Gabriel Emerson g teoros HW3 1.) Bandgap in eV=[1.12V] Due 1/29 eV-> J= 1.12.1.6×10°2 = 1.7925 2) $N_0 = 2 \times 10^{17} \text{cm}^3$ $N_0 = 3 \times 10^{17} \text{cm}^3$ ND>NA N= ND-NA+ - (ND-NA)+4ni2 ni = 100/cm3 n=1×10"/em3 P= 11 = (10'0/cm3) 1x10 /en3 p= 1000/cm3 NA>NO P=NA-NO+ (NA-NO)241.2 3.) NA=5x13' /cm3 P=4x1016/cm3 n = 100/cm3 $n = \frac{10^{\circ}/\text{cm}^3}{P} = \frac{10^{\circ}/\text{cm}^3}{4 \times 10^{\circ}/\text{cm}^3} = 2500 \text{ cm}^3 = n$ 4.) a.) ND=1×105/cm3 } Silicon 4.) a.) ND=1×105/cm3 } Silicon Dollar Silicon 111 C.) Boron must be added to convert to P type. This is due to Boron hourng 3 valence electrons. This has one bonding place that is unsatisfied and will accept an electron from Si atoms, leaving a whole in its place.

N= n:2 = (1x10 km3)2 Gabriel Emerson ate0002 4.) d) P=5x10'59cm3 0 NP= 1x 101/cm3 NA=1x100/cm3 / n=>20000/cm3 P-n= NA-NA 5x10'5/cm3-20,000/cm3=NA-(1x10'6/cm3) NA= 1.5 × 10 6/cm3 e.) In (drift) = -MNE = -500 x 20,000 = -1 x 10 cm/s Jn(dift) = -anVn = -1.6×10-9 x 20,000 -1×107cm => 3,2×10-8 A/cm2 Vp(drift) = -MpE = 100cm = 20,000 = [2×106 cm/s] JP(driff) = ap VP = 1.6×10"A.s × 5×10" x 2×10"cm =>1600 A/cm2 5.) P-type Si Water SX10 /cmb 6.) 200 cm² . x 10.1.107 = 1×10° cm -> 10 cm²/s 10cm = 200cm = x=0.05 V

Gabriel Emerson glecoop HW3 7.) E=he/2 h=6.626×10345 C= 3×10'0 cm/s Energy required = 1.12(1.6 x 10"5)=1.79 x 10"5 E=hc/2 ->)= hc => 6.626×10-34 J (3×10'°cm/s) 1.79×10-195 X= 1.049x10 m 2=1,649 mm 8.)a.) Germanium has one more valence electron than Indian, SO I would expect it to be a donor. b.) Germanium has one less valence electron than phospherous, so I would expect Germonium to act as an acceptor.