# A collaborative LaTeX document

# Class of ID2090, Third Trimester of 2021 batch $\label{eq:June 14} \text{June 14, 2022}$

### Contents

1	Introduction	3
2	AE21B003	4
3	AE21B028	5
4	AE21B045	6
5	AE21B056	7
6	AE21B062	8
7	AE21B107	9
8	BE21B016	10
9	BE21B040	11
10	CE19B020	12
11	CE21B021	13
<b>12</b>	CE21B088	14
13	CE21B097	15
14	CE21B112	16
15	CE21B115	17
16	CH21B067	18
17	CH21B079	19
18	CH21B101	20
19	ME21B050	21
20	ME21B060	22
21	ME21B065	23

22 ME21B079	24
23 ME21B088         23.1 Heisenberg Uncertainty Principle          23.1.1 Equations          23.1.2 Explanation          23.1.3 Figure 1          23.1.4 Significance	$   \begin{array}{r}     25 \\     25 \\     26   \end{array} $
24 ME21B091	27
25 ME21B186	28
26 ME21B190	<b>2</b> 9
27 ME21B196	30
28 ME21B204	31
29 ME21B217	32
$30~\mathrm{MM21B012}$	33
31 MM21B024	34
32 MM21B032	35
33 MM21B044	36
34 MM21B046	37
35 MM21B059	38
36 MM21B063	39
37 NA21B002	40
38 NA21B005	41
39 NA21B006	42
40 NA21B007	43
41 NA21B020	44
42 NA21B048	45
43 NA21B052	46
44 Conclusions	47
45 References	47

# List of Figures

### List of Tables

### 1 Introduction

This file includes tex files from the folders of each student. The students are expected to update the file named after their roll number and place any images in the same folder. Students do not have to edit this master document. Once the student has sent a pull request which is accepted and processed successfully, his/her assignment submission is deemed to be complete.

You are also welcome to add references and cite them. Examples on how to do that are on the course repository [?].

### 8 BE21B016

### 9 BE21B040

# 10 CE19B020

# 16 CH21B067

# 17 CH21B079

# 18 CH21B101

Assignment 4 Khobragade Kashish Vinod me21b088 June 2022

### 23.1 Heisenberg Uncertainty Principle

#### 23.1.1 Equations

$$\Delta x \cdot \Delta p \ge \frac{h}{4\pi} \tag{1}$$

$$\Delta x \cdot m \Delta v \ge \frac{h}{4\pi} \tag{2}$$

$$\Delta E \cdot \Delta t \ge \frac{h}{4\pi} \tag{3}$$

Symbols	Meaning of Symbols
$\Delta x$	uncertainty in displacement
$\Delta p$	uncertainty in momentum
$\Delta v$	uncertainty in velocity
$\Delta E$	uncertainty in energy
$\Delta t$	uncertainty in time
h	planck's constant value is $6.63 * 10^{-34}$ Jsec
m	mass

According to Heisenberg, it is impossible to measure both the position and momentum of moving particle with accuracy.

- 1. If value of position is small, it can be measured accurately but not momentum.
- 2. If value of momentum is small it is measured accurately but not the position.

#### 23.1.2 Explanation

- 1. Suppose we need to measure position accurately, then we need to use light.
- 2. So, that the photon of light must strike the electron and reflected photon is seen with microscope.
- 3. Due to hitting, the position and velocity of electron is changed.
- 4. But to pin point position, the light of shorter wavelength should be used.
- 5. The shorter wavelength means high frequency and high energy.
- 6. So, this high energy photon may change the speed and direction of particle.

### 23.1.3 Figure 1

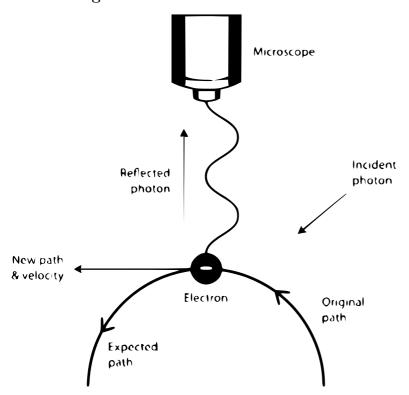


Figure 1. Heisenberg Uncertainty Principle: The observation of an electron with a microscope requires reflection of a photon off of the electron. This reflected photon causes a change in the path of the electron.

#### 23.1.4 Significance

This holds good only for microscopic particles, as energy of photon is enough to change the position and velocity of bigger bodies, So, in our daily routine it has no significance.

### 44 Conclusions

If this master tex file could be compiled successfully, it means that the class has learnt the concepts of Git as well as LaTeX properly.

### 45 References

### References

[1] Repository for id2090 course. https://github.com/gphanikumar/mm2090. Accessed: 2022-06-13.