D

## Useful Numerical Values

FUNDAMENTAL CONSTANTS		
Universal gas constant (R*)	8.3143 J K <sup>-1</sup> mol <sup>-1</sup>	
Boltzmann's constant (k)	$1.38 \times 10^{-23} \mathrm{J \ K^{-1}}$	
Stefan–Boltzmann constant ( $\sigma$ )	$5.67 \times 10^{-8}  \mathrm{W \ m^{-2} \ K^{-4}}$	
Planck's constant (h)	$6.63 \times 10^{-34} \mathrm{J \ s}$	
Speed of light (c*)	$2.998 \times 10^8  \mathrm{m \ s^{-1}}$	
Gravitational constant	$6.67 \times 10^{-11} \mathrm{N} \;\mathrm{m}^2 \mathrm{kg}^{-2}$	
SUN		
Luminosity	$3.92 \times 10^{26} \mathrm{W}$	
Mass	$1.99 \times 10^{30} \mathrm{kg}$	
Radius	$6.96 \times 10^{8} \mathrm{m}$	
EARTH		
Average radius (a)	$6.37 \times 10^6 \mathrm{m}$	
Equatorial radius	$6.378 \times 10^6 \mathrm{m}$	
Polar radius	$6.357 \times 10^6 \mathrm{m}$	
Standard gravity (g)	$9.80665~{\rm m~s^{-2}}$	
Mass of Earth	$5.983 \times 10^{24} \mathrm{kg}$	
Mass of ocean	$1.4 \times 10^{21}  \mathrm{kg}$	
Mass of atmosphere	$5.3 \times 10^{18}  \mathrm{kg}$	
Mean angular rotation rate ( $\Omega$ )	$7.292 \times 10^{-5}  \mathrm{rad \ s^{-1}}$	
Total solar irradiance $(S_0)$	$1360.8 \pm 0.5 \text{ W m}^{-2}$	
Mean distance from Sun ( <i>d</i> )	$1.496 \times 10^{11} \mathrm{m}$	

DRY AIR		
Average molecular weight (m <sub>a</sub> )	28.97 g mol <sup>-1</sup>	
Gas constant $(R) - R^*/m_a$	$287  \mathrm{J  K^{-1}  kg^{-1}}$	
Density at 0°C and 101325 Pa	1.293 kg m <sup>-3</sup>	
Specific heat at constant pressure ( $c_p$ )	$1004\mathrm{JK^{-1}kg^{-1}}$	
Specific heat at constant volume $(c_v)$	$717  \mathrm{J  K^{-1}  kg^{-1}}$	
WATER		
Molecular weight $(m_{\rm w})$	18.016 g mol <sup>-1</sup>	
Gas constant for vapor ( $R_v = R^*/m_w$ )	$461  \mathrm{J  K^{-1}  kg^{-1}}$	
Density of pure water at 0°C	$1000 \text{ kg m}^{-3}$	
Density of ice at 0°C	917 kg m <sup>-3</sup>	
Specific heat of vapor at constant pressure	$1952 \mathrm{J} \mathrm{K}^{-1} \mathrm{kg}^{-1}$	
Specific heat of vapor at constant volume	$1463  \mathrm{J  K^{-1}  kg^{-1}}$	
Specific heat of liquid water at 0°C	$4218  \mathrm{J  K^{-1}  kg^{-1}}$	
Specific heat of ice at 0°C	$2106  \mathrm{J  K^{-1}  kg^{-1}}$	
Latent heat of vaporization at 0°C	$2.5  imes 10^6  \mathrm{J  kg^{-1}}$	
Latent heat of vaporization at 100°C	$2.25 \times 10^6 \mathrm{Jkg^{-1}}$	
Latent heat of fusion at 0°C	$3.34 \times 10^5 \mathrm{Jkg^{-1}}$	