

CSC258 Winter 2016

Computer Organization

Lecture 1

Thanks to Andrew Petersen, Myrto Papadopoulou and Steve Engels for previous course materials.

Larry Zhang

Office: DH-3076

Email: ylzhang@cs.toronto.edu

Today's outline

- **Why** CSC258
- **What** is in CSC258
- **How** to do well in CSC258

- Start learning

Why take CSC258?

As a computer science student, it is **embarrassing** to not know how a computer works.

You have to take CSC258, then CSC369 to know how computers work.

More specifically...

- Why are computers built with 1's and 0's and boolean logics?
- How does the computer do everything with just 1's and 0's?
- What is stored in that "minecraft.exe" file, what exactly happens when I double-click on it?
- How does the CPU run an if-statement, or for loop, or recursion?

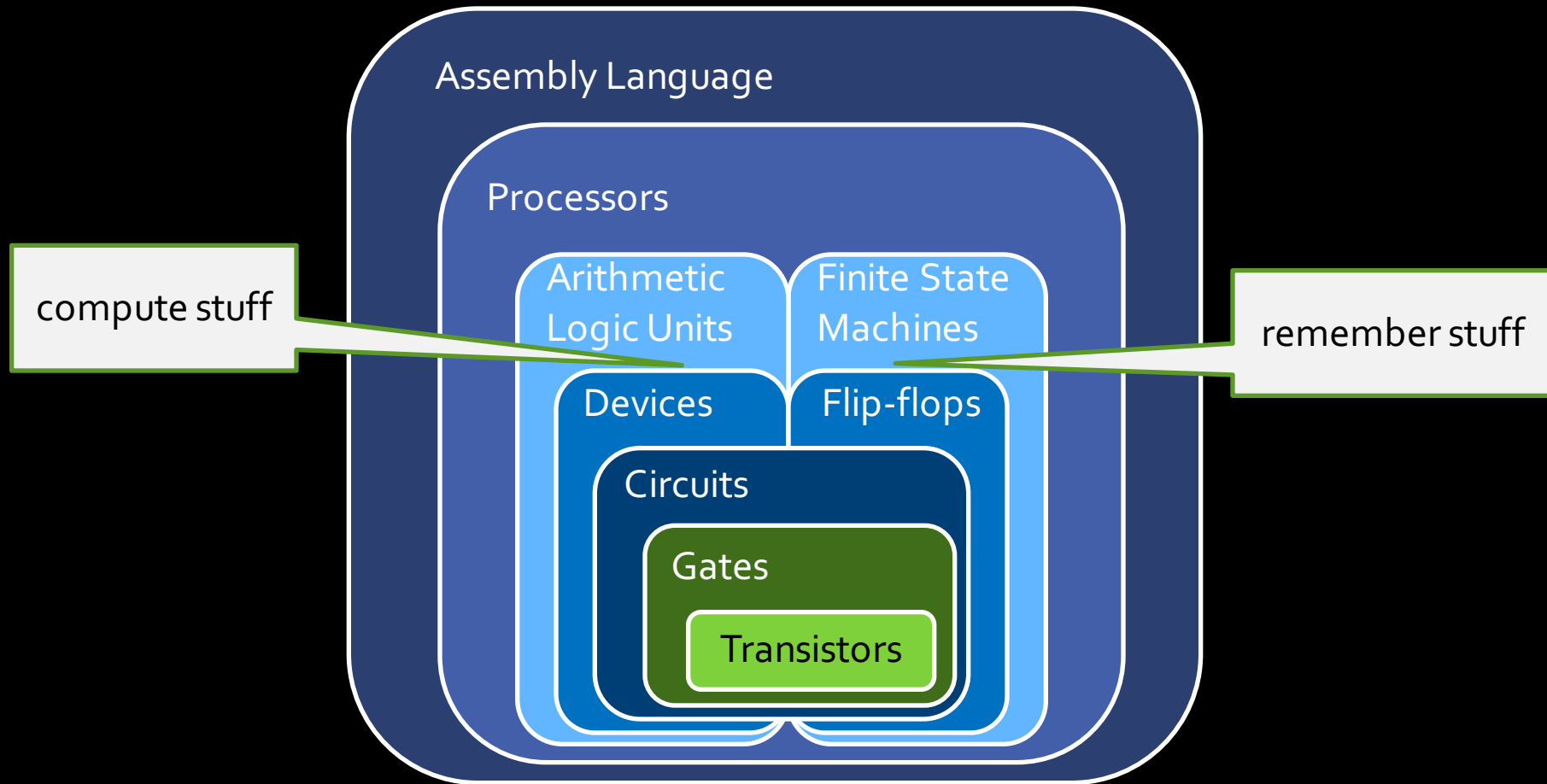
CSC258 has all the answers!

CSC258 Course Goals

- You'll know exactly how a computer is physically built, and you can build one if you want.
- Based on your hardware knowledge, you will be able to engineer the performance of your software like never before.
- Basically, after taking this course, the computer will never look the same to you again.

What's in CSC258?

The architecture of a computer **hardware**, level by level, bottom-up



We learn the whole real deal

- From atom level to assembly level
- Above the assembly level is the Operating System, whose main job is **virtualization**, i.e., create convenient illusions.
- Everything you learn from every CS course are all **illusions** except for **CSC258**

How to do well in CSC258

First of all ...

Be interested

Course website

<http://www.cs.toronto.edu/~ylzhang/csc258w16/>

All course materials are here.

Lectures (Monday 3-5pm, IB-345)

- Learn the concepts and theories.

A tip for lectures

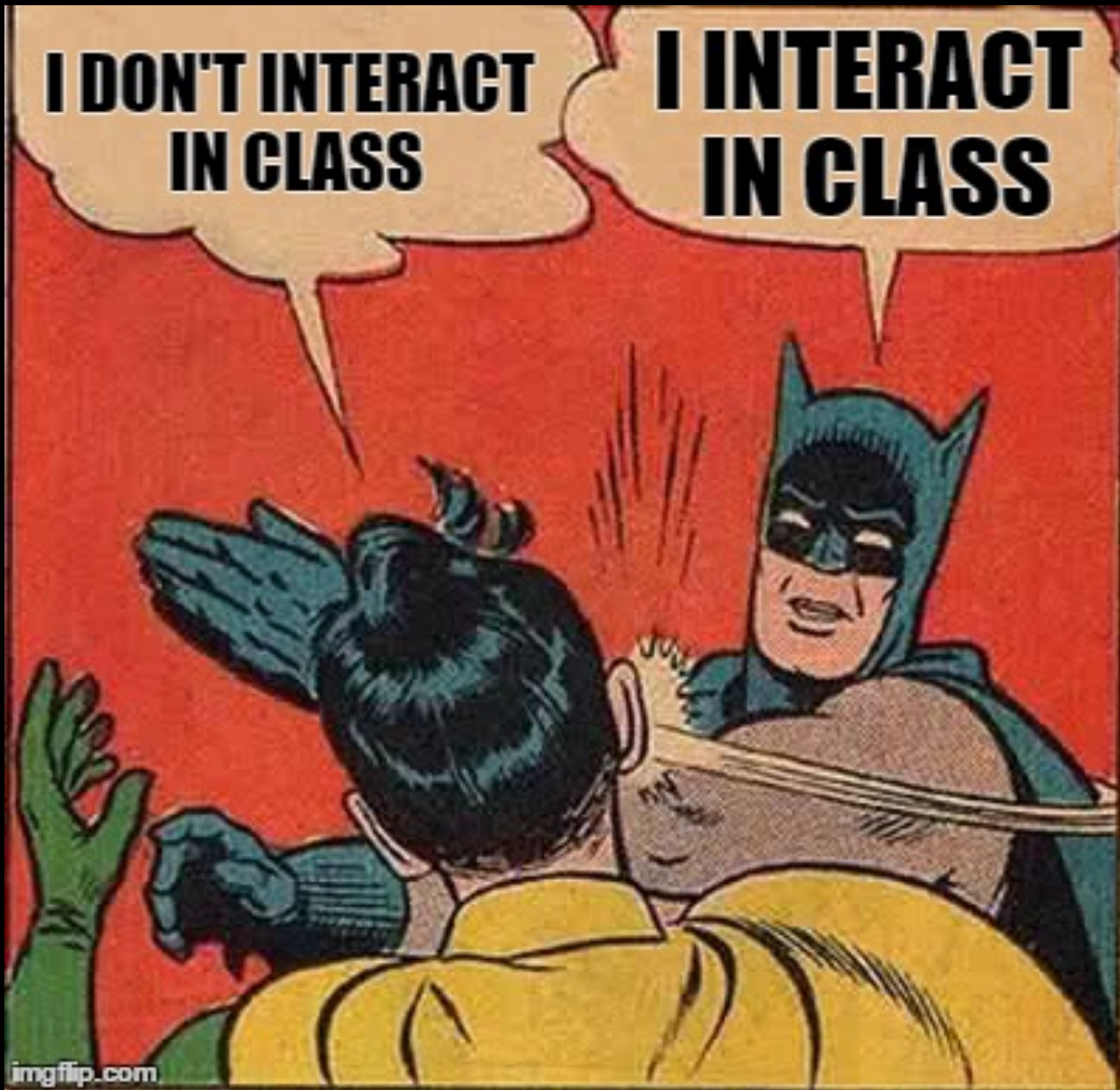
Get involved in classroom interaction

- Answering a question
- Making a guess / bet / vote
- Back-of-envelope calculations

Emotional involvement makes the brain remember better!

**I DON'T INTERACT
IN CLASS**

**I INTERACT
IN CLASS**



Labs ! (start from Week 2)

- Hands-on exercises in which you will build real pieces of hardware.
- Work in **pairs**.
- ONLY go to the tutorial section that you are registered to on ROSI/ACORN. If you want to switch lab section, find someone who is willing to change (use Piazza) and let me know.
- Most of the labs involve work **before** the lab.
- **Lab marks are NOT easy free giveaways!**

If you are on the waiting list right now, and still hope to get in, send me an email this week to inform me, so I can make arrangement about labs.

Exams !

- Midterm
 - In class 50 minutes, Feb 22
- Final exam
 - Some time in April

Marking scheme

- Labs: $40\% \times 10 = 40\%$
 - Midterm: 20%
 - Final exam: 40%
- 100%

Must get 40% of final exam to pass

Bonus Mark!

There is a **quiz** at the end of each week's lecture. About what's learned in this week and/or last week. (start from Week 2)

Each quiz earns you some **points**.

At the end of the terms, bonus marks will be awarded according to the **ranking** of quiz points earned.

Top 20% students gets 3% bonus in final grade; top 50% gets 2% bonus in final grade.

The purpose of this is to give you some incentive to keep up with the lecture material, so that you can succeed in exams.

Discussion board (Piazza)

<https://piazza.com/utoronto.ca/winter2016/csc258h5>

- All course announcements will be posted on Piazza.
- **Daily** reading is required.

Office hours

Monday 5-6:30pm, priority for CSC258 students

Tuesday 5-6:30pm, priority for CSC309 students

Friday 5-6:30pm, priority for CSC263 students

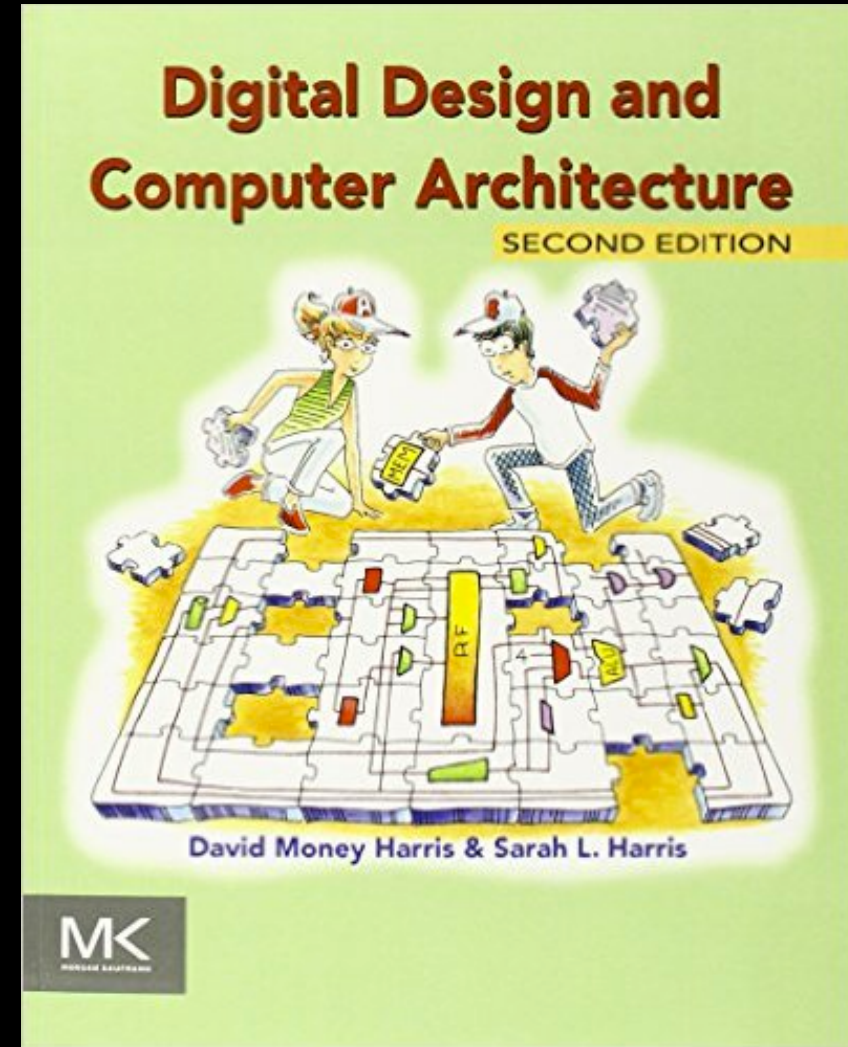
Going to office hours is very helpful, also it is a good habit to foster in college.



Textbook: DDCA

Digital Design and Computer Architecture, 2nd edition, 2012 by David Harris, Sarah Harris

Available online at UofT library (link in course info sheet)



Weekly feedback form

<http://goo.gl/forms/o248ETWgnS>

Good feedbacks directly improves your learning experience.

Have your issues addressed weekly rather than termly.

Checklist: How to do well

- Be interested
- Check course website and Piazza regularly
- Go to lectures, interact in class
- Be prepared for the quiz
- Read the slides, read the book
- Work hard on the labs
- Go to office hours
- Discuss on Piazza
- Give weekly feedback
- Do well in midterm and final

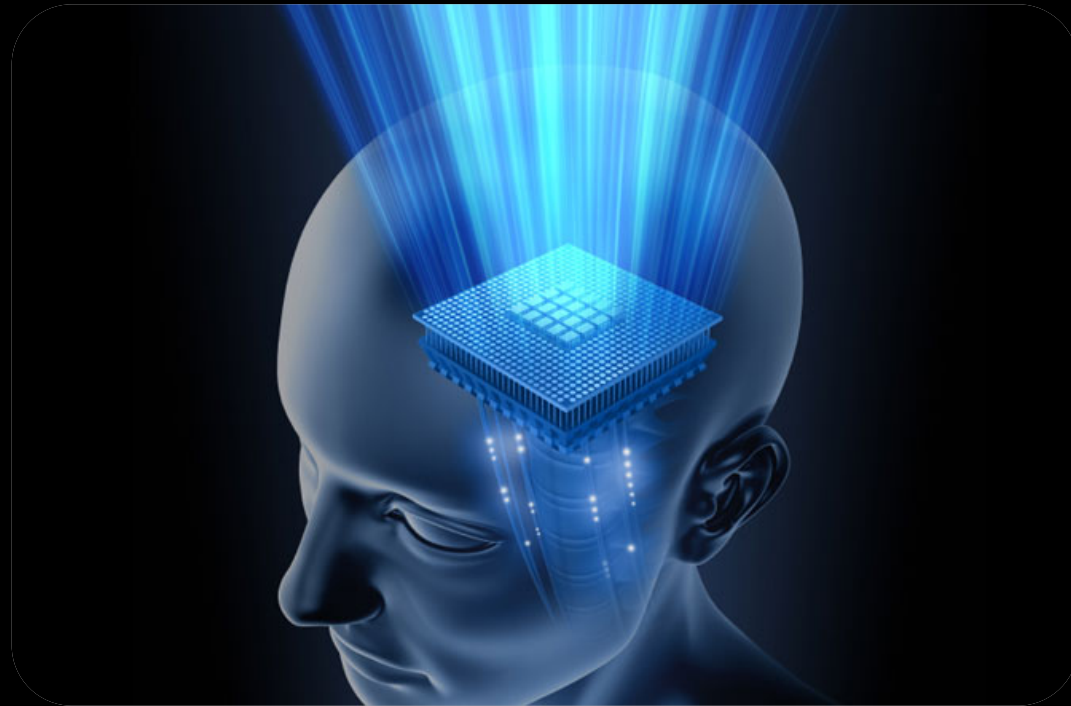
It will be a lot of work, and a lot of fun!

Let the learning begin



Basic Logic Gates

You already know something...

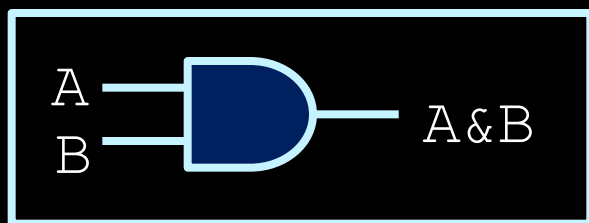


Logic from math course

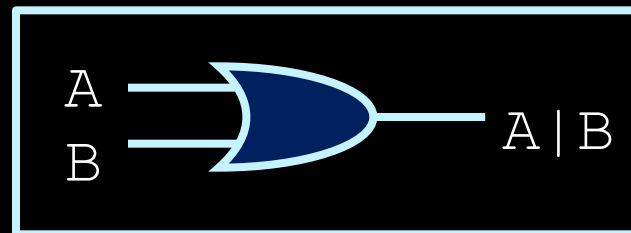
- Create an expression that is true iff the variables A and B are true, or C and D are true.

$$G = (A \ \& \ B) \ | \ (C \ \& \ D)$$

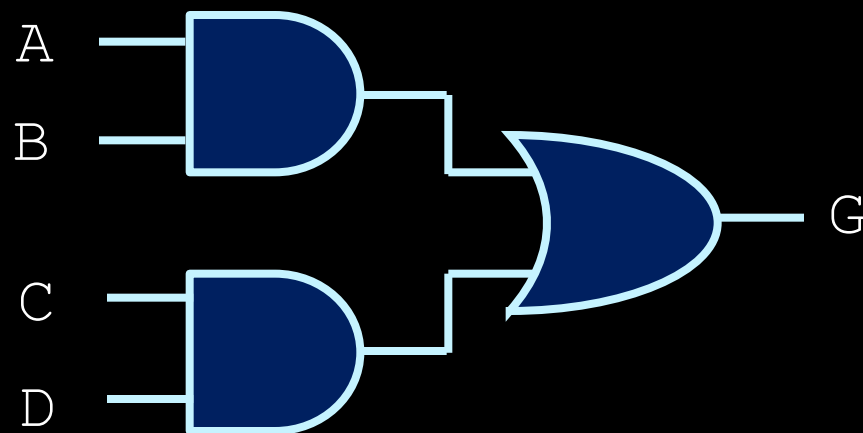
$$G = (A \ \& \ B) \ | \ (C \ \& \ D)$$



AND Gate



OR Gate



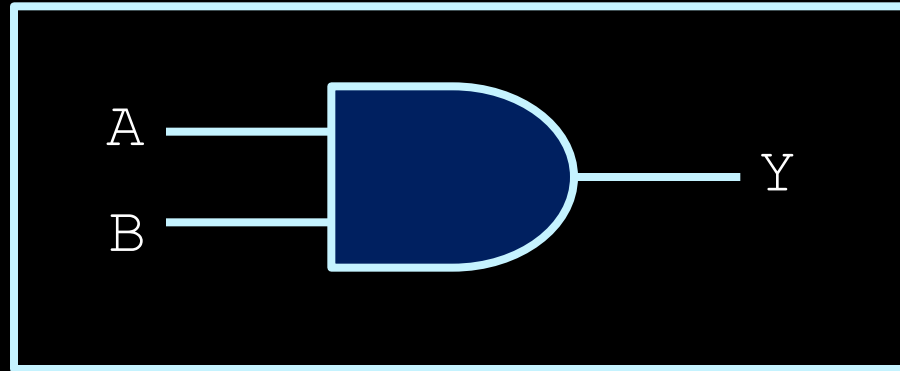
You just designed your first circuit in CSC258!

Gates = Boolean logic

- If we know the logical expression, we already know how to put logic gates together to form a circuit.
- Just need to know which logic operations are represented by which gate!

Let's meet all the gates.

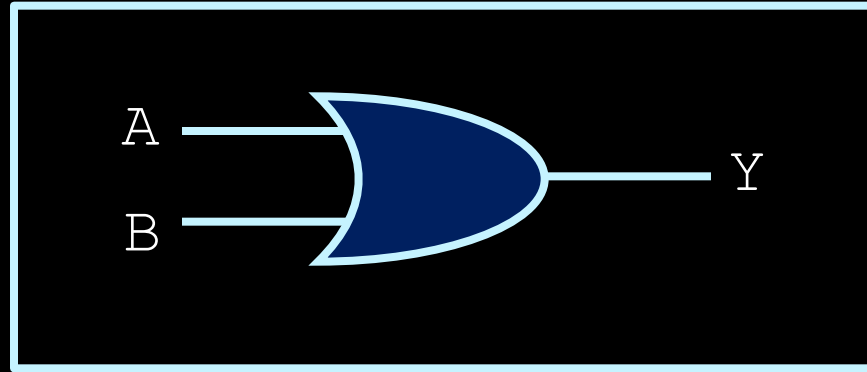
AND Gates



Truth table

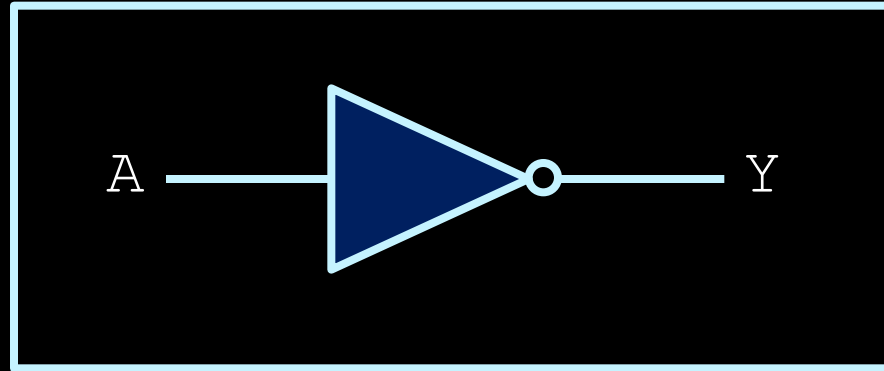
A	B	Y
0	0	0
0	1	0
1	0	0
1	1	1

OR Gates



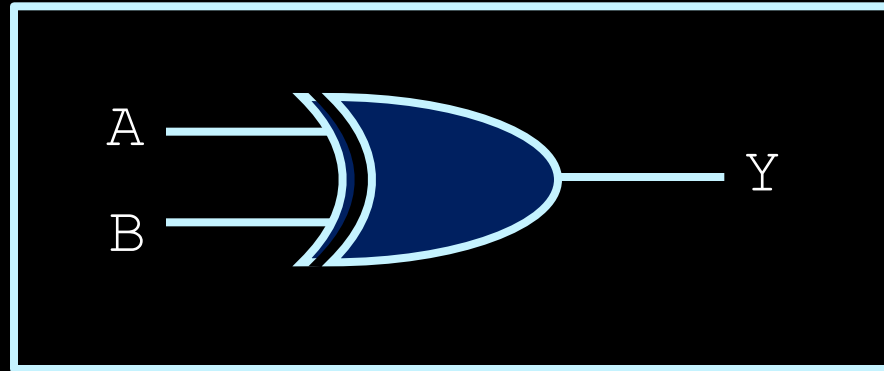
A	B	Y
0	0	0
0	1	1
1	0	1
1	1	1

NOT Gates



A	Y
0	1
1	0

XOR Gates

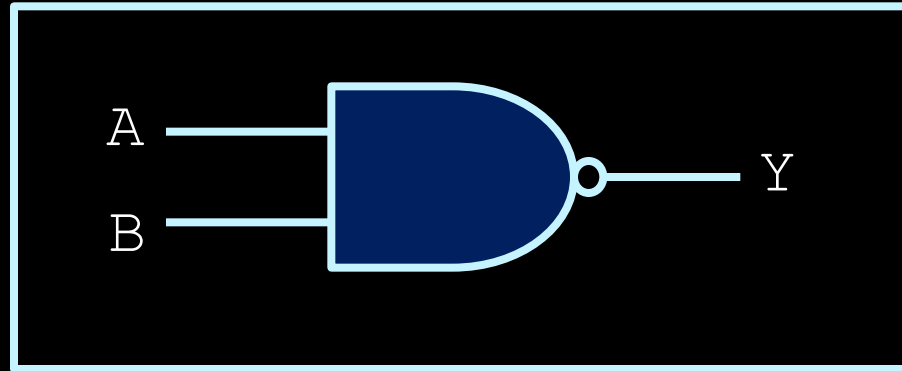


A	B	Y
0	0	0
0	1	1
1	0	1
1	1	0

Bill Gates

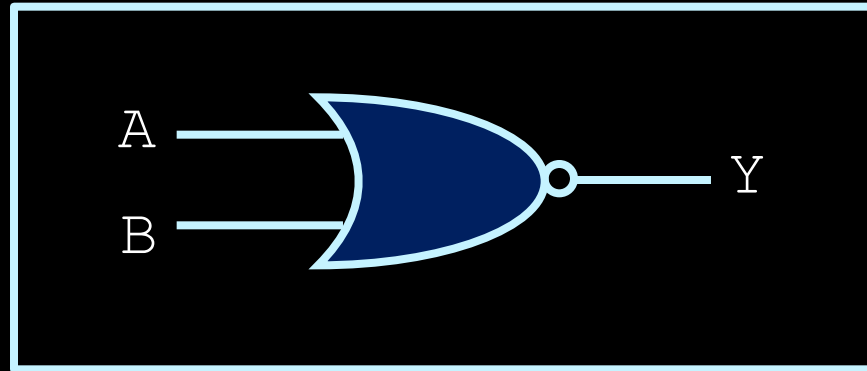


NAND Gates



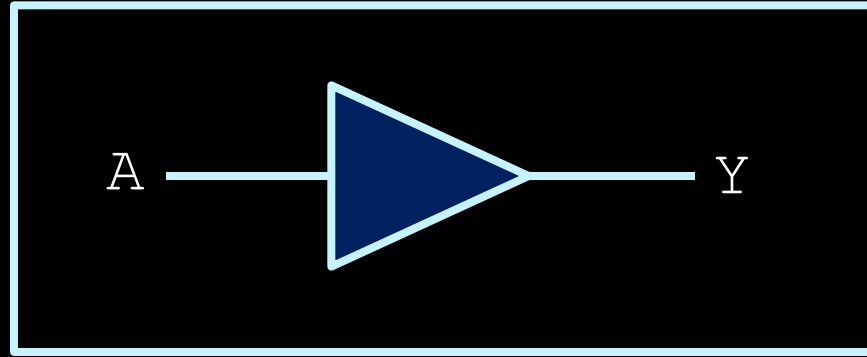
A	B	Y
0	0	1
0	1	1
1	0	1
1	1	0

NOR Gates



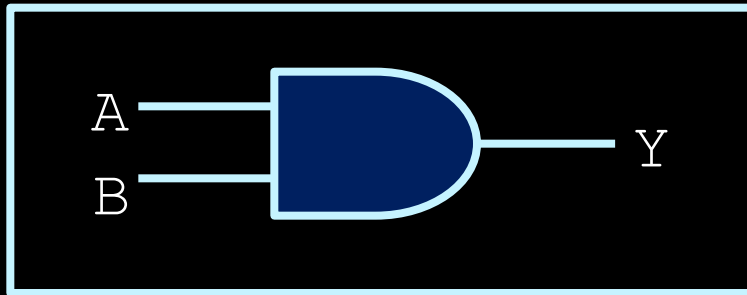
A	B	Y
0	0	1
0	1	0
1	0	0
1	1	0

Buffer

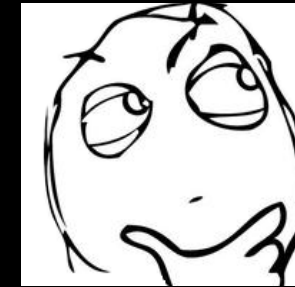


This is not as silly as
you might think now,
as we'll see later...

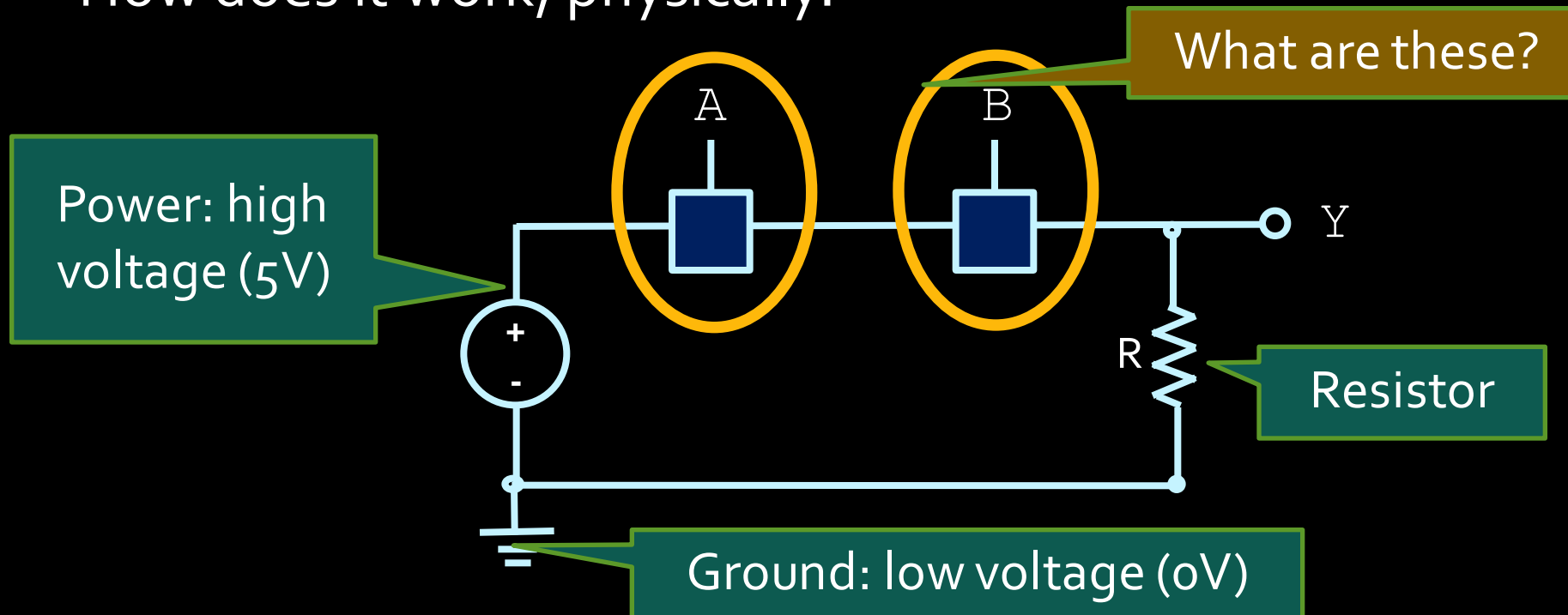
A	Y
0	0
1	1

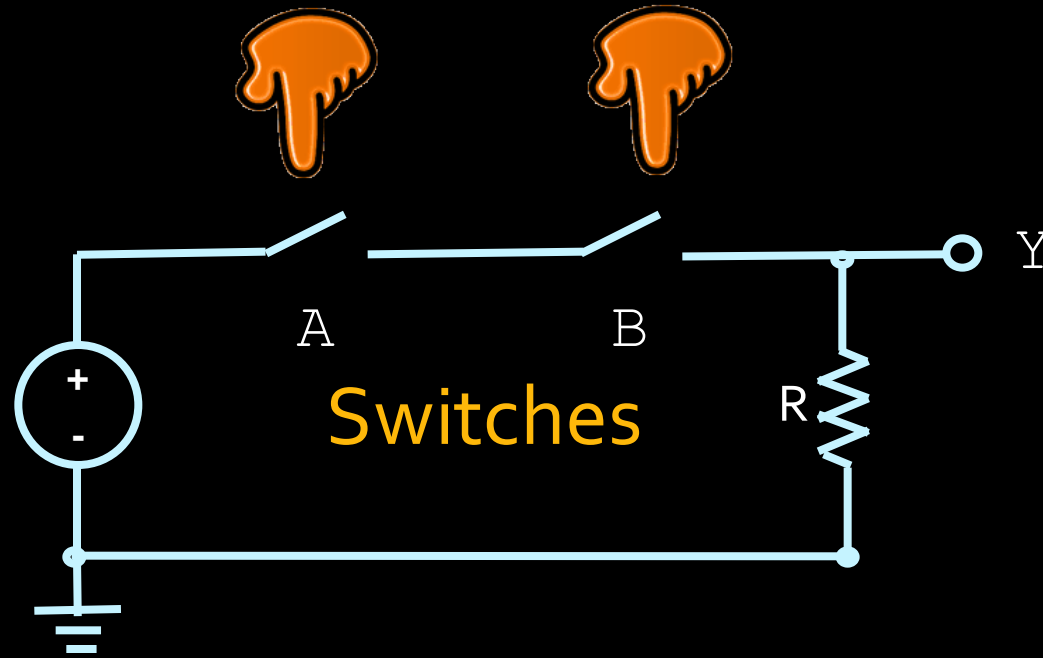


AND Gate

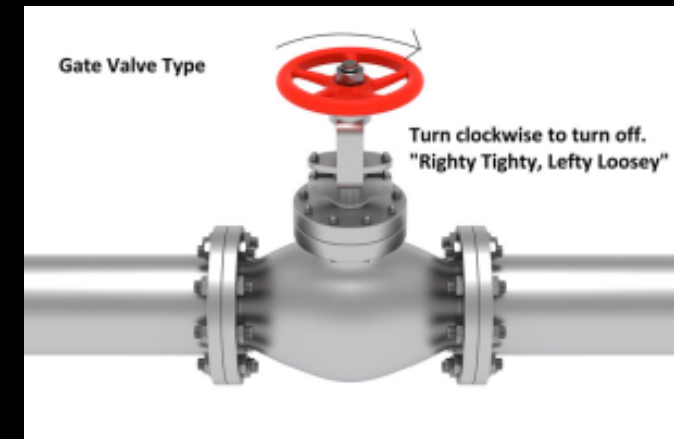
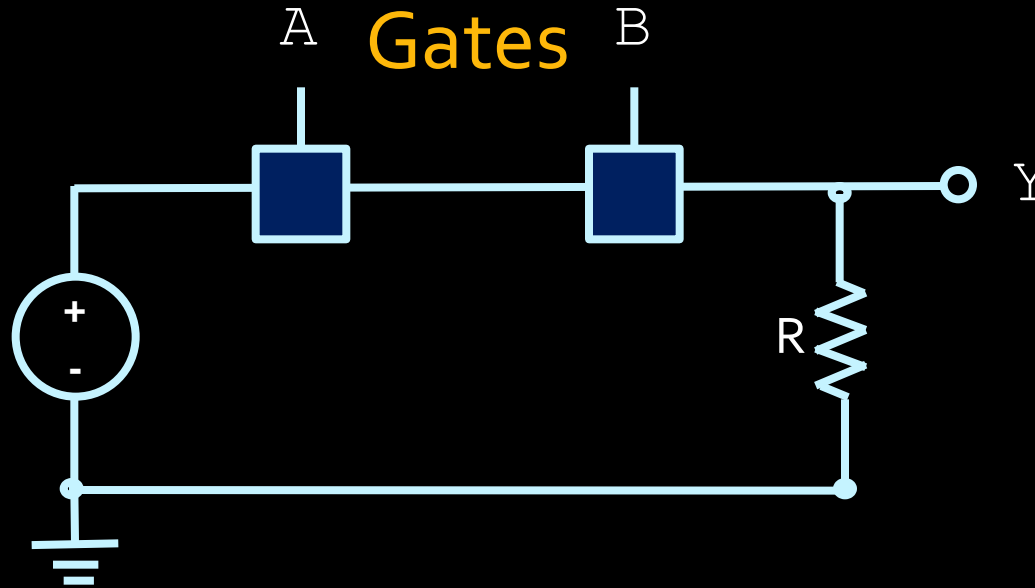


This is just a symbol...
What does it really look like, inside?
How does it work, physically?

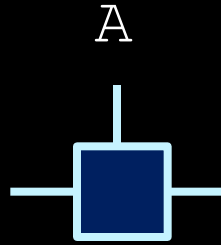




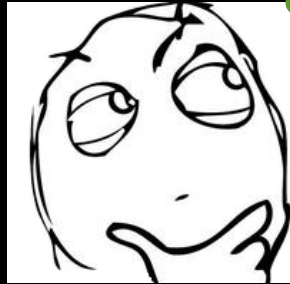
When and only when **both** A are B are switched **ON**, Y has **high** voltage.



- Gate is like a switch, but controlled by the voltage of the input signal, instead of by a finger.
- Gate A is switched **ON** when signal A is of **high** voltage.
- When and only when **both** A and B have **high** voltage, Y has **high** voltage.
- High voltage is **1 (True)**, low voltage is **0 (False)**.
- **Y is True iff both A and B are True ($Y = A \& B$).**



Gate is switched ON
when signal A is of
high voltage ...



Why?

How?

What does the inside of a gate look like?

Answer: There are **transistors**.



Transistors

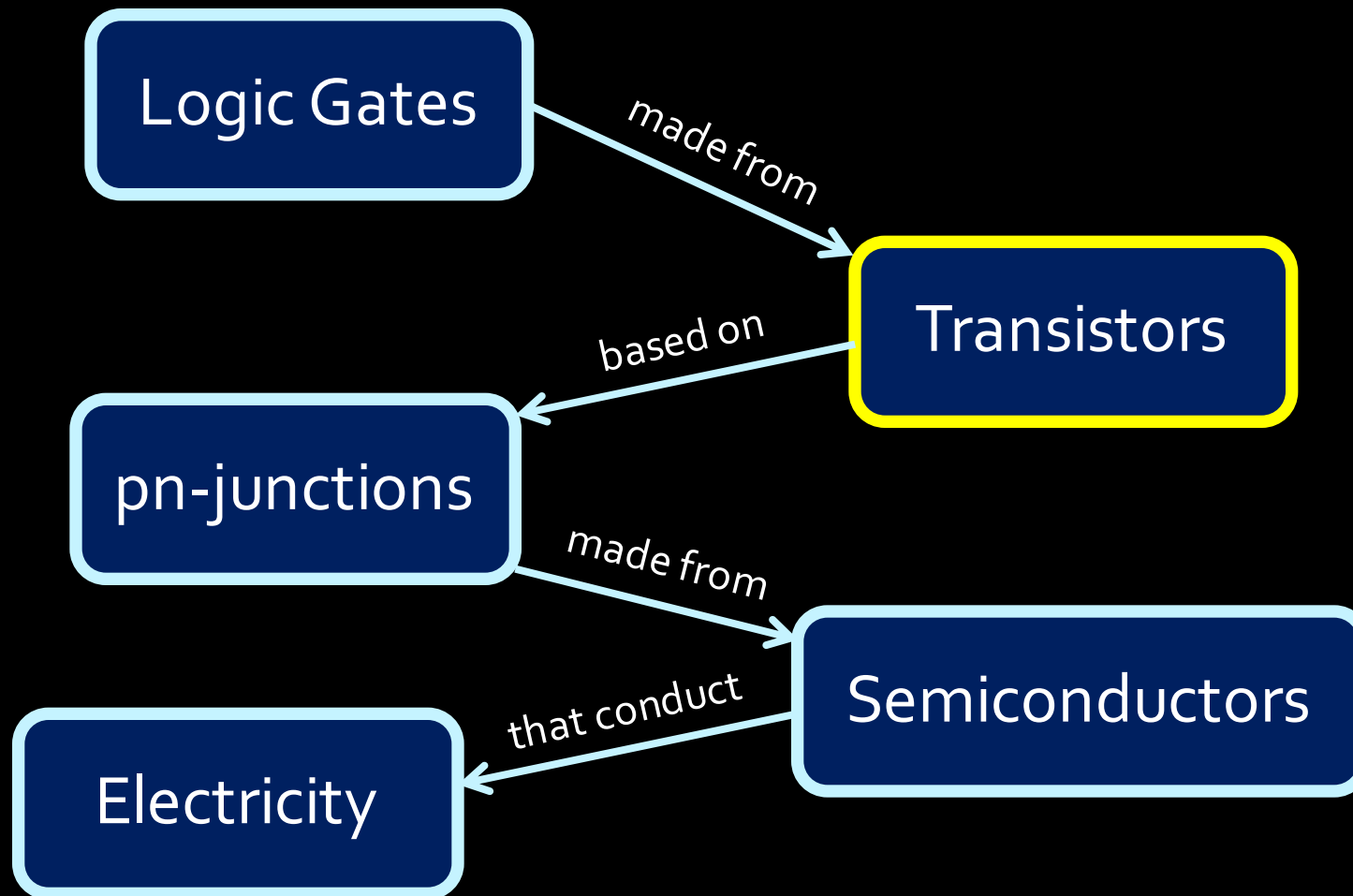
One of the greatest inventions of the 20th century

- Invented by William Shockley, John Bardeen and Walter Brattain in 1947, replacing previous vacuum-tube technology.
 - ▣ Nobel Prize for Physics in 1956.



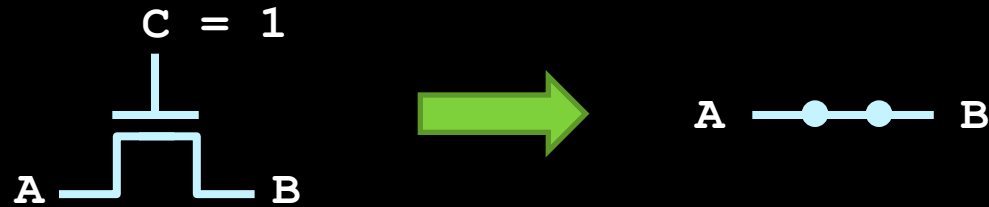
Building block for the hardware of all your computers and electronic devices.

Where do transistors fit?

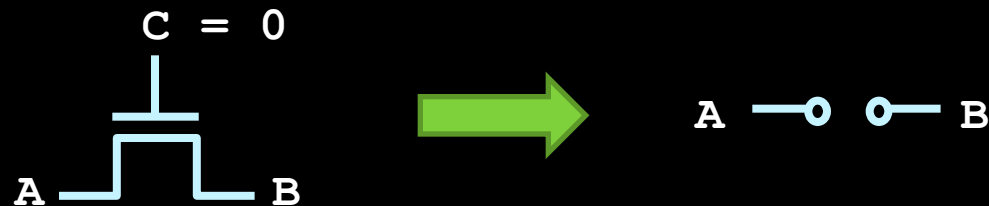


What do transistors do?

- Transistors connect Point A to Point B, based on the value at Point C.
 - If the value at Point C is high, A and B are connected.

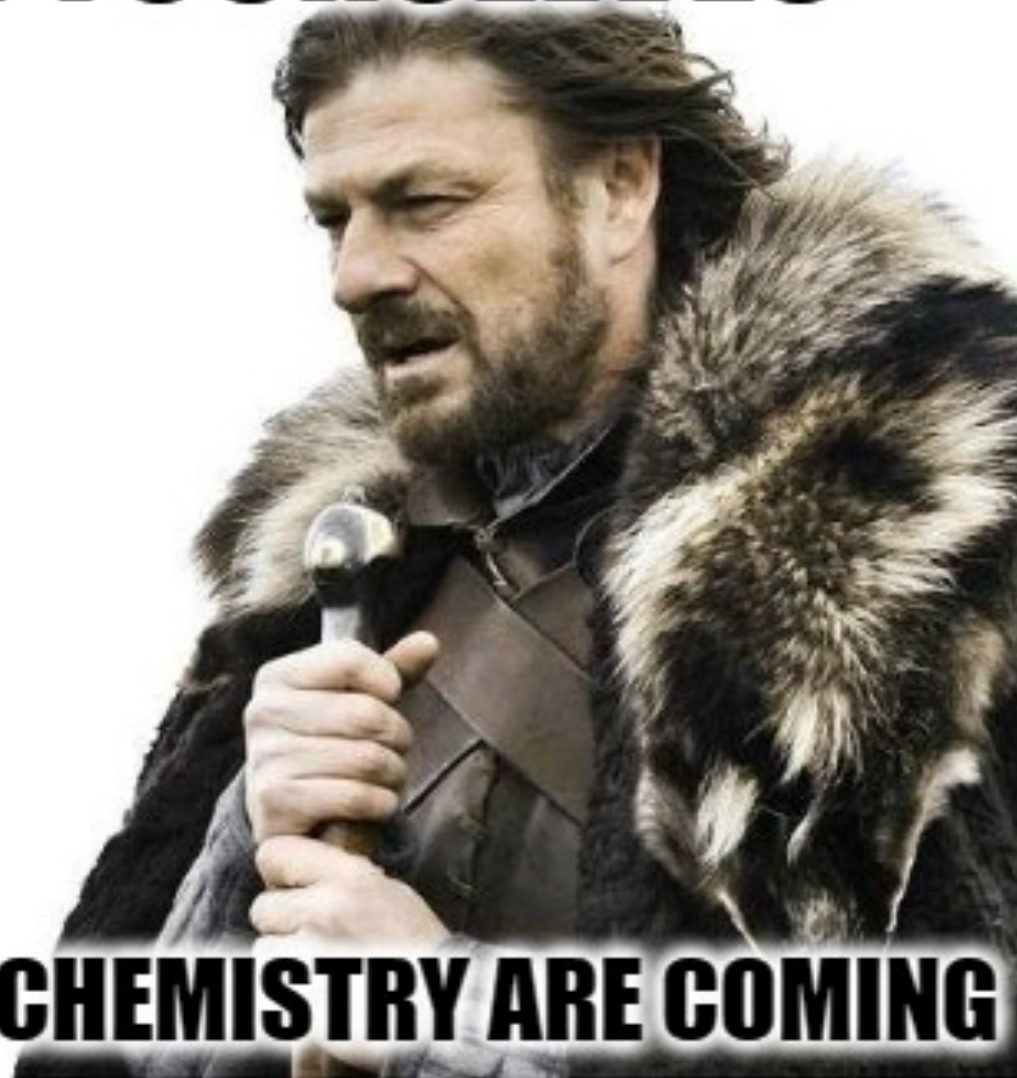


- And if the value at Point C is low, A and B are not.



- Need to know a little about electricity now....

BRACE YOURSELVES



PHYSICS AND CHEMISTRY ARE COMING

imgflip.com

Outline of the story

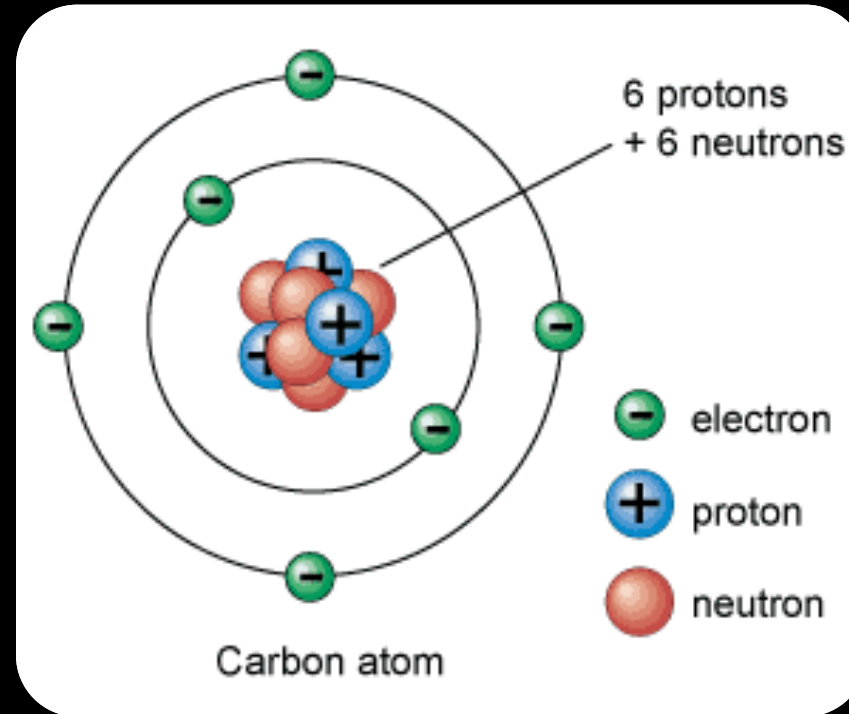
- Electricity, basic concepts
- Insulators, conductors, in between ..., Semiconductors
- Impure semiconductors, p-type / n-type
- Put p-type and n-type together -- pn-junction
- Apply voltage to a pn-junction – principle of transistors
- A real-world manufacturing of transistor -- MOSFET



Electricity Basics

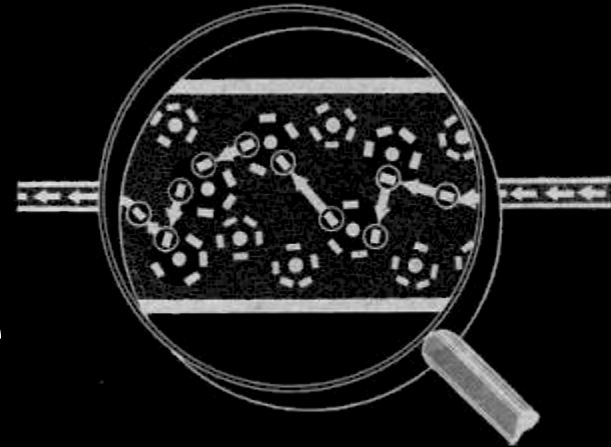
Everything is made out of atoms ...

- **Protons** are big (hardly move) and positively charged.
- **Electrons** are small (easily move) and negatively charged.
- **Neutrons** are big and of course, neutral.
- Overall, an atom is **neutral**.



What is Electricity?

- Electricity is the **flow** of charged particles (usually electrons) through a material.
 - Electricity could be caused by the flow of protons as well, but since they're so much bigger than electrons, we usually assume that it's the electrons flowing.



How do electrons flow?

They flow ...



Keep this analogy in mind...



How do electrons flow?

- Electrons want to flow from regions of **high electrical potential** (many electrons) to regions of **low electrical potential** (fewer electrons).
 - Like water flows from high to low.
- This potential is referred to as **voltage (V)**.
- The rate of this flow is called the **current (I)**.
- **Resistance ($I = V / R$)** is like how narrow the pipe is.
 - Narrower water pipe has higher resistance.


Note

The direction of the current is **opposite** to the direction of the electron movement, because electrons are **negatively** charged.

More on Resistance

- Electrical resistance indicates how well a **material** allows electricity to flow through it:
 - High resistance (aka **insulators**) don't conduct electricity at all.
 - Low resistance (aka **conductors**) conduct electricity well, and are generally used for wires.
- **Semiconductors** are somewhere in between conductors and insulators, which makes it interesting...

Outline of the story

- Electricity, basic concepts
- Insulators, conductors, in between ...
-  **Semiconductors**
- Impure semiconductors, **p-type / n-type**
- Put p-type and n-type together -- **pn-junction**
- Apply voltage to a pn-junction – **principle of transistors**
- A real-world manufacturing of transistor -- **MOSFET**



Semiconductors

Here comes the chemistry

Periodic Table of Elements

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H Hydrogen 1.00794																	2 He Helium 4.002602
3 Li Lithium 6.941	4 Be Beryllium 9.012182																
11 Na Sodium 22.98976928	12 Mg Magnesium 24.3050																
19 K Potassium 39.0983	20 Ca Calcium 40.078	21 Sc Scandium 44.955912	22 Ti Titanium 47.887	23 V Vanadium 50.9415	24 Cr Chromium 51.9961	25 Mn Manganese 54.938045	26 Fe Iron 55.845	27 Co Cobalt 58.933195	28 Ni Nickel 58.6934	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.64	33 As Arsenic 74.92160	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.798
37 Rb Rubidium 85.4678	38 Sr Strontium 87.62	39 Y Yttrium 88.90585	40 Zr Zirconium 91.224	41 Nb Niobium 92.90638	42 Mo Molybdenum 95.96	43 Tc Technetium (97.9072)	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.90550	46 Pd Palladium 106.42	47 Ag Silver 107.8682	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.760	52 Te Tellurium 127.60	53 I Iodine 126.90447	54 Xe Xenon 131.293
55 Cs Caesium 132.9054519	56 Ba Barium 137.327	57-71	72 Hf Hafnium 178.49	73 Ta Tantalum 180.94788	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.217	78 Pt Platinum 195.084	79 Au Gold 196.966569	80 Hg Mercury 200.59	81 Tl Thallium 204.3833	82 Pb Lead 207.2	83 Bi Bismuth 208.98040	84 Po Polonium (209.9824)	85 At Astatine (210.9871)	86 Rn Radon (222.0176)
87 Fr Francium (223)	88 Ra Radium (226)	89-103	104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (266)	107 Bh Bohrium (264)	108 Hs Hassium (277)	109 Mt Meitnerium (268)	110 Ds Darmstadtium (271)	111 Rg Roentgenium (272)	112 Uub Ununbium (285)	113 Uut Ununtrium (284)	114 Uuq Ununquadium (289)	115 Uup Ununpentium (288)	116 Uuh Ununhexium (292)	117 Uus Ununseptium	118 Uuo Ununoctium (294)

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>

Ptable
.com

57 La Lanthanum 138.90547	58 Ce Cerium 140.116	59 Pr Praseodymium 140.90765	60 Nd Neodymium 144.242	61 Pm Promethium (145)	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.92535	66 Dy Dysprosium 162.500	67 Ho Holmium 164.93032	68 Er Erbium 167.259	69 Tm Thulium 168.93421	70 Yb Ytterbium 173.054	71 Lu Lutetium 174.9668
89 Ac Actinium (227)	90 Th Thorium 232.03806	91 Pa Protactinium 231.03688	92 U Uranium 238.02891	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)

Periodic Table of Elements

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H Hydrogen 1.00794																	2 He Helium 4.002602
3 Li Lithium 6.941	4 Be Beryllium 9.012182											5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.0064	8 O Oxygen 15.9994	9 F Fluorine 18.9984032	10 Ne Neon 20.1797
11 Na Sodium 22.98976928	12 Mg Magnesium 24.3050											13 Al Aluminum 26.9815386	14 Si Silicon 28.0855	15 P Phosphorus 30.973762	16 S Sulfur 32.065	17 Cl Chlorine 35.453	18 Ar Argon 39.948
19 K Potassium 39.0983	20 Ca Calcium 40.078	21 Sc Scandium 44.955912	22 Ti Titanium 47.887	23 V Vanadium 50.9415	24 Cr Chromium 51.9961	25 Mn Manganese 54.938045	26 Fe Iron 55.845	27 Co Cobalt 58.933195	28 Ni Nickel 58.6934	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.64	33 As Arsenic 74.9216	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.798
37 Rb Rubidium 85.4678	38 Sr Strontium 87.62	39 Y Yttrium 88.90585	40 Zr Zirconium 91.224	41 Nb Niobium 92.90638	42 Mo Molybdenum 95.96	43 Tc Technetium (97.9072)	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.90550	46 Pd Palladium 106.42	47 Ag Silver 107.8682	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.760	52 Te Tellurium 127.6	53 I Iodine 126.90447	54 Xe Xenon 131.293
55 Cs Caesium 132.9054519	56 Ba Barium 137.327	57–71 Lanthanoids	72 Hf Hafnium 178.49	73 Ta Tantalum 180.94788	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.222	78 Pt Platinum 195.084	79 Au Gold 196.96657	80 Hg Mercury 200.59	81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.9804	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222.0176)
87 Fr Francium (223)	88 Ra Radium (226)	89–103 Actinoids	104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (266)	107 Bh Bohrium (264)	108 Hs Hassium (277)	109 Mt Meitnerium (268)	110 Ds Darmstadtium (271)	111 Rg Roentgenium (272)	112 Cn Copernicium (285)	113 Nh Nihonium (284)	114 Fl Flerovium (289)	115 Mc Moscovium (288)	116 Uuh Ununhexium (292)	117 Uus Ununseptium (294)	118 Uuo Ununoctium (294)

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>

57 La Lanthanum 138.90547	58 Ce Cerium 140.116	59 Pr Praseodymium 140.90765	60 Nd Neodymium 144.242	61 Pm Promethium (145)	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.92535	66 Dy Dysprosium 162.500	67 Ho Holmium 164.93032	68 Er Erbium 167.259	69 Tm Thulium 168.93421	70 Yb Ytterbium 173.054	71 Lu Lutetium 174.9668
89 Ac Actinium (227)	90 Th Thorium 232.03806	91 Pa Protactinium 231.03688	92 U Uranium 238.02891	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)

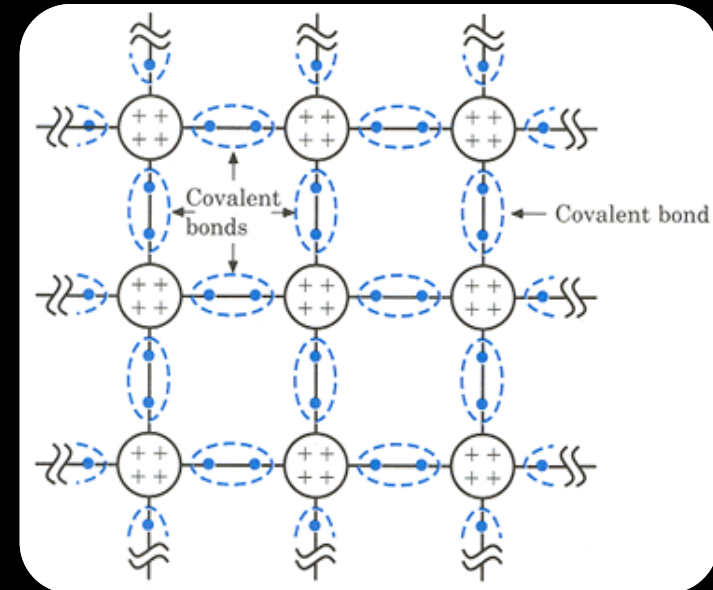
Conductivity of Semiconductors

- Semiconductor materials (e.g., silicon and germanium) straddle the boundary between **conductors** and **insulators**, behaving like one or the other, depending on factors like temperature and **impurities** in the material.

Impurity

Pure semiconductor is pretty stable

- Each atom has **4 electrons**, forming **bonds** with other atoms, and the structure is pretty stable.
- At room temperature, a weak current will flow through the material, much less than that of a conductor.

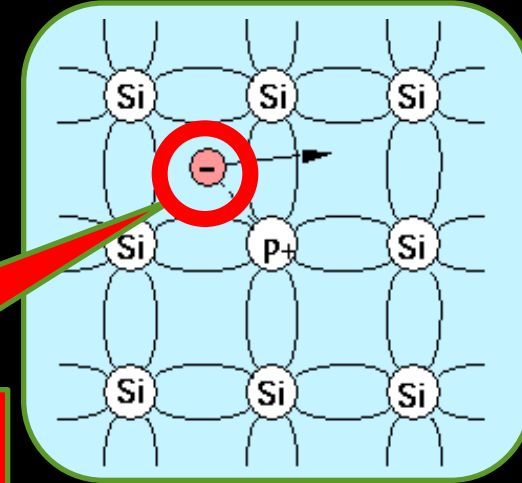


Encourage semiconductor's conductivity

N-type:

Add some atoms with **5** valence electrons, such as **Phosphorus**.

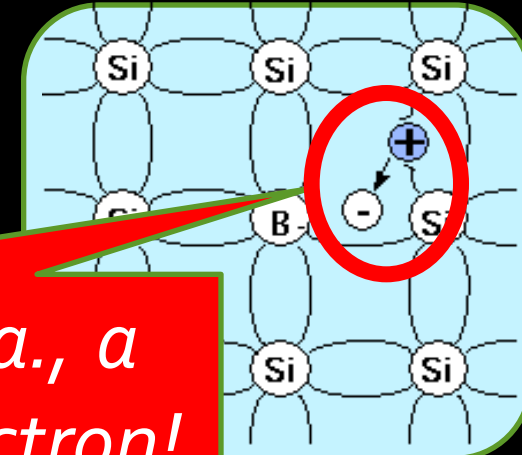
An extra electron!



P-type:

Add some atoms with **3** valence electrons, such as **Boron**.

A missing electron, a.k.a., a "hole", like a positive electron!

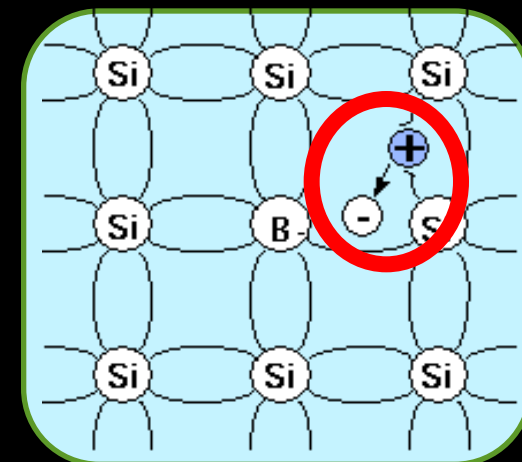
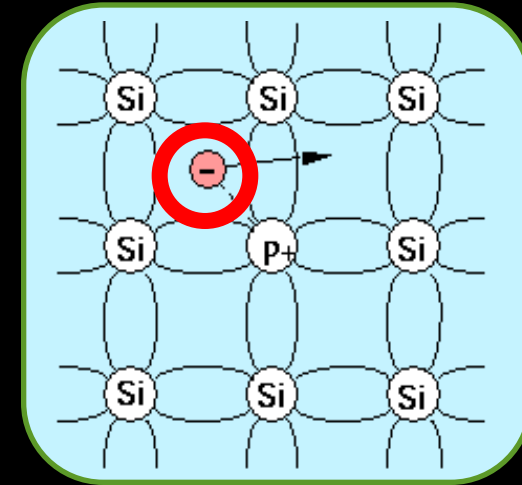


Encourage semiconductor's conductivity

The extra electrons and the holes are **charge carriers**, which can move **freely** through the materials.

Thus the conductivity is encouraged.

This process of adding stuff is called **doping**, (n or p type).




Free electrons
move like



Free holes
move like



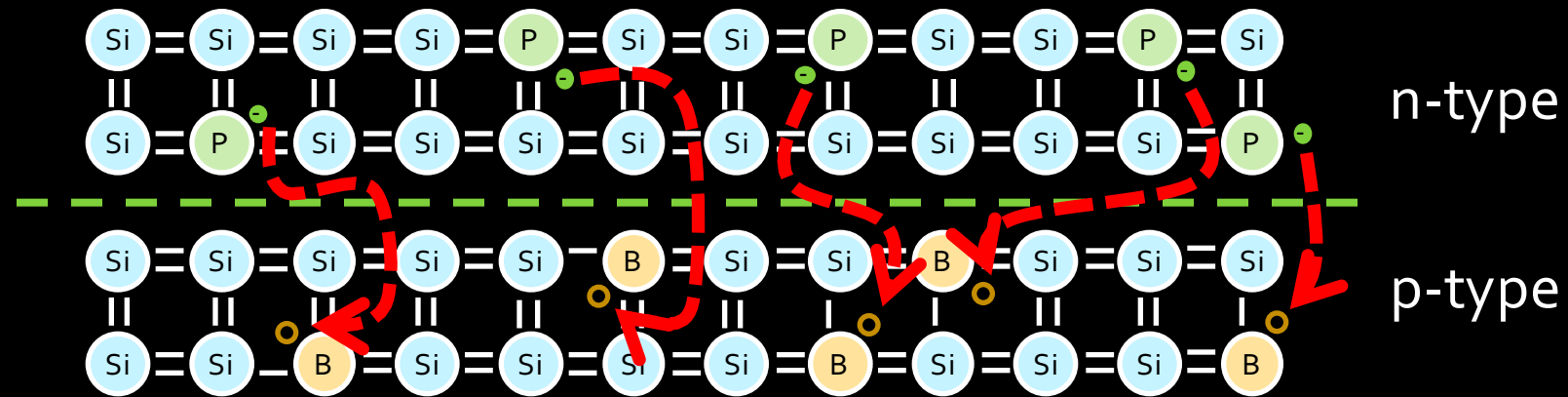
Outline of the story

- Electricity, basic concepts
- Insulators, conductors, in between ...,
Semiconductors
- Impure semiconductors, **p-type / n-type**
-  Put p-type and n-type together -- **pn-junction**
- Apply voltage to a pn-junction – **principle of transistors**
- A real-world manufacturing of transistor --
MOSFET

PN-junctions

Bringing **p** and **n** together

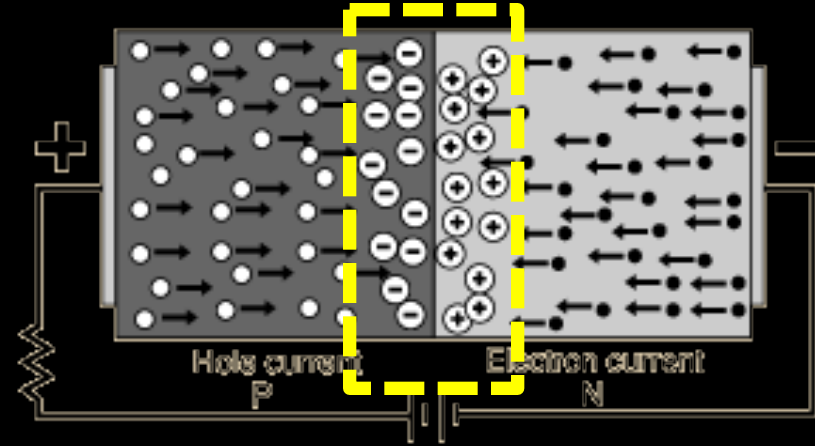
- What happens if you brought some p-type material into contact with some n-type material?

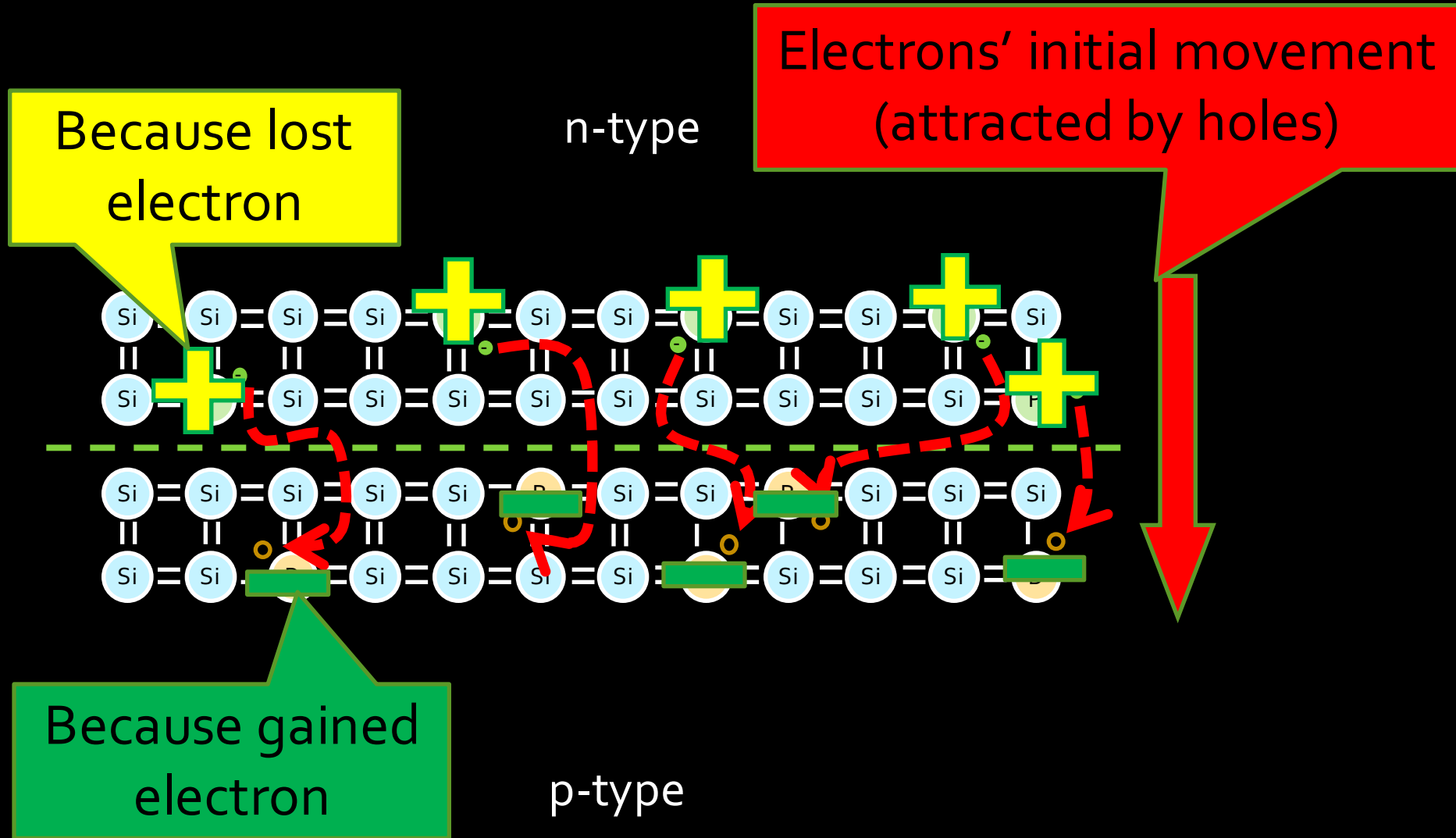


- The **electrons** at the surface of the n-type material are **drawn** to the **holes** in the p-type.

p-n Junctions

- When left alone, the **electrons** from the **n** section of the junction will fill the **holes** of the **p** section, **cancelling** each other and create a section with **no free carriers** called the **depletion layer**.
- Once this depletion layer is wide enough, the doping atoms that remain will create an **electric field** in that region.

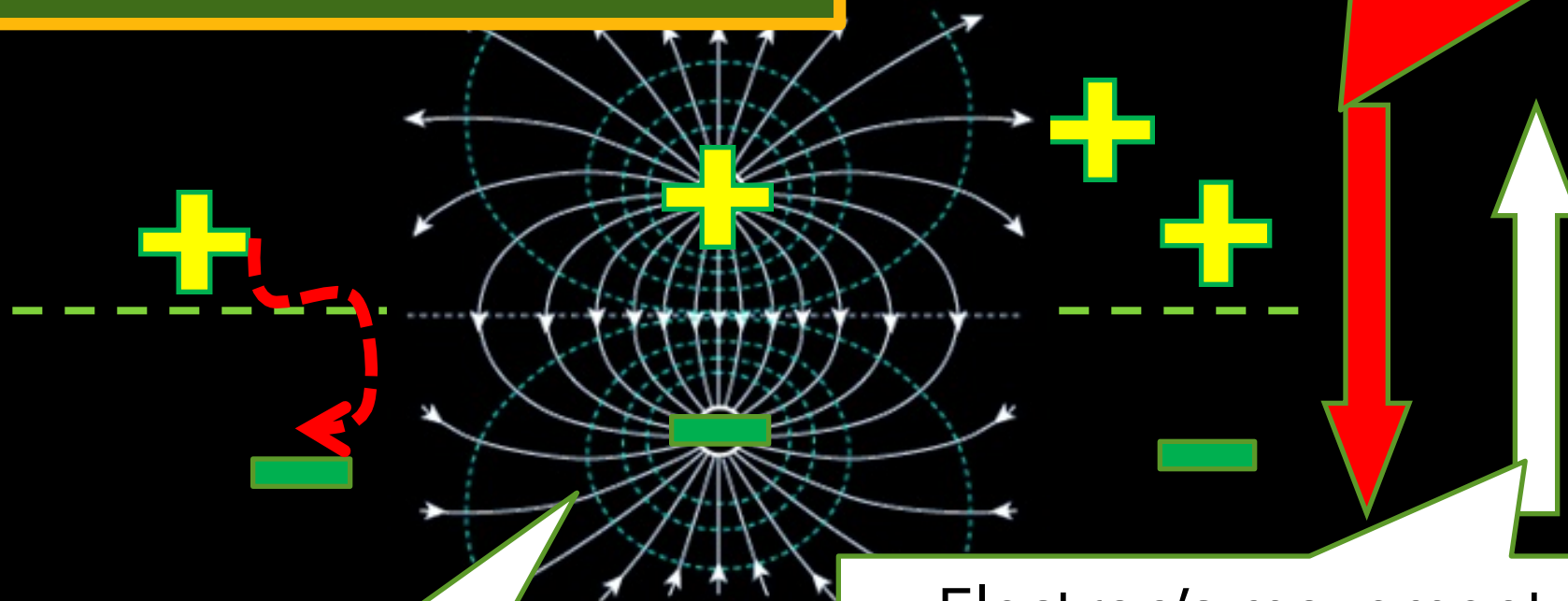




Diffusion increases the width of depletion layer, and drift draws it back. An **equilibrium** is reached, when the depletion layer is of a certain width.

"Diffusion"

Electrons' initial movement
(attracted by holes)



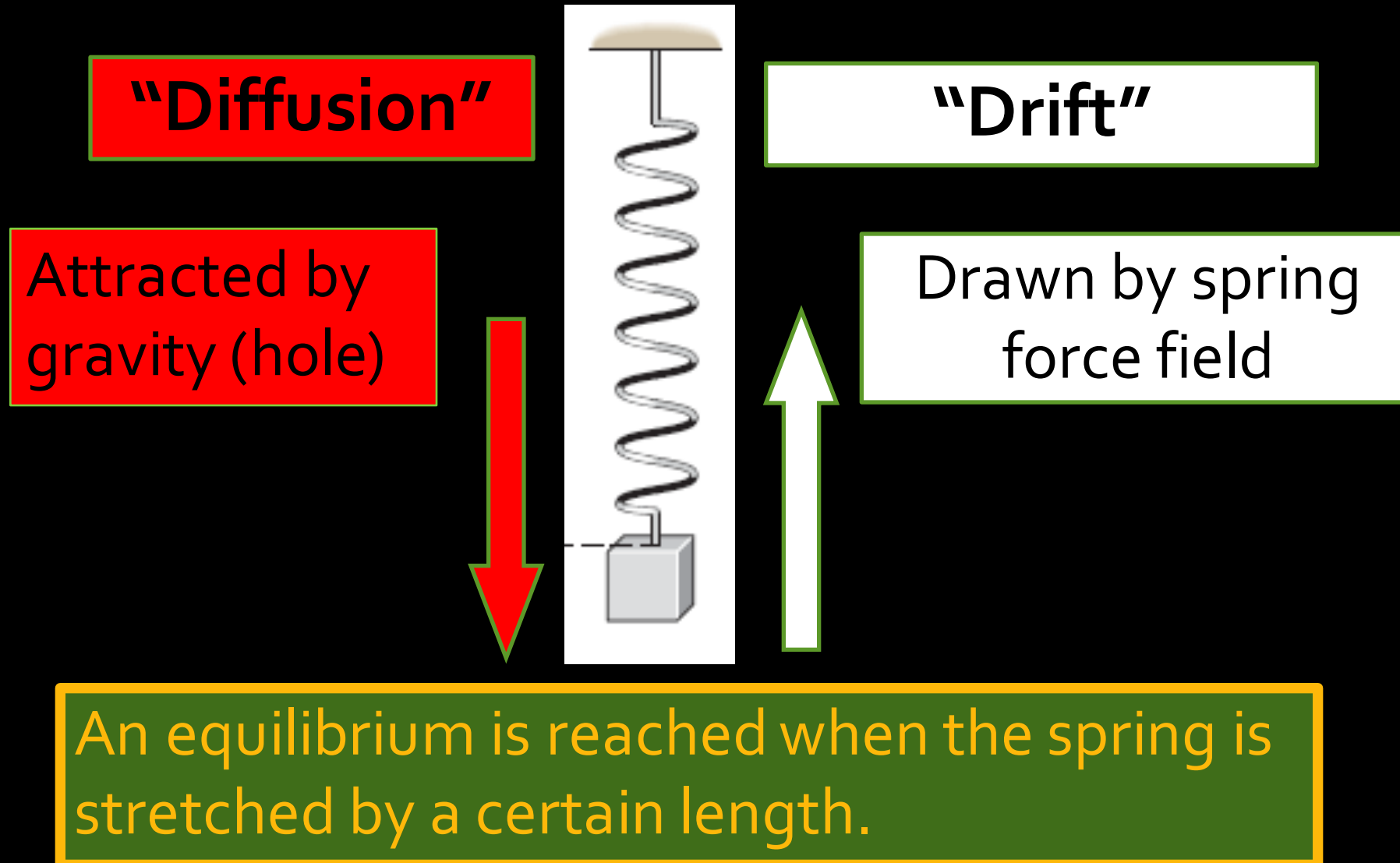
Electric field

p-type

Electron's movement
drawn by the electric field

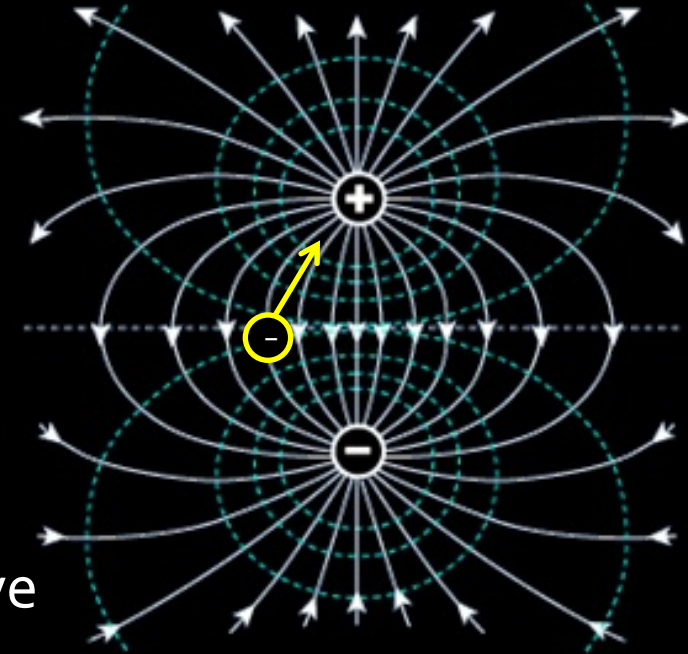
"Drift"

Analogy: Spring with weight



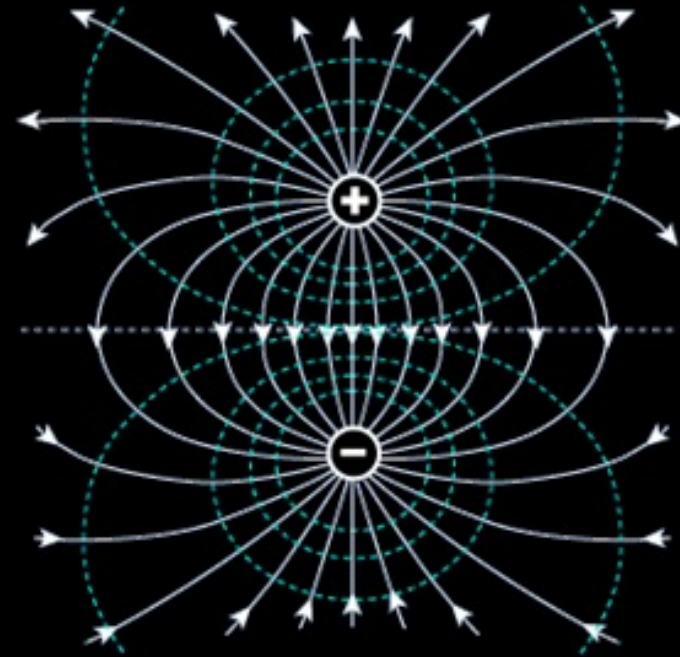
Electric fields (after class reading)

- What is an electric field?
 - When a phosphorus atom loses its electron, the atom develops an overall positive charge.
 - Similarly, when a boron atom takes on an extra electron, that atom develops an overall negative charge.
 - If an electron was dropped between the two, it would be attracted to the phosphorus atom, and repelled by the boron atom.
 - This effect is called an **electric field**.

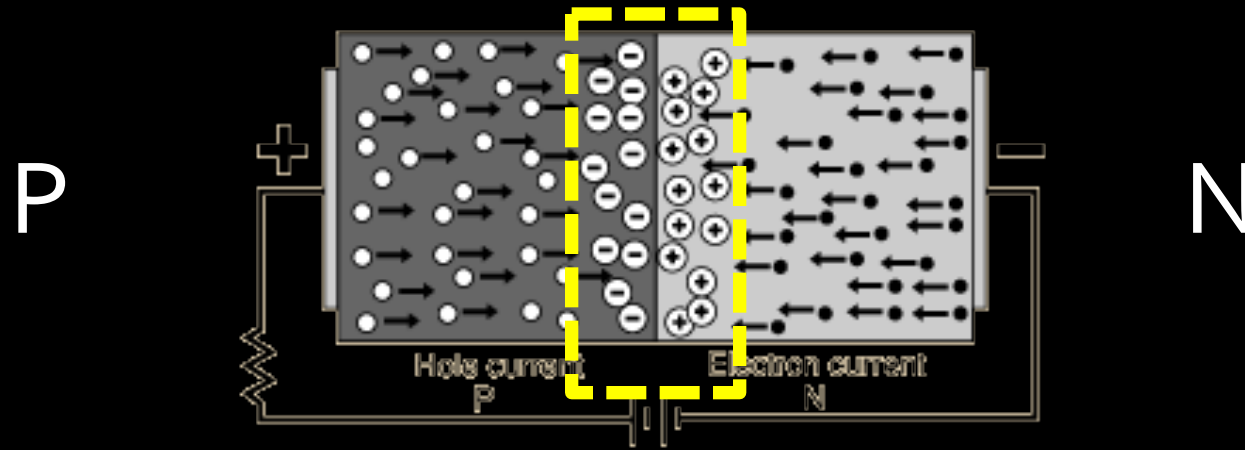


Electric fields (after class reading)

- A depletion layer is made up of many of these electrically imbalanced phosphorus and boron atoms.
- The electric field caused by these atoms will cause holes to flow back to the p section, and electrons to flow back to the n section.
 - The current caused by this electric field is called **drift**.
 - The current caused by the initial electron/hole recombination is called **diffusion**.
- At rest, these two currents reach **equilibrium**.




Summary of pn-junction



When we put **p** and **n** together, they will form a depletion layer with electric field in it.

The depletion layer grows up to a certain **width**, until equilibrium is reached.

Outline of the story

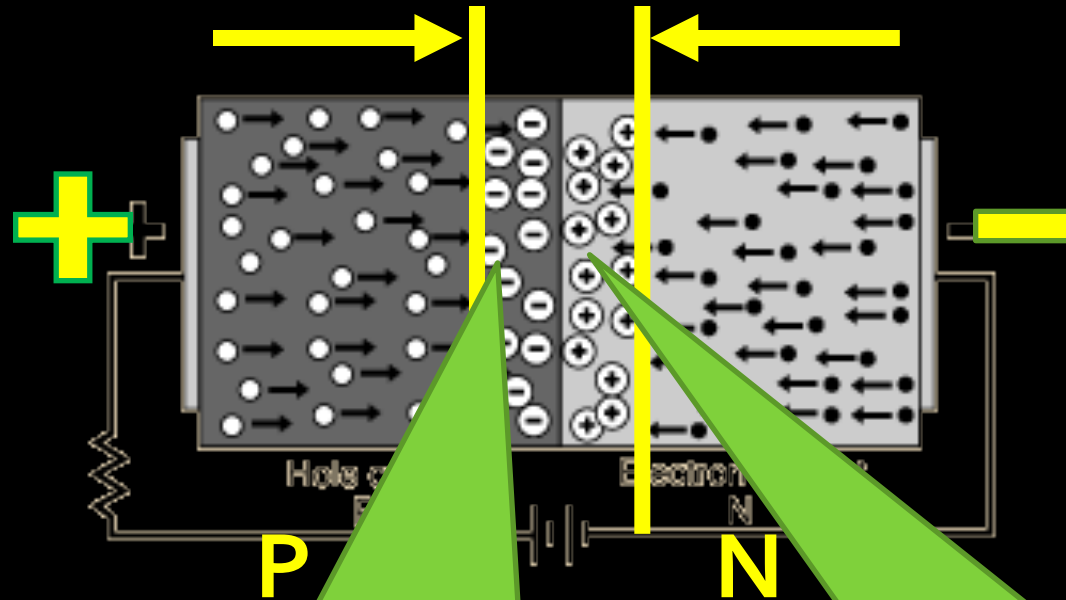
- Electricity, basic concepts
- Insulators, conductors, in between ...,
Semiconductors
- Impure semiconductors, **p-type / n-type**
- Put p-type and n-type together -- **pn-junction**
-  Apply voltage to a pn-junction – **principle of transistors**
- A real-world manufacturing of transistor --
MOSFET

Apply voltage to a PN-junction

It could be applied in two possible **directions**

- **P**ositive voltage **to** the **P** side
- **P**ositive voltage **to** the **N** side

Forward Bias (Positive voltage to P)

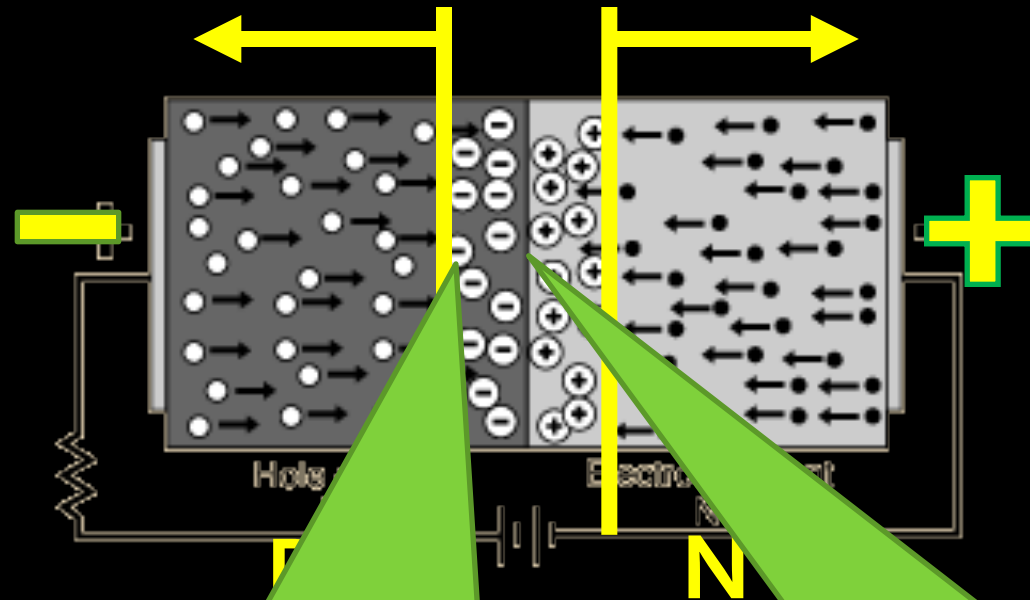


Negative charges sucked out of depletion layer

Positive charges sucked out of depletion layer

Depletion layer becomes **narrower**.

Reverse Bias (Positive voltage to N)



Negative charges **injected**
into depletion layer

Positive charges **injected**
into depletion layer

Depletion layer becomes **wider**.

Apply **forward bias**


- Depletion layer narrower
- Easier to travel through
- Better conductivity
- Like switch **connected**

Apply **reverse bias**

- Depletion layer wider
- Harder to travel through
- Worse conductivity
- Like switch **disconnected**

That's how transistors work!

Outline of the story

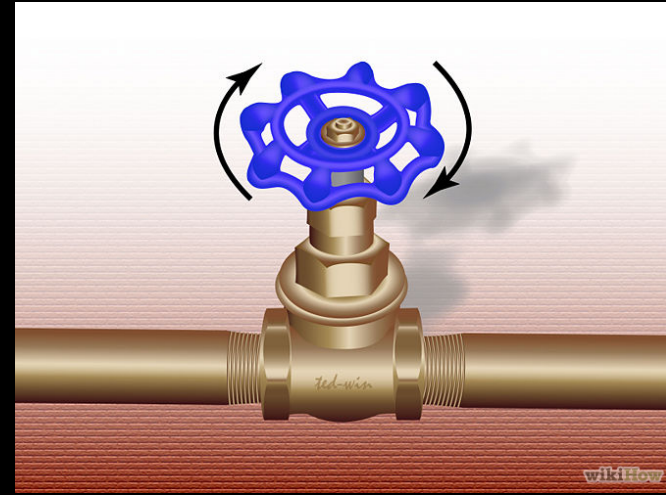
- Electricity, basic concepts
 - Insulators, conductors, in between ...,
Semiconductors
 - Impure semiconductors, **p-type / n-type**
 - Put p-type and n-type together -- **pn-junction**
 - Apply voltage to a pn-junction – **principle of transistors**
-  A real-world manufacturing of transistor --
MOSFET

Creating transistors

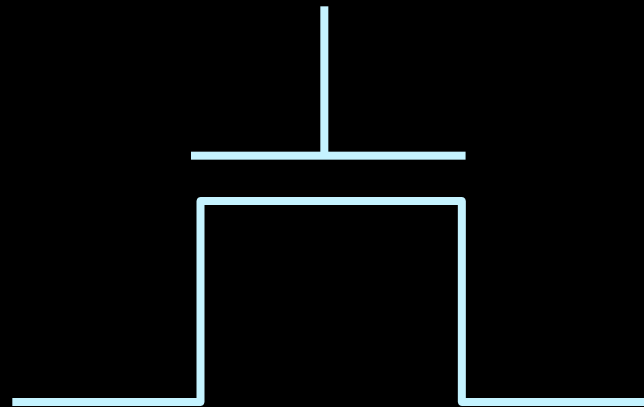
- Transistors use the characteristics of p-n junctions to create more interesting behaviour.
- Three main types:
 - Bipolar Junction Transistors (BJTs)
 - Metal Oxide Semiconductor Field Effect Transistor (MOSFET)
 - Junction Field Effect Transistor (JFET)
- The last two are part of the same family, but we'll only look at the MOSFET for now.

Metal Oxide Semiconductor Field Effect Transistor

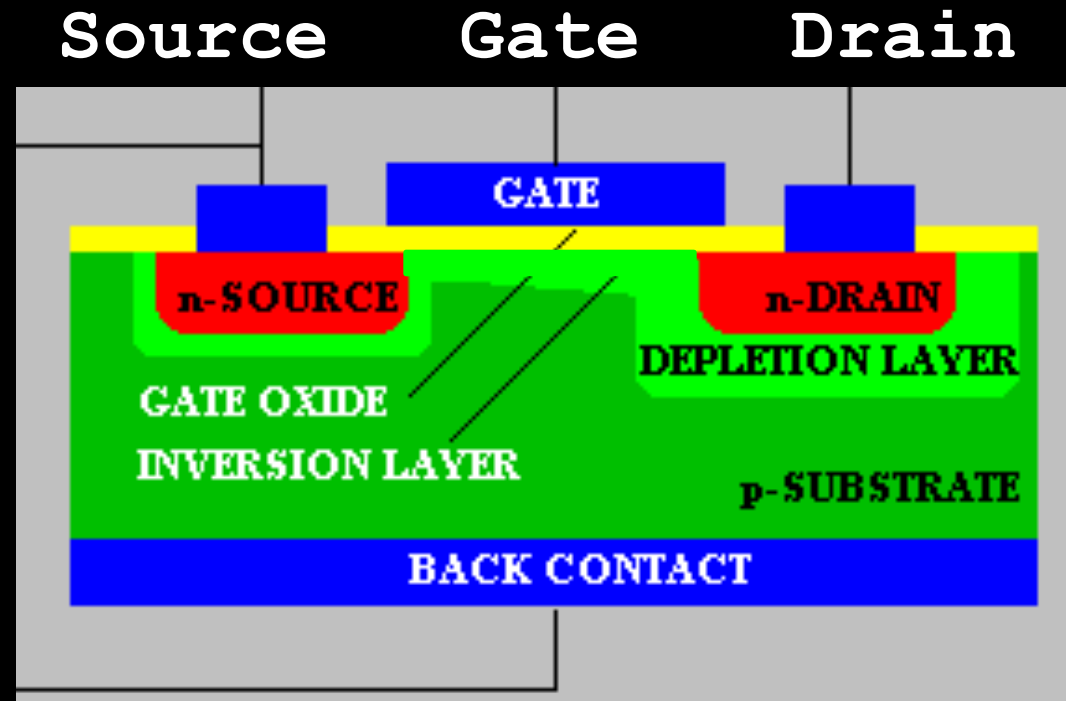
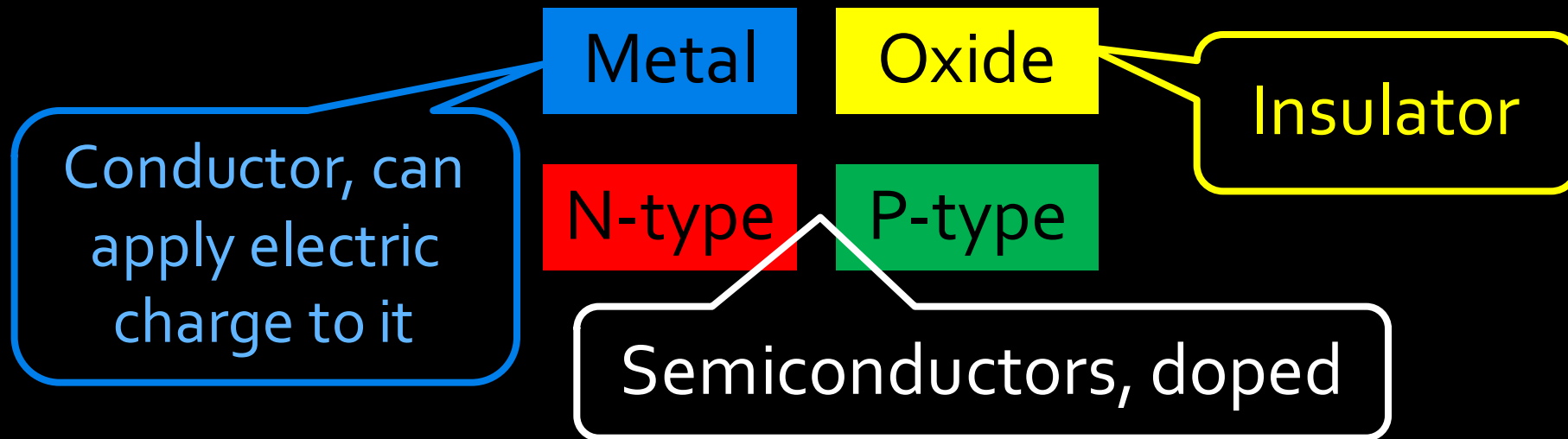
Gate



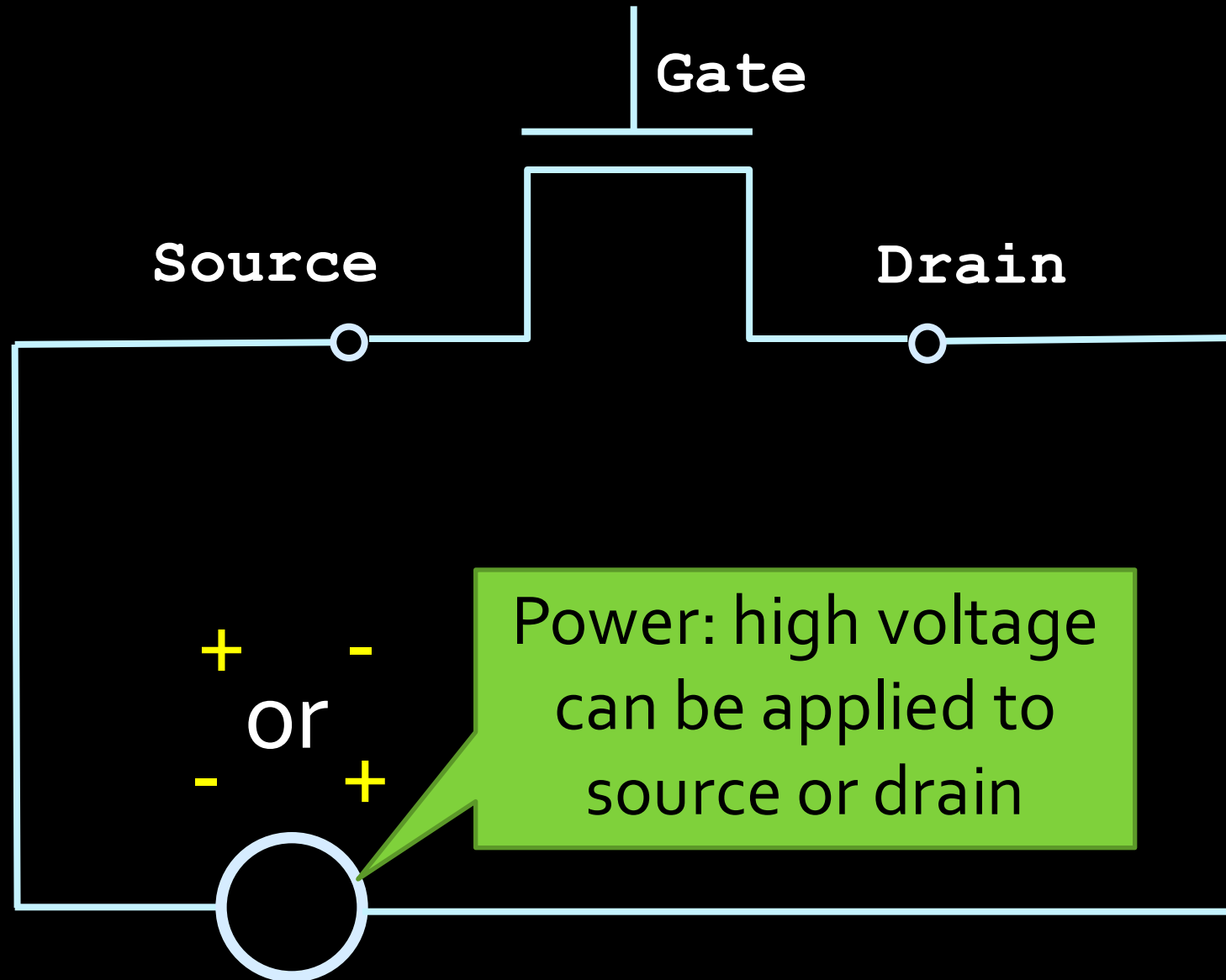
Source



Drain



Put a MOSFET into a circuit



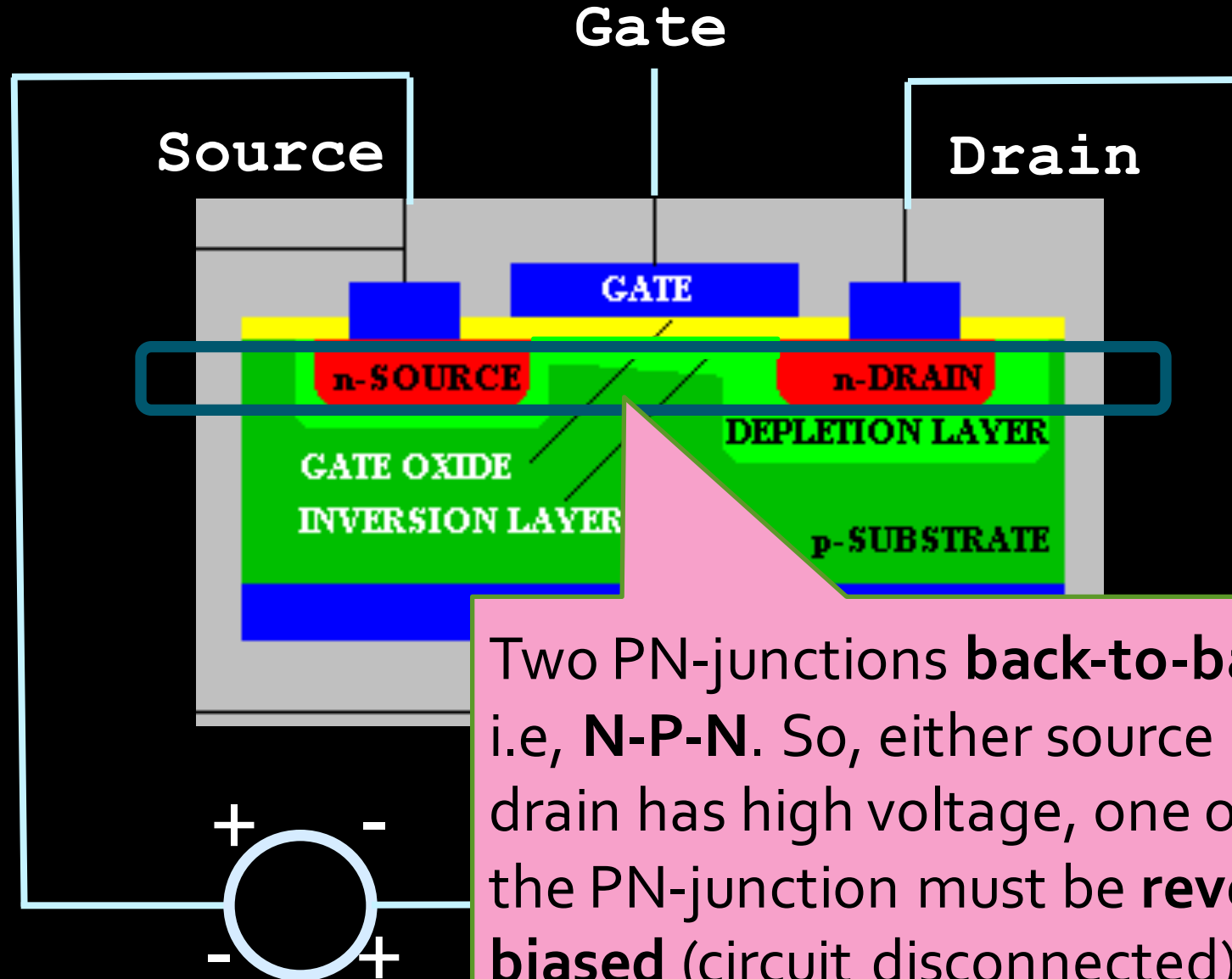
Metal

Oxide

N-type

P-type

Put it into a circuit



Two PN-junctions **back-to-back**, i.e, **N-P-N**. So, either source or drain has high voltage, one of the PN-junction must be **reverse biased** (circuit disconnected).

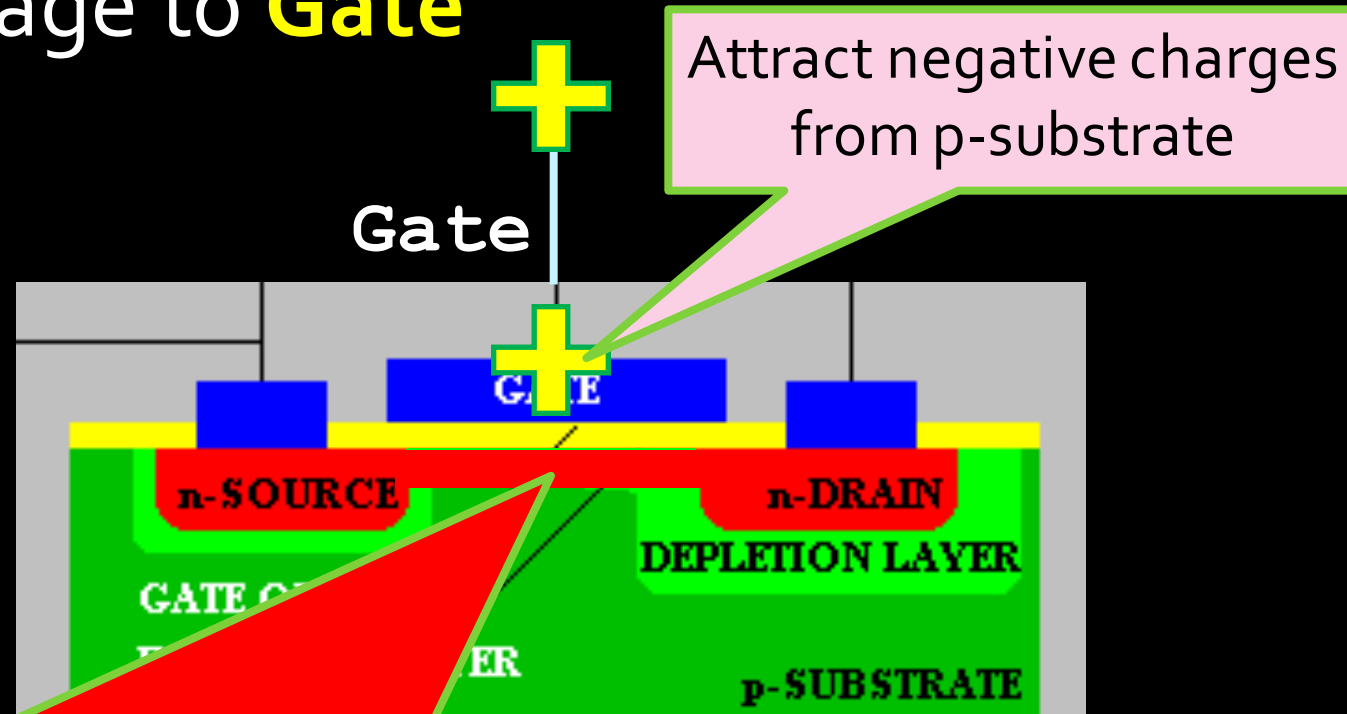
Metal

Oxide

N-type

P-type

But things change if we apply
high voltage to **Gate**

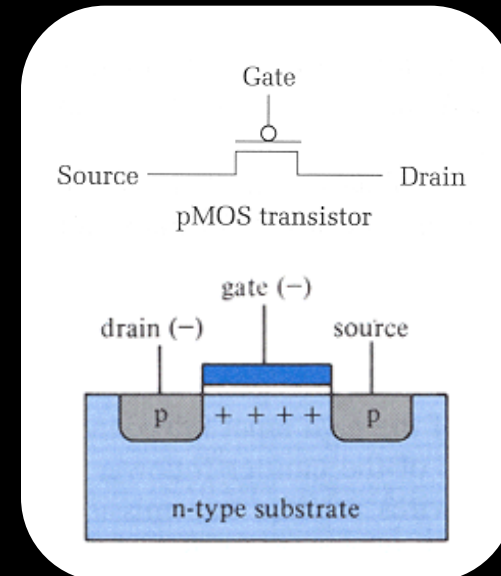
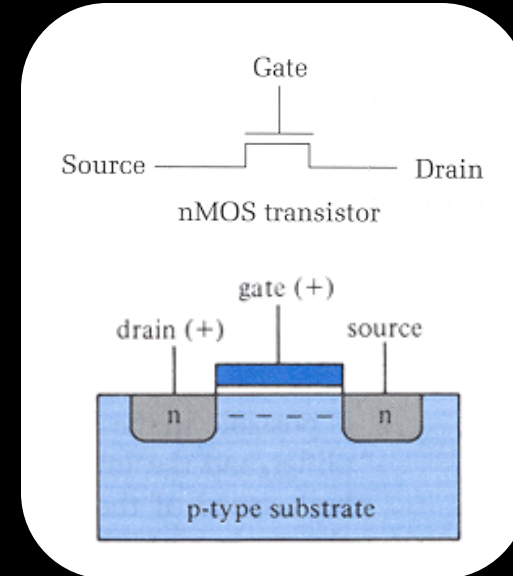


Create n-type **channel** between source and
drain, **CIRCUIT CONNECTED**

The wider the channel, the higher the current

Two types of MOSFET

- **nMOS** (what we just describe)
 - N-P-N
 - Gate high, connected
 - Gate low, disconnected
- **pMOS** (opposite to nMOS)
 - P-N-P
 - Gate low, connected
 - Gate high, disconnected



Outline of the story

- Electricity, basic concepts
- Insulators, conductors, in between ...,
Semiconductors
- Impure semiconductors, **p-type / n-type**
- Put p-type and n-type together -- **pn-junction**
- Apply voltage to a pn-junction – **principle of transistors**
- A real-world manufacturing of transistor –
MOSFET



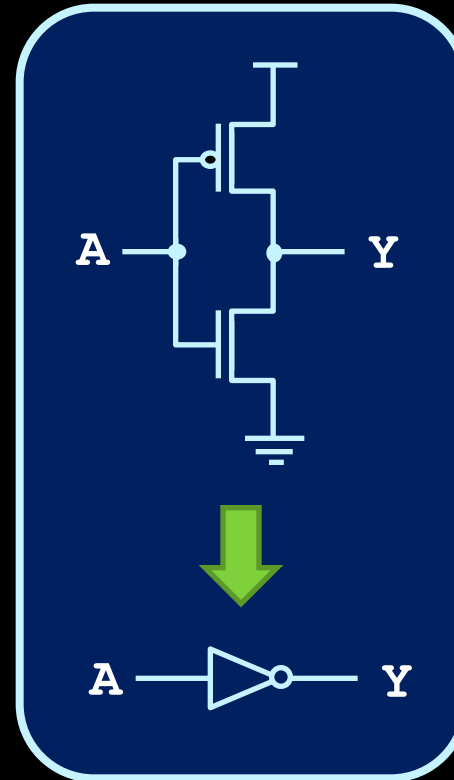
Use transistors build Logic Gates

Transistors to Logic Gates

Create gates using a combination of transistors

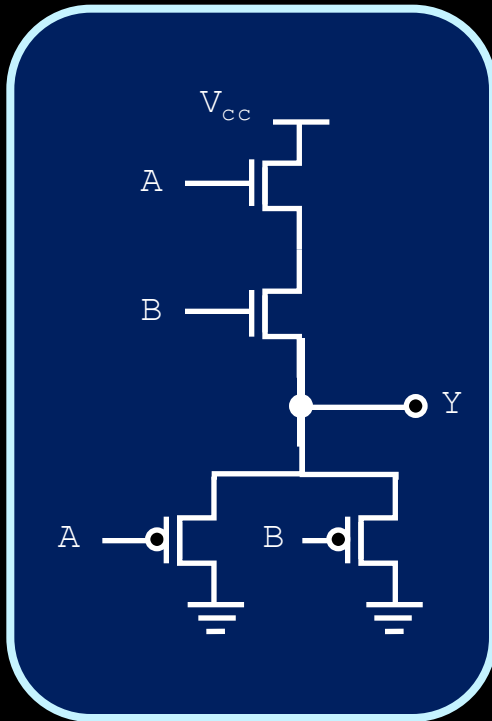
Physical data:

- "High" input = 5V
- "Low" input = 0V
- Switching time ≈ 120 picoseconds
- Switching interval ≈ 10 ns

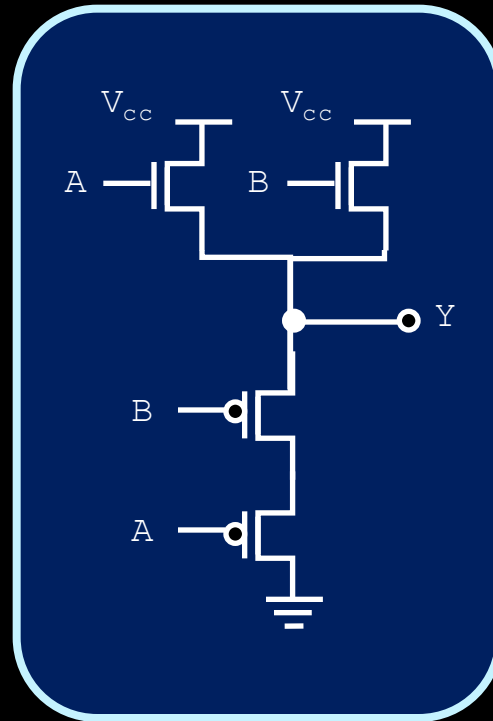


NOT Gate

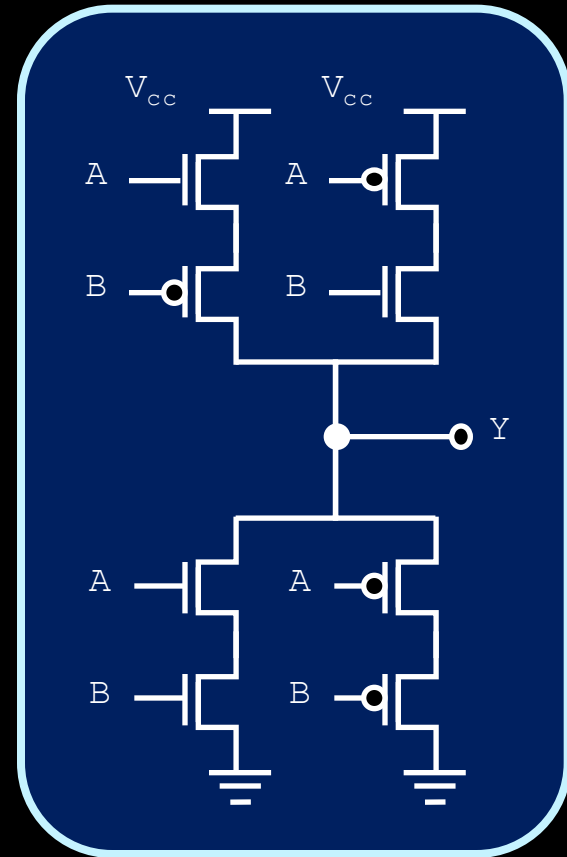
Transistors into gates



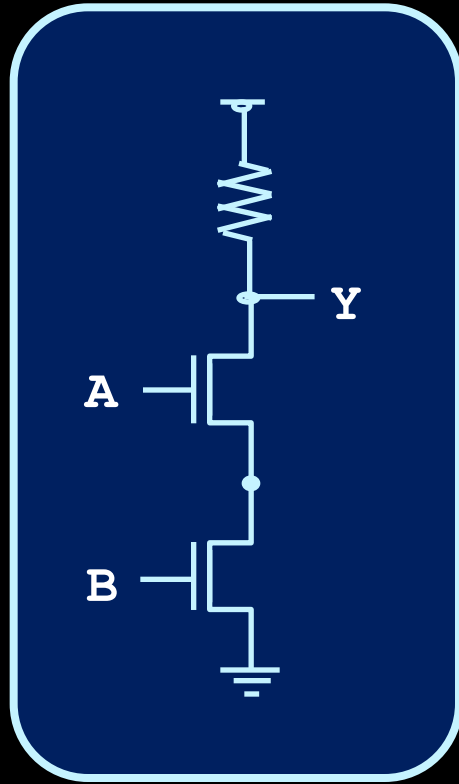
AND



OR



XOR



NAND is the most awesome (and common) logic gate

- It uses **fewer** transistors than other gates
- **All** other logic functions (AND, OR, ...) can be implemented using **only** NAND.



Challenge for home: implement AND, OR, NOT, XOR using only NAND.

Next week:

- Circuit creation