

1.1. Heating of a Copper Ball

- heating from 100°C to 150°C
- time: 30 minutes
- density $\rho = 8950 \text{ kg/m}^3$
- specific heat $c_p = 0,385 \text{ kJ/kg} \cdot ^{\circ}\text{C}$
- ball diameter: 10 cm $\leftrightarrow D = 10 \text{ cm}$

- Determine:**
- the total amount of heat transfer to the ball
 - the average rate of heat transfer to the ball
 - the average heat flux.

a) it is simply the change in its internal energy
energy transfer to system = energy increase of the system

$$Q = \Delta U = m \cdot c_{\text{avg}} (T_2 - T_1)$$

* density $\rho = \frac{m}{V}$, so $m = \rho V$

* volume of ball: $V = \frac{\pi}{6} D^3$

$$Q = \rho \cdot \frac{\pi}{6} D^3 \cdot c \cdot 50^{\circ}\text{C} = 92,6 \text{ kJ}$$

b) $\dot{Q}_{\text{avg}} = \frac{Q}{\Delta t} = \frac{92,6 \text{ kJ}}{1800 \text{ s}} = 51,4 \text{ W}$

c) $\dot{q}_{\text{avg}} = \frac{\dot{Q}}{A}$, $A = \pi D^2$, so $\dot{q}_{\text{avg}} = \frac{51,4 \text{ W}}{(0,1 \text{ m})^2 \cdot \pi} = 1636 \frac{\text{W}}{\text{m}^2}$

1 FINITE CHANGES $\Delta U = m \cdot c_{\text{avg}} \cdot \Delta T$
* for solids and liquids we can use just "c"

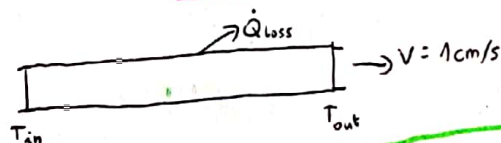
2 $Q = \int_0^{\Delta t} \dot{Q} dt$, but $\dot{Q} = \text{const}$ so $Q = \dot{Q} \Delta T$
so $\dot{Q} = \frac{Q}{\Delta T}$

2 Note that heat flux may vary with location on a surface. This is average value
A = surface area, not volume!

1.2. Cooling a stainless steel sheets.

- convection at a constant speed 1 cm/s into a chamber to be cooled.
- sheet: 5 mm thick \times 2 m wide
- temperature: $500 \text{ K} \rightarrow 300 \text{ K}$

- Determine:**
- the rate of heat loss from the stainless steel sheet inside the chamber.



a) Assumptions:

1. STEADY OPERATING CONDITIONS EXISTS
2. THE STEEL SHEET HAS CONSTANT PROPERTIES
3. CHANGES IN POTENTIAL & KINETIC ENERGY ARE NEGLIGIBLE

* the constant pressure specific heat of this steel
AVERAGE TEMPERATURE $\frac{500+300}{2} \text{ K} = 400 \text{ K}$
so $c_p = 515 \frac{\text{J}}{\text{kg} \cdot \text{K}}$

* density $\rho = 7900 \text{ kg/m}^3$

* rate of steel sheet entering and exiting the chamber

$$\dot{m} = \rho V A_c = (7900 \frac{\text{kg}}{\text{m}^3}) (0,01 \frac{\text{m}}{\text{s}}) (2 \text{ m}) (0,005 \text{ m}) = 0,79 \frac{\text{kg}}{\text{s}}$$

* the rate of heat loss

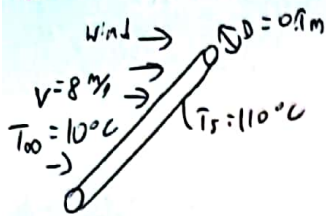
$$\dot{Q}_{\text{loss}} = \dot{m} \cdot c_p \cdot (T_{\text{in}} - T_{\text{out}}) = (0,79 \frac{\text{kg}}{\text{s}}) (515 \frac{\text{J}}{\text{kg} \cdot \text{K}}) (500 - 300 \text{ K}) = 81370 \text{ J/s} = 81,4 \text{ kW}$$

3 no change with time at specified location

3 mass flow rate

1 $\Delta U = m \cdot c_{\text{avg}} \cdot \Delta T$ implies that:
energy balance for steady-flow system:
 $\dot{Q} = \dot{m} \cdot \Delta h = \dot{m} \cdot c_p \cdot \Delta T$ (3)

7.5 Heat Loss from a Steam Pipe



I Film temperature $T_f = \frac{(T_s + T_\infty)}{2} = \frac{110 + 10}{2} = 60^\circ\text{C}$, 1 atm :

II Properties for T_f : $k = 0.02802 \frac{\text{W}}{\text{m}\cdot\text{K}}$, $Pr = 0.702$, $\nu = 1.836 \cdot 10^{-5} \frac{\text{m}^2}{\text{s}}$

III Reynolds $Re = \frac{v \cdot D}{\nu} = \frac{(8 \frac{\text{m}}{\text{s}})(0.1 \text{ m})}{1.836 \cdot 10^{-5}} = 4.299 \cdot 10^4$

IV Nusselt $\rightarrow h$

$$Nu = \frac{hD}{k} = 0.3 + \dots = 124 \quad \sim \text{simplest and less accurate}$$

$$Nu = 0.027 Re^{0.805} Pr^{\frac{1}{4}}$$

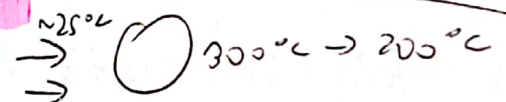
$$h = \frac{k}{D} Nu = 34.8 \frac{\text{W}}{\text{m}^2\cdot\text{K}}$$

V Newton

$$A_s = \pi DL = \pi (0.1 \text{ m}) (1 \text{ m}) = 0.314 \text{ m}^2$$

$$\dot{Q} = h \cdot A_s (T_s - T_\infty) = 1093 \text{ W}$$

7.6 Cooling of a Steel Ball by Forced Air



$\approx T_f \rightarrow \text{props.} \rightarrow \text{Reynolds} \rightarrow \text{Nusselt} \rightarrow \text{Newton}$

$$\dot{Q}_{avg} = h \cdot A_s (T_{avg} - T_\infty) = 610 \text{ W} \quad - \text{average rate of heat transfer}$$

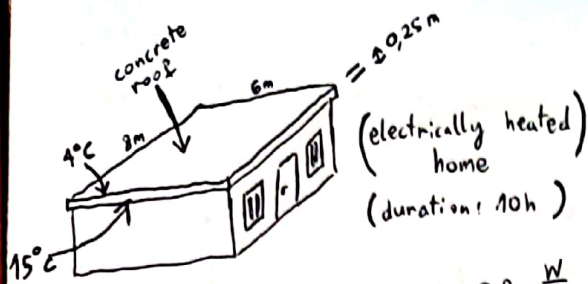
I Total heat transferred from the ball (change of its energy as it cools from 300 °C to 200 °C)

$$m = \rho V = \rho \cdot \frac{4}{3} \pi R^3 = 65.9 \text{ kg}$$

$$Q_{total} = m \cdot c_p (T_2 - T_1) = 3163000 \text{ J}$$

$$Q = \int_0^{\Delta t} \dot{Q} dt, \dot{Q}_{const} \rightarrow Q = \dot{Q} \int_0^{\Delta t} dt \rightarrow Q = \dot{Q} \Delta t \rightarrow \Delta t = \frac{Q}{\dot{Q}} = \frac{3163000 \text{ J}}{610 \text{ W}} = 5185 \text{ s} = 1.44 \text{ h}$$

1.5. The cost of heat loss through a roof



- concrete thermal conductivity $k = 0,8 \frac{W}{mK}$

Assumptions

- 1: STEADY OPERATING CONDITIONS (constant surface temperatures during night)
- 1: CONSTANT ROOF PROPERTIES

a) $\dot{Q} = kA \frac{T_1 - T_2}{L} = 0,8K \cdot 48m^2 \cdot \frac{(15-4)^\circ C}{0,25m} = 1690W = 1,69kW$

b) The amount of heat lost through the roof during a 10-hour period:

$$Q = \dot{Q} \Delta t = 1,69kW \cdot 10h = 16,9kWh$$

$$\text{Cost} = 16,9kWh \cdot \$0,08/kWh = \$1,35$$

a) the rate of heat loss through the roof that night

b) the cost of that heat loss to the home owner if the cost of electricity is \$0,08/kWh.

Determine:

4 Fourier's law

2 $Q = \int_0^{\Delta t} \dot{Q} dt \rightarrow Q = \dot{Q} \Delta t$

1.11 Heat Transfer Between Two Isothermal Plates



$T_1 = 300K$

1. Average temperature $= \frac{300+200[K]}{2} = 250K$

2. k from Table

a) air $k = 0,0219 \frac{W}{mK}$, b) vacuum, c) urethane insulation $k = 0,026 \frac{W}{mK}$, d) superinsulation $k = 0,0002 \frac{W}{mK}$

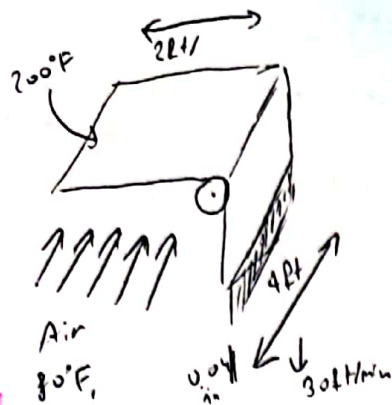
only radiation!

$$\begin{cases} \dot{Q}_{cond} = kA \frac{T_1 - T_2}{L} = 219W \\ Q_{rad} = \epsilon \sigma A (T_1^4 - T_2^4) = 369W \\ Q_{total} = 219W + 369W = 588W \end{cases}$$

$Q_{rad} = \dots$

7.3 Cooling of plastic sheets by Forced Air

FORCED



Determine:
a) the rate of heat transfer from the plastic sheet to air by forced convection and radiation

b) the temperature of plastic sheet at the end of the cooling section

$$\rho = 75 \text{ lbm/ft}^3, C_p = 0.4 \text{ Btu/lbm}^\circ\text{F}, \epsilon = 0.9$$

air - ideal gas

a)

I Film temperature and properties

$$T_f = \frac{T_s + T_\infty}{2} = \frac{200 + 80}{2} = 140^\circ\text{F}, \text{ 1 atm pressure}$$

$$k = 0.01623 \text{ Btu/h}^\circ\text{F}$$

$$\mu = 0.7202 \text{ lbm/ft}^\circ\text{F}$$

$$\nu = 0.204 \cdot 10^{-3} \text{ ft}^2/\text{s}$$

II Reynolds - at the end of the air flow across the plate

$$Re_L = \frac{VL}{\nu} = \frac{(10 \text{ ft/s})(4 \text{ ft})}{0.204 \cdot 10^{-3} \text{ ft}^2/\text{s}} = 1.961 \cdot 10^5 < 5 \cdot 10^5 \Rightarrow \text{Laminar flow.}$$

III Nusselt for laminar flat plate flow \rightarrow calculate h

$$Nu = \frac{hL}{k} = 0.664 Re_L^{0.5} \cdot \mu^{1/3} = 263.6$$

then

$$h = \frac{k}{L} Nu = 1.07 \text{ Btu/h}^\circ\text{F} \cdot \text{ft}^2$$

$$A_s = (2 \text{ sides})(4 \text{ ft})(2 \text{ ft}) = 16 \text{ ft}^2$$

IV \dot{Q}_{conv} & \dot{Q}_{rad} for the whole plate

$$\dot{Q}_{\text{conv}} = h \cdot A_s (T_s - T_\infty) = (1.07)(16 \text{ ft}^2)(200 - 80^\circ\text{F}) = 2054 \text{ Btu/h}$$

$$\dot{Q}_{\text{rad}} = \epsilon \cdot \sigma \cdot A_s (T_s^4 - T_\infty^4) = 2585 \text{ Btu/h}$$

$$\dot{Q}_{\text{total}} = \dot{Q}_{\text{conv}} + \dot{Q}_{\text{rad}} = 2054 + 2585 \text{ (Btu/h)} = 4639 \text{ Btu/h}$$

\dot{Q}_{total}
FROM THE PLASTIC
SHEET TO AIR

b) mass of the plastic rolling out per unit time (mass flow rate)

$$\dot{m} = \rho \cdot A_c \cdot V_{\text{plastic}} = (75 \text{ lbm/ft}^3) \left(\frac{4 \cdot 0.04 \text{ (ft}^2\text{)}}{12} \right) \left(\frac{30 \text{ (ft/s)}}{60} \right) = 0.5 \text{ lbm/s}$$

II Energy balance on the cooled section

$$\dot{Q} = \dot{m} C_p (T_2 - T_1) \rightarrow T_2 = T_1 + \frac{\dot{Q}}{C_p \dot{m}}$$

\dot{Q} is negative \Rightarrow HEAT LOSS

$$T_2 = 200^\circ\text{F} + \frac{-4639}{0.5 \cdot 0.4} \left(\frac{1 \text{ h}}{3600 \text{ s}} \right) = 193.6^\circ\text{F}$$

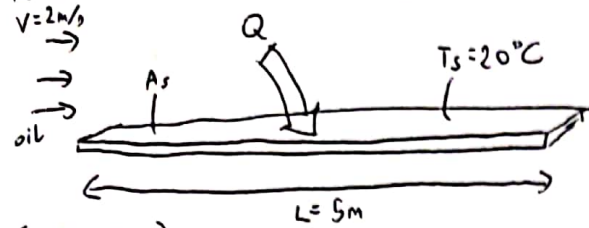
7.1 Flow of Hot Oil over a Flat Plate

Determine:

FORCED

$$T_{\infty} = 60^{\circ}\text{C}$$

$$V = 2 \text{ m/s}$$



- a) The total drag force?
- b) The rate of heat transfer per unit width of the entire plate?

(Critical Reynolds) = flow becomes turbulent \rightarrow flow over a flat plate $Re = \frac{V_{\infty} L}{\nu} = 5 \cdot 10^5$

1. Properties of oil at film temperature $\left(\frac{60+20}{2}\right) = 40^{\circ}\text{C}$:

$$\rho = 876 \text{ kg/m}^3, \quad Pr = 2962, \quad k = 0.1444 \frac{\text{W}}{\text{mK}}, \quad \nu = 2.485 \cdot 10^{-4} \frac{\text{m}^2}{\text{s}}$$

2. Reynolds:

$$Re_L = \frac{VL}{\nu} = \frac{(2 \text{ m/s})(5 \text{ m})}{2.485 \cdot 10^{-4} \frac{\text{m}^2}{\text{s}}} = 4.024 \cdot 10^4, \quad \text{so } Re_L < Re_{\text{critical}} \Rightarrow \text{LAMINAR FLOW}$$

(a) so average friction coefficient:

$$C_f = 1.33 Re_L^{-0.5} = 0.00663 \quad \text{and} \quad F_D = C_f A \frac{\rho V^2}{2} = 58.1 \text{ N}$$

(b)

The Nusselt Number - laminar flow for a flat plate

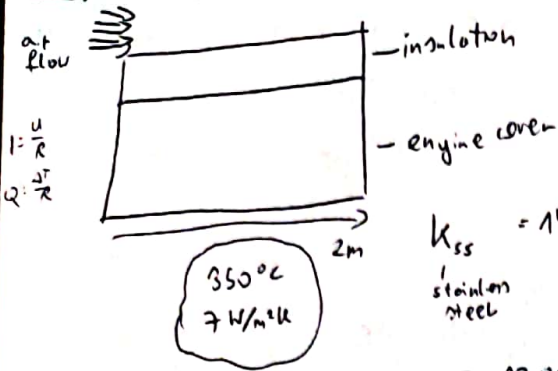
$$Nu = \frac{hL}{k} = 0.664 Re_L^{0.5} Pr^{\frac{1}{3}} = 1913$$

then:

$$h = \frac{Nu \cdot k}{L} = 55.25 \frac{\text{W}}{\text{m}^2\text{K}}$$

$$\dot{Q} = h A_s (T_{\infty} - T_s) = 11,050 \text{ W}$$

7.2. Prevention of Fire Hazard in the event of Oil Leakage



Test: increase the cooling air velocity by 10%.

$$k_{ss} = 14 \frac{W}{mK} \quad , \quad k_{ins} = 0.5 \frac{W}{mK}$$

stainless steel insulation

$$V = 7 \frac{m}{s}$$

1. The properties of air: at $T_f = 170^\circ C \Rightarrow \mu = 0.03235 \frac{W}{mK}$, $\nu = 2.522 \cdot 10^{-5} \frac{m^2}{s}$, $Pr = 0.7073$

2. Reynolds for 10% cooling air velocity

$$Re_c = \frac{VL}{\nu} = \frac{(2.7 m/s) \cdot 2m}{2.522 \cdot 10^{-5}} = 610,621 > 5 \cdot 10^5, \text{ but } < 10^7,$$

so

$$Nu = \frac{hL}{k} = (0.037 Re_c^{0.8} - 871) \cdot Pr^{1/4} = 625,77.$$

then

$$h = h_o = \frac{Nu \cdot k}{L} = \frac{625,77 \cdot 0.03235}{2} = 10,122 \frac{W}{m^2K}.$$

3. Thermal resistances

$R_{conv,o}$	$= \frac{1}{h_o A}$
R_{ins}	$= \frac{L_{ins}}{k_{ins} \cdot A}$
R_{ss}	$= \frac{L_{ss}}{k_{ss} \cdot A}$
$R_{conv,i}$	$= \frac{1}{h_i A}$

$$R_{total} = R_{conv,i} + R_{ss} + R_{ins} + R_{conv,o} = \frac{1}{h_i} + \frac{L_{ss}}{k_{ss}} + \frac{L_{ins}}{k_{ins}} + \frac{1}{h_o}$$

$$AR_{total} = \frac{1}{7 W/m^2K} + \frac{0.01m}{14 W/mK} + \frac{0.005m}{0.5 W/mK} + \frac{1}{10,122 \frac{W}{m^2K}} =$$

$$0.25237 m^2K/W$$

$$AR_{conv,o} = \frac{1}{h_o} = \frac{1}{10,122 W/m^2K} = 0.09879 \frac{m^2K}{W}$$

4. Heat flux through the layers.

$$\dot{q} = \frac{\dot{Q}}{A} = \left(\frac{R_{total}}{R_{total}} \right) \cdot \frac{\dot{Q}}{A} = \frac{T_{o,i} - T_{o,o}}{A \cdot R_{total}} = \frac{T_{s,o} - T_{o,o}}{A \cdot R_{conv,o}} \Rightarrow T_{s,o} = \frac{R_{conv,o}}{R_{total}} (T_{o,i} - T_{o,o}) + T_{o,o}$$

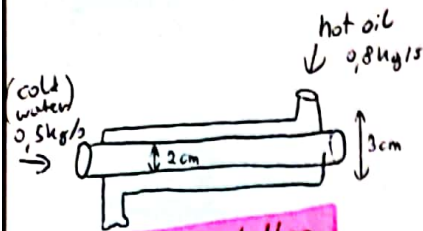
$$\dot{Q} = \frac{\Delta T}{R}$$

$$T_{s,o} = \frac{0.09879}{0.25237} (350 - 60)^\circ C + 60^\circ C = 173.5^\circ C$$

< 180°C



11.1 Overall Heat Transfer Coefficient of a Heat Exchanger.



- negligible inner tube thickness

$$T_{\text{water}} = 45^\circ\text{C}$$

$$T_{\text{oil}} = 80^\circ\text{C}$$

& Properties of oil at 80°C

$$\rho, \mu, k, \nu$$

I Read from tables
Properties for water at 45°C
 ρ, μ, k, ν

II U can be determined from: $\frac{1}{U} \approx \frac{1}{h_i} + \frac{1}{h_o}$

III Find h_i and h_o

$$Re = \frac{VD}{\nu}, \text{ and you don't know } V? \text{ Don't worry, use } \dot{m} \left(\dot{m} = \rho A_c V \right) \rightarrow V = \frac{\dot{m}}{\rho A_c}$$

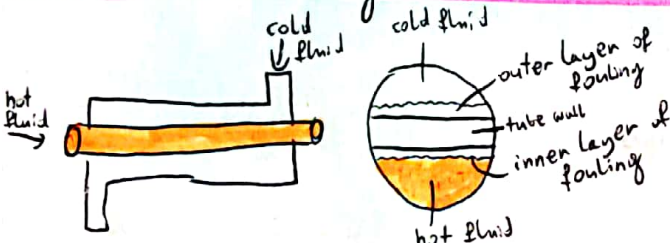
\rightarrow Nusselt $\rightarrow h$

IV Paste h_i & h_o to II

$$U = \frac{1}{\frac{1}{h_i} + \frac{1}{h_o}} = \frac{1}{\frac{1}{7663 \frac{\text{W}}{\text{m}^2\text{K}}} + \frac{1}{75.2 \frac{\text{W}}{\text{m}^2\text{K}}}} = 74.5 \frac{\text{W}}{\text{m}^2\text{K}}$$

\leftarrow very close to smaller h

11.2. Effect of Fouling on the Overall Heat Transfer Coefficient



inner tube:

$$D_i = 1.5 \text{ cm inner diameter}, h_i = 800 \text{ W/m}^2\text{K}$$

$$D_o = 1.9 \text{ cm outer diameter}, h_o = 1200 \text{ W/m}^2\text{K}$$

outer shell \rightarrow inner diameter 3.2 cm

Fouling factor $R_{f,i} = 0.0004 \text{ m}^2\text{K/W}$ on the tube side

$R_{f,o} = 0.0001 \text{ m}^2\text{K/W}$ on the shell side

Find a) the thermal resistance of the heat exchanger per unit length
b) the overall heat transfer coefficient U_i, U_o (based on inner, outer) (surfaces of the tube)

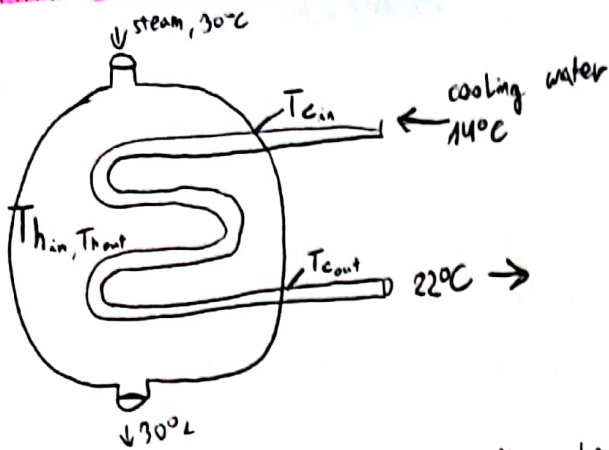
I Use this eq.

$$R = \frac{1}{UA} = \frac{1}{U_i A_i} = \frac{1}{U_o A_o} = \frac{1}{h_i A_i} + \frac{R_{f,i}}{A_i} + \frac{\ln(D_o/D_i)}{2\pi k L} + \frac{R_{f,o}}{A_o} + \frac{1}{h_o A_o}, \text{ where } A_i = \pi D_i L$$

$$\text{to } R = 0.0532^\circ\text{C/W}$$

$$R = \frac{1}{U_i A_i} \rightarrow U_i = \frac{1}{R \cdot A_i} = 339 \frac{\text{W}}{\text{m}^2\text{K}}$$

11.3 The Condensation of Steam in a Condenser



Determine - the mass flow rate of the cooling water needed
- the rate of condensation of the steam in the condenser

* can be treated as a counter-flow heat exchanger, since the temperature of one of the fluids remains constant (the steam)

The temperature difference between the steam and the cooling water at the two ends of the condenser is:

$$\Delta T_1 = T_{h,in} - T_{c,out} = 30 - 22 = 8 [^{\circ}\text{C}]$$

$$\Delta T_2 = T_{h,out} - T_{c,in} = 30 - 14 = 16 [^{\circ}\text{C}]$$

but the proper average temperature is... (not the arithmetic!), a little less.

$$\Delta T_{lm} = \frac{\Delta T_1 - \Delta T_2}{\ln(\Delta T_1 / \Delta T_2)} = \frac{8 - 16}{\ln(8/16)} = 11.5^{\circ}\text{C}$$

$$\text{Then } \dot{Q} = UA_s \Delta T_{lm} = (2100 \frac{\text{W}}{\text{m}^2\text{K}})(45\text{m}^2)(11.5^{\circ}\text{C}) = 1087\text{ kW}$$

* so, steam will lose heat at a rate of 1087 kW as it flows through the condenser, and the cooling water will gain practically all of it, since the condenser is well insulated

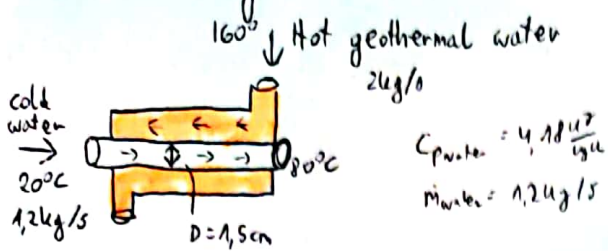
$$\dot{Q} = [\dot{m} c_p (T_{out} - T_{in})]_{\text{cooling water}} = [\dot{m} h_{fg}]_{\text{steam}}, \text{ so:}$$

$$\dot{m}_{\text{cooling water}} = \frac{\dot{Q}}{c_p (T_{out} - T_{in})} = \frac{1087\text{ kJ/s}}{(4.184\text{ kJ/kgK})(22-14)^{\circ}\text{C}} = 32.5\text{ kg/s}$$

$$\dot{m}_{\text{steam}} = \frac{\dot{Q}}{h_{fg}} = \frac{1087\text{ kJ/s}}{2431\text{ kJ/kg}} = 0.45\text{ kg/s}$$

so we have to circulate about 72 kg of cooling water for each kg of steam condensing to remove the heat released during the condensation process.

11.4 Heating Water in a Counter-Flow Heat Exchanger



Determine: the length of the heat exchanger required to achieve the desired heating

$$c_{p, \text{water}} = 4.18 \frac{\text{kJ}}{\text{kg} \cdot ^\circ\text{C}}$$

$$\dot{m}_{\text{water}} = 1.2 \text{ kg/s} \quad \dot{m}_{\text{geothermal}} = 2 \text{ kg/s}$$

I The rate of heat transfer we can determine from:

$$\dot{Q} = [\dot{m} c_p (T_{\text{out}} - T_{\text{in}})]_{\text{water}} = (1.2 \text{ kg/s}) (4.18 \frac{\text{kJ}}{\text{kg} \cdot ^\circ\text{C}}) (80 - 20)^\circ\text{C} = 301 \text{ kW}$$

II The outlet temperature of geothermal water:

$$\dot{Q} = [\dot{m} c_p (T_{\text{in}} - T_{\text{out}})]_{\text{geothermal}}$$

$$T_{\text{out}} = T_{\text{in}} - \frac{\dot{Q}}{\dot{m} c_p} = 125^\circ\text{C}$$

III Temperatures

$$\Delta T_1 = T_{\text{h, in}} - T_{\text{c, out}} = 160 - 80 = 80^\circ\text{C}$$

$$\Delta T_2 = T_{\text{h, out}} - T_{\text{c, in}} = 125 - 20 = 105^\circ\text{C}$$

$$\begin{aligned} & \left[\text{hot IN} - \text{cold OUT} \right] \\ & \left[\text{hot OUT} - \text{cold IN} \right] \end{aligned}$$

IV Logarithmic mean temperature

$$\Delta T_{\text{lm}} = \frac{\Delta T_1 - \Delta T_2}{\ln(\Delta T_1 / \Delta T_2)} = \frac{80 - 105}{\ln(80/105)} = 91.9^\circ\text{C}$$

V Surface Area of Heat Exchanger

$$\text{from } \dot{Q} = U A_s \Delta T_{\text{lm}} \rightarrow A_s = \frac{\dot{Q}}{U \Delta T_{\text{lm}}} = \frac{301000}{(640 \frac{\text{W}}{\text{m}^2 \cdot ^\circ\text{C}}) (91.9^\circ\text{C})} = 5.12 \text{ m}^2$$

$$\text{then: } A_s = \pi D L \rightarrow L = \frac{A_s}{\pi D} = \frac{5.12 \text{ m}^2}{\pi \cdot 0.015 \text{ m}} = 109 \text{ m.}$$