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Green Metal from Thin Air: Electrowinning Iron Using Direct Air Capture CO₂ as Reductant in a Solar-Powered Reactor

Grev Williem Manurung¹

¹Department of Metallurgical Engineering, Institut Teknologi Del, Laguboti, Indonesia Corresponding author: tms23036@students.del.ac.id

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Abstract

This paper presents a study on the development of a carbon-negative steelmaking process that utilizes captured CO₂ as a reducing agent in an electrometallurgical system. The main objective is to demonstrate a low-emission pathway for converting Fe²⁺ into solid iron while simultaneously sequestering atmospheric CO₂. The methodology includes the design and operation of a 1 L tubular electrochemical reactor equipped with a 3D-printed porous titanium cathode, operating at 60 °C and 200 A m⁻², powered entirely by a 5 kWp solar panel. The results show that the process produces 99.7% pure iron with an energy consumption of 3.8 MWh per ton, representing a 35% reduction compared to the conventional blast furnace route, while capturing 0.4 tonnes of CO₂ per ton of iron produced as a calcium carbonate by-product. This research is expected to contribute to decarbonizing the steel industry, promoting sustainable metallurgy, and advancing green technologies aligned with CBAM and US-IRA initiatives.

Keywords: electrowinning iron, CO₂ utilization, carbon-negative steel, solar reactor, urban mining