NANYANG TECHNOLOGICAL UNIVERSITY

SEMESTER 1 EXAMINATION 2018-2019

EE3013 - SEMICONDUCTOR DEVICES AND PROCESSING

November / December 2018

Time Allowed: 2 hours

INSTRUCTIONS

- 1. This paper contains 4 questions and comprises 9 pages.
- 2. Answer ALL questions.
- 3. All questions carry equal marks.
- 4. This is a closed book examination.
- 5. Unless specifically stated, all symbols have their usual meanings.
- 6. A List of Formulae is provided in Appendix A on pages 7 and 8. The Table of Physical Constants is provided in Appendix B on page 9:
- 1. (a) The total junction capacitance of a GaAs p-n junction at T = 300 K is found to be 100 pc at a reversed bias voltage of 1 V. The doping concentration in one region is found to be $8 \times 10^{16} \text{ cm}^{-3}$, and the built-in potential is found to be $V_{bl} = 1.20 \text{ V}$. Given the intrinsic carrier concentration (n_l) and relative permittivity (s_l) of GaAs are $1.8 \times 10^6 \text{ cm}^{-3}$ and 13.1, respectively, determine:
 - (i) The doping concentration in the other region of the p-n junction. $V_{ki} = \frac{kT}{N} \ln \frac{N_A N_D}{N_I^2}$ (3 Marks)
 - (ii) The cross-sectional area of the p-n junction diode.

(3 Marks)

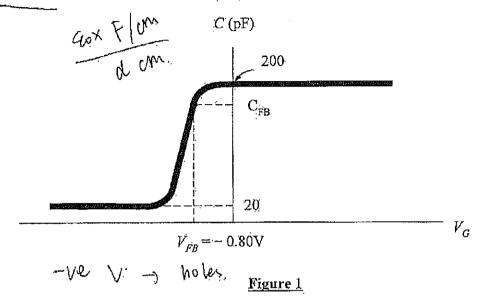
(iii) What will happen to the junction capacitance value if the reverse-biased voltage is increased.

(3 Marks)

Note: Question No. 1 continues on page 2.

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Figure 1 shows the high-frequency C-V characteristic curve of a silicon MOS (b) capacitor. The area of the device is 2×10^{-3} cm². The metal-semiconductor work function difference $\phi_{ms} = -0.50$ V, the relative permittivity of the oxide $\varepsilon_{\rm K}$ = 3.9, and the semiconductor doping concentration is 2×10^{16} cm⁻³.



Is the semiconductor n-or p-type? (i)

(2 Marks)

What is the oxide thickness? (ii)

(3 Marks)

What is the equivalent trapped oxide charge density (Q_o) where (iii) $(Q_o = Q_f + Q_m + Q_{ol})?$

(3 Marks)

Explain why the base-emitter should be forward biased and the base-collector (c) (j)

State two ways to increase the cutoff frequency (f_T) of a BJT. (ii)

(4 Marks)

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2. (a) (i) A silicon dioxide layer was grown at 1050 °C under wet oxidation on (100) silicon for 120 minutes. Assume that $A=0.226~\mu m$ and $B=0.287~\mu m^2/hr$. If the final thickness of the oxide is found to be $0.67~\mu m$, what is the thickness of the initial oxide?

(4 Marks)

(ii) Name two applications of oxide in semiconductor devices. Explain if wet or dry oxidation is preferred in growing this oxide.

(4 Marks)

(b) (i) A p-n junction is formed by diffusing pre-deposited boron atoms into the n-type silicon substrate at 1100 °C for 1 hour. The dose for the pre-deposition is 2×10^{13} atoms/cm². Assume that the diffusion is Gaussian, the doping concentration of the silicon substrate is 1.5×10^{16} cm⁻³, and $D_B = 9.2 \times 10^{-13}$ cm²/s at 1100 °C, what is the junction depth?

(5 Marks)

(ii) Name two advantages of thermal diffusion over implantation.

(4 Marks)

(c) (i) A silicon wafer with a diameter of 20 cm is implanted with phosphorus ions at an implant energy of 150 keV to form a p-n junction. The implantation time is 10 seconds and the average current used is 5 μA. Under these implant conditions, the values of the projected range (R_p) is 0.14 μm and the straggle is found to be 1.6 × 10¹⁴ cm⁻³, what is the junction depth?

(5 Marks)

(ii) Explain how a mass spectrometer in an ion-implantation system is used to select the desired ions for implant.

(3 Marks)

3. (a) Figure 2 shows the schematic of a lithography system used in many research labs to prototype new devices.

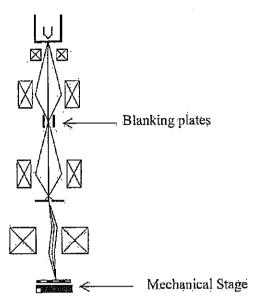


Figure 2

- (i) Name the lithography system. List the advantages and disadvantages of such a system as compared to optical lithography.
 - (5 Marks)
- (ii) State, giving reasons, the type of resist that is usually preferred for the above lithography system.

(3 Marks)

An engineer deploys such a lithography system with step and repeat imaging features to print a submicron pattern. Given that the resists used have a k_j value of 0.7, and that the minimum line-width is 8 nm, while the gap between the mask and the wafer is 10 µm determine the energy required for the system.

(4 Marks)

- (b) Define in a few sentences the differences between the following:
 - (i) Hard and soft bake deployed in lithography processing steps.
 - (ii) Mask and reticle used for lithography.
 - (iii) X-ray and E-beam lithography.

(6 Marks)

Note: Question No. 3 continues on page 5.

(c) Figure 3 shows the objective lens of a projection lithography system.

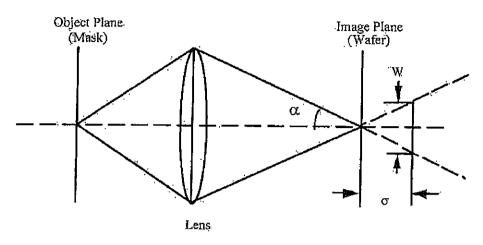


Figure 3

- (i) Assuming $\tan \alpha \sim \sin \alpha$, determine the Numerical Aperture (NA), given that the diameter of the objective lens and its focal length are 1.5 cm and 1 cm, respectively.
- (ii) Hence, determine the depth of focus of the system given that k_2 value is 0.85 and the wavelength of the optical source is 193 nm.
- 4. (a) In the plasma of a RIE system, consider an argon atom that is conized during a collision in the plasma. Over a given mean free path, one of the particles will achieve a much higher speed. Explain, with a diagram, how that will affect the potential (voltage) of the plasma relative to that of either electrode.

 (3 Marks)
 - (b) What are the etch rate results and selectivity of a RIE system if the following parameters are changed?
 - (i) RF frequency is decreased. ht
 - (ii) RF power is decreased.
 - (iii) DC bias is increased.
 - (iv) Cathode size is decreased.

(4 Marks)

(7 Marks)

Note: Question No. 4 continues on page 6.

 $\subseteq \mathfrak{h}_2$

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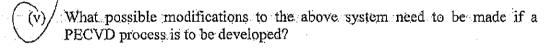
The mixture of SF₆ and O₂ is commonly employed in the Si dry etching process. List the main volatile etch-product of this etching process and explain the roles played by the O₂ plasma. List two other types of commonly used gases for the Si dry etching process. Explain why an effective dry etching process for SiO₂ and Si₃N₄ cannot be developed using SF₆ and O₂ mixture.

(5 Marks)

(d) Define the residence time of a gas molecule in a plasma system. Discuss what factors can affect the average residence time of the gas molecule of the plasma.

(3 Marks)

- (e) In a LPCVD process system, reaction gas is metered into a 100 litter reactor at a flow of 10 ml/sec. The resulting steady state pressure is 0.2 torr. Assume that 1 torr litre per second = 79 secm.
 - (i) Determine the pumping speed of the pump.
 - (ii) Determine the residence time of a gas molecule in the reactor.
 - (iii) If the flow is increased to 20 ml/sec, find the pressure and residence time.
 - (iv) If a throttle valve is placed on the pump to cut the pumping speed to half, find the pressure and the residence time:



(10 Marks)

APPENDIX A

List of Selected Formulae

P-n junction

$$\begin{split} V_{bl} &= \frac{kT}{q} \ln \frac{N_A N_D}{n_p^2}; \qquad N_A x_p = N_D x_n; \qquad W = x_p + x_n; \qquad C_J = \frac{\mathcal{E}_s}{W}; \\ V_{bl} &= \sqrt{\frac{2\mathcal{E}_s}{q}} \left[\frac{1}{N_A} + \frac{1}{N_D} \right] (V_{bl} - V); \qquad L_p = \sqrt{D_p \tau_p} \;. \end{split}$$

Bipolar junction transistors

$$\begin{split} \gamma &= \frac{I_{Ep}}{I_E} = \frac{I_{Ep}}{I_{Ep} + I_{En}}; \quad \alpha_T = \frac{I_{Cp}}{I_{Ep}}; \quad \alpha_0 = \gamma \alpha_T; \quad \beta_0 = \frac{\alpha_0}{1 - \alpha_0}; \quad I_C = \alpha_0 I_E + I_{CBO}; \\ I_{CEO} &= (1 + \beta_0) I_{CBO}; \quad p_n(x) = p_{no} e^{qV_{EB}/kT} (1 - \frac{x}{W}); \quad \gamma = \frac{1}{1 + \frac{D_E}{D_p} \cdot \frac{N_B}{N_E} \cdot \frac{W}{I_E}}; \\ I_{Ep} &= qA \frac{D_p p_{n0}}{W} e^{(qV_{EB}/kT)}; \quad I_{En} = qA \frac{D_E n_{E0}}{I_E} (e^{qV_{EB}/kT} - 1); \quad I_{Cp} = qA \frac{D_C n_{C0}}{I_C}; \\ p_{no} \cdot N_B &= n_{E0} \cdot N_E = n_{Co} \cdot N_C = n_i^2; \quad \tau_B = \frac{W^2}{2D_p}; \quad f_T = \frac{1}{2\pi \tau_0}. \end{split}$$

MOS devices

$$\begin{split} \psi_{s} &= 2\psi_{B} = \frac{2kT}{q} \ln(\frac{N_{A}}{n_{l}}); \ W_{m}^{2} = \frac{2\varepsilon_{s}(2\psi_{B})}{qN_{A}} = \frac{4\varepsilon_{s}kT}{q^{2}N_{A}} \ln(\frac{N_{A}}{n_{l}}); \ V_{T} = \frac{qN_{A}W_{m}}{C_{o}} + 2\psi_{B}; \\ \frac{C}{C_{0}} &= \frac{1}{\sqrt{1 + \frac{2\varepsilon_{ox}^{2}V}{qN_{A}\varepsilon_{s}d^{2}}}}; \qquad \frac{1}{C_{\min}} = \frac{d}{\varepsilon_{ox}} + \frac{W_{m}}{\varepsilon_{s}}; \qquad V_{FB} = \phi_{ms} - \frac{(Q_{f} + Q_{m} + Q_{ol})}{C_{0}}. \\ I_{D} &= K_{n}[(V_{GS} - V_{T})V_{DS} - \frac{V_{DS}^{2}}{2}] \text{ for } V_{DS} \leq V_{GS} - V_{T}; \qquad V_{T} = \frac{qN_{A}W_{m}}{C_{0}} + 2\psi_{B} \text{ when } V_{FB} = 0; \\ I_{D} &= \frac{K_{n}}{2}(V_{GS} - V_{T})^{2} \text{ for } V_{DS} \geq V_{GS} - V_{T}; \qquad K_{n} = \mu_{n}C_{ox}\frac{W}{L}. \end{split}$$

Thermal oxidation

$$t_{ax}^2 + At_{ax} = B(t+\tau); \quad \tau = \frac{t_{axt}^2}{B} + \frac{t_{axt}}{B/A}, \quad t_{ax} = \frac{-A + \sqrt{A^2 + 4B(t+\tau)}}{2}$$

Thermal diffusion

$$D = D_0 \exp(-\frac{E_0}{kT})$$

Constant source diffusion:

Limited source diffusion:

$$N(z,t) = N_s erfc(\frac{z}{2\sqrt{Dt}})$$

$$N(z,t) = N_s erfc(\frac{z}{2\sqrt{Dt}}) \qquad N(z,t) = \frac{Q}{\sqrt{\pi Dt}} \exp[-\frac{z^2}{4Dt}], \quad Q = \frac{2}{\sqrt{\pi}} N_s \sqrt{Dt}.$$

Ion implantation

Before Annealing

After annealing

$$N(x) = \frac{Q}{\sqrt{2\pi} \Delta R_{\rho}} \exp\left[-\frac{(x - R_{p})^{2}}{2\Delta R_{\rho}^{2}}\right] \qquad N(x) = \frac{Q}{\sqrt{2\pi} (\Delta R_{\rho}^{2} + 2Dt)^{1/2}} \exp\left[-\frac{(x - R_{p})^{2}}{2(\Delta R_{\rho}^{2} + 2Dt)}\right]$$

$$Q = \int_{0}^{\infty} N(x) dx = \sqrt{2\pi} N_{\rho} \Delta R_{\rho}$$

APPENDIX B

Table of Physical Constants

Physical Constant	Symbol	Value	Units
Electronic charge	g	1.6×10^{-19}	C
Boltzmann's constant	ķ	8.62×10^{-5}	eV/K
		1.38066 ×10 ⁻²³	J/K
Planck's constant	.h	6.626 × 10 ⁻³⁴	J·s
Permittivity of free space	En	8/85 × 10 ⁻¹⁴	F/cm
Dielectric constant of Si	ધ્યા	11,7	
Dielectric constant of SiO ₂	Eox	3,9	-
Electron Mass	m	9.11 × 10 ⁻³¹	kg
Speed of Light	С	3 × 10 ⁸	m/s·
Bandgap of Si at 300 K	Eġ	1.12	eV
Intrinsic carrier concentration in Si at 300 K	n_{i}	1 × 10 ¹⁰	cm ⁻³

END OF PAPER

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Q1 (a) (i)
$$V_{bi} = \frac{kT}{q} \ln \frac{NAND}{n_i^2} = 0.075^q \ln \frac{8 \times 10^{16} \times N}{(1.8 \times 10^6)^2} = 1.2V$$

$$N = \frac{4 \times P(\frac{1.2}{0.075^q}) \times (0.8 \times 10^6)^2}{9 \times 10^{16}} = 5.76 \times 10^{15} \text{ cm}^{-3}$$

= 0.02953 cm

$$= 0.02957 \text{ cm}$$

$$C_{10}^{2} = \frac{8185 \times 10^{-1} \times 1.8 \times 10^{6}}{0.02953} = 5.2965 \times 10^{-1} \text{ F}$$

$$S = \frac{C_1 + otal}{C_1 \circ o} = \frac{1}{5.39} \frac{10^{-12}}{5.39} = \frac{1}{5.39} \frac{0.04 \times 10^{-1}}{5.39} \frac{0.0$$

reverse hias voltage in crease - (Vb) - V) in crease. (m)

Junction capacitance will decrease.

(i) the semi conductor 13 ntype.

(i) the semi conductor 13 n type.

(ii)
$$C_0 = 200 \times 10^{-12} = \frac{3.9 \times 9.95 \times 10^{-14}}{200 \times 10^{-12}} = \frac{G_0 \times 10^{-14}}{200 \times 10^{-12}} = \frac{1.126 \times 10^{-3} \text{ cm}}{100}$$

(iii) $C_0 = 200 \times 10^{-12} = \frac{3.9 \times 9.85 \times 10^{-14}}{200 \times 10^{-12}} = \frac{1.126 \times 10^{-3} \text{ cm}}{100}$

$$-0.8 = -0.5 - \frac{000 \times 10^{-17}}{0.00 \times 10^{-17}}$$

(c) (i) base - emitter should be forward brossed because in active mode, we want more holes to be injected from emitter to base.

base - collector should be reversed blased because. me want to increase the energy burnier to prevent holes being impected back to base. and more holes can be drift to collector. the efficiency with be higher.

- (17) O decrease doping concentration NA or No.
 - 3) increase the reverse was voltage for B-c junction. This two ways can both increase the depletion region. thus decrease the neutral base Since $CB = \frac{W^2}{2Pp}$ When W decrease. To decrease for $TB = \frac{W^2}{2Pp}$ with increase.

(a) (i) tox + Atox = B(t+t) 0,672+0,726×0,67=0,287(2+t) Z= 010917 hr.

$$toxi = \frac{-0.776 + \sqrt{6.748 + 4x0.287 \times 0.0917}}{7} = 0.0576 \text{ Am}.$$

- dry oxidation; because it requires thin, high quality layer. (ii) Waate oxide
 - Wet oxidation because it requires high rate large area O. masking but not high quality needed.

$$2.(b)^{(1)}N = \frac{Q}{\sqrt{nDt}} \cdot \exp\left(-\frac{Z^2}{4Dt}\right) = 1.5 \times 10^{15}.$$

$$\frac{2 \times 10^{17}}{\sqrt{nDt}} \cdot \exp\left(-\frac{Z^2}{4Dt}\right) = 1.5 \times 10^{15}.$$

$$\frac{2 \times 10^{17}}{\sqrt{nDt}} \cdot \exp\left(-\frac{Z^2}{4Dt}\right) = 1.5 \times 10^{15}.$$

$$(11) \quad (0) \quad \text{cheap.}$$

$$(2) \quad (0) \quad N(x) = \frac{Q}{\sqrt{nDt}} \cdot \exp\left(-\frac{(x-Pt)^2}{2DPt}\right) = NB.$$

$$(0) \quad \frac{1}{\sqrt{nDt}} \cdot \frac{10}{\sqrt{nDt}} \cdot \frac{10}{\sqrt{nDt}}$$

Q3. (a) (i) E-beam lithography system. Advantages.

Opprecise control of energy and dose

- O. Small Wavelength.) good resolution.
- beam focusing achieved.
- able to register accurately over small areas
- 9 1000 defects densities Disadvantages.

require high vacuum system.

@ sensitive to electronic and mechanical noise.

3 proximity effect.

Lit) use electron resist which is sensitive to a beam.

(171) Whin = $\sqrt{1009}$. 8 = $\sqrt{0.77 \cdot 10000}$.

$$7 = \frac{h}{\sqrt{12mev}} \quad v = \frac{\left(\frac{h}{\sqrt{1}}\right)^{2}}{\sqrt{2me}}$$

= 1807777V.

E= 1807 . Trev. required.

(b) (i) soft bake improve the light absorbance for photoresist.

hard bake hardens the resist for etching or mosk contain the full graphic pattern used for single emposure.

While reficle contain part of the emposure.

pattern use for step and repeat process.

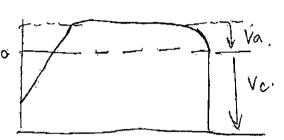
pattern use for step and repeat process.

A ray 14thrography system use x ray for exposure.

e- bean system use electron beam for exposure.

 $(NA = Sind \approx fom k = \frac{A}{f} = \frac{115}{2} = 175$ $=. 0.85 \times 193. =. 291.64 nm.$

Q4 (a) (not gure finin. duning exam)



Distance. higher speed - higher energy - higher voltage.

- (b) (i) etch rate 1. selectivity no change.
 - etch rate & select Wity no change.
 - citi) etch rate t. selectivity no change.
 - civ) γ (re) = $V_{A}(\frac{Aa}{Ac})$ in one ase energy, etch rate γ . Selectivity no change.
- Or react with Str. Str. to form sof and F. reduce the recombination rate of the active. F. prevent SFs. Lunreactive)
 - the time a gas molecule can stay in the chamber before being pumped out. (O);

tr= 3. The chamber volume V and pumping speed Es can offect tr.

4(e) (i)
$$P = \frac{F(760)}{S}$$
 $S = \frac{10 \times 10^{-3} \times 760}{0.12}$
 $= .38 \text{ liter/S}.$

(iii) $tv = \frac{V}{S} = \frac{100}{38} = .2163.5.$

(iii) $P = \frac{20 \times 10^{-3} \times 760}{38} = 0.14 + 0.000.$
 $tv = \frac{V}{S} = \frac{100}{38} = 0.14 + 0.000.$
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 $tv = \frac{100}{38} = 0$

(v) O connect electrode with RF generator.

to generate plasma.

to generate plasma.

Wafer horizontally placed on the other

electrode

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