Spin-orbit coupling in an f-electron tight-binding model: Electronic properties of Th, U, and Pu

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We extend a tight-binding method to include the effects of spin-orbit coupling and apply it to the study of the electronic properties of the actinide elements Th, U, and Pu. These tight-binding parameters are determined for the fcc crystal structure using the equivalent equilibrium volumes. In terms of the single-particle energies and the electronic density of states, the overall quality of the tight-binding representation is good and of the same quality as without spin-orbit coupling. The values of the optimized tight-binding spin-orbit coupling parameters are comparable to those determined from purely atomic calculations.

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I. INTRODUCTION

The accurate determination of interatomic forces is crucial for almost all aspects of modeling the fundamental behavior of materials. Whether one is interested in static equilibrium properties using Monte Carlo methods or timedependent phenomena using molecular dynamics, the essential feature remains the origin, applicability, and transferability of the forces acting on the fundamental unit being modeled (atoms or molecules in most cases). First-principles methods based on density-functional theory (DFT) have gained wide acceptance for their ease of use, relatively accurate determination of fundamental properties, and high transferability. These techniques, however, are limited in their application by current computing technology to systems of a few hundred atoms or less (most commonly a few dozen atoms). Potentials that are classically derived (i.e., pair potentials) lack directional bonding (or at best add some bond angle information) and other quantum-mechanical effects but are computationally far more tractable for larger simulations. Recent advances in tight-binding (TB) theory, which include directional bonding but treat only the most important valence electrons shells, therefore show a great deal of promise.

TB models have become a useful method for the computational modeling of material properties, thanks to their ability to incorporate quantum mechanics in a greatly simplified theoretical treatment, making large accurate simulations possible on modern digital computers. 1,2 Another advantage of these TB models is their ability to treat a general class of problems that include directional bonding between valence electrons of particular importance for transition-metal and f-electron materials. Finally, TB models are widely used in many-body formalisms for the one-electron part of the Hamiltonian. It is therefore a useful representation of the band structure for a more sophisticated treatment of electronic correlation and has so been used,³ for example, in dynamical mean-field theory applications for Pu. TB models can also be fitted to experimental results (resulting in an empirical TB model) or a combination of theoretical and experimental data (semiempirical). As will be discussed below, the present study uses highly accurate theoretical calculations as a starting point, but this choice needs not be a constraint other than to recognize possible limitations in the original data used to derive the TB model.

In this paper we present recent developments toward a transferable tight-binding total-energy technique applicable to heavy metals. With the addition of spin-orbit coupling effects for angular momentum up to (and including) f character, we demonstrate the applicability of this technique for the elements Th, U, and Pu, of particular interest for its position near the half-filling point of the 5f subshell in the actinide sequence and the boundary between localized and delocalized f electrons.⁴

II. TB METHOD

The TB model used in this paper is similar to that used in the handbook by Papaconstantopoulos.⁵ We have extended the calculations to include f electrons⁶ and spin-orbit coupling,⁷ of particular interest for the actinides.^{8–11} As such, in this paper we will elaborate only on those aspects of the technique that are unique to this work. A very brief recapitulation of the underlying TB method and its approximations is included to create the proper context for the addition of f electrons and spin-orbit coupling.

The Slater-Koster method¹² consists of solving the secular equation,

$$H\psi_{i,v} = \epsilon_{i,v} S\psi_{i,v},\tag{1}$$

for the single-particle eigenvalues and orbitals under the following restrictions: terms involving more than two centers are ignored, terms where the orbitals are on the same atomic site are taken as constants, and the resulting reduced set of matrix elements is treated as variable parameters. The Hamiltonian, H, includes the labels for orbitals having generic quantum numbers α and β localized on atoms i and j where the effective potential is assumed to be spherical and can be represented as a sum over atomic centers,

1=0 1 = 11 = 21 = 3 $|s\pm\rangle = \sqrt{1/4\pi} |\pm\rangle$ $|p_1\pm\rangle = \sqrt{3/4\pi}f_1(r)x|\pm\rangle$ $|d_1\pm\rangle = \sqrt{5/16\pi}f_2(r)xy|\pm\rangle$ $|f_1\pm\rangle=2\sqrt{105/16\pi}f_3(r)xyz|\pm\rangle$ $|p_2\pm\rangle = \sqrt{3/4\pi}f_1(r)y|\pm\rangle$ $|d_2\pm\rangle = 2\sqrt{15/16\pi}f_2(r)yz|\pm\rangle$ $|f_2\pm\rangle = \sqrt{7/16\pi}f_3(r)x(5x^2-3r^2)|\pm\rangle$ $|d_3\pm\rangle=2\sqrt{15/16\pi}f_2(r)zx|\pm\rangle$ $|f_3\pm\rangle = \sqrt{7/16\pi}f_3(r)v(5v^2-3r^2)|\pm\rangle$ $|p_3\pm\rangle = \sqrt{3/4\pi}f_1(r)z|\pm\rangle$ $|d_4\pm\rangle = \sqrt{15/16\pi}f_2(r)(x^2-y^2)|\pm\rangle$ $|f_4\pm\rangle = \sqrt{7/16\pi}f_3(r)z(5z^2-3r^2)|\pm\rangle$ $|d_5\pm\rangle = \sqrt{5/16\pi}f_2(r)(3z^2-r^2)|\pm\rangle$ $|f_5\pm\rangle = \sqrt{105/16\pi}f_3(r)x(v^2-z^2)|\pm\rangle$ $|f_6\pm\rangle = \sqrt{105/16\pi}f_3(r)y(z^2-x^2)|\pm\rangle$ $|f_7\pm\rangle = \sqrt{105/16\pi}f_3(r)z(x^2-v^2)|\pm\rangle$

TABLE I. TB basis functions used for an $sp^3d^5f^7$ calculation. Note that $f_l(r)=1/r^l$.

$$H_{\alpha i,\beta j} = \langle \alpha,i | -\nabla^2 + \sum_k V_k^{\rm eff} |\beta,j\rangle, \eqno(2)$$

which we further decompose into "on-site" and "intersite" terms,

$$H_{\alpha i,\beta j} = e_{\alpha} \delta_{\alpha \beta} \delta_{ij} + E_{\alpha i,\beta j \neq i}, \tag{3}$$

where the on-site terms, e_{α} , represent terms in which two orbitals share the same atomic site and

$$E_{\alpha i,\beta j\neq i} = \sum_{n} e^{i\mathbf{k}\cdot(\mathbf{R}_{n}+\mathbf{b}_{j}-\mathbf{b}_{i})} \int d\mathbf{r} \psi_{\alpha}(\mathbf{r}-\mathbf{R}_{n}-\mathbf{b}_{i}) H \psi_{\beta}(\mathbf{r}-\mathbf{b}_{j})$$
(4)

are the remaining energy integrals involving orbitals located on different atomic sites, and we have used translational invariance to reduce the number of sums over bravais lattice points $\{\mathbf{R}_n\}$, and the \mathbf{b}_i denote atomic basis vectors within the repeated lattice cells. Note that terms in which both orbitals are located on the same site, but the effective potential (V^{eff}) on other sites, have been ignored. These contributions are typically taken to be "environmental" corrections to the onsite terms and are not accounted for in the usual Slater-Koster formalism.¹² For the intersite terms, the two-center approximation also consists of ignoring these additional terms in which the effective potential, V^{eff} , does not lie on one of the atomic sites [k differs from both i and j in Eq.](2)]. Once this approximation has been made, the interatomic $(i \neq j)$ matrix elements reduce to a simple sum over angular functions, $G_{ll'm}(\Omega_{i,i})$, and functions which depend only on the magnitude of the distances between atoms,

$$H_{\alpha i,\beta j} = \sum h_{ll'm}(r_{ij})G_{ll'm}(\Omega_{i,j}), \qquad (5)$$

where we have now adopted the usual convention of using the familiar l,m angular-momentum quantum numbers and the axis connecting the atoms is the quantization axis. An equivalent expression for $s_{ll'm}$ terms exists when nonorthogonal orbitals are used, and these overlap terms are fitted in the same fashion as the Hamiltonian, as will be discussed below. The basis set used for the α and β quantum states is the cubic harmonics¹³ whose functional forms are given in Table I (with appropriate normalization factors) where $|\pm\rangle$ denotes the spin state, which we will need for spin-orbit coupling.

The Slater-Koster tables 12 for the sp^3d^5 matrix elements can be found in standard references, 14 and we have used the tabulated results of Takegahara et~al. 15 for the additional matrix elements involving f electrons. Typical TB applications are then reduced to using TB as an interpolation scheme; the matrix elements $(h_{ll'm}, s_{ll'm},$ and $e_{\alpha})$ are determined by fitting to ab~initio calculated quantities such as the total energy and band energies.

In this study we restrict ourselves to the determination of optimal TB parameters at the neighbor distances in the face-centered-cubic crystal structure (often used as a surrogate for the more complex ground-state crystal structure of the actinides) near the equilibrium volume. Such tabulations have been extensively used⁵ in the study of materials with lower atomic number. Parameters are presented for light actinide elements Th, U, and Pu that include the f-electron orbitals (although similar parameters have been determined for the elements Ac and Th in an sp^3d^5 basis⁵). The TB parameter values so derived are available (upon request) from the authors.

A. Spin-orbit coupling

The primary impact of spin-orbit coupling is to nontrivially couple electrons of different spin states, thus doubling the size of the TB Hamiltonian. The spin-orbit contribution to the Hamiltonian is given by

$$H^{\text{so}} = \xi(r)\mathbf{L} \cdot \mathbf{S},\tag{6}$$

where $\xi(r) = [\alpha^2/(2r)](\partial V/\partial r)$, V is the total (crystal) potential [we use L-S coupling to more closely reflect the scalar-relativistic treatment of the linearized augmented plane-wave (LAPW) method to which we will ultimately fit]. We neglect contributions from more than one center. A new Hamiltonian matrix can then be defined in terms of the spinless one,

$$\mathcal{H} = H + H^{\text{so}} = \begin{pmatrix} H + \frac{1}{2}\xi L_z & \frac{1}{2}\xi L_- \\ \frac{1}{2}\xi L_+ & H - \frac{1}{2}\xi L_z \end{pmatrix}, \tag{7}$$

where

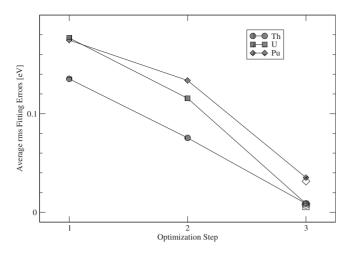


FIG. 1. TB fit quality in terms of the cumulative rms errors at various steps of the optimization procedure. Step 1 relaxes the spin-orbit parameters (ξ_{nl}), step 2 relaxes the remaining on-site parameters, and step 3 is a full relaxation of all parameters. Open symbols at step 3 indicate the original scalar-relativistic fit quality. Note that the cumulative rms error is over all of the fitted bands (20 bands for Th, U, and Pu). Although the spin-orbit coupling is an atomic quantity, the improvement of our results in step 3 (which relaxes intersite parameters) that indicates some environmental effects should also be taken into account.

$$\xi_{nl} = \hbar \int_{0}^{\infty} \xi(r) [R_{nl}^{0}(r)]^{2} r^{2} dr$$
 (8)

is the spin-orbit coupling parameter between orbitals of orbital angular-momentum l and primary quantum numbers n located on the same atom, L_{\pm} are the usual raising and lowering operators, and L_z the azimuthal angular-momentum operator,

$$L_{\pm}Y_{lm}(\theta,\phi) = \hbar \sqrt{l(l+1) - m(m \pm 1)} Y_{lm \pm 1},$$

$$L_{\tau}Y_{lm}(\theta,\phi) = \hbar m Y_{lm}$$
.

The functions $R_{nl}^0(r)$ are the nonrelativistic radial wave functions (which are fitted according to the TB scheme that will be discussed below). The spin-orbit contributions to the Hamiltonian matrix can then be expressed in terms of the TB basis functions listed in Table I. Rather than list contributions for the 32×32 (16 basis functions from Table I and two electron spins) matrix, here we list the matrices in the subblocks corresponding to each orbital angular momentum,

$$H_p^{\text{so}} = \frac{\xi_{np}}{2} \begin{pmatrix} 0 & -i & 0 & 0 & 0 & 1\\ i & 0 & 0 & 0 & 0 & -i\\ 0 & 0 & 0 & -1 & i & 0\\ 0 & 0 & -1 & 0 & i & 0\\ 0 & 0 & -i & -i & 0 & 0\\ 1 & i & 0 & 0 & 0 & 0 \end{pmatrix}, \tag{9}$$

$$H_d^{\text{so}} = \frac{\xi_{nd}}{2} \begin{pmatrix} 0 & 0 & 0 & 2i & 0 & 0 & 1 & -i & 0 & 0 \\ 0 & 0 & i & 0 & 0 & -1 & 0 & 0 & -i & -i\sqrt{3} \\ 0 & -i & 0 & 0 & 0 & i & 0 & 0 & -1 & \sqrt{3} \\ -2i & 0 & 0 & 0 & 0 & 0 & i & 1 & 0 & 0 \\ 0 & 0 & 0 & 0 & 0 & 0 & i\sqrt{3} & -\sqrt{3} & 0 & 0 \\ 0 & -1 & -i & 0 & 0 & 0 & 0 & 0 & -2i & 0 \\ 1 & 0 & 0 & -i & -i\sqrt{3} & 0 & 0 & -i & 0 & 0 \\ i & 0 & 0 & 1 & -\sqrt{3} & 0 & i & 0 & 0 & 0 \\ 0 & i & -1 & 0 & 0 & 2i & 0 & 0 & 0 & 0 \\ 0 & i\sqrt{3} & \sqrt{3} & 0 & 0 & 0 & 0 & 0 & 0 \end{pmatrix},$$

$$(10)$$

$$H_f^{so} = \frac{\xi_{nf}}{4}$$

$$0 \quad 0 \quad 0 \quad 0 \quad 0 \quad 0 \quad 2i \quad 0 \quad 0 \quad 0 \quad 2i \quad 2 \quad 0$$

$$0 \quad 0 \quad \frac{3i}{2} \quad 0 \quad 0 \quad it \quad 0 \quad 0 \quad 0 \quad 0 \quad 0 \quad \frac{3i}{2} \quad 0 \quad 0 \quad t$$

$$0 \quad -\frac{3i}{2} \quad 0 \quad 0 \quad it \quad 0 \quad 0 \quad 0 \quad 0 \quad 0 \quad \frac{3i}{2} \quad 0 \quad 0 \quad it$$

$$0 \quad 0 \quad \frac{3}{2} \quad -\frac{3i}{2} \quad 0 \quad t \quad it \quad 0$$

$$0 \quad 0 \quad -it \quad 0 \quad 0 \quad -\frac{i}{2} \quad 0 \quad -2i \quad 0 \quad 0 \quad -t \quad 0 \quad 0 \quad \frac{1}{2}$$

$$0 \quad -it \quad 0 \quad 0 \quad \frac{i}{2} \quad 0 \quad 0 \quad -2 \quad 0 \quad 0 \quad -it \quad 0 \quad 0 \quad -\frac{i}{2}$$

$$-2i \quad 0 \quad -t \quad -it \quad 0 \quad -\frac{1}{2} \quad 0 \quad 0$$

$$0 \quad 0 \quad 0 \quad \frac{3i}{2} \quad 0 \quad 0 \quad -t \quad 0 \quad 0 \quad -\frac{3i}{2} \quad 0 \quad 0 \quad -it \quad 0$$

$$0 \quad 0 \quad 0 \quad \frac{3i}{2} \quad 0 \quad 0 \quad -t \quad 0 \quad 0 \quad -\frac{3i}{2} \quad 0 \quad 0 \quad -it \quad 0 \quad 0$$

$$0 \quad 0 \quad 0 \quad \frac{3i}{2} \quad 0 \quad -t \quad it \quad 0 \quad 0 \quad 0 \quad 0 \quad 0 \quad 0 \quad 0$$

$$-2i \quad 0 \quad 0 \quad t \quad 0 \quad 0 \quad -\frac{1}{2} \quad 0 \quad 0 \quad it \quad 0 \quad 0 \quad \frac{i}{2} \quad 0$$

$$0 \quad t \quad -it \quad 0 \quad \frac{1}{2} \quad \frac{i}{2} \quad 0 \quad 2i \quad 0 \quad 0 \quad 0 \quad 0 \quad 0 \quad 0$$

where $t = \sqrt{15}/2$. The p and d contributions have been previously discussed in relation to the tight-binding formalism; 16,17 to the best of our knowledge no f contribution has yet appeared in the literature. For completeness we detail the spin-orbit contribution for all values of the angular momentum up to l = 3 (expressed in the basis set of Table I).

B. Fitting the parameters

The values of the TB parameters were determined using standard nonlinear least-squares optimization routines by matching energy-band values derived from highly accurate first-principles DFT calculations. The technique is described in detail in a previous work, where the DFT calculations in this case used a generalized gradient approximation DFT functional and the improved tetrahedron scheme for Brillouin-zone integrations. In this study we use as a starting point high quality fits to the scalar-relativistic energy bands and approximate atomic values of

the spin-orbit parameters. The first step is to then use this fit for fitting the relativistic energy bands including spin-orbit coupling. Successive optimization steps then relax only the spin-orbit coupling parameters (step 1), the remaining on-site parameters (step 2), and finally the intersite terms (step 3). The fit quality through these steps is shown in Fig. 1. Note that the quality of the final fit is comparable to the original fit quality (open symbols in step 3) when only scalar-relativistic effects were taken into account.

III. APPLICATION TO THE LIGHT ACTINIDES, TH, U, AND PU

A. Energy bands including spin-orbit coupling

The first comparison between the TB fit and full-potential augmented planewave (FLAPW) calculations is the energy bands shown in Fig. 2. Note the agreement between the two sets of calculations [the cumulative root-mean-square errors in the TB fits to the first 20 energy bands in the irreducible Brillouin zone are 0.013, 0.013, and 0.072 Ry]. Also note

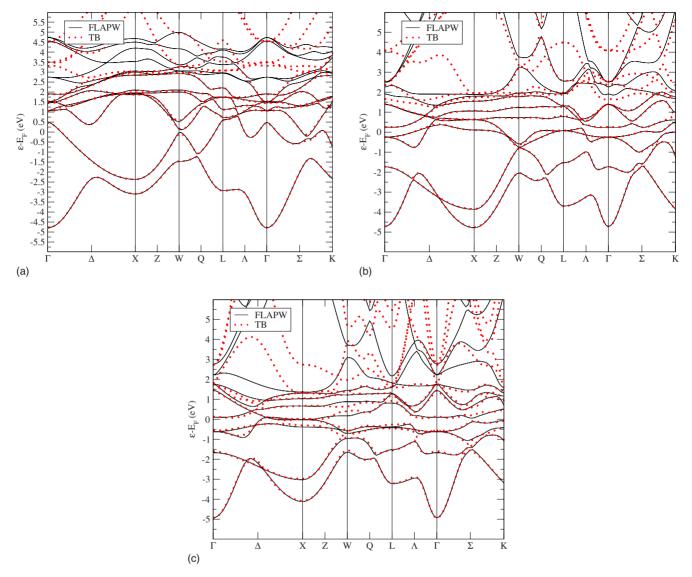


FIG. 2. (Color online) TB energy bands for Th (a=9.61), U (a=8.22), and Pu (a=8.14) shown in comparison with FLAPW valence energy bands (dotted lines). Note the good agreement. The abscissa for each calculation has been shifted such that the Fermi energy is at zero. Higher valence states (above the first 20) are not fit. In addition, such states would require additional basis states to be represented properly. Hence the fit quality is reduced in the energy range well above the Fermi level.

that we have included the "semicore" 6p states in the fit to better fix the available p states in the TB basis. To expand the energy scale comparing the valence bands, the fit quality for the semicore 6p states is shown separately in Fig. 3 for Pu (all three elements have similar fit quality for the more localized 6p states). Note that higher energy bands (well above the Fermi level) are not fit. Such states would also require additional basis sets (orbitals) in order to be treated properly. For these reasons, there are larger discrepancies for those levels. Consistent with most TB studies on materials, the bands of interest are almost always those that lie below the Fermi surface, and, in the interests of efficiency (particularly for large numbers of atoms), a reduced basis set is used that limits accuracy at higher energies.

B. Density of states including spin-orbit coupling

We also compare the total density of states (DOS) between TB and FLAPW methods in Fig. 4. The TB method shown in the figure used a simple Fermi-Dirac temperature smearing method (with $k_BT=500$) for integrating over the irreducible wedge of the Brillouin zone, while the FLAPW calculations used the improved tetrahedron²⁰ method with Gaussian smearing. From the comparison between the TB and FLAPW methods shown in the above figure, we note that the agreement is good, with all major features in the DOS reproduced by the TB calculations. There is a slight reduction in the height of some of the larger peaks in the DOS for the TB technique, most likely due to the inability of the temperature smearing technique to represent the

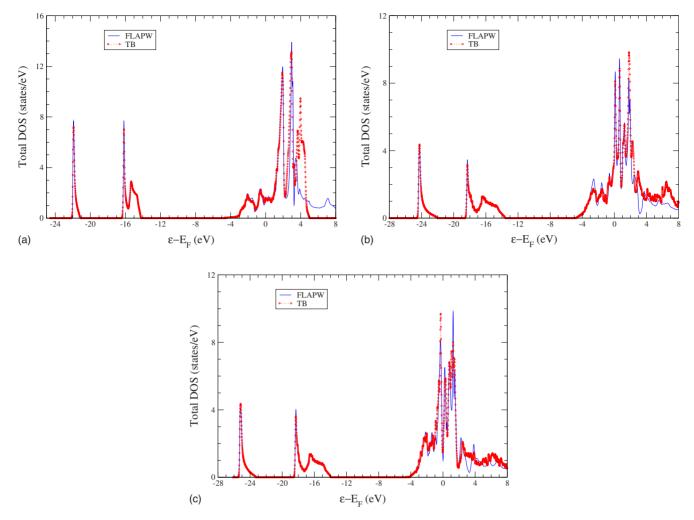


FIG. 3. (Color online) TB energy bands (dotted lines) for Pu semicore 6p states, compared with FLAPW values (solid lines).

finer grained features as well as the improved tetrahedron method.

C. Spin-orbit coupling terms

It is interesting to compare the spin-orbit coupling parameters, ξ_{nl} , predicted by TB theory for the various valence shells relative to the values predicted by accurate Hartree-Fock-Slater (HFS) calculations of isolated atoms.²³ This comparison is shown in Table II.

Note the overall agreement between the TB fitted parameters and the atomic values. The overall shift of a few tenths of an eV for the TB values is interesting, and this trend could be representative of crystal-field effects (this speculation could be checked by performing equivalent fits at different densities). Equivalently, one can compare the spin-orbit splitting of the electronic energy levels with the purely atomic case. This comparison is also shown in Table II.

IV. CONCLUSIONS

We have included *f*-electron and spin-orbit effects in a standard tight-binding method for solids in order to advance

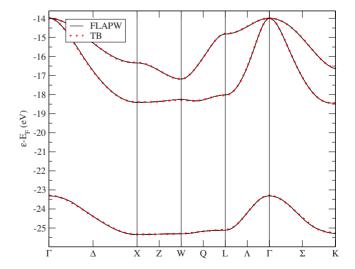


FIG. 4. (Color online) TB (dotted lines) and FLAPW (solid lines) total DOS, including spin-orbit coupling. Note that the TB calculation is in quite good agreement with the FLAPW results despite using a different BZ integration method. The abscissa for each calculation has been shifted such that the Fermi energy is at zero.

TABLE II. Values of spin-orbit coupling strength, ξ_{nl} , and spin-orbit splittings, $\Delta_{nl} = (2l+1)\xi_{nl}/2$, for the various valence electron shells predicted by the TB fit compared with purely atomic values using relativistic DFT (Ref. 21), Dirac-Slater atomic code (Dirac) (Ref. 22), and relativistic HFS (Ref. 23) atomic calculations. Dashed entries are used for orbitals not populated in the atomic calculations. Values are in eV.

Method	ξ_{6p}	Δ_{6p}	ξ_{5d}	Δ_{5d}	ξ_{5f}	Δ_{5f}
			Th			
DIRAC	5.29	7.94	0.20	0.51	0.19	0.66
DFT	5.24	7.86	0.21	0.52		
HFS	4.09	6.14	0.30	0.75		
TB	4.19	6.29	0.20	0.51	0.18	0.62
			U			
DIRAC	5.96	8.94	0.19	0.47	0.24	0.83
DFT	5.90	8.85	0.20	0.50	0.24	0.84
HFS	4.38	6.57	0.30	0.75	0.35	1.24
TB	4.64	6.96	0.23	0.58	0.42	1.48
			Pu			
DIRAC	6.92	10.38	0.20	0.51	0.31	1.10
DFT						
HFS	4.60	6.90			0.41	1.43
TB	5.23	7.84	0.59	1.46	0.54	1.90

simpler simulation methods that are capable of ensuring the accuracy of more expensive full-potential density-functional techniques. We have applied this TB technique to elemental fcc Th, U, and Pu and have achieved good agreement with the electronic properties predicted using a highly accurate FLAPW method. The fitted spin-orbit coupling parameters match very well the values independently predicted by atomic electronic structure calculations. This methodology bodes well for further TB investigations, especially for the study of defects, phonons, and dynamical properties. In future work we intend to develop a more transferable model based on a TB total-energy formalism, which should allow the straightforward calculation of detailed material properties.

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¹C. M. Goringe, D. R. Bowler, and E. Hernandez, Rep. Prog. Phys. **60**, 1447 (1997).

²D. A. Papaconstantopoulos and M. J. Mehl, J. Phys.: Condens. Matter **15**, R413 (2003).

³ J.-X. Zhu, A. K. McMahan, M. D. Jones, T. Durakiewicz, J. J. Joyce, J. M. Wills, and R. C. Albers, Phys. Rev. B **76**, 245118 (2007).

⁴R. C. Albers, Nature (London) **410**, 759 (2001).

⁵D. A. Papaconstantopoulos, *Handbook of the Band Structure of Elemental Solids* (Plenum, New York, 1986).

⁶M. D. Jones and R. C. Albers, Phys. Rev. B **66**, 134105 (2002).

⁷M. Lach-hab, M. J. Mehl, and D. A. Papaconstantopoulos, J. Phys. Chem. Solids **63**, 833 (2002).

⁸ K. T. Moore and G. van der Laan, arXiv:0807.0416, Rev. Mod. Phys. (to be published).

⁹ K. T. Moore, M. A. Wall, A. J. Schwartz, B. W. Chung, D. K. Shuh, R. K. Schulze, and J. G. Tobin, Phys. Rev. Lett. **90**, 196404 (2003).

¹⁰K. T. Moore, G. van der Laan, M. A. Wall, A. J. Schwartz, and R. G. Haire, Phys. Rev. B **76**, 073105 (2007).

¹¹M. T. Butterfield, K. T. Moore, G. van der Laan, M. A. Wall, and R. G. Haire, Phys. Rev. B 77, 113109 (2008).

¹²J. C. Slater and G. F. Koster, Phys. Rev. **94**, 1498 (1954).

¹³F. von Der Lage and H. A. Bethe, Phys. Rev. **71**, 612 (1947).

¹⁴W. A. Harrison, Electronic Structure and the Properties of Solids (Freeman, San Francisco, CA, 1980).

¹⁵K. Takegahara, Y. Aoki, and A. Yanase, J. Phys. C **13**, 583 (1980).

¹⁶J. Friedel, P. Lenglart, and G. Leman, J. Phys. Chem. Solids 25, 781 (1964).

¹⁷D. J. Chadi, Phys. Rev. B **16**, 790 (1977).

¹⁸P. Blaha, K. Schwarz, G. K. H. Madsen, D. Kvasnicka, and J. Luitz, WIEN2K, An Augmented Plane Wave+Local Orbitals Program for Calculating Crystal Properties (Karlheinz Schwartz, Techn. Universität Wien, Austria, 2001).

¹⁹J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. **77**, 3865 (1996).

²⁰P. E. Blöchl, O. Jepsen, and O. K. Andersen, Phys. Rev. B 49, 16223 (1994).

²¹S. Kotochigova, Z. H. Levine, E. L. Shirley, M. D. Stiles, and C. W. Clark, http://math.nist.gov/DFTdata, 1996.

²² ADF2004 01, SCM, Theoretical Chemistry, Vrije Universiteit, Amsterdam, The Netherlands (http://www.scm.com).

²³F. Herman and S. Skillman, *Atomic Structure Calculations* (Prentice-Hall, Englewood Cliffs, NJ, 1963).