

## Q2. Sources, atoms, and spectra

### A. Light source

In its proper reference frame, a point source emits light in the form of a divergent conical beam, with the angular width of  $90^\circ$  (from  $-45^\circ$  to  $+45^\circ$  with respect to the cone axis). In a reference frame which moves towards the source with an unknown speed  $v$ , the angular width of the beam is of only  $60^\circ$  (from  $-30^\circ$  to  $+30^\circ$  with respect to the same cone axis). The light speed in vacuum is  $c = 2.998 \cdot 10^8 \frac{\text{m}}{\text{s}}$ .

<b>A</b>	Determine the speed $v$ of the source.	2.50 p.
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### B. Balmer emission spectrum

The spectral resolving power of a spectrometer is  $R = 5 \cdot 10^5$ . The spectrometer is used to observe the Balmer series in the emission spectrum of the hydrogen atom (the visible domain).

*Note: The possible mechanisms of broadening of the spectral lines (Lorentzian, Gaussian, etc.) will not be considered.*

<b>B.1</b>	Express the mathematical definition of the spectral resolving power of the instrument.	0.25 p.
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<b>B.2</b>	Determine the highest value for the principal quantum number $n$ of the energy level for which the spectral line emitted by an atom for the transition to the level $n' = 2$ can still be distinctly resolved by the instrument, with respect with its neighbours.	2.25 p.
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### C. Absorption spectra

The energy levels of an atom are given by  $E_n = -\frac{A}{n^2}$ , where  $n$  is an integer and  $A$  is a positive constant. Among the adjacent spectral lines which, at room temperature, the atom can absorb, two have the wavelengths  $97.5 \text{ nm}$  and  $102.8 \text{ nm}$ , respectively. The elementary electric charge is  $e = 1.602 \cdot 10^{-19} \text{ C}$ , the speed of light in vacuum is  $c = 2.998 \cdot 10^8 \frac{\text{m}}{\text{s}}$ , and Planck's constant is  $h = 6.626 \cdot 10^{-34} \text{ J}\cdot\text{s}$ .

<b>C.1</b>	Find the values of the quantum numbers $n$ of the energy levels implied in the transitions.	3.00 p.
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<b>C.2</b>	Determine the value of the constant $A$ in joule and in electron-volt.	1.50 p.
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<b>C.3</b>	Identify the nature of the atom and justify the choice made.	0.50 p.
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**Q3. Sources, atoms, and spectra**

**Answer sheet**

<b>A</b>	Final expression for the speed of the source	Numerical value for the speed of the source	2.50 p.
	$v =$	$v =$	

<b>B.1</b>	$R =$	0.25 p.
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<b>B.2</b>	$n =$	2.25 p.
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<b>C.1</b>	$n =$	3.00 p.
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<b>C.2</b>	in joule	$A =$	1.00 p.
	in electron-volt	$A =$	0.50 p.

<b>C.3</b>		0.50 p.
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