

Lecture 1: Introduction to Molecular Simulation

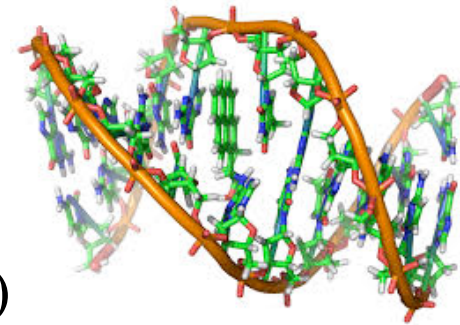
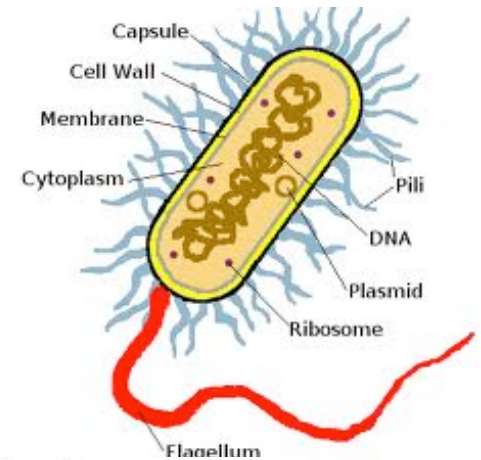
August 29, 2016

Outline

- Syllabus and introductions
- Perspective: Why molecular simulation?
- Computer logistics
- Break, fill out survey.
- Simulation techniques
- Files related to the course can be found on Blackboard.

What is Nano?

- 1 meter
 - about 3.3 feet
 - a very tall person is 2 m tall
- 1 millimeter = 0.001 meters
 - smallest division on your ruler
 - a typed period is about 0.5 mm in diameter
- 1 micrometer (micron) = 0.000001 meters
 - average human hair diameter is 80 microns
 - a bacterium cell is 1 micron in length
- 1 nanometer = 0.000000001 meters
 - diameter of a DNA molecule is 2 nm
 - atomic bonds are less than 1 nm long
- Also interested in short time scales (fs-ns)

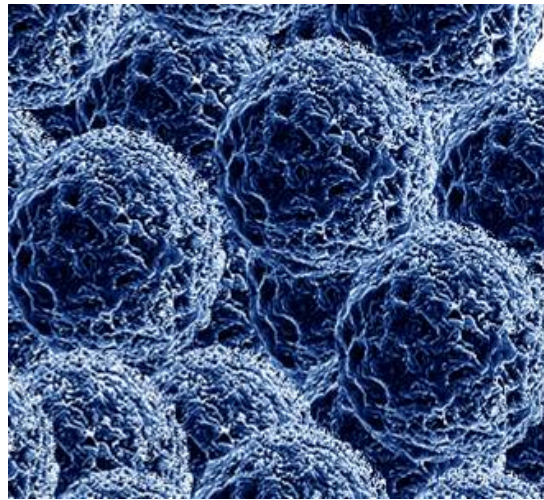
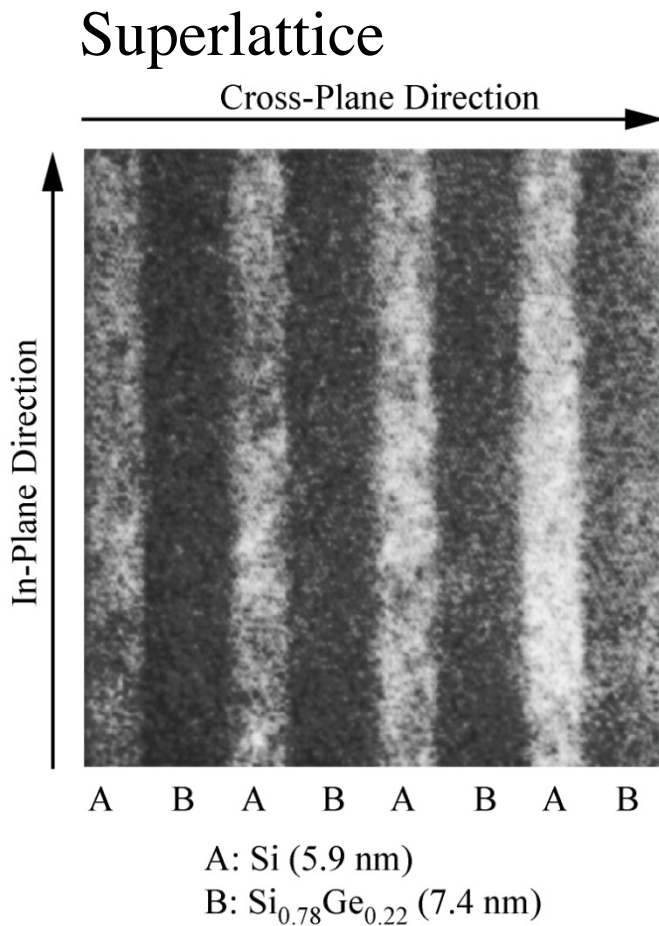


[http://en.wikipedia.org/wiki/Orders_of_magnitude_\(length\)](http://en.wikipedia.org/wiki/Orders_of_magnitude_(length))
<http://scaleofuniverse.com/>

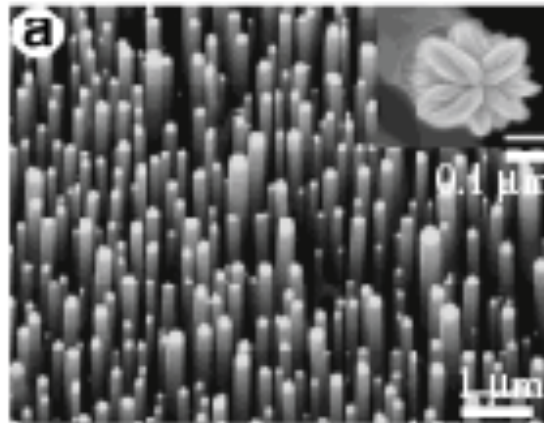
Energy Scale

- Energy = mass x length² / time²
 - Mass scale $\sim 1\text{E-}26$ kg
 - Results are often reported in electron-Volts (eV)
 - $1 \text{ eV} = 1.60219\text{E-}19 \text{ J}$
 - You may also see energies in terms of $k_{\text{B}}T$
(thermal fluctuation reference point)
 - k_{B} = Boltzmann constant = $1.3806\text{E-}23 \text{ J/K}$
(related to the ideal gas law)
 - at $T = 300 \text{ K}$, $k_{\text{B}}T = 4.14\text{E-}21 \text{ J} = 0.025 \text{ eV} = 1/40 \text{ eV}$
 - Atomic-level energies are typically between 0.01 eV and 100 eV .
-

Nanomaterials

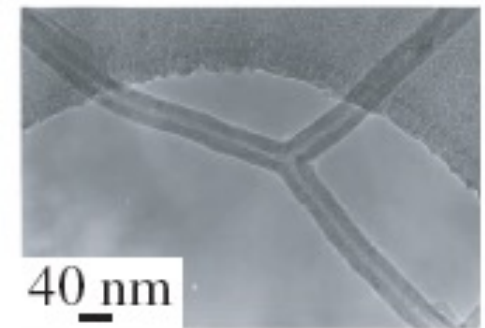
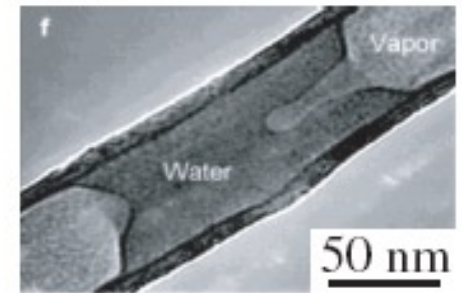


Metallic Nanoparticles



Silicon-Germanium
Nanowires

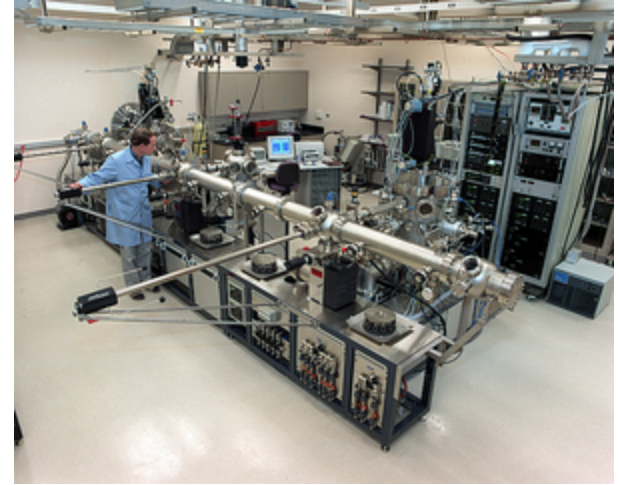
Carbon Nanotubes



... quantum dots, nanofluids, nanoribbons, nanopipes ...

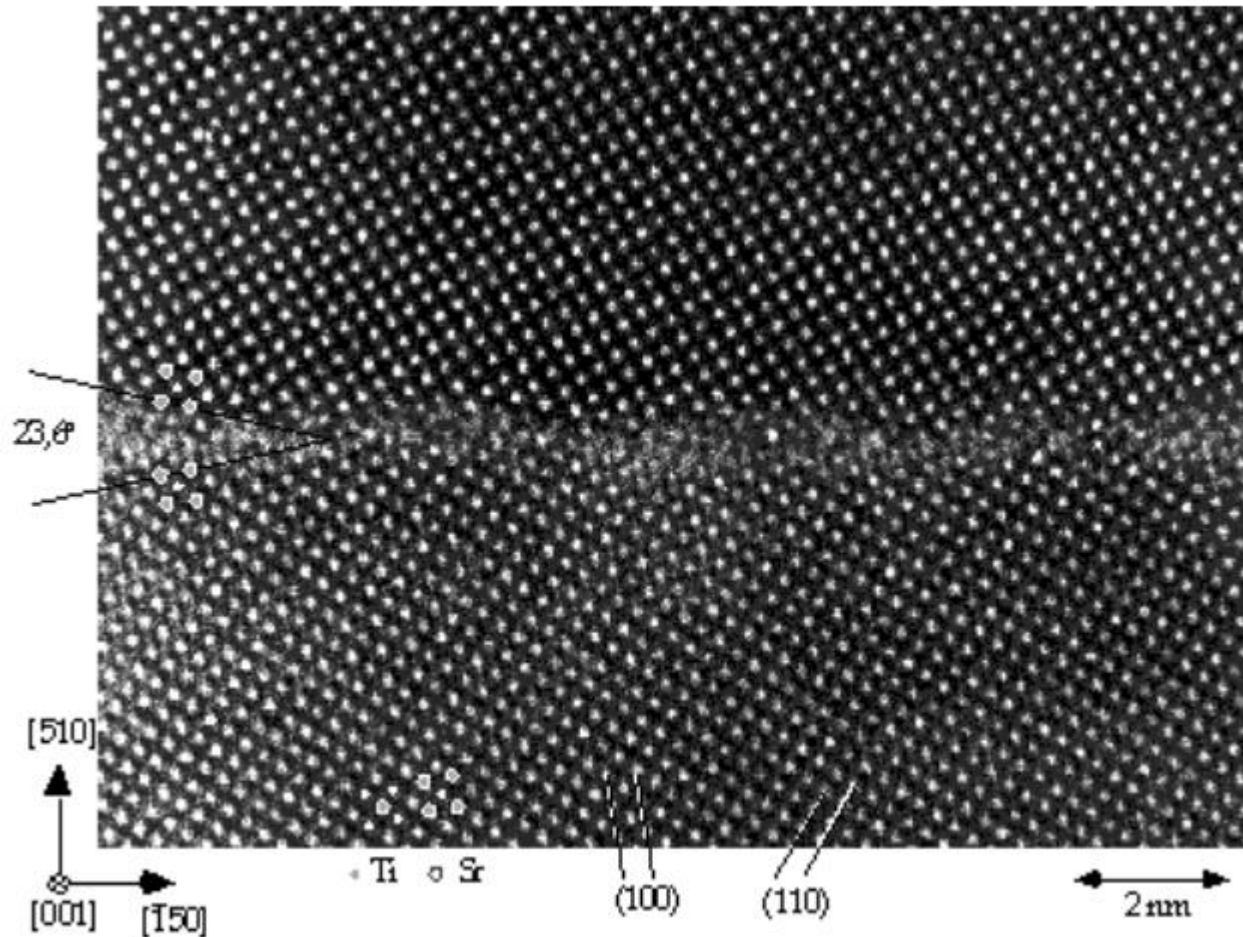
Fabrication and Characterization

- Structures are built from the bottom-up:
 - chemical vapor deposition
 - molecular beam epitaxy
 - sputtering
 - ...
- We can't see things this small. Images are obtained using *electron microscopy*.
- Sample preparation is challenging.



What About Experiments?

- Electron microscopy can generate images with sub-micron resolution
 - *In situ* time resolution \sim ms (maybe)

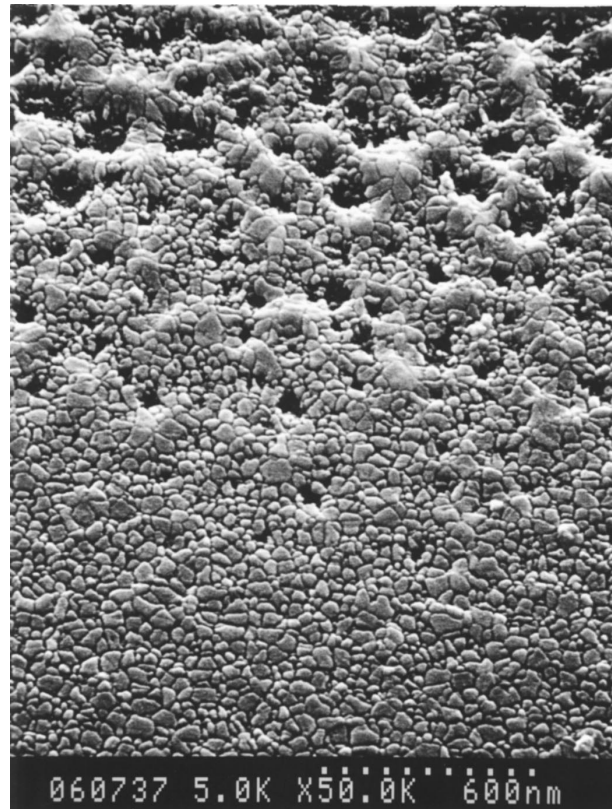
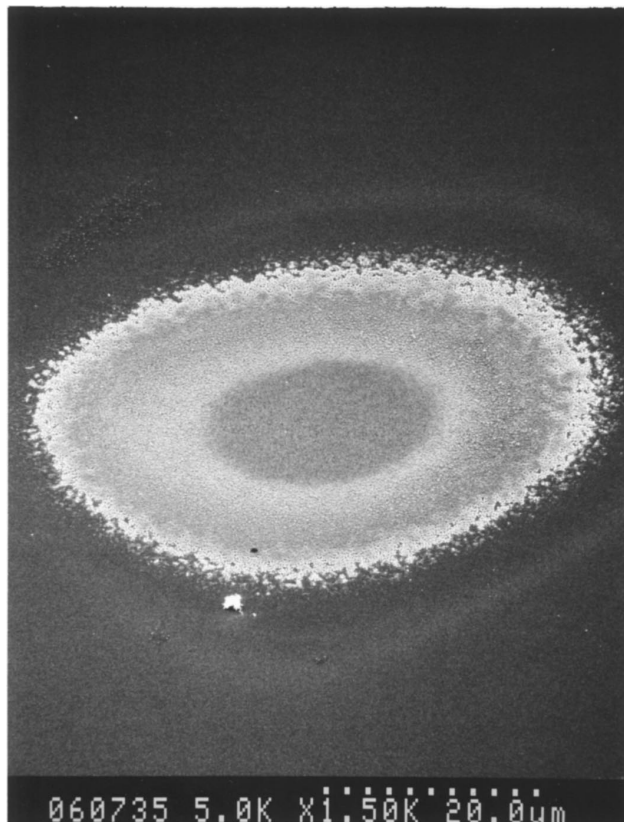


Grain boundary in
strontium titanate

[pruffle.mit.edu/
~ccarter/NANOAM/
images/](http://pruffle.mit.edu/~ccarter/NANOAM/images/)

What About Experiments? (2)

- Lasers can resolve phenomena down to picoseconds (10^{-12} s), femtoseconds (10^{-15} s)
 - measuring thermal conductivity (Malen), materials processing (ablation)
 - spatial resolution $\sim \mu\text{m}$



Femtosecond pulsed laser
recrystallization of a-Si
film on quartz substrate

[http://
www.me.berkeley.edu/ltl/
research/fs.html](http://www.me.berkeley.edu/ltl/research/fs.html)

Why Modeling?

- Experiments can be time- and resource-intensive
- Nanoscale phenomena are challenging to observe experimentally
 - Small length (10^{-10} m) and time (10^{-15} s) scales
 - Difficult to measure properties and elucidate physics
- Properties not available in handbooks or textbooks
 - New bulk materials and nanostructures, extreme conditions (geophysics)
- Fundamental understanding
 - Why does one material have a higher thermal conductivity than another?
 - How to design materials with tailored thermal properties?

Why Simulation?

- Exact solutions are the exception. Need numerical solutions!

$$m_i a_i = \sum_{j=1, j \neq i}^N F_{ij}$$

- Closed form (algebraic) solution only for $N = 1$ or 2 .
- We are interested in hundreds, thousands, millions of atoms.

Why at the Molecular Level?

- A route from the microscopic (atoms and their interactions) to the macroscopic (equation of state, transport coefficients).
- Length and time scales not describable by continuum approaches (e.g., energy equation, Navier Stokes, beam equations, ...).

Perspective

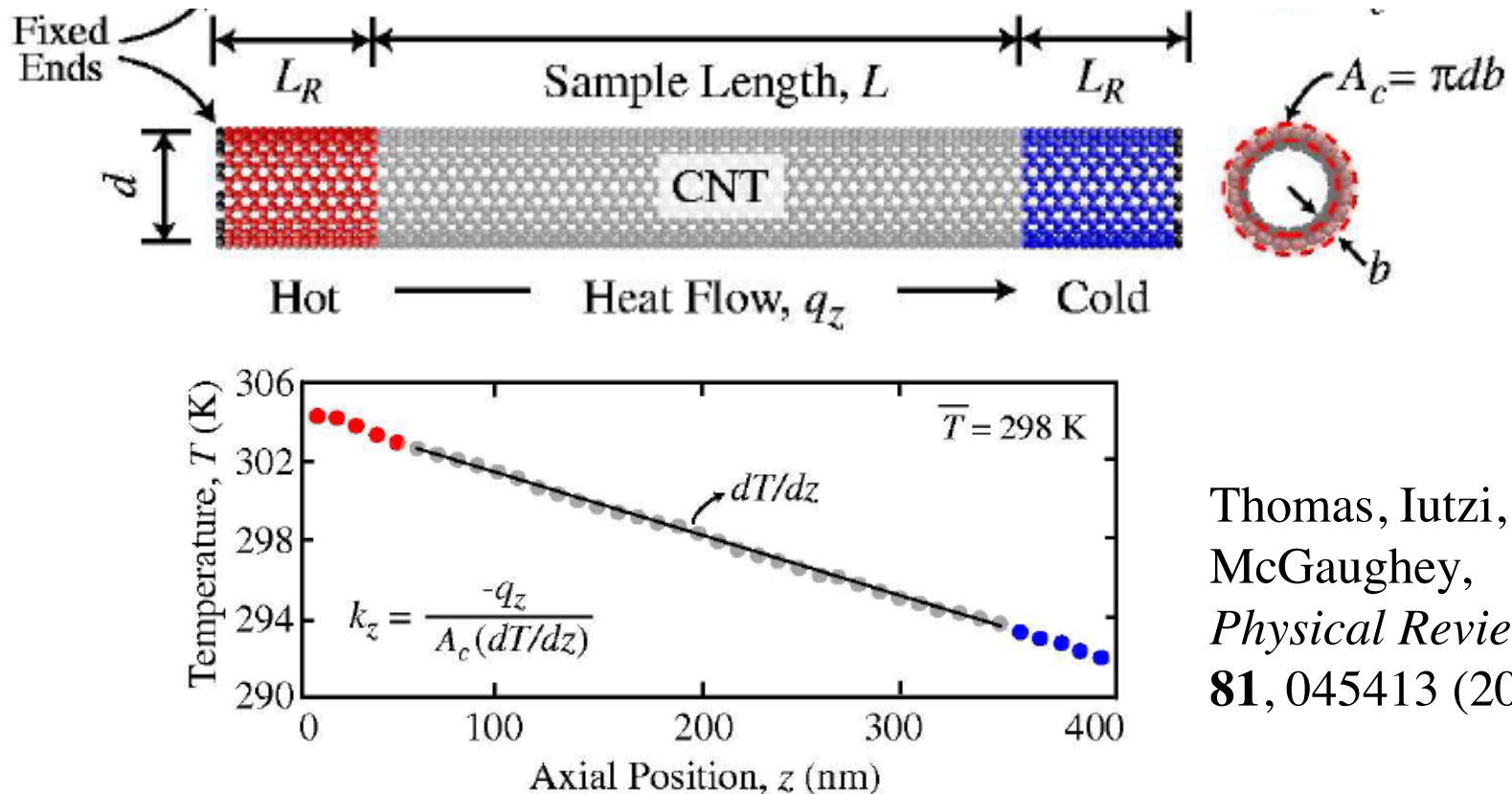
- **A simulation is a computer experiment.**
 - Make measurements, identify mechanisms.
 - Not just a calculation, like taking the Fourier transform
 - Inherent uncertainty due to finite sampling
(we will see this point a lot)
- As with an experiment, you need to understand the capabilities and limitations of your simulation technique.

Which Class to Take? Both!

24-623/12-623 (this class)	06-640 (John Kitchin in ChemE)
Classical molecular dynamics/Monte Carlo	Quantum mechanics-based methods (density functional theory)
Classical limit: no quantum effects, limited to “high” temperature	All temperature accessible, but trajectories in dynamics are classical
Millions (billions?) of atoms	Hundreds/thousands of atoms (size effects may still be present)
Nanoseconds of simulation	Picoseconds of simulation
Energetics, transport, mechanics Reaction (breaking/forming bonds) is difficult	Predicting energies and energy barriers, relaxing structures, reaction pathways Limited access to dynamics

Predicting the Thermal Conductivity of CNT

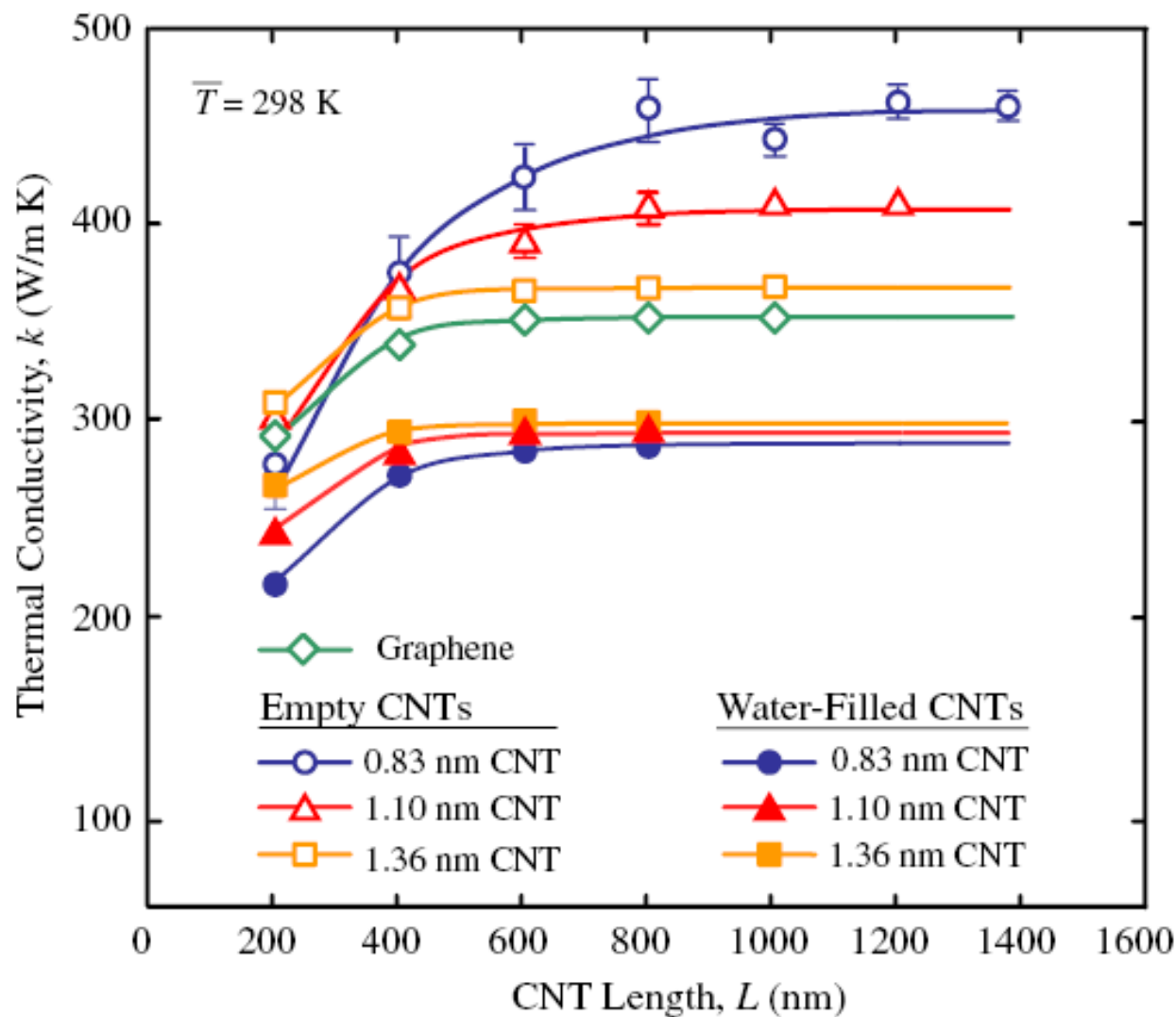
- How long does the CNT need to be?
- How much time is needed to get the system to steady state?
- How much time is needed to obtain enough data to get good averages?
- How do these quantities depend on temperature and CNT diameter?
- What is the resulting uncertainty?



Thomas, Iutzi, and McGaughey,
Physical Review B.
81, 045413 (2010).

Results

You won't know the answers to any of these questions when you start.



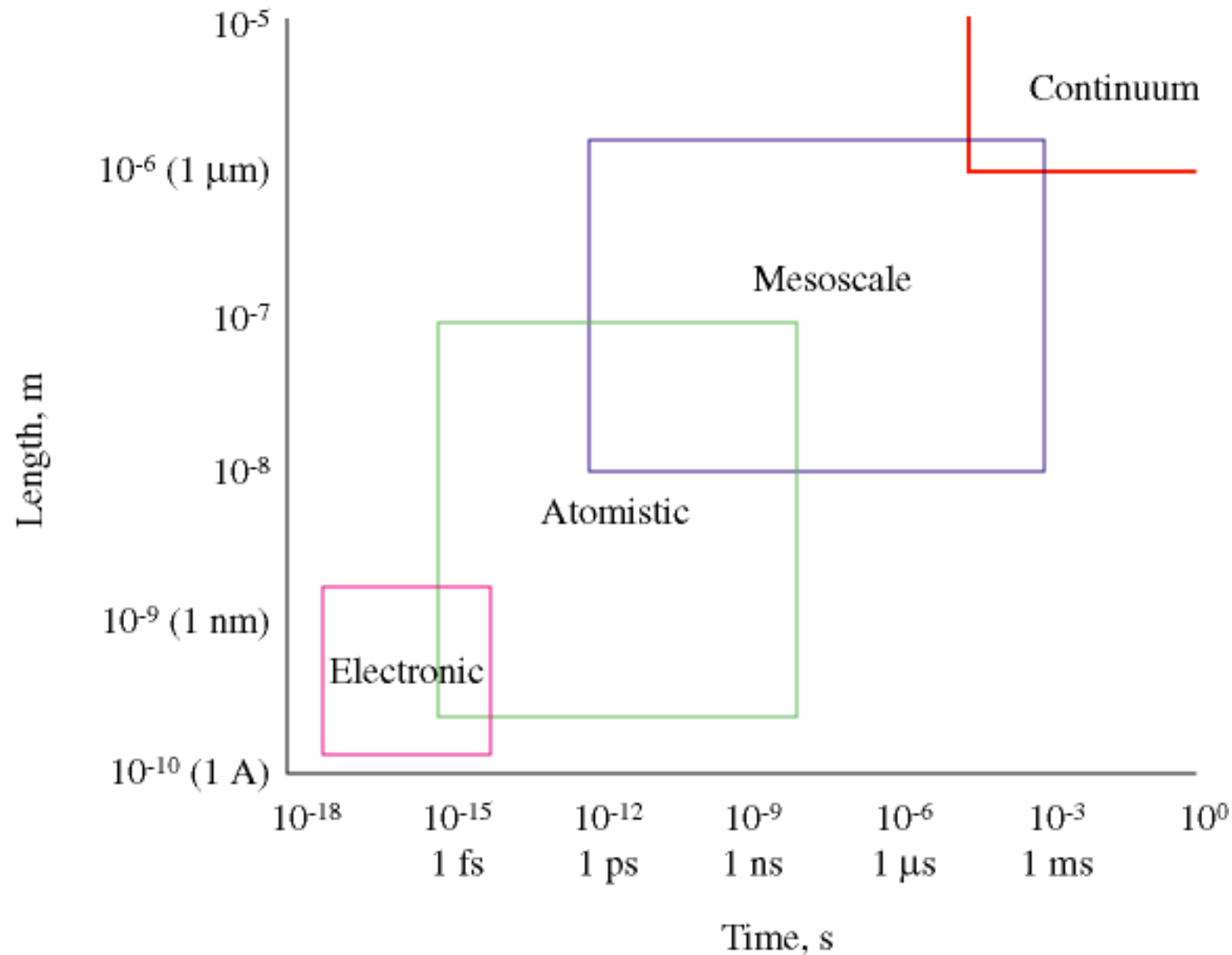
Computer Logistics

- Writing code from scratch (don't use parts of a code from your research group). You will be submitting your code. It must compile and run.
- Options: C++, Fortran. DO NOT USE A MATH PROGRAM.
- Class examples will be in C++.
- Don't make code any more complicated than it needs to be.
- Comment! (you will forget how/why you did something).
- Spreadsheet program (Excel) and/or math program (Matlab, Mathematica).
- See the file computer.pdf on the website for more details.
- *Establish your computer presence.*

Wednesday Coding Workshop

- Bring your laptop to class if you have one.
- No problem if you don't, we will be working in pairs.
- Before: Download and install an editor and compiler. Test it out if you know how.
- We will go through how to compile and run C++ on a Mac.
- Explain a simple code line-by-line.
- Then you will modify.

Simulation Techniques



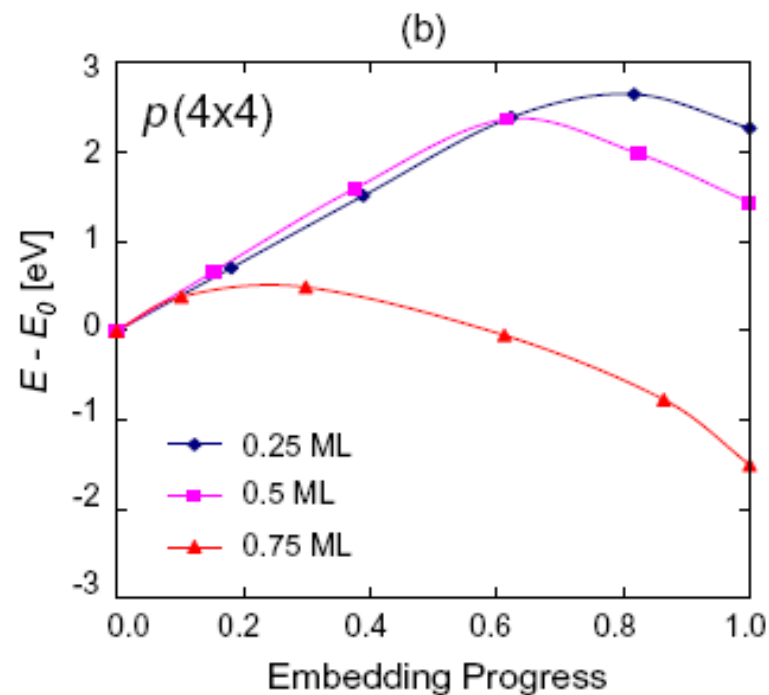
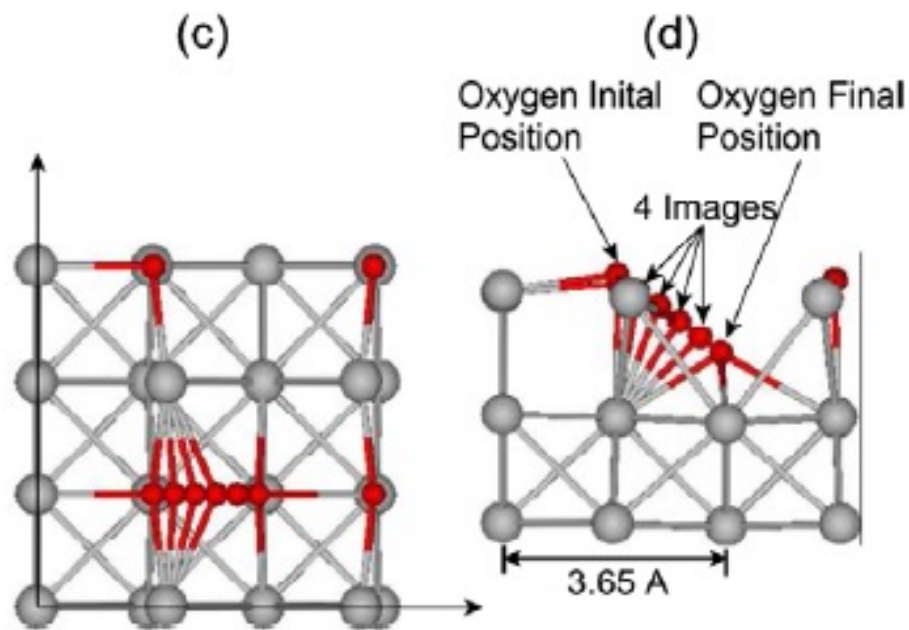
More Perspective

- Purpose of a simulation: qualitative or semi-quantitative comparison to experiments.
 - Simulations can be tuned but this can cause a loss of generality.
 - Identify trends and mechanisms
- Compromise between accuracy and computational time.
 - Depends on what you are trying to do

Electronic Structure Calculations

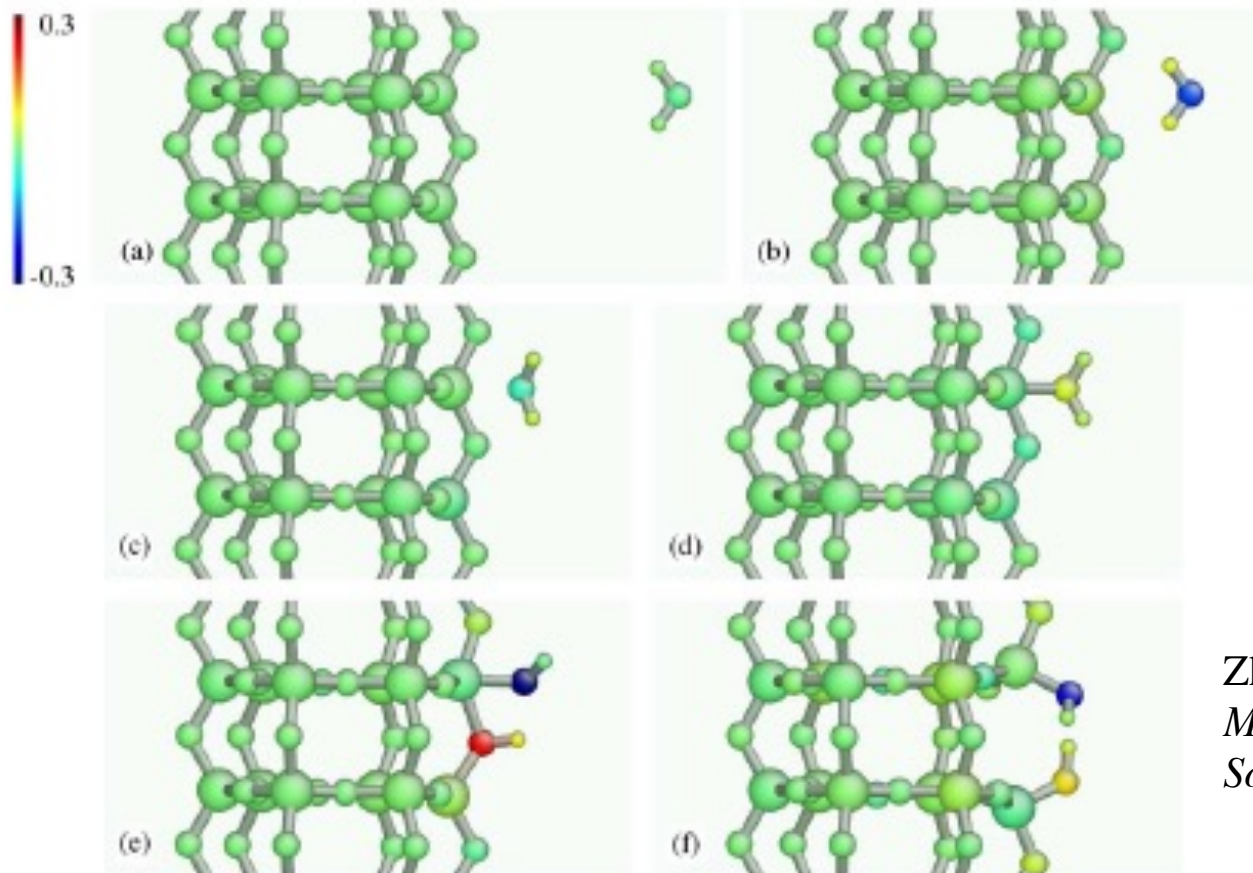
- Electronic degrees of freedom are considered
 - Quantum mechanics -> Solve the Schrödinger Equation.
- Theory is very complicated.
 - All-electron, density functional theory, tight-binding
 - Commercial/academic codes (would take far too long to write one yourself). Quantum Espresso (free), VASP (license), GAUSSIAN (license), ABINIT (free), SIESTA (free)
 - courses at CMU (06-640) and Pitt (Chemistry Dept.)
- Can be easy to use, but garbage in = garbage out!
- Typically scales as the cube of the number of electrons.
 - Limited access to dynamics.
 - Systems with more than a few thousand atoms rarely considered.

Oxidation of Copper Surfaces (DFT calculations)



Lee and McGaughey, *Surface Science* **603** (2009) 3404-3409.

Water Interacting with a Silica Nanowire (Semi-Empirical Molecular Orbital Theory)



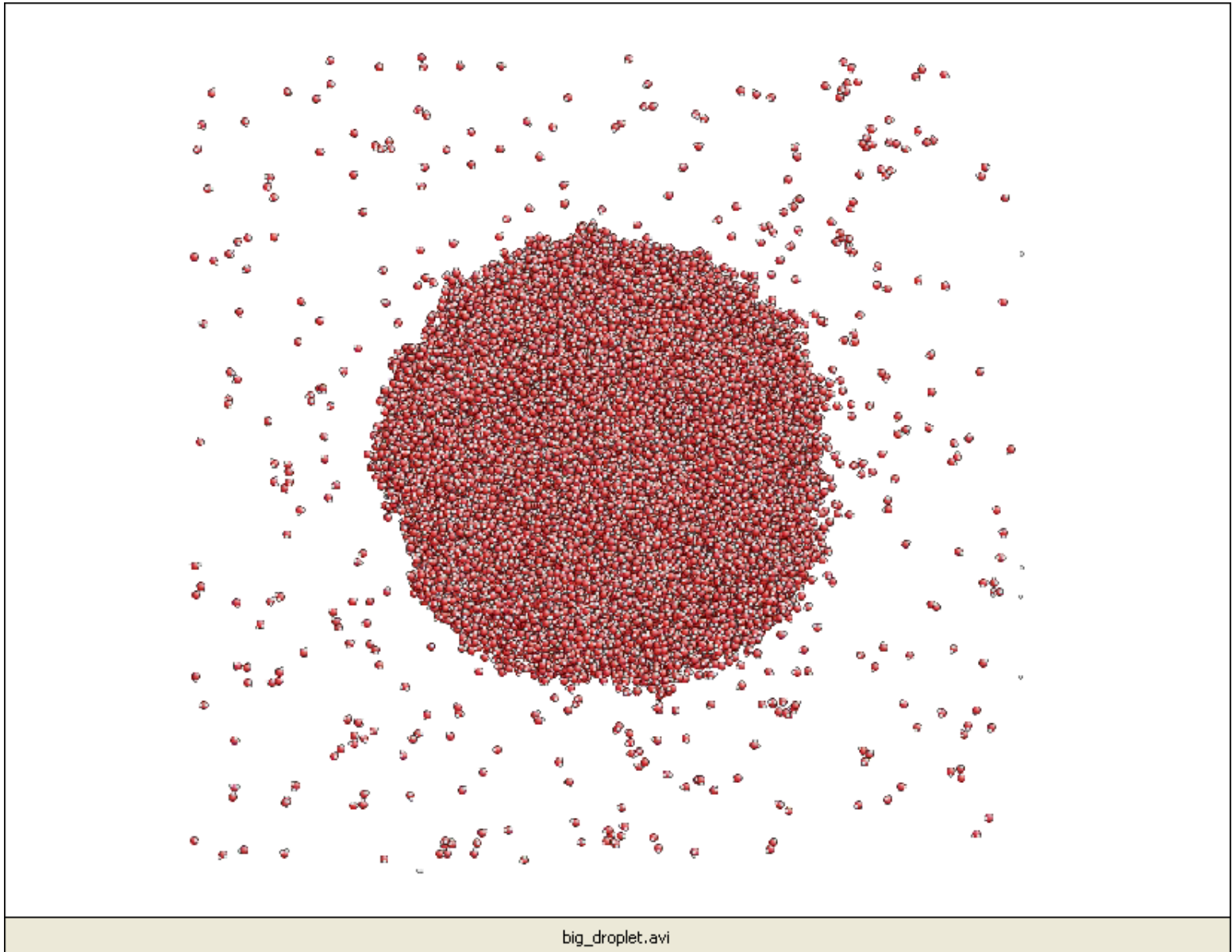
Zhu et al., *Journal of the Mechanics and Physics of Solids* **53** (2005) 1597-1623.

Fig. 6. Atomic configurations along the transition pathway of hydrolysis under a stress of $0.3\sigma_{cr}$. Mechanism II: (a) initial state, (b) physisorbed state, (c) the first saddle-point configuration, (d) metastable molecularly adsorbed state, (e) the second saddle-point configuration, and (f) final chemisorbed state. Atoms are color-coded by charge variation relative to the initial configuration (a).

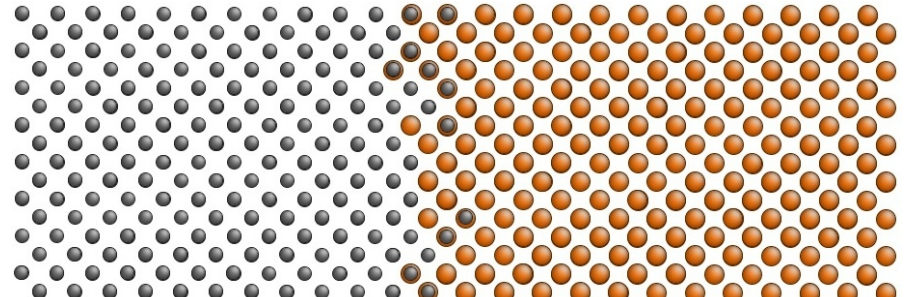
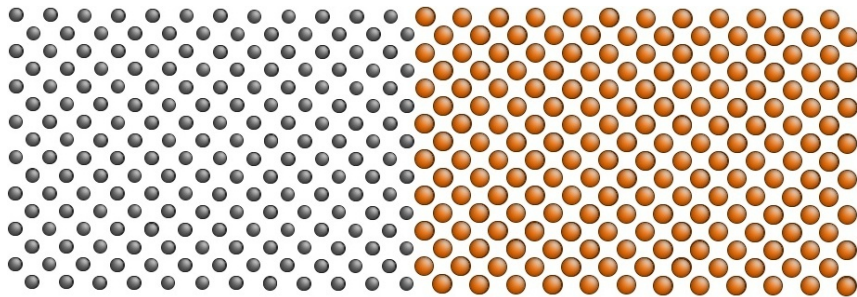
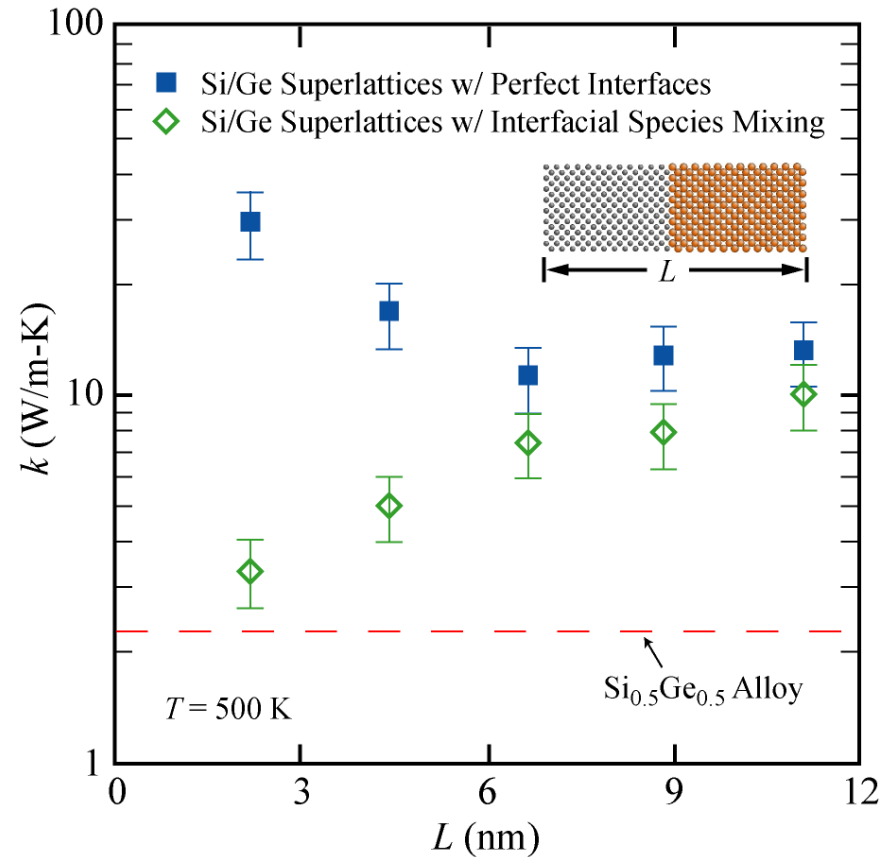
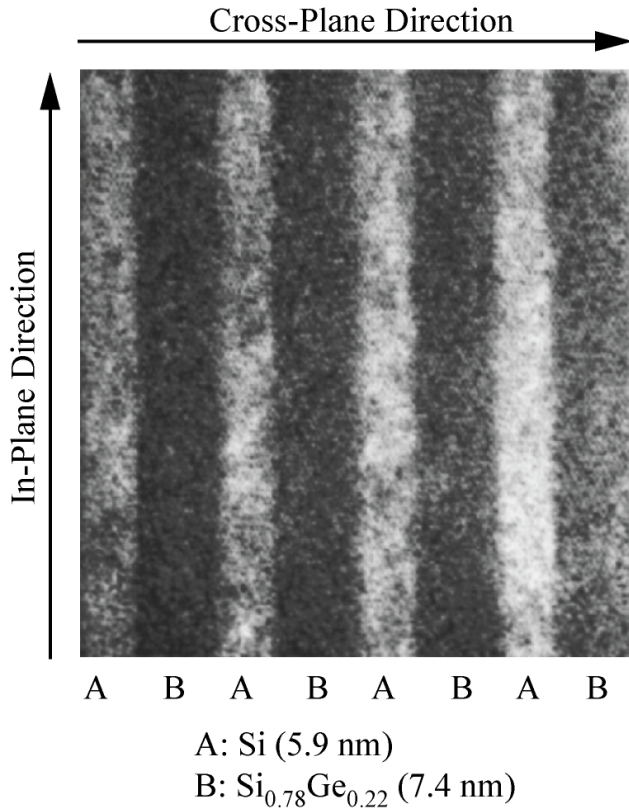
Atomistic Simulation (MD and MC)

- Electrons not explicitly considered, atom-by-atom
 - Larger lengths and times than ES methods.
The adiabatic (Born-Oppenheimer) approximation.
- Most common approaches: Molecular dynamics (MD) and Monte Carlo (MC).
- Techniques based in classical statistical mechanics.
- Quantum effects cannot be considered (that's the whole point!)
- Need a way to predict how the atoms will interact:
an interatomic potential (Lecture 3)

MD Simulation of a Water Droplet



Heat Transfer in Superlattices



Landry and McGaughey, *Physical Review B* **79** (2009) 075316.

Reality?

- MD/MC have nothing to do with modern physics.
 - They directly contradict relativity and quantum mechanics!
 - Relativity predicts the finite times needed to transfer information. In MD, it is instantaneous.
 - QM says that it is impossible to know both a particle's position and momentum at the same time (the uncertainty principle), while this is exactly what we do in MD.
- Don't worry. For what we will do, these approaches are fine.

Mesoscale Simulation

- Information about every atom may not be necessary.
- There may be a larger length/time scale that is appropriate to the problem (e.g., in granular flow).
- May not need to explicitly model the fluid surrounding a molecule. Still require discrete information, though.
- Introduce random, dissipative forces, and effective potentials.

Continuum Modeling

- Traditional engineering analysis tools: CFD, FEA.
- Do not have specific information about atoms.
- Material properties must be specified in advance.
- Some hybrid methods exist. Significant challenges at interfaces.

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