



Review

Medically Useful Plant Terpenoids: Biosynthesis, Occurrence, and Mechanism of Action

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Academic Editors: Ewa Swiezewska, Liliana Surmacz and Bernhard Loll Received: 3 October 2019; Accepted: 30 October 2019; Published: 1 November 2019



Abstract: Specialized plant terpenoids have found fortuitous uses in medicine due to their evolutionary and biochemical selection for biological activity in animals. However, these highly functionalized natural products are produced through complex biosynthetic pathways for which we have a complete understanding in only a few cases. Here we review some of the most effective and promising plant terpenoids that are currently used in medicine and medical research and provide updates on their biosynthesis, natural occurrence, and mechanism of action in the body. This includes pharmacologically useful plastidic terpenoids such as *p*-menthane monoterpenoids, cannabinoids, paclitaxel (taxol®), and ingenol mebutate which are derived from the 2-C-methyl-p-erythritol-4-phosphate (MEP) pathway, as well as cytosolic terpenoids such as thapsigargin and artemisinin produced through the mevalonate (MVA) pathway. We further provide a review of the MEP and MVA precursor pathways which supply the carbon skeletons for the downstream transformations yielding these medically significant natural products.

Keywords: isoprenoids; plant natural products; terpenoid biosynthesis; medicinal plants; terpene synthases; cytochrome P450s

1. Plant Terpenoids

Terpenoids, or isoprenoids, are isoprene-based natural products with fundamental roles in the metabolism of all organisms [1]. Terpenoid chemical diversity is especially high in plants where many can be considered secondary metabolites. Such non-essential, specialized plant terpenoids underlie many ecological interactions between plants and animals [2,3], acting as allelochemicals to attract pollinators, repel herbivores, or attract herbivore predators [4]. The evolution of terpenoid secondary metabolism in plants began with the recruitment of genes from primary metabolism [5] and accelerated due to the proliferation of cytochrome P450 and terpene synthase gene families in the genomes of plants [6,7]. Terpenoid chemical diversity partly reflects a natural history marked by herbivory stress and other selective pressures imposed by animals, resulting in a broad array of functionalized terpenoids in the plant kingdom pre-selected for their potent biological activities towards animals [8]. This selective process may have been facilitated by the general similarity of protein folds and motifs between plant and animal proteins, resulting in plant secondary metabolites with natural affinity for animal proteins by virtue of having been produced by plant enzymes composed of the same amino acids [9]. Thus, pre-selection and emergence of plant secondary metabolites with biological activity towards animals may have both an evolutionary and a biochemical basis.

As a consequence, many plant terpenoids have found fortuitous uses in medicine, and the terpenoid family of natural products has been a valuable source of medical discoveries [10–12]. Hundreds more allegedly possess medicinal properties, but the testing process is painstaking and

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resource intensive. The true number of plant terpenoids in nature which could potentially be screened for the applications is unknown but is likely $>10^5$, including >12,000 from the diterpenoid group alone [13]. While this number is small compared to modern combinatorial methods, the lead compound discovery rate may be significantly higher for plant natural products due to the aforementioned pre-selection effects. But owing to their high degree of chemical functionalization and metabolic specialization, many are produced in small amounts, only in response to elicitation, or accumulate exclusively in specialized tissues, necessitating microbial production or major advances by plant breeding and genetic improvement to obtain sufficient quantities to investigate clinical potential [14,15]. This in turn necessitates a detailed understanding of the biosynthetic enzymes, genes, and regulatory programs of a given candidate compound. Here we review the current understanding of the biosynthesis, occurrence, and mechanism of action of a select group of plant terpenoids of medical significance. Due to the large number of plant terpenoids described in the literature in various health contexts, we have excluded those of a purely nutritional, nutraceutical, or anti-oxidant nature and instead focused on compounds with specific pharmacological activity. Cardenolides such as digoxin, which inhibit cardiac Na+/K+ ATPase and improve cardiac muscle contraction and stroke volume, have been reviewed recently [16,17] and therefore were not covered here.

2. Terpenoid Precursor Pathways

The carbon backbone of highly functionalized terpenoid natural products is formed through the condensation of the central metabolic intermediates of terpenoid metabolism, isopentenyl and dimethylallyl diphosphate (IDP and DMADP) (Figures 1 and 2). Two distinct biochemical pathways in nature synthesize them: the 2C-methyl-D-erythritol-4-phosphate (MEP) pathway and the mevalonic acid (MVA) pathway [18–20]. The cytosolic MVA pathway supplies IDP and DMADP in essentially all eukaryotes and the archaea, whereas the MEP pathway is functional in eubacteria and in the plastids of algae, higher plants, and protists [21]; photosynthetic organisms utilize both precursor pathways but separate them into different subcellular compartments. In plants, there is limited evidence of exchange of IDP/DMADP pools between the plastid and cytosol, usually in the context of secondary metabolism [22–26]. In general, the MEP pathway supplies C_5 prenyl diphosphates for the synthesis of C_{10} monoterpenes, C_{20} diterpenes, and C_{40} tetraterpenes while the MVA pathway provides the same universal precursors for the synthesis of C_{15} sesquiterpenes, C_{27-29} sterols, C_{30} triterpenes, and their saponin derivatives.

Figure 1. Biosynthesis of isopentenyl (IDP) and dimethylallyl diphosphate (DMADP) via the mevalonic acid (MVA) pathway and alternative MVA pathway (boxed) in the cytosol. Enzyme abbreviations are as follows; AACT—acetoactl-CoA thiolase, HMGS—hydroxymethylglutaryl-CoA synthase, HMGR—hydroxymethylglutaryl-CoA reductase, MVK—mevalonate kinase, PMK—phosphomevalonate kinase, MVD—mevalonate 5-phosphate decarboxylase, IDI—isopentenyl diphosphate isomerase, PMD—phosphomevalonate decarboxylase, IPK-isopentenyl phosphate kinase, NDX—Nudix hydrolase, AcCoA—acetyl-CoA. **OP** and **OPP** signify mono- and diphosphate moieties, respectively. P_i represents inorganic phosphate.

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Figure 2. Biosynthesis of IDP and DMADP via the MEP pathway in the plastid. Enzyme abbreviations are as follows: DXS—1-deoxy-p-xylulose 5-phosphate synthase, DXR—1-deoxy-p-xylulose 5-phosphate reductoisomerase, MCT—2*C*-methyl-p-erythritol 4-phosphate cytidyltransferase, CMK—4-(cytidine 5'-diphospho)-2*C*-methyl-p-erythritol kinase, MDS—2*C*-methyl-p-erythritol-2,4-cyclodiphosphate synthase, HDS—4-hydroxy-3-methylbut-2-enyl diphosphate synthase, HDR—4-hydroxy-3-methylbut-2-enyl diphosphate reductase. For intermediates, GAP—p-glyceraldehyde-3-phosphate, DXP—1-deoxy-p-xylulose 5-phosphate, MEP—2*C*-methyl-p-erythritol 4-phosphate, CDP-cytidyl diphosphate, MEcDP—2*C*-methyl-p-erythritol-2,4-cyclodiphosphate, HMBDP—1-hydroxy-2-methyl-2-(*E*)-butenyl-4-diphosphate, IDP—isopentenyl diphosphate, DMADP—dimethylallyl diphosphate; **OP** and **OPP** signify mono- and diphosphate groups, respectively.

2.1. The Mevalonate Pathway

The cytosolic MVA pathway produces one molecule of IDP in six enzymatic steps with the consumption of 3 acetyl-CoAs, 3ATPs, and 2 NADPH reducing equivalents [27,28] (Figure 1). The pathway begins with the Claisen condensation of two acetyl-CoA units to form acetoacetyl-CoA, a reaction carried out by acetoacetyl-CoA thiolase [29], followed by condensation with a third acetyl-CoA unit by 3-hydroxy-3-methylglutaryl-CoA synthase (HMGS) to yield HMG-CoA [30]. HMG-CoA reductase (HMGR) then reduces HMG-CoA to mevalonate at the endoplasmic reticulum (ER) surface [31], the main regulatory step of the MVA pathway, consuming 2 NADPH molecules in the process. Mevalonate kinase then phosphorylates mevalonate into mevalonate-5-phosphate [32] which is followed by a second phosphorylation into mevalonate 5-diphosphate by phosphomevalonate kinase. The final step of the MVA pathway is the ATP-dependent decarboxylation of mevalonate diphosphate into IDP by mevalonate diphosphate decarboxylase [33]. There is increasing evidence that the final steps of the MVA pathway take place in the peroxisome [34,35]. Following IDP formation, it can be reversibly isomerized into DMADP by IDP isomerase [36] and further converted into the C₁₅ prenyl diphosphate farnesyl diphosphate (FDP) by FDP synthase [37], a member of the prenyl transferase family of enzymes which condense DMADP with a variable number of IDP units to produce prenyl diphosphates, usually of 10, 15, or 20 carbons, although longer chain prenyl diphosphates are also known. FDP is the last common intermediate among most MVA-derived plant terpenoids.

These reactions are highly conserved within plants; however, an alternative MVA route that operates simultaneously was recently identified (Figure 1). It resembles the ancestral MVA pathway in archaea wherein the last two steps are reversed, i.e., mevalonate 5-phosphate is first decarboxylated to isopentenyl phosphate, which is then phosphorylated to IDP [38]. In plants, IDP and DMADP can evidently be interconverted with their monophosphates forms via NUDIX hydrolase [39] and isopentenyl monophosphate kinase activities [40]. This process is thought to play a regulatory function to control terpenoid precursor availability, but its broader biological significance is still being explored.

2.2. The MEP Pathway

The MEP pathway provides IDP and DMADP for terpenoid biosynthesis in the plastid. Over a sequence of seven reactions (Figure 2), D-glyceraldehyde-3-phosphate (GAP) and pyruvate undergo condensation and reduction to yield IDP and DMADP at the expense of 3 ATP and 3 NADPH equivalents [19]. 1-Deoxy-D-xylulose 5-phosphate (DXP) synthase (DXS) condenses pyruvate and GAP

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into DXP in the committing step, an enzyme which also controls flux through the MEP pathway [41]. Most plant genomes encode multiple DXS isoforms belonging to distinct phylogenetic groups, denoted type I and type II [42]. While type I genes are expressed in photosynthetic tissue, the type II DXS clade is frequently associated with the biosynthesis of terpenoid secondary metabolites [43–45]. This class of DXS may therefore represent a biotechnological target for the manipulation of specialized terpenoids of medical interest. Type II DXS genes are often preferentially expressed in tissues where secondary metabolites accumulate, such as trichomes [46,47] or roots [48,49].

Following DXP formation, DXP reductoisomerase (DXR) then reduces it to MEP through an NADPH-dependent isomerization [50]. Following activation of MEP with a cytidine diphospho linkage at C4 [51] and phosphorylation of the alcohol at C2, the enzyme 2C-methyl-p-erythritol-2,4-cyclodiphosphate (MEcDP) synthase (MDS) then cyclizes this activated intermediate to the unusual 8-membered phosphodiester ring, MEcDP [52,53]. MEcDP is then reduced to 1-hydroxy-2-methyl-2-(E)-butenyl-4-diphosphate (HMBDP) by the iron-sulfur protein HMBDP synthase (HDS) [54,55] and then reduced again by HMBDP reductase (HDR), which produces both IDP and DMADP in a ratio of about 5:1 [56,57]. IDP isomerase is also present in the plastid, presumably to maintain IDP/DMADP ratios at optimal levels [58]. In analogy with FDP synthase noted above, the principal prenyl transferases which supply plastidic terpene biosynthesis include geranyl diphosphate (GDP) and geranylgeranyl diphosphate (GGDP) synthase, which provide C₁₀ and C₂₀ prenyl diphosphates for the synthesis of mono- and diterpenes, respectively. In chloroplasts, the MEP pathway mainly provides precursors for the synthesis of carotenoids and phytol for photosynthesis, along with other primary metabolites [59]. Secondary metabolites, such as those with uses in medicine, are often made in specialized, pigment-free plastids known as leucoplasts located in non-photosynthetic glandular trichomes [60,61].

3. Medicinal Monoterpenes

Monoterpenes are C_{10} , plastid derived terpenoids, often with appreciable volatility. As such, they are frequent constituents of plant essential oils. Their anti-microbial action has been attributed to their general membrane disrupting properties [62]. Below we described two class of monoterpenoids with well documented pharmacological properties, the volatile p-menthane (–)-menthol and the cannabinoids, whose carbon skeleton includes portions derived from GDP as well as a polyketide fragment and a fatty acid contribution of variable length.

3.1. (–)-Menthol and Related p-Menthanes from the Lamiaceae

The *Mentha* genus (Lamiaceae) provides several p-menthane monoterpenoids of pharmacological interest. (–)-Menthol (Figure 3), a major constituent of the essential oil of peppermint (*Mentha* x *piperita*), has been known since the 1950s to act as a full agonist of the Cold and Menthol Receptor 1 (CMR1) [63]. The CMR1 receptor, also known as 'transient receptor potential melastatin 8' (TRPM8), is a cold-sensitive, homotetrameric Na⁺/Ca²⁺ ion channel expressed in sensory neurons [64]. Its activation by cold, positive membrane potential, phosphatidylinositol-4,5-bisphosphate, or (–)-menthol produces a cold sensation, although the gating mechanism is different in each case [65,66]. This stands in contrast to other plant natural products such as capsaicin, which activates the heat sensitive TRP vanilloid 1 receptor to produce the opposite effect [67]. The Ca²⁺ blocking properties of (–)-menthol also confer mild pain relieving and analgesic effects through the indirect activation of the κ opioid receptor [68], relaxation of smooth muscle, and positive allosteric activation of the GABA_A receptor [69]. These pharmacological attributes may explain its traditional and current use in relieving muscle pain and in the treatment of irritable bowel syndrome [70], in addition to its better known role as a flavoring agent in oral hygiene products and as a food additive.

The biosynthesis of (–)-menthol begins with GDP (Figure 3), the immediate precursor to most monoterpenes [71] with rare exceptions [72]. GDP itself is formed through the condensation of IDP and DMADP through the action of GDP synthase (GDS), a heterodimer with a large catalytic and smaller chain length specificity subunit first cloned from peppermint (Mentha~x)

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piperita L.) and spearmint (*Mentha spicata* L.) [73]. (–)-Limonene synthase acts on GDP to produce the cyclic, olefinic monoterpene (–)-limonene with loss of pyrophosphate as the first committed intermediate [74], to which (–)-limonene-3-hydroxylase then introduces an alcohol function at C3 to form (–)-*trans*-isopiperitenol [75]. (–)-*trans*-Isopiperitenol dehydrogenase next oxidizes this to (–)-*trans*-isopiperitenone [76] followed by its reduction to (+)-*cis*-isopulegone [77]. (+)-*cis*-Isopulegone isomerase then yields (+)-pulegone [78], the branch-point intermediate leading alternately to (+)-menthofuran or the menthones [79]. In the formation of (–)-menthol, (+)-pulegone reductase reduces (+)-pulegone to (–)-menthone [77]. (–)-Menthol is then formed by a final reduction by (–)-menthone:(–)-menthol reductase [80]. These seven enzymatic steps begin in the plastid, move to the ER, continue in the mitochondria, and the final steps are completed in the cytosol [81]. Ultimately, the products are secreted into the subcuticular storage space of the oil gland. Developmental regulation of (–)-menthol biosynthesis in glandular trichomes of peppermint occurs primarily at the transcriptional level [82].

The metabolic control of this process has been the subject of extensive mathematical modeling for the purpose of engineering oil quality [14,83–85]. Since peppermint is a sterile hybrid, plant breeding is not available as a means to improve oil yields and quality, for instance, by reducing (+)-menthofuran content, an undesirable side product resulting from the oxidation of (+)-pulegone by (+)-menthofuran synthase, a cytochrome P450 monooxygenase [79]. Indeed, the toxicity of pennyroyal (*M. pulegium*) is due to its high (+)-pulegone content, which when ingested is similarly converted to (+)-menthofuran by a different P450 enzyme in the human liver [86]. The uterine contraction-inducing, hepatotoxic properties of pennyroyal are due to this bioactivation and explain the historical use of its essential oil as an abortifacient and emmenagogue in folk medicine dating to at least 421 B.C. [87], despite its obvious toxicity in concentrated form. Biotechnological approaches have successfully reduced (+)-menthofuran content in peppermint through antisense silencing of (+)-menthofuran synthase while simultaneously increasing yields of the desirable (–)-menthol through upregulation of DXR, the second enzyme of the MEP pathway [88].

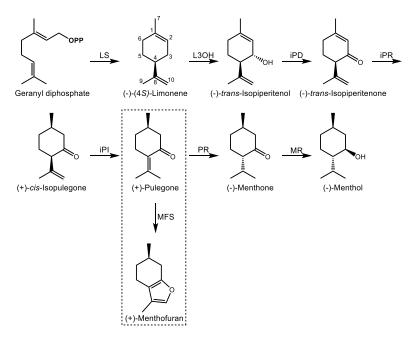


Figure 3. Steps in the biosynthesis of (–)-menthol and related p-menthanes in peppermint (M. x piperita). Enzyme abbreviations are as follows; LS–(–)-limonene synthase, L3OH—(–)-limonene 3-hydroxylase, iPD—(–)-trans-isopiperitenol dehydrogenase, iPR—(–)-trans-isopiperitenone reductase, iPI—(+)-cis-isopulegone isomerase, PR—(+)-pulegone reductase, MR—(–)-menthone reductase, MFS—(+)-menthofuran synthase. **OPP** signifies a pyrophosphate group. The box indicates the branch pathway leading to (+)-menthofuran.

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3.2. Medicinal Monoterpene Derived Meroterpenes: Cannabinoids

'Cannabinoid' originally referred to the group of prenylated phenolic compounds from *Cannabis spp.* (Cannabaceae) but now includes any ligand capable of specifically binding to the human cannabioid receptors, even endogenously produced cannabinoids with no structural similarity to their plant-derived, terpenophenolic counterparts [89]. Here we focus exclusively on cannabinoids of plant origin. *C. sativa* L. and *C. indica* Lam. (marijuana) have been used for their psychoactive, anxiolytic and anesthetic effects for thousands of years [90]. Like *p*-menthane monoterpenoids in the Lamiaceae, cannabinoids accumulate in glandular trichomes, but in *Cannabis spp.*, cannabinoid-rich trichomes occur primarily in calyces and bracts of female flowers [91].

The structure of cannabinoids includes a C_{10} GDP derived monoterpene backbone, a polyketide resorcinol ring, and an acyl chain of variable length, usually 3- or 5-carbons [92]. The two best known cannabinoids from C. sativa include Δ^9 -tetrahydrocannabinol (THC) and cannabidiol (CBD) (Figure 4), known for their psychoactive and pain relieving properties, respectively. They share a common set of early biosynthetic steps and diverge at the central intermediate cannabigerolic acid (CBGA), a central intermediate in cannabinoid biosynthesis. Geranylpyrophosphate:olivetolate geranyl-transferase (GOT) produces CBGA through the prenylation of olivetolic acid (OA) [93]. OA is the product of the condensation of hexanoyl-CoA with 3 malonyl-CoA units whose formation requires the combined action of olivetol synthase [94] as well as olivetolic acid cyclase [95]. A gene responsible for hexanoyl-CoA formation has not yet been isolated. CBGA can then undergo oxidative cyclization to CBD acid (CBDA) or THC acid (THCA) through the action of one of two related enzymes. In the first, CBDA synthase, a flavin-bearing enzyme, oxidatively cyclizes the geranyl group through a mechanism involving deprotonation of the terminal methyl group on the geranyl chain, leaving a free propylene group [96]. Conversely, CBGA may also undergo a similar oxidative cyclization by THCA synthase [97]. In this case, deprotonation of the carbocationic intermediate occurs on the C1' hydroxyl group of the alkylresorcinol ring which then attacks the propylene tail, leading to formation of a third ring [98]. This additional ring distinguishes THCA from CBDA, although the enzyme mechanisms are nearly identical. Both CBDA and THCA then undergo spontaneous decarboxylation to form the biologically active components CBD and THC, respectively [92].

CBDA and THCA bear a characteristic pentyl side chain on C5′ of the resorcinolic acid ring. However, other pharmacologically active cannabinoids containing acyl groups of different lengths [99] are also known. Propyl substituted analogs of CBDA and THCA exist in the form of cannabidivarinic acid (CBDVA) and tetrahydrocannabivarinic acid (THCVA) (Figure 4). It is likely they are formed by the same enzymes yielding their bi- and tri-cyclic pentyl counterparts (CBDAS and THCAS, respectively), although this area continues to be investigated. The resorcinolic acid component of propyl cannabinoids is divarinolic acid in place of OA [100], and butanoyl-CoA is the presumed acyl thioester contributing the alkyl group [101].

The principal target of THC is the cannabinoid 1 (CB1) receptor, the primary cannabinoid receptor in the central nervous system [102] whose discovery followed from the affinity of the plant compound towards CB1 [103]. This G-protein coupled receptor functions as part of an inhibitory pathway to block calcium influx in the pre-synaptic nerve needed for release of excitatory neurotransmitters such as glutamate into the synaptic cleft [104]. The targets of CBD are less clear [105,106] but both have demonstrated efficacy towards treating epilepsy [107].

Pharmacologically active terpenophenolics from *Cannabis spp.* are not limited to cannabinoids *sensu stricto*. Two prenylated flavones, cannflavin A and B [108], result from the transfer of either a geranyl or dimethylallyl group, respectively, to the A ring of the methylated flavone chrysoeriol (Figure 5). The position of the alcohol groups on the aromatic ring relative to the site of prenylation is identical to that of the resorcinol ring of OA during the formation of CBGA. Rea et al. recently described two cDNAs involved in the formation of cannflavins [109], namely an *O*-methyl transferase converting luteolin to chrysoeriol and an aromatic prenyltransferase accepting either GDP or DMADP as prenyl donor to produce cannflavin A and B, respectively. Their analgesic and anti-inflammatory activity [110]

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derive from inhibition of two enzymes in the E2 prostaglandin inflammatory pathway [111]. The recent isolation of the cDNAs for the cannflavin pathway will enable a more detailed evaluation of their clinical potential for pain relief and treatment of chronic inflammation.

Figure 4. Steps in the biosynthesis of major cannabinoids in *C. sativa* and *C. indica*. Metabolites are as follows: GDP—geranyl diphosphate, OA–olivetolic acid, CBGA–cannabigerolic acid, CBDA—cannabidiolic acid, CBD—cannabidiol, THCA— Δ^9 -tetrahydrocannabinic acid, THC— Δ^9 -tetrahydrocannabinol, DVA—divarinic acid, CBGVA—cannabigerovaric acid, CBDVA—cannabidivarinic acid, CBV—cannabivarin, THVA—tetrahydrocannabivarinic acid, THV—tetrahydrocannabivarin. Enzymes are as follows: GOT—geranylpyrophosphate:olivetolate geranyl transferase, THCAS—tetrahydrocannabinolic acid synthase, CBDAS—cannabidiolic acid synthase, THCAS—tetrahydrocannabinolic acid synthase, CBDAS—cannabidiolic acid synthase. The traditional name Δ^9 -THC is used in the text while the monoterpene numbering scheme is used in this figure. **OPP** indicates a diphosphate group.

Figure 5. Biosynthetic pathway for the formation of cannflavin A and B in *C. sativa* and *C. indica*. This terpenophenolic compound is produced from geranyl diphosphate (GDP) or dimethylallyl diphosphate (DMADP) and the flavonoid chrysoeriol derived from naringenin in *C. sativa* and *C. indica*. SAM—S-adenosylmethionine, SAhc—S-adensoylhomocysteine. Enzymes are as follows: CHS—chalcone synthase, CHI—chalcone isomerase, FNS—flavone synthase, F3'H—flavanoid 3' hydroxylase, CsOMT21—*O*-methyltransferase 21, CsPT3—prenyltransferase 3.

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4. Medicinal Sesquiterpenes

Sesquiterpene biosynthesis normally begins in the cytosol via the C_{15} prenyl diphosphate FDP, which itself is derived from mevalonic acid and the MVA pathway. Sesquiterpene synthases act on FDP to yield hundreds of sesquiterpene hydrocarbons and alcohols based on dozens of carbon skeletons. They are frequently responsible for familiar scents and flavors, for instance ginger (gingerol), clove, cannabis, rosemary (β -caryophyllene), patchouli (patchoulol), sandalwood (α -santalene), and rain (geosmin, a bacterial sesquiterpene). They are the heaviest of the volatile terpenes under standard conditions (diterpene hydrocarbons ordinarily require heating to form gases). Plant families known to be principal producers of sesquiterpene volatiles include the Lamiaceae, Geraniaceae, Rutaceae, Myrtaceae, Cannabaceae, and Gingeraceae. The use of these essential oils in traditional herbal medicine such as aromatherapy and Ayurvedic medicine is well documented [112]. There is currently little evidence their use confers actual medical benefits, yet their use remains widespread due to their aesthetic, historical, and cultural significance. On the other hand, a number of highly functionalized, non-volatile sesquiterpenes of plant origin have demonstrated highly specific biological activity and have therefore been investigated for their clinical potential. A selection of these medically useful plant sesquiterpenoids follows.

4.1. Artemisinin

Wormwood (*Artemisia annua* L., or qinghaosu, Asteraceae) is native to China and is the source of the sesquiterpene endoperoxide artemisinin, shown to be highly effective at treating malaria [113,114]. In the early 1970s, phytochemists such as Youyou Tu investigated plants used in traditional Chinese herbal medicine and identified the usefulness of the traditional remedy qinghaosu in eradicating *Plasmodium spp.* infections, the causative agent of malaria, for which she received a Nobel Prize in Medicine in 2015 [115]. It is more effective against a broader range of life cycle stages of the apicomplexan parasite than traditional anti-malarials such as quinine [116].

Artemisinin is a sesquiterpene lactone with an unusual endoperoxide ring which imparts its anti-malarial properties (Figure 6). The mechanism of action is still unclear [117], and several possibilities have been proposed [118–120]. Most involve the degradation of the endoperoxide bridge in a heme-dependent process to form carbon centered radicals which then alkylate multiple targets including heme and proteins. The involvement of heme is consistent with the specificity of artemisinin towards *Plasmodium spp*. [116]. The direct cause of toxicity to the parasite is unclear but may involve interference with the conversion of heme to hemozoin [121], although this hypothesis has been disputed [122] (and reviewed in reference [123]).

The early stages of artemisinin biosynthesis in glandular trichomes of $A.\,annua$ are now understood at the biochemical and molecular levels, although the exact mechanism of the later stages remains unclear. Current efforts now focus on the regulation of this pathway in an effort to improve artemisinin production directly in plants. The pathway begins with the conversion of FDP to amorpha-4,11-diene by amorpha-4,11-diene synthase (Figure 6) [124,125]. Biochemical characterization and metabolite profiling of plant tissue indicated the presence of this olefin as well as its oxygenated derivatives artemisinic alcohol and aldehyde, dihydroartemisinic alcohol and aldehyde, and dihydroartemisinic acid [126]. Following the cloning and expression of CYP71AV1, which converts amorpha-4,11-diene to artemisinic alcohol and aldehyde [127], Zhang et al. characterized the artemisinic aldehyde Δ 11(13) reductase (Double bond reductase 2, Dbr2), which preferentially reduces artemisinic aldehyde to dihydroartemisinic aldehyde [128]. This substrate is then converted to dihydroartemisinic acid by aldehyde dehydrogenase 1 (Aldh1) [129]. From dihydroartemisinic acid, the remaining steps involve the photooxidative formation of the endoperoxide ring. However, the details of this process, such as the potential involvement of additional enzyme activities, are currently unclear.

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Figure 6. Steps in the biosynthesis of the sesquiterpene lactone artemisinin in *A. annua*. Enzyme abbreviations are as follows: ADS–amorpha-4,11-diene, CYP71AV1–cytochrome P450 71AV1, DBR2–double bond reductase 2, Aldh1–aldehyde dehydrogenase 1. **OPP** signifies a diphosphate moiety and FDP represents farnesyl diphosphate. The open arrow indicates photooxidative steps involved in the formation of the endoperoxide ring which remain uncharacterized.

4.2. Thapsigargin

Thapsigargin is a highly functionalized 6,12-guianolide sesquiterpene lactone produced by Thapsia garganica (Apiaceae), a perennial shrub native to the western Mediterranean basin. Its resin elicits contact dermatitis [130]. As such, it is sometimes referred to as 'deadly carrot'. However, it has long been utilized in European and North African herbal medicine traditions for rheumatism, sterility, and colds [131] and its use in Algeria for circulatory illnesses dates to the 19th century [132]. Patkar, et al. reported its histamine releasing properties in the late 1970s [133], and its protein target in mammalian cells was later identified as the sarcoplasmic reticulum Ca^{2+/}ATPase pump (SERCA) [134]. Following additional characterization of its properties as a SERCA inhibitor [135], its apoptosis inducing potential became apparent [136]. SERCA is responsible for pumping cytosolic Ca²⁺ into the sarcoplasmic reticulum in myocytes during muscle contraction/relaxation cycles. Thapsigargin mediated inhibition of SERCA therefore blocks muscle relaxation by preventing removal of Ca²⁺ ions from the cytosol. Thapsigargin induces endoplasmic reticulum stress generally, and one consequence of this is the inhibition of autophagosome fusion with the lysosome during autophagy, resulting in apoptosis [137]. Thus, its ability to disrupt Ca²⁺ homeostasis in mammalian cells suggested a potential role in cancer therapy, yet its high cytotoxicity and lack of specificity for tumor cells precluded further development as an anti-cancer drug until recently. Denmeade et al. successfully coupled thapsigargin to a prostate specific antigen (PSA) [138], effectively generating a prostate cancer targeted prodrug. Mipsigargin, a thapsigargin-based variant of this PSA-coupling approach targeting solid tumors [139], is now in phase II clinical trials for hepatocellular carcinoma [140]. Yields of thapsigargin are low from T. garganica (<0.5% by dry weight), and the plant is difficult to cultivate, making a detailed characterization of the relevant genes and enzymes instrumental to achieving industrial scale production.

Our current knowledge of the biosynthesis of thapsigargin is limited, and complete cDNA cloning and biochemical characterization have been accomplished for only the initial steps to date (reviewed in [141]). In the committing step of the pathway, the sesquiterpene synthase TgTPS2 converts FDP into

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the sesquiterpene alcohol *epi*-kunzeaol (Figure 7) [142]. Screening of *T. garganica* cytochrome P450s by transient expression in Nicotiana benthamiana followed by mass spectrometry analysis demonstrated that only one enzyme, TgCYP76AE2, was capable of transforming epi-kunzeaol into a product, identified as the germacrenolide sesquiterpenes lactone *epi*-dihydrocostunolide [143]. Unlike the biosynthesis of costunolide, which requires two separate cytochrome P450s to form the alcohol and acid groups prior to lactonization, the preexistence of a C6 hydroxyl group in epi-kunzeaol allows for direction lactone ring formation following a triple oxidation at C12 by TgCYP76AE2, a reaction similar to that of germacrene A oxidase [144]. Andersen et al. also reported detection of transcripts for TgTPS2 and TgCYP76AE2 in the epithelial cells lining secretory ducts of *T. garganica* root sections. Imaging mass spectrometry detection of thapsigargin in structures resembling secretory ducts corroborated these observations, strongly implicating epithelial cells in the early steps of thapsigargin biosynthesis [143]. Additional reactions to reach thapsigargin that currently remain to be elucidated include, at a minimum, closure of the 10-membered epi-dihydrocostunolide ring into five and seven membered rings, six additional hydroxylations, and four acylations. The difficulty in detecting additional intermediates in root extracts which might provide clues as to the predominant reaction sequence in vivo could be an indication of the involvement of a metabolon structure which efficiently transfers metabolic intermediates between active sites [145]. Improvements in bioinformatics approaches and transient expression technologies which mitigate the necessity of obtaining synthetic substrates will no doubt play a role in the elucidation of the remaining steps of the pathway.

Figure 7. Initial biosynthetic steps for the formation of the sesquiterpene lactone thapsigargin in *T. garganica*. Enzyme abbreviations are as follows: TgTPS2–*epi*-kunzeaol synthase, TgCYP76AE2–cytochrome P450 76AE2. **OPP** signifies a diphosphate moiety. The open arrow indicates uncharacterized steps between *epi*-dihydrocostunolide and thapsigargin.

5. Medicinal Diterpenoids

In addition to C_{10} monoterpenoids, the plastid is also the site of several C_{20} plant diterpenoids of medical significance. Although not all reactions in these complex pathways take place in the plastid, the initial steps leading to the prenyl diphosphate precursor of diterpenes, GGDP, and the cyclization to an olefinic hydrocarbon by a terpene synthase occur here.

5.1. Paclitaxel

The noted anti-neoplastic agent paclitaxel (also known by its trademarked name taxol®) was first isolated from the bark of the pacific yew tree (*Taxus brevifolia*) and structurally elucidated in 1971 [146]. A substantial effort was required to elucidate its structure (Figure 8) due to the complexity of its functional groups. Paclitaxel possesses eight oxygen functional groups, including two with acetylations, a benzoylation, an oxetane ring, and a side chain attachment at C13 composed of a β-phenylalanyl group which is further modified by hydroxylation and *N*-benzoylation [147]. Evidence for its mechanism of action as a microtubule stabilizer and inhibitor of mitosis [148] emerged a few years later, but FDA approval for use in refractory ovarian cancer and metastatic breast cancer was not issued until 1992 and1994, respectively [149]. Much of this delay was due to the exceptionally low abundance of paclitaxel in plant tissue, a restriction which to this day limits the world wide availability of what is now the most commercially successful anti-cancer agent with combined annual sales of derivatives and generics approaching ten billion USD [150,151].

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Figure 8. Simplified pathway for the biosynthesis of paclitaxel (taxol) in *T. brevifolia* (pacific yew). GGDP–geranylgeranyl diphosphate, **OPP**–diphosphate, Bz–benzyl, Ac–acetyl, CoA–Coenzyme A. Enzymes: TS–taxa-4(5),11(12)-diene, T5 α OH–taxadiene-5 α -hydroxylase, TAT–taxadiene-5 α -ol-O-acetyl transferase, T9 α DH–taxane-9 α -dehydrogenase, DBAT–10-deacetylbaccatin III-10-O-acetyl transferase, BAPT–baccatin III:3-amino,13-phenylpropanoyl transferase, T2′- α OH–taxane-2′ α -hydroxylase, DBTNBT–debenzoyltaxol *N*-benzoyl transferase, PAM–phenylalanine aminomutase, PCL– β -phenylalanyl-CoA ligase. The open arrow represents multiple oxidative steps to arrive at the hypothetical intermediate.

Taxus cell cultures, microbial fermentation, and semi-synthesis are now the principal avenues for producing pharmaceutical grade paclitaxel since harvesting yew bark is destructive and yields only ~1 g product from three adult trees, a slow growing and threated species which cannot meet the global demand without risk of extinction [152]. The biosynthesis *in planta* has been extensively studied, and most of the 19 enzymatic steps have been characterized. However, uncertainty regarding the order of several hydroxylations and acetylations remain. Indeed, feeding studies suggest multiple routes to paclitaxel operate in parallel in the plant, along with several metabolic dead ends resulting in taxoids with lesser or no biological activity such as (+)-taxusin, an abundant, non-bioactive taxoid which nonetheless provides a generic, readily available substrate to test the activities of candidate cytochrome P450s [153]. The order of some reactions in the natural pathway may be somewhat variable. Below, we highlight what is generally considered to be the dominant path to paclitaxel.

GGDP is generally considered the precursor to all diterpenoids. It results from GGDP synthase activity which combines one unit of DMADP with 3 units of IDP from the MEP pathway. The first committed reaction in paclitaxel biosynthesis is the cyclization of GGDP to form taxa-4(5),11(12)- diene, carried out by taxadiene synthase [154] (Figure 8). This olefin is then substrate for taxadiene- 5α -hydroxylase, an ER localized cytochrome P450 which employs a radical mechanism that abstracts a H atom from the C20 methyl group, generating an allylic radical to which oxygen is added to produce taxa-4(20),11(12)-dien- 5α -ol [155]. Acetylation of this alcohol in the cytosol by taxadiene- 5α -ol-O-acetyl transferase yields taxa-4(20),11(12)-dien- 5α -yl-acetate [156].

Considerable uncertainty regarding the order of the next eight steps leads to multiple possible intermediates which converge on a hypothetical intermediate (Figure 8) which now includes all eight oxygen functional groups. As shown here, this intermediate, now contains the four-membered oxetane ring which confers binding affinity to β -tubulin and is therefore critical to the biological activity of

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paclitaxel. Studies on the oxetane ring formation have provided two mechanistic explanations for its formation: one involves an epoxidation of the ene-acetoxy group of taxa-4(20),11(12)-dien- 5α -yl-acetate to yield a 5-acetoxy-4(20)-epoxy intermediate, which then rearranges to the oxetane ring [152]. The other suggests that a downstream acetyltransferase, 10-deactylbaccatin III 10β -O-acetyltransferase (DBAT), may also acetylate the tertiary C4 position on the oxetane ring separately [157].

There is little doubt that following taxane-9 α -dehydrogenase activity, which converts the alcohol function at C9 to a ketone group (uncharacterized but described by [150]), DBAT then converts 10-deacetylbaccatin III into baccatin III [158]. Attachment of the eventual phenylisoserine side chain at C13 follows. This begins with conversion of α -phenylalanine to β -phenylalanine, a reaction catalyzed by phenylalanine aminomutase [159], and its ligation to coenzyme A [160]. Following the transfer of this acyl group to C13 of baccatin III [161], the final steps include a 2'-hydroxylation of the phenylpropyl side chain (an as of yet uncharacterized enzyme) and N-benzoylation of the amino group to yield paclitaxel [162]. As of this date, a total of five cDNAs encoding biosynthetic enzymes in the paclitaxel pathway remain to be isolated and characterized. A complete cohort of biosynthetic genes for paclitaxel formation will enable improvements to sustainable cell culture and microbial production systems to ensure a regular supply of this drug without compromising threatened populations of the pacific yew tree from which it was originally isolated.

5.2. Ingenol Mebutate and Related Phorbol Esters

The family Euphorbiaceae features a wide variety of toxic and pharmacologically useful diterpenoids in the phorbol ester and ingenol subclasses. Most, if not all, of these secondary metabolites appear to be derived from the diterpene olefin casbene (Figure 9), an early prediction made upon the initial characterization of casbene synthase from *Ricinus communis* [163] that has since been corroborated by the isolation of an *R. communis* cDNA encoding casbene synthase [164] and transcriptome mining, cloning, and expression of diterpene synthases from multiple Euphorbiaceae species. With one exception, all appear to use casbene synthase as the singular diterpene synthase in their respective genomes beyond those of primary metabolism [165]. The use of a common terpene synthase product as a starting point for the chemical diversification of diterpenes in the Euphorbiaceae stands in contrast to the more common strategy for generating terpenoid diversity in plants at the terpene synthase step. Three diterpenoids from the Euphorbiaceae display significant pharmacological activity: ingenol mebutate (described further below), the latent HIV-1 activator prostratin [166], and the analgesic resiniferatoxin with properties similar to capsaicin [167].

Figure 9. Proposed pathway for the initial steps leading to the *Euphobia* factor ingenol mebutate in *Euphorbia spp.*, as per [168]. GGDP represents geranylgeranyl diphosphate and OPP signifies a diphosphate group. Enzyme abbreviations: CS-casbene synthase, CYP726A27–cytochrome P450 726A27, CYP71D445–cytochrome P450 71D445, ADH1–alcohol dehydrogenase 1.

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Ingenol 3-mebutate, from *Euphorbia peplus* and *E. lathyrus*, has been approved for treatment of actinic keratosis [169] due to its ability to interact with protein kinase C (PKC) [170]. Most phorbol ester PKC-activators such as phorbol 12-myristate 13-acetate are potent mitogens [171,172]. However, ingenol 3-mebutate displays anti-tumor [173] and anti-leukemic effects [174] while also promoting PKC.

Little is known regarding the biosynthesis of ingenol 3-mebutate, however, cyclization of the C₂₀ prenyl diphosphate precursor, GGDP, to casbene has been proposed as the first committed step in this pathway [165]. However, to obtain the 5/7/7/3 ring system of the polycyclic ingenane carbon skeleton (Figure 9), this macrocyclic intermediate would need to form two C-C bonds to generate the four ring ingenane system. The absence of casbene-like structures containing only 6 or 7 membered rings in this genus and the large number of lathyrane type skeletons bearing a 5-membered ring [175] have led some investigators to speculate that the 6,10 ring closure to yield a lathyrane skeleton is the next C-C bond formed en route to the ingenane system seen in ingenol mebutate [165]. Luo et al. point out that polycyclic diterpenoids display a higher degree of oxygenation than bicyclic casbene types [168], suggesting that cyclization may depend on the regiospecific addition of oxygen groups, particularly at C9, which may represent an early bifurcation step between polycyclic skeletons of the lathryane, tigliane, and ingenane classes and macrocyclic casbene type diterpenoids. Using cDNAs obtained from seeds of E. lathyris, they then present co-expression evidence using casbene synthase, two cytochrome P450s, and an alcohol dehydrogenase transiently expressed in tobacco which demonstrate the dehydrogenase-dependent cyclization in this pathway to form the 5-membered lathyrane jolkinol C, which naturally occurs in E. jolkini [176]. The authors propose the following biosynthetic steps for ingenol 3-mebutate: following the formation of casbene, two regiospecific hydroxylations carried out by CYP71D445 and CYP726A27 introduce alcohol groups at C9 and C5, respectively. One of the aforementioned cytochromes then adds a third hydroxylation at C6 to produce a triol. ADH1 then dehydrogenates the alcohol function at C5 or C6, leading to the formation of the C6-C10 bond, possibly through an enol-tautomer mechanism. Additional steps leading from jolkinol C to ingenol 3-mebutate are presently unknown, but efforts to elucidate the remaining steps of this pathway using transient expression in tobacco as a screening system as well as traditional biochemical characterization are currently underway.

6. Conclusions

The evolution of highly functionalized plant terpenes was punctuated by rapid proliferation of terpene synthase and cytochrome P450 gene families [6], resulting in chemical diversification across the plant kingdom through the introduction of oxygen functional groups into the olefinic backbone of terpene hydrocarbons (for example, see (-)-menthol-Section 3.1 and artemisinin-Section 4.1) and condensations with derivatives from other natural products families, such as short and branched chain fatty acids (cannabinoids–Section 3.2, ingenol mebutate–Section 5.2, and thapsigargin–Section 4.2), amino acids (paclitaxel-Section 5.2), and phenolic compounds (cannabinoids and cannflavin A and B-Section 3.2). Both the cytosolic MVA pathway and the plastid localized MEP pathway contribute terpene skeletons leading to functionalized and biologically active secondary metabolites. While some plant terpenes such as β -carotene, phylloquinone and tocopherols constitute essential (pro)vitamins in the human diet, others, such as phytosterols and essential oils, provide nutritional benefits or act as beneficial anti-oxidants. Here, however, we have focused on compounds with highly specific pharmacological activities. We have also endeavored to highlight lesser known plant terpenes whose clinical evaluation has been hampered by poor availability and intractable biochemistry, but whose clinical potential could be facilitated by elucidation of the corresponding biosynthetic pathways in coming years.

A major question surrounding plant terpenes of medical relevance concerns why this group of plant natural products yields clinically valuable compounds at such a high rate compared to unbiased combinatorial synthesis. Since plants are sessile organisms, their survival depends partly on the ability of specialized metabolites to manipulate animals in their environment for pollination and defense,

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and some of these natural products fortuitously exhibit properties useful in medicine although this is uncommon. It is likely that the medically useful properties of plant terpenes are exaptive, and largely a reflection of their chemical ecological roles as plant defense compounds against insect herbivores [4,177]. The emergence of highly potent and medically useful plant terpenes may therefore exploit natural similarities between human and insect physiology as well as a higher compatibility with proteins compared to synthetic compounds. As defense compounds, high potency and specificity are essential for effectiveness as feeding deterrents, particularly in light of the high metabolic cost of producing and storing such secondary metabolites [178]. Future efforts aimed both at optimizing screening strategies for plant terpenoid lead compounds as well as new approaches to elucidate complex biochemical pathways of existing candidates will undoubtedly improve the discovery rate and availability of plant terpenoids for use in medicine.

Funding: This research was funded by the Natural Sciences and Engineering Research Council of Canada (grant number RGPIN-2017-06400, to M.P.).

Acknowledgments: The authors wish to acknowledge the support of the administration of the Department of Biology at the University of Toronto–Mississauga, which provided material support towards the development of the undergraduate course "Medicinal Plants and Human Health" which led to the conceptual development of this article.

Conflicts of Interest: The authors declare no conflict of interest.

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