

**FIGURE 2-3** The scientific method is not a stepwise process. Scientists may repeat steps many times before there is sufficient evidence to formulate a theory. You can see that each stage represents a number of different activities.

## **Theorizing**

When the data from experiments show that the predictions of the hypothesis are successful, scientists typically try to explain the phenomena they are studying by constructing a model. A **model** in science is more than a physical object; it is often an explanation of how phenomena occur and how data or events are related. Models may be visual, verbal, or mathematical. One of the most important models in chemistry is the atomic model of matter, which states that matter is composed of tiny particles called atoms.

If a model successfully explains many phenomena, it may become part of a theory. The atomic model is a part of the atomic theory, which you will study in Chapter 3. A **theory** is a broad generalization that explains a body of facts or phenomena. Theories are considered successful if they can predict the results of many new experiments. Examples of the important theories you will study in chemistry are kinetic-molecular theory and collision theory. Figure 2-3 shows where theory fits in the scheme of the scientific method.

## SECTION REVIEW

- 1. What is the scientific method?
- 2. Which of the following are quantitative?
  - a. the liquid floats on water
  - b. the metal is malleable
  - c. the liquid has a temperature of 55.6°C

- 3. How do hypotheses and theories differ?
- **4.** How are models related to theories and hypotheses?
- **5.** What are the components of the system in the graduated cylinder shown on page 38?