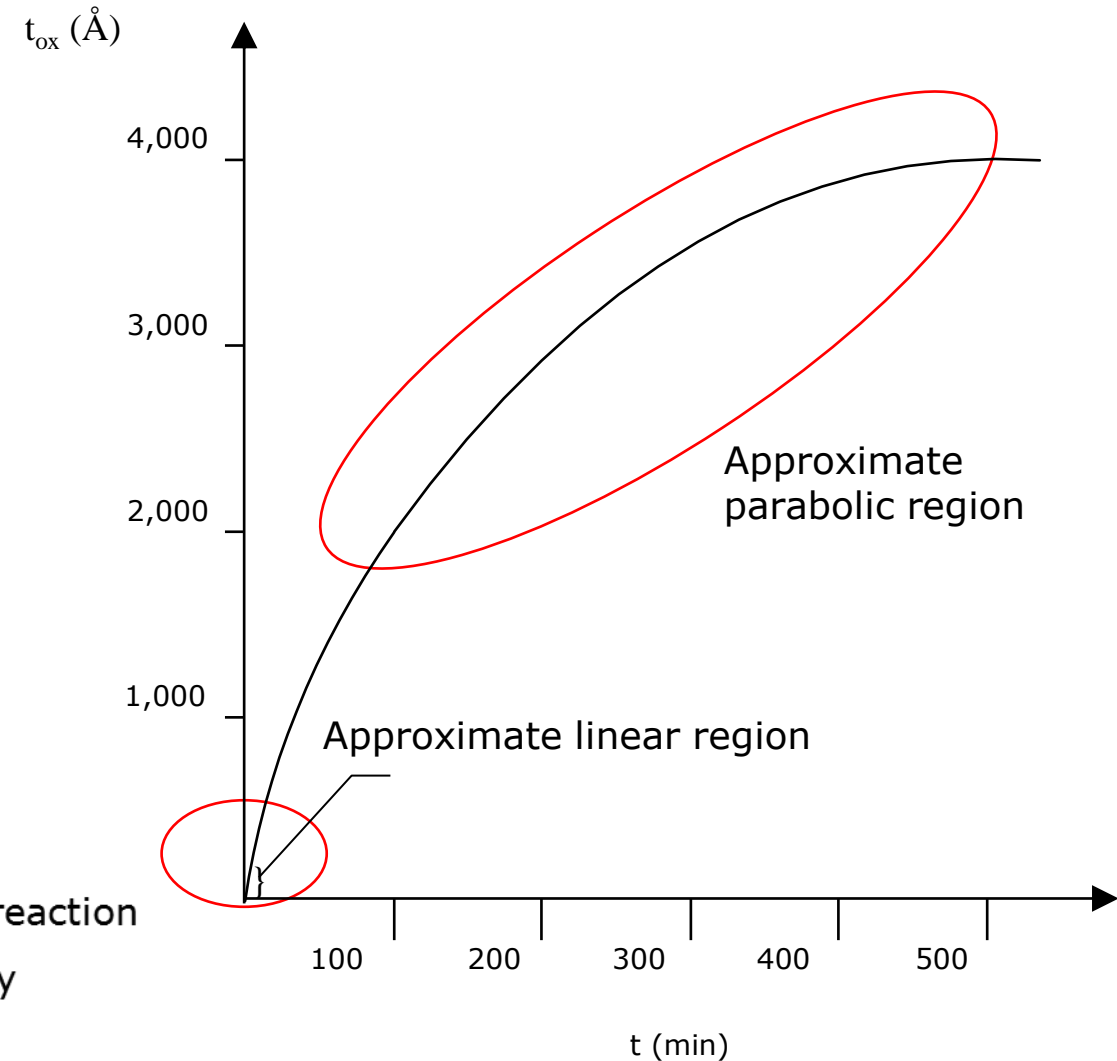


Figure 7.40

Factors affecting Oxidation Rate

- Growth techniques: wet versus dry
- Growth temperature
- Pressure
- Crystal orientation: (100) versus (111)
- Impurity doping

Oxide Thickness Measurement Techniques

- Color Chart
- Ellipsometry
- TEM (Transmission Electron Microscopy)