

Welcome to my master thesis

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This JupyterBook constitutes my master thesis. It contains multiple Jupyter Notebooks in which I applied the workflow for kinetic parameter estimation of enzyme reactions research scenarios of the EnzymeML project.

Except for this page and the capability to execute Jupyter Notebooks, the printed version is identical to this JupyterBook.



University of Stuttgart
Institute of Biochemistry and Technical
Biochemistry

Master Thesis

**From raw data to kinetic parameters:
an EnzymeML-based workflow for
reproducible enzyme kinetics**

Technical Biology Degree Course

Submitted by
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Submitted on: December 22, 2022
Matriculation Number: 3502013

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Declaration of authorship

I, Max Häußler, hereby declare that the master thesis submitted is my own work. It was made unaided by third parties and only using the quoted sources and tools. All sources used are labelled as references. The thesis, in equal or similar form, was not previously presented to another examination board, and the electronic copy matches the other copies.

Acknowledgements

First, I would like to thank Prof. Dr. Jürgen Pleiss for the opportunity to work on this project and the support throughout the work of this thesis. Furthermore, I would like to thank Prof. PhD. Marilize Le Roes-Hill together with Prof. Dr. Jürgen Pleiss for organizing and enabling my research visit to South Africa, which fueled the collaboration on the projects of this thesis.

My special thanks go to my project partners Alaric Prins, Marwa Mohamed, Chantal Daub, and Paulo Durão who provided the experimental data for the projects of this thesis as well as the fruitful and pleasant project discussions.

I would like to thank Jan Range and Torsten Gieß for their support as well as the whole group for inspiring and motivating seminar discussions.

Abstract

This dataset builds the master thesis with the title 'From raw data to kinetic parameters: an EnzymeML-based workflow for reproducible enzyme kinetics'. The thesis contains four different research scenarios, in which kinetic parameter were estimated by a reproducible modeling workflow based on EnzymeML.

In this work, a workflow for kinetic parameter estimation of single substrate enzyme reactions was developed, enabling a continuous data flow from raw data of absorption measurements to kinetic parameters. For this purpose, two Python modules were developed and published on PyPI: CaliPytion facilitating precise calculation of concentrations by identifying the best linear or non-linear calibration equations, which best describes the relation between analytical signal and analyte concentration. EnzymePynetics is a Python module for kinetic parameter estimation. Thereby, ...inhibition studies

Von Rohdaten zu kinetischen Parametern: ein EnzymeML basierter Workflow für reproduzierbare Enzymkinetik

Alman

Introduction

Reproducible research is the cornerstone of scientific advances and builds the core principle of the scientific method itself. Within the last decade, an increasing number of reports raised attention on unreproducible results across various scientific discipline [\[Begley and Ioannidis, 2015\]](#). Reportedly, more than 70 % of researchers were not able to reproduce the results of another researcher, whereas more than 50 % were not able to reproduce their own findings [\[Baker, 2016\]](#). Since reproducibility is a non-standardized term, it often serves as a vague umbrella term for mingling (i) repeatable analysis based on given information by the same scientist, (ii) reproducible analysis based on given information by another scientist, or (iii) replicable analysis with new information by another scientist. Therefore, reproducibility is not an absolute term but rather relative, depending on the scope of the notion [\[Goodman et al., 2016\]](#). Goodman et. al. proposed to combine the word reproducibility with its relative scope, hence subdividing reproducibility in methods reproducibility, results reproducibility, and inferential reproducibility. According to Goodman, 'methods reproducibility refers to the provision of enough detail about study procedures and data so the same procedures could, in theory or in actuality, be exactly repeated.'. Results reproducibility requires methodological reproducibility, whereas same results are obtained from an independent study. Inferential reproducibility is given, if qualitatively similar conclusions are drawn based on the results of independent studies or reanalysis of a given study. Therefore, inferential reproducibility constitutes the highest level of reproducibility.

Experimental reproducibility advances in data handling

With scope on biocatalysis, all laboratory as well as data analytic procedures, need to be described for methodological reproducibility. Thereby, laboratory procedures include protocols how reagents as well as biological materials were created, as well as protocols on measurement procedures. For that purpose, the STRENDa guidelines were developed, which define minimum reporting standards for enzyme experiments. The [guidelines]<https://www.beilstein-institut.de/en/projects/strenda/guidelines/> specify

[Pleiss, 2021]

Based on the detail of information transparency generates trust in scientific findings

ranging from reagent and protocols how biological species were created

Same applies for kinetic parameter results from enzyme kinetics experiments. Kinetic parameters stand at the end of a long chain of experimental and analytical steps. Whereas upcoming methods open new possibilities in biochemical research. In consequence, latter part is becoming more important. Still in scientific reporting not allowing reproduce the results, since journals

The lack of raw data is therefore described as one of the reasons for the reproducibility crisis [Miyakawa, 2020].

Whereas procedures of wet-lab experiments are commonly described in a sufficient manner to technically repeat the experiment, data analysis procedures are not.

- misspractise of science HARKing p-hacking, selective reporting, poor research design
- lack of raw data
- definition of reproducibility repeatability
- Biology and big data

transparency generates trust in scientific findings

Methods

EnzymeML

EnzymeML serves as an exchange format for biocatalytic data between experimentalist, modelers and database providers. EnzymeML is based on the ontology of Systems Biology Markup Language [Hucka *et al.*, 2003] and adheres to STRENDa guidelines and thus suited for storing biocatalytic data sets in a FAIR data compliant fashion. Within an EnzymeML document information on the reaction conditions, obtained measurement data, as well as modeling results are stored. Thereby, reaction conditions contain information on the pH value, temperature, and the reaction vessel. Apart from a custom name, each represented species is uniquely labeled with an identifier, which allows referencing the specific species in databases. Thereby, proteins are labeled with their UniProtID, and reactants are labeled with their respective SMILES or InChI code.

Modeling results and information on model parameters and model equations can be stored alongside the respective measurement data on which the results are based on.

In this work, EnzymeML documents were read, edited, and written via the Python API PyEnzyme [Range *et al.*, 2021], providing the possibility to integrate PyEnzyme in Python-based data analysis workflows.

Thereby, EnzymeML documents can be used as a data source, or modeling results can be written back to the document. Within PyEnzyme,

Creation of EnzymeML documents

EnzymeML documents were created using the EnzymeML Excel template, in combination with the `.fromTemplate()` method of PyEnzyme. Within the spreadsheet, data and metadata of an experiment were filled in the respective sheets. Alternatively, only metadata was entered to the spreadsheet, whereas measurement data was parsed by a custom Python function from the output file of the analytical device to the measurement data section of the EnzymeML document.

Saving of modeling results

Modeling results from parameter estimation were written back to the EnzymeML document, which contained the measurement data by using PyEnzyme

CaliPytion

CaliPytion was developed to provide an easy way to find the best model by which the relationship between analytical signal to analyte concentration of photometric measurements can be determined. The resulting **StandardCurve** is created based on a provided analyte standard. Besides a linear calibration equation, non-linear calibration models were implemented. All implemented calibration equations (1), (2), (3), (4), (5) are listed below, whereas the A denotes the absorption, x the concentration, and a, b, c the respective parameters of each equation. During the fitting process, all calibration equations are fitted against the analyte standard by non-linear least-square minimization. Therefore, the Python library Imfit [Newville et al., 2016] was utilized. The best fitting model is then preselected based on the lowest Akaike information criterion (AIC). Furthermore the fit of the models can be visually assessed by calling the `.visualize()` method of the **StandardCurve** class. After the calibration curves are fitted, concentration of an unknown sample are determined by calculating the root of the fitted calibration model.

$$A = ax \quad (1)$$

$$A = ax^2 + bx \quad (2)$$

$$A = ax^3 + bx^2 + cx \quad (3)$$

$$A = ae^{\frac{b}{x}} \quad (4)$$

$$A = \frac{ax}{b + x} \quad (5)$$

Data model

The **StandardCurve** functionality of CaliPytion is based on a [data model](#), structuring data and metadata of calibration measurements. Thereby, calibration conditions like temperature and pH, as well as information of the analytical device can be provided. Furthermore, the name as well as an ID can be specified for the analyzed substance. All of this information is stored in the **Calibration** root object. Additionally, the root object can contain a **Standard** and a **Spectrum**.

A **Standard** contains measurements of multiple predefined concentrations and a wavelength at which the measurement was conducted. Additionally, a **Spectrum** can be defined by providing measurement data as well as the respective wavelengths at which the data was measured. The data model was generated using [sdRDM](#), a python tool allowing the creation and versioning of data models.

Initialization

A **StandardCurve** can be initialized directly from an Excel file by calling `StandardCurve.from_Excel()`. Therefore, the first column must contain concentration data, whereas all following columns contain the according measurement data. The first row is not considered and should be used to label the data. Alternatively a **StandardCurve** can be initialized by providing a **Calibration** object, which contains the calibration data.

After **StandardCurve** initialization, concentrations can be calculated by calling the `get_concentration()` method. Alternatively, a **StandardCurve** can be directly applied to an **EnzymeMLDocument** by calling the `.apply_to_EnzymeML()` method. The code cell below demonstrates schematically, how absorption data in an **EnzymeMLDocument** is transformed to concentration data by applying **StandardCurve**.

```
enzmldoc_absorption = EnzymeMLDocument()

# Create standard curve
product_standard = StandardCurve()

# Apply standard curve to species 's1' of the EnzymeML document
enzmldoc_concentration = product_standard.apply_to_EnzymeML(
    enzmldoc=enzmldoc_absorption,
    species_id="s1")
```

Code availability

CaliPytion was published on the Python packaging index ([PyPI](#)) whereas the source code is accessible on [GitHub](#)

EnzymePynetics

EnzymePynetics is a python package for kinetic parameter estimation of single-substrate enzyme reactions. The **ParameterEstimator** of EnzymePynetics estimates the turnover number and Michaelis-Menten constant by fitting time-course measurement data of enzyme reactions to different Michaelis-Menten models. Thereby, the residuals between measurement data and integrated Michaelis-Menten rate equations are minimized through a non-linear least-squares algorithm utilizing the `lmfit` library. Additionally, the inhibition constant K_i can be assessed for potential substrate or product inhibition. Furthermore, K_i of an enzyme inhibitor apart from substrate or product can be determined for inhibitor studies.

Data model

EnzymePynetics is based on a data model, resembling the experimental design of an enzyme kinetics assay. Thereby, all relevant data and metadata of an kinetic experiment are structured in the **EnzymeKineticsExperiment** base object. Thereby, reaction temperature with its respective unit, pH value, and the name of substance from which the measurement data originates. An **EnzymeKineticsExperiment** contains one or multiple **Measurements**, which contain the measurement data. Each measurement stores the information of an experimental conditions the enzyme reaction. Therefore, each **Measurement** contains information on the initial substrate concentration, enzyme concentration, and inhibitor concentration, along with the respective concentration units. Each **Measurement** contains measured data, which itself consist of one or multiple replicates of the respective experimental condition. The data model was generated using [sdRDM](#), a python tool allowing the creation and versioning of data models.

An extensive documentation of the data model can be accessed in the [specifications](#) of the the software package.

Kinetic models

Besides the irreversible Michaelis-Menten rate equation (6) inhibition models for competitive (7), uncompetitive (8), non-competitive (9), and partially competitive inhibition (10) were implemented. Thereby, S , E , and I respectively denote the concentration of substrate, enzyme, and inhibitor. In terms of kinetic parameters, k_{cat} denotes the turnover number, K_m the Michaelis-Menten constant of the substrate, whereas K_{ic} and K_{iu} respectively describe the competitive and uncompetitive inhibition constant.

$$\frac{dS}{dt} = -\frac{k_{cat} * E * S}{K_m + S} \quad (6)$$

$$\frac{dS}{dt} = -\frac{k_{cat} * E * S}{K_m * (1 + \frac{I}{K_{ic}}) + S} \quad (7)$$

$$\frac{dS}{dt} = -\frac{k_{cat} * E * S}{K_m * (1 + \frac{I}{K_{iu}}) * S} \quad (8)$$

$$\frac{dS}{dt} = -\frac{k_{cat} * E * S}{K_m * (1 + \frac{I}{K_{ic}}) + (1 + \frac{I}{K_{iu}}) * S} \quad (9)$$

$$\frac{dS}{dt} = -\frac{k_{cat} * E * S}{K_m * \frac{(1 + \frac{I}{K_{ic}})}{(1 + \frac{I}{K_{iu}})} + S} \quad (10)$$

By default, E is assumed to be constant throughout the reaction. If required, the kinetic models can be extended by the parameter k_{inact} , which describes the time-dependent inactivation rate of the enzyme. The decrease in active enzyme concentration is modeled by (11):

$$\frac{dE}{dt} = -k_{inact} * E \quad (11)$$

Each kinetic models therefore consists of a set of ordinary differential equations to calculate the change in substrate, product, and enzyme concentration concentration for each time step.

Run parameter estimation

Initialization

Whereas the **EnzymeKineticsExperiment** object solely serves as a data container, the **ParameterEstimator** harbors the functionalities for parameter estimation. Data can be provided as an **EnzymeKineticsExperiment** object. Alternatively, an **EnzymeMLDocument** documents can be provided as the

data source via PyEnzyme by calling the `.from_EnzymeML()` method. Within the method, the id of the measured species needs to be specified according to the nomenclature of the `EnzymeMLDocument`. If the document contains information on an inhibitor, the respective id needs to be provided as well, if the inhibition constant should be estimated for the respective inhibitor.

Initially, missing data for the fitting process is calculated based on the assumption of mass conservation. Thereby, missing product concentration P is calculated based on the given substrate measurement data and the specified initial substrate concentration S_0 for each measurement t ((12)). If product data is provided, missing substrate concentrations are calculated accordingly ((13)).

$$P_t = S_0 - S_t \quad (12)$$

$$S_t = S_0 - P_t \quad (13)$$

Fitting of models

After the `ParameterEstimator` is initialized, all kinetic models are initialized by calling the `.fit_models()` method. Furthermore, it can be determined whether enzyme inactivation should be considered for modeling. Depending on whether an inhibitor was specified, the models are initialized accordingly. If no inhibitor was specified, product and substrate inhibition models are initialized besides the irreversible Michaelis model. If inhibitor data was provided, the inhibition models are initialized with concentration data of the inhibitor. Thereafter, kinetic parameters of each model are initialized with estimates based on the highest reaction rate in the dataset. For minimization the parameter space is limited to ± 1000 -fold of the initial parameter estimates.

After the all models are fitted, an overview table is printed which lists all the kinetic parameters of each model together with the respective 1σ standard deviation in percent. The table is sorted by ascending AIC.

Visualization

After model fitting, the modeling results can be visualized by calling the `.visualize()` method, which was implemented utilizing Matplotlib [Caswell et al., 2020]. Thereby, the measurement data is visualized together with the fitted kinetic model. If the experiment was carried out in replicates, mean values of the measurement data with the corresponding standard deviations are shown. If the experiment data contained reactions with different inhibitor concentrations, individual inhibitor concentrations are denoted by unique markers.

By default the best fitting model according AIC is visualized. Different models can be visualized by passing the name of the respective model to the function call. Besides the figure, an detailed statistical fit report is printed, if not specified otherwise.

Data units So far, EnzymePynetics neither converts nor validates units of the measurement data. Hence, all data needs to be provided within the same molar or mass unit. Only the concentration unit of the inhibitor may differ from all other data units.

Code availability

EnzymePynetics was published on the Python packaging index (PyPI) whereas the source code is available on [GitHub](#)

Model comparison

The kinetic model, which describes the experimental data the best, was selected based on AIC, standard deviation of parameter estimates, and visual fit between measurement data and fitted model. AIC served as a statistical metric for information loss, which allows to relatively compare different models with differing number of parameters for a given data set. Hence, AIC can be applied for model selection [Arnold, 2010], [Akaike, 1998]. Thereby, models with a lower AIC indicate less information loss. Since AIC is based on the chi-square statistic, it does not consider the standard deviation of the estimated parameters of a model. Therefore, standard deviation was additionally considered in model selection. Models with low standard deviation were therefore preferred over models with a high standard deviation. Furthermore, fit quality was assessed visually by confirming that the model describes the progress curve of the experimental data.

Jupyter Notebook

All developed tools were deployed in Jupyter Notebooks [Kluyver et al., 2016] for data analysis within a Python environment. A Jupyter Notebook is a digital document containing code cells and text cells. Thereby, executable code with its corresponding output can be supplemented with narrative text, figures, and equations. Hence, providing an scientific programming environment with a low entry barrier. Besides

the documentation capabilities, notebooks can be uploaded to platforms like GitHub or be deployed with Binder [Ragan-Kelley *et al.*, 2018]. GitHub is a cloud-based Git repository, which enables versioning as well as collaborative working on projects involving any kind of code. Besides GitHub, sharing of Jupiter Notebooks is possible with the open-source project Binder, which allows hosting of notebooks. Thus, Jupiter Notebooks can be shared in an executable form, which is independent from the operating system and the location of the user. Only an internet connection and a browser is necessary. In result, Jupiter Notebooks, which are executable through Binder, form a user-friendly environment for scientific data analysis. Due to the extensive documentation capabilities of Jupiter Notebooks, the analysis workflow is comprehensible and accessible for programmers as well as for non-programmers.

Jupyter Book

The written thesis, except for the cover page, was entirely conceptualized as a Jupyter Book, enabling reproducible data analysis of enzyme reaction which is compliant to FAIR data principles. A Jupyter Book is an open-source tool to incorporate computational material in publication-quality books [Community, 2020]. Thereby, Jupyter Notebooks can natively be integrated in Jupyter Books. Thus, making all advantages of the notebook also available in the book. Jupyter Books allow text formatting with Markdown, which is a simple markdown language. Furthermore, Jupyter Book supports cross-referencing, citations, and numbering of equations. Along with all data, the final and interactive book was deployed on GitHub Pages whereas the print version was generated by exporting the html-version of the book to pdf.

Results

Workflow for enzyme kinetics parameter estimation

In this work, a Python-based analysis workflow for kinetic parameter estimation of enzyme reactions was developed. The workflow itself consists of the steps (i) data acquisition, (ii) raw data preparation, (iii) concentration calculation, (iv) quality control, (v) modeling, and (vi) saving of results. The workflow was deployed in Jupyter Notebooks, which combine scientific computing with extensive documentation capabilities and are thus conclusive for programmers as well as for non-programmers. For precise concentration calculations as well as kinetic parameter estimation the two Python modules CaliPytion and EnzymePyntetics were respectively developed over the course of this work. The two packages were designed to seamlessly interact with EnzymeML documents which were modified using PyEnzyme and thus allowing an continuous data stream from analytical raw data to kinetic parameters.

Starting from raw data of photometric measurements, which are stored in an EnzymeML document, concentration values are calculated by CaliPytion, whereupon kinetic parameters are determined by EnzymePyntetics. Finally, the modeling results are written back to the EnzymeML document via PyEnzyme (Fig. 1).

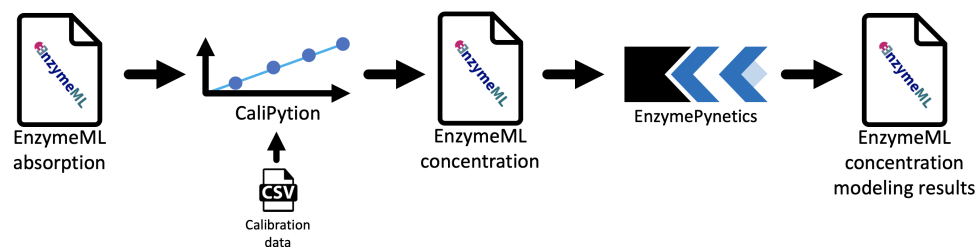


Fig. 1: Components of the kinetic parameter estimation workflow for enzyme reaction data.

A key aspect of the workflow design was to enable reproducible analysis. Therefore, documentation of raw data treatment prior to modeling is vital. As a result, Jupyter Notebooks were used as a platform, since they allow to write text and code within the same document. In consequence, data analysis as well as documentation of data analysis are unified within the same document. Ultimately, each data treatment step from raw data to kinetic parameters is comprehensively documented.

CaliPytion

CaliPytion was developed to facilitate robust conversion of measurement data to concentration data. Based on an analyte standard, a calibration curve is created. Since linear and non-linear relations between analyte concentration and analytic signal may occur, [Hsu and Chen, 2010], [Martin *et al.*, 2017], linear, polynomial, exponential, and rational calibration equations were implemented. After fitting, the best model is

preselected based on Akaike information criterion [Akaike, 1998], offering a metric to compare models with different number of parameters. Thereby, AIC describes the information loss relative to alternative models, while penalizing the use of additional parameters. CaliPytion was designed to work with PyEnzyme. Hence, a **StandardCurve** of CaliPytion can natively be applied to an **EnzymeMLDocument** of PyEnzyme. As a result, absorption data of an EnzymeML document can be converted into concentration data based on the previously generated standard curve.

EnzymePynetics

EnzymePynetics was developed to enable easy applicable parameter estimation of single substrate enzyme reaction directly from EnzymeML documents. In its core, the **ParameterEstimator** of EnzymePynetics fits experimental data to different Michaelis-Menten type models. Besides the irreversible Michaelis-Menten model, competitive, non-competitive, uncompetitive, and partially competitive inhibition models were implemented. Hence, the inhibition constant of potentially inhibiting substrate or product can be estimated. Furthermore, the inhibition constant of an applied inhibitor can be quantified. Since, the experimental data is fitted against multiple kinetic models, the best fitting model is suggested based on AIC. Hence, a proposal of the enzyme's kinetic mechanism is given based on the best fitting model. For quality control, visualization of experimental data as well as the fitted kinetic model was a priority during development. Thereby, detection of systematic deviation between model and measurement data is possible. In consequence, potential experimental errors can be identified. Furthermore, the **ParameterEstimator** allows to subset the measurement data by time as well as initial substrate concentration without deleting data. Hence, identified systematic deviations (e.g. lag-phases or measurements with incorrect enzyme concentrations) can be excluded from parameter estimation.

Like the **StandardCurve** of CaliPytion, the **ParameterEstimator** of EnzymePynetics natively supports **EnzymeMLDocuments** as a data source.

Scenario-driven workflow development

The development of the workflow was driven by different research scenarios of EnzymeML project partners. Goal of all projects was to reliably determine kinetic parameters of enzyme reactions. The collaboration consisted of multiple rounds of lab experiments (performed by the experimental partners) and kinetic modeling of the experimental data. Each round included (i) wet lab experiments, (ii) kinetic modeling, (iii) design of follow-up experiments, and (iv) discussion of results with the respective project partners. Hence, a short feedback loop between lab experiments and modeling-based experimental suggestions was established.

In parallel, the Python modules CaliPytion and EnzymePynetics were developed. Thereby, individual requirements of each research scenario fostered the implementation of various functionalities, resulting in a feature-rich yet generic workflow.

FAIR data analysis with Jupyter Notebooks

All data analysis of the project was carried out in Jupyter Notebook by applying the developed workflow for kinetic parameter estimation. Within the next chapters, the latest results of all four experimental scenarios are shown. Thereby, the applicability of the workflow is demonstrated while analyzing and discussing the results of the respective scenario. Since the analysis was conducted in Jupyter Notebooks, each of the following chapter is an executed Jupyter Notebook. Hence all figures and tables for data visualization are generated at runtime of the analysis. Thereby, the applicability of Jupyter Notebooks for reproducible and comprehensive computational workflows in a scientific environment is demonstrated. Every scenario consists of a short description of the project's background and methodology carried out by the project partners in the wet lab. Additionally, data preparation, kinetic modeling steps, as well as the results are shown and project specific results are discussed. All notebooks can be launched interactively by clicking on the 📄-icon in the upper section of each scenario page. As a result, the analysis itself is transparent, repeatable and compliant with FAIR data principles. Therefore, it is highly recommended to [read this work in its native form](#).

Scenario A:

Chymotrypsin inhibition by a *in silico* designed albumin fusion protein

Data provided by Marwa Mohamed (Institute of Cell Biology and Immunology, University of Stuttgart, Germany)

Project background

In this scenario, the binding of an *in silico* designed protein to an enzyme was assessed by determining the inhibitory constant K_i . Thereby, the efficiency of the newly developed approach for computational design of protein binders to deliberately chosen targets was demonstrated. This was done by comparing K_i of the designed human serum albumin variant (HSA(M3)) to the wild-type (HSA(WT)) by respectively applying the proteins to chymotrypsin enzyme reactions. Both designed proteins were individually dimerized into fusion proteins through a huFc.

Experimental design

Enzyme activity was monitored by measuring the product formation of p-Nitroanilin (p-NA) photometrically at 410 nm for 30 min at 30°C. Therefore, Succinyl-gly-gly-phe-p-nitroanilide (SGGPpNA) was applied as substrate in a concentration range of 0.25 - 2 mM. For concentration calculations, p-NA standard was prepared in the range of 0 - 0.3 mM in duplicates. K_i of HSA(WT)-huFc and HSA(M3)-huFc on chymotrypsin were investigated in independent experiments. Each experiment consisted of enzyme reactions with and without the respective HSA variant. Each enzyme reaction contained 0.2 µM of enzyme and 26.88 µM of the respective HSA variant, if inhibitor was applied. All enzyme reactions were prepared in duplicates.

Data management

Experimental data and meta data was filled in EnzymeML Excel templates for each of the two inhibition experiments respectively. Calibration data was stored as Excel files. Measurement data was already blanked.

Data preparation

Imports

```
import numpy as np
import pandas as pd
import matplotlib.pyplot as plt
import pyenzyme as pe
import copy
import string
from IPython.display import display
from EnzymePynetics.tools.parameterestimator import ParameterEstimator
from CaliPytion .tools.standardcurve import StandardCurve

import warnings
warnings.filterwarnings('ignore')
```

Concentration calculation

Product standard data was imported directly from an Excel file. Then, a standard curve was created.

```
product_standard = StandardCurve.from_excel(
    path="../../../data/chymotrypsin_inhibition/pNA-standard.xlsx",
    reactant_id="s1",
    wavelength=410,
    sheet_name="csv",
    concentration_unit="mmole / l",
    temperature=30,
    temperature_unit="C")

product_standard.visualize()
```

Calibration data was automatically blanked.

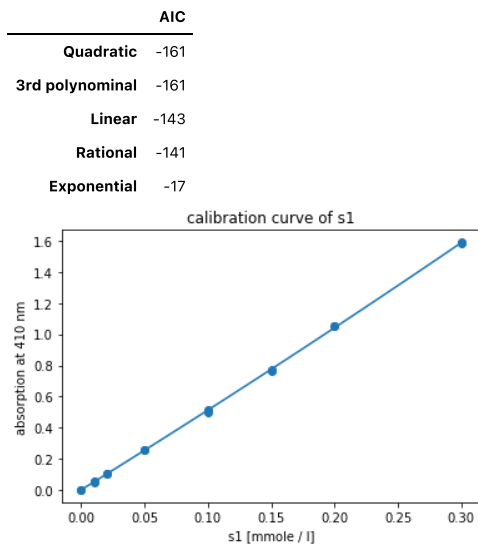


Fig. 2: Fitted quadratic calibration model to standard data of *p*-NA.

Based on the Akaike information criterion (AIC), the relation between concentration and absorption is best described by a quadratic function. The calibration measurements and the fitted calibration model are shown in Fig. 2.

Experimental data

The EnzymeML documents of each experiment were loaded and the standard curve was applied to the absorption data to calculate concentrations.

```
# Load data from
chymo_HSAwt =
pe.EnzymeMLDocument.fromTemplate("../data/chymotrypsin_inhibition/chymo_HSAwt.xlsx")
chymo_HSAM3 =
pe.EnzymeMLDocument.fromTemplate("../data/chymotrypsin_inhibition/chymo_HSA(M3).xlsx")

# Apply standard curve to 'EnzymeMLDocument'
chymo_HSAwt = product_standard.apply_to_EnzymeML(chymo_HSAwt, "s1")
chymo_HSAM3 = product_standard.apply_to_EnzymeML(chymo_HSAM3, "s1")
```

Comparability of the experiments

Since experimental data with HSA(WT)-huFc and HSA(M3)-huFc originate from independent experiments, the control reactions without the respective inhibitor were compared by performing a parameter estimation (Fig. 3). Thereby, catalytic efficiency $\frac{k_{cat}}{K_m}$ was used to assess comparability between the data sets, since k_{cat} and K_m were highly correlated (corr > 0.98). High correlations between parameters indicate that the parameters cannot be determined independently with certainty. In this case, the highest initial substrate concentration is presumably too low, compared to the true K_m of the enzyme under the given experimental conditions. However, higher substrate concentration were not applied for multiple reasons. On the one hand dimethyl sulfoxide (DMSO) was used as a co-solvent of the substrate, which inhibits enzyme activity Busby *et al.* [1999]. Hence, higher initial substrate concentrations would have led to higher enzyme inhibition, which would have distorted the assessment of K_i . On the other hand, high substrate viscosity denied the application of higher concentrations without sacrificing pipetting precision.

```

# Create copies of the data sets and delete measurements with inhibitor.
wt_control = copy.deepcopy(chymo_HSAwt)
del wt_control.measurement_dict["m4"]
del wt_control.measurement_dict["m5"]
del wt_control.measurement_dict["m6"]
del wt_control.measurement_dict["m7"]

m3_control = copy.deepcopy(chymo_HSAM3)
del m3_control.measurement_dict["m4"]
del m3_control.measurement_dict["m5"]
del m3_control.measurement_dict["m6"]
del m3_control.measurement_dict["m7"]

# Estimate kinetic parameters of the control reactions of the HSA wild-type data
set.
kinetics_wt_control = ParameterEstimator.from_EnzymeML(wt_control, "s1",
"product")
kinetics_wt_control.fit_models(stop_time_index=-1, display_output=False)
print("Kinetic parameters of HSA(wt) chymotrypsin control reactions:")
display(kinetics_wt_control.result_dict\
        .style.set_table_attributes('style="font-size: 12px"'))

# Estimate kinetic parameters of the control reactions of the HSA(M3) data set.
kinetics_m3_control = ParameterEstimator.from_EnzymeML(m3_control, "s1",
"product")
kinetics_m3_control.fit_models(stop_time_index=-1, display_output=False)
print("\nKinetic parameters of HSA(M3) chymotrypsin control reactions:")
display(kinetics_m3_control.result_dict\
        .style.set_table_attributes('style="font-size: 12px"'))

fig, axes = plt.subplots(1,2, figsize=(12.8,4.8), sharey=True, sharex=True)
for e, (doc, ax, title) in enumerate(zip([kinetics_wt_control
,kinetics_m3_control], axes.flatten(), ["chymotrypsin control reactions HSA(wt)",
"chymotrypsin control reactions HSA(M3)"])):
    doc.visualize(ax=ax, title=title)
    ax.set_ylabel("4-nitroanilin [mM]")
    ax.set_xlabel("time after reaction start [min]")
    ax.set_xticks([5, 10, 15, 20])
    ax.text(0, 1.1, string.ascii_uppercase[e], transform=ax.transAxes,
           size=20, weight='bold')

handles, labels = ax.get_legend_handles_labels()

fig.legend(handles, labels, loc="lower center", ncol=4, title="initial SGGPpNA
[mM]", bbox_to_anchor=(0.5,-0.15))
plt.tight_layout()

```

Kinetic parameters of HSA(wt) chymotrypsin control reactions:

	AIC	kcat [1/min]	Km [mmole / l]	kcat / Km [1/min * 1/mmole / l]	Ki competitive [mmole / l]	Ki uncompetitive [mmole / l]
irreversible Michaelis Menten	-1857	56.440 +/- 4.52%	2.226 +/- 7.28%	25.353 +/- 8.57%	-	-
competitive product inhibition	-1856	60.225 +/- 9.01%	2.342 +/- 9.88%	25.714 +/- 13.37%	0.725 +/- 112.65%	-
uncompetitive product inhibition	-1855	57.529 +/- 8.30%	2.276 +/- 10.65%	25.272 +/- 13.50%	-	3.175 +/- 428.92%
substrate inhibition	-1855	56.585 +/- 29.14%	2.233 +/- 35.51%	25.339 +/- 45.93%	-	997.715 +/- 12275.11%
non-competitive product inhibition	-1854	60.110 +/- 9.50%	2.331 +/- 10.00%	25.787 +/- 13.79%	0.702 +/- 117.81%	179.118 +/- 358.38%

Kinetic parameters of HSA(M3) chymotrypsin control reactions:

	AIC	kcat [1/min]	Km [mmole / l]	kcat / Km [1/min * 1/mmole / l]	Ki competitive [mmole / l]	Ki uncompetitive [mmole / l]
irreversible Michaelis Menten	-549	48.996 +/- 2.23%	1.978 +/- 3.74%	24.776 +/- 4.35%	-	-
competitive product inhibition	-549	51.227 +/- 4.39%	2.043 +/- 4.76%	25.077 +/- 6.47%	0.908 +/- 81.63%	-
uncompetitive product inhibition	-547	49.384 +/- 5.33%	1.996 +/- 6.71%	24.743 +/- 8.57%	-	7.445 +/- 631.41%
non-competitive product inhibition	-547	51.241 +/- 5.62%	2.044 +/- 6.62%	25.073 +/- 8.69%	0.914 +/- 83.56%	121.556 +/- 2156.85%
substrate inhibition	-545	50.368 +/- 3.78%	2.046 +/- 5.12%	24.623 +/- 6.36%	-	87.549 +/- 121.66%

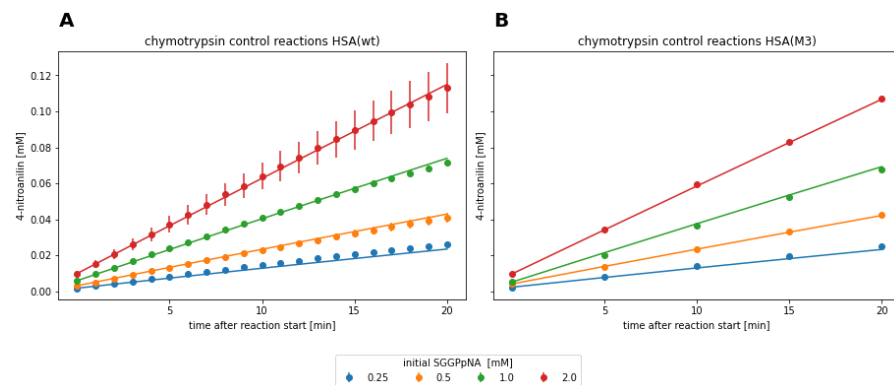


Fig. 3: Measurement data and fitted irreversible Michaelis-Menten model for chymotrypsin reactions without HSA inhibitor.

The two dataset were fitted against kinetic models, which are displayed in the tables above. Each dataset is best described by the irreversible Michaelis-Menten model in terms of AIC and standard deviation on the estimated parameters. Models with product or substrate inhibition resulted in large uncertainties above 80 % on the parameter estimates. Therefore, irreversible Michaelis-Menten model was utilized for comparison of parameters. $\frac{k_{cat}}{K_m}$ was estimated to be $25.353 \text{ min}^{-1}\text{mM}^{-1} \pm 8.57\%$ for the control reaction of the HSA(WT)-huFc data set and $24.776 \text{ min}^{-1}\text{mM}^{-1} \pm 4.35\%$ for the HSA(M3)-huFc data set. As a result, the two experiments showed to be comparable, since the catalytic efficiency differs less than 3 % between the two data sets.

Determination and comparison of K_i

Experimental data of chymotrypsin inhibition by HSA(M3)-huFc contained negative absorption values for the first measurement point. Presumably sourcing from an incorrect blank measurement. Therefore, only measurement data from the second data point (minute 5 and onward) was considered for parameter estimation. Parameter estimates for all applied kinetic models are displayed in the output below and visualized in Fig. 4.

```

# Parameter estimation for HSA(wt) data set
kinetics_HSAwt = ParameterEstimator.from_EnzymeML(chymo_HSAwt, reactant_id="s1",
inhibitor_id="s2", measured_species="product")
kinetics_HSAwt.fit_models(initial_substrate_concs=[0.25, 0.5, 1, 2],
stop_time_index=-1, start_time_index=5, display_output=False)
print("Kinetic parameters estimates for all models of chymotrypsin inhibition by
HSA(WT)-huFc:")
display(kinetics_HSAwt.result_dict.drop(columns=["kcat [1/min]", "Km [mmole /
l]"])\
.style.set_table_attributes('style="font-size: 12px"'))

# Parameter estimation for HSA(M3) data set
kinetics_HSAM3 = ParameterEstimator.from_EnzymeML(chymo_HSAM3, reactant_id="s1",
inhibitor_id="s3", measured_species="product")
kinetics_HSAM3.fit_models(initial_substrate_concs=[0.25, 0.5, 1, 2],
stop_time_index=-1, start_time_index=1, display_output=False)
print("\nKinetic parameters estimates for all models of chymotrypsin inhibition by
HSA(M3)-huFc:")
display(kinetics_HSAM3.result_dict.drop(columns=["kcat [1/min]", "Km [mmole /
l]"])\
.style.set_table_attributes('style="font-size: 12px"'))

# Visualize experimental data and fitted models
fig, axes = plt.subplots(1,2, figsize=(12.8,4.8), sharey=True, sharex=True)
for e, (doc, ax, title) in enumerate(zip([kinetics_HSAwt, kinetics_HSAM3],
axes.flatten(), ["chymotrypsin inhibition by HSA(WT)-huFc", "chymotrypsin
inhibition by HSA(M3)-huFc"])):
    doc.visualize(ax=ax, title=title)
    ax.set_ylabel("4-nitroanilin [mM]")
    ax.set_xlabel("time after reaction start [min]")
    ax.set_xticks([5, 10, 15, 20])
    ax.text(0, 1.1, string.ascii_uppercase[e], transform=ax.transAxes,
size=20, weight='bold')

handles, labels = ax.get_legend_handles_labels()
fig.legend(handles, labels, loc="lower center", ncol=2, title="initial SGGpNA
[mM]", bbox_to_anchor=(0.5,-0.2))
plt.tight_layout()

```

Kinetic parameters estimates for all models of chymotrypsin inhibition by HSA(WT)-huFc:

	AIC	kcat / Km [1/min * 1/mmole / l]	Ki competitive [mmole / l]	Ki uncompetitive [mmole / l]
competitive inhibition	-3100	22.253 +/- 6.97%	0.460 +/- 27.24%	-
non-competitive inhibition	-3098	22.403 +/- 7.00%	0.449 +/- 27.64%	79.193 +/- 516.52%
irreversible Michaelis Menten	-3088	21.622 +/- 7.15%	-	-
uncompetitive inhibition	-3087	21.631 +/- 7.63%	-	0.696 +/- 88.99%
partially competitive inhibition	-3084	21.643 +/- 8.23%	796.196 +/- 1075.02%	992.685 +/- 48.74%

Kinetic parameters estimates for all models of chymotrypsin inhibition by HSA(M3)-huFc:

	AIC	kcat / Km [1/min * 1/mmole / l]	Ki competitive [mmole / l]	Ki uncompetitive [mmole / l]
competitive inhibition	-808	23.031 +/- 10.48%	0.059 +/- 8.51%	-
non-competitive inhibition	-806	23.002 +/- 10.89%	0.060 +/- 9.04%	44.965 +/- 302.77%
uncompetitive inhibition	-751	19.297 +/- 21.77%	-	0.035 +/- 21.33%
irreversible Michaelis Menten	-713	18.512 +/- 24.41%	-	-
partially competitive inhibition	-709	18.544 +/- 27.64%	569.354 +/- 57.96%	925.096 +/- 11733.22%

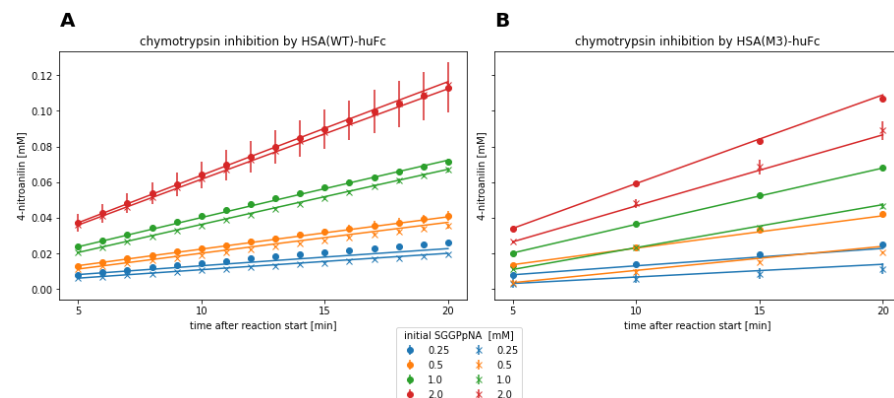


Fig. 4: Measurement data and fitted product inhibition model for chymotrypsin reactions with respective HSA inhibitor.

Both reaction systems are best described by the competitive inhibition model, which is indicated by the lowest AIC and standard deviation on the estimated parameters. Thereby, a K_i of $0.460 \text{ mM} \pm 27.24\%$ was estimated for HSA(WT)-huFc and $0.059 \text{ mM} \pm 8.51\%$ for HSA(M3)-huFc. This resembles a roughly 7-fold increase in affinity of HSA(M3) to the enzyme compared to the HSA(wt). Since the competitive inhibition model describes the data the best, HSA(M3)-huFc presumably interacts with the enzyme in the active site region.

Save modeling results

Lastly, the modeling results are written to the EnzymeML documents and the files are exported.

```
for result, enzmldoc in zip([kinetics_HSAwt, kinetics_HSAM3], [chymo_HSAwt,
chymo_HSAM3]):

    # Write modeling results to kinetic parameters
    k_cat = pe.enzymeml.models.KineticParameter(
        name="k_cat",
        value=result.get_model_results()["k_cat"].value,
        unit=f"1 / {result.data.time_unit}")

    K_m = pe.enzymeml.models.KineticParameter(
        name="K_m",
        value=result.get_model_results()["Km"].value,
        unit=result.data.data_conc_unit)

    K_ic = pe.enzymeml.models.KineticParameter(
        name="K_ic",
        value=result.get_model_results()["K_ic"].value,
        unit=result.data.data_conc_unit)

    # Define kinetic model
    model = pe.KineticModel(
        name="competitive inhibition",
        equation="-k_cat * p0 * s0 / (K_m*(1+(s2 / K_ic))+s0)",
        parameters=[k_cat, K_m, K_ic])

    enzmldoc.getReaction("r0").model = model

    # Export EnzymeML documents
    export = False
    if export:
        enzmldocToFile()
```

Scenario B: α -glucosidase inhibition by fucoidan

Data provided by Chantal Daub (Biochemistry, Rhodes University, Makhanda, South Africa)

Project background

In this scenario, the inhibitory properties of fucoidan on α -glucosidase from *Saccharomyces cerevisiae* were analyzed. Fucoidan is a sulfated polysaccharide found in various brown algae. The polysaccharide is investigated as a potential active compound in the fields of anti-cancer, anti-inflammation, and anti-coagulate research, among others (Li *et al.* [2008]). Furthermore, fucoidan showed to be a promising drug for diabetes mellitus treatment, since it effectively inhibits α -glucosidase. Recently, Daub *et al.* [2020] showed that α -glucosidase inhibition by fucoidan from *E. maxima* significantly stronger inhibition, compared to the established diabetes drug acarbose.

In the following analysis, α -glucosidase was exposed to fucoidan from *Ecklonia maxima*, *Ecklonia radiata*, *Fucus vesiculosus*, and *Schimmelmanna elegans* to test their respective capability of α -glucosidase inhibitor by assessing their respective K_i values.

Experimental design

α -glucosidase reactions, catalyzing the hydrolysis of p-nitrophenyl glucopyranoside (p-NPG) to p-nitrophenol (p-NP) were conducted with and without fucoidan from each seaweed species as well as acarbose. Thereby, fucoidan was applied in two different concentrations. p-NPG was applied in a range from 0.1 mM to 5 mM, to enzyme reactions containing 9.19 μM α -glucosidase. Product formation was recorded photometrically at 405 nm and 37°C for 20 min. Product concentrations were calculated utilizing a photometric p-NP standard. Additionally, control reaction without enzyme were prepared to subtract the absorption contribution of the respective inhibitor, buffer, enzyme and substrate from each measurement.

Data preparation

Imports and parser function

```
from typing import Dict
import pandas as pd
import numpy as np
import matplotlib.pyplot as plt
import matplotlib.colors as mcolors
import re
import os
import pyenzyme as pe
from IPython.display import display
from CalIPython.tools.standardcurve import StandardCurve
from EnzymePynetics.tools.parameterestimator import ParameterEstimator

import warnings
warnings.filterwarnings('ignore')

colors = list(mcolors.TABLEAU_COLORS.values())

# Parser
def measurement_data_to_EnzymeML(
    template_path: str,
    measurement_data: np.ndarray,
    species_id: str,
    time: np.ndarray,
    data_unit: str,
    time_unit: str
) -> pe.EnzymeMLDocument:
    enzmldoc: pe.EnzymeMLDocument =
    pe.EnzymeMLDocument.fromTemplate(template_path)

    for IDs, concentration in zip(enzmldoc.measurement_dict.keys(),
    measurement_data):
        for counter, replicate in enumerate(concentration):
            enzmldoc.getMeasurement(IDs).addReplicates(pe.Replicate(
                id=f"Measurement{counter}",
                species_id=species_id,
                data=list(replicate),
                data_unit=data_unit,
                time=list(time),
                time_unit=time_unit), enzmldoc)

    return enzmldoc

# Ignore hidden files in file system
def listdir_nohidden(path):
    for f in os.listdir(path):
        if not f.startswith('.'):
            yield f
```

Measurement data was provided as an Excel file, whereas metadata was filled in EnzymeML Excel templates for each fucoidan seaweed species and acarbose respectively. In preliminary experiments, p-NPG showed to slightly absorb at the product detection wavelength. Therefore, the absorbance contribution of substrate at the product detection wavelength was subtracted as well as the the contributions of enzyme, buffer and inhibitor. Then, the blanked absorbance data was written to the EnzymeML documents by a parser function.

```

dataset_path = "../../../data/glucosidase_inhibition/experimental_data_real.xlsx"
template_directory = "../../../data/glucosidase_inhibition/EnzymeML_templates"

enzml_docs = []
# Load experimental data from Excel
excel_sheets = sorted(pd.ExcelFile(dataset_path).sheet_names)
inhibitors = excel_sheets[:-1]
substrate_controls = excel_sheets[-1]
initial_substrates = [0.1, 0.25, 0.5, 1, 2.5, 5] # mM

# Blank data
## Absorption contribution from substrate
substrate_absorption_data = pd.read_excel(dataset_path,
sheet_name=substrate_controls).set_index("time")
buffer_enzyme_absorption = np.mean(substrate_absorption_data.iloc[:,0])
substrate_absorptions =
substrate_absorption_data.subtract(buffer_enzyme_absorption).drop(columns=
["Buffer+ Enzyme"])
substrate_absorptions = substrate_absorptions.values.T.reshape(2,6,21)
substrate_absorptions = np.mean(substrate_absorptions, axis=0)
substrate_absorptions = np.mean(substrate_absorptions, axis=1)
mapper_substrate_enzyme_absorption = dict(zip(initial_substrates,
substrate_absorptions))

for inhibitor, template in zip(sorted(inhibitors),
sorted(listdir_nohidden(template_directory))):
    df = pd.read_excel(dataset_path, sheet_name=inhibitor).set_index("time")
    time = df.index.values
    inhibitor_controls = df.iloc[:,4]
    inhibitor_concs = np.unique([float(conc.split(" ")[-2]) for conc in
inhibitor_controls.columns])
    inhibitor_absorptions = inhibitor_controls.values.T.reshape(2,2,21)
    inhibitor_absorptions = np.mean(inhibitor_absorptions, axis=1)
    inhibitor_absorptions = np.mean(inhibitor_absorptions, axis=1)
    mapper_inhibitor_absorption = dict(zip(inhibitor_concs,
inhibitor_absorptions))
    mapper_inhibitor_absorption[0.0] = buffer_enzyme_absorption
    df = df.iloc[:,4:]
    for column in df.columns:
        init_substrate = float(column.split(" ")[4])
        inhibitor_conc = float(column.split(" ")[1])
        df[column] = df[column] -
mapper_substrate_enzyme_absorption[init_substrate]
        df[column] = df[column] - mapper_inhibitor_absorption[inhibitor_conc]

    data = df.values.T.reshape(3,2,6,21)
    data = np.moveaxis(data,1,2).reshape(18,2,21)

# Parse measurement data to EnzymeML documents
enzml_docs.append(measurement_data_to_EnzymeML(
    template_path=f"{template_directory}/{template}",
    measurement_data=data,
    time=time,
    species_id="s1",
    data_unit="mmole / l",
    time_unit="min"
))

```

Data quality

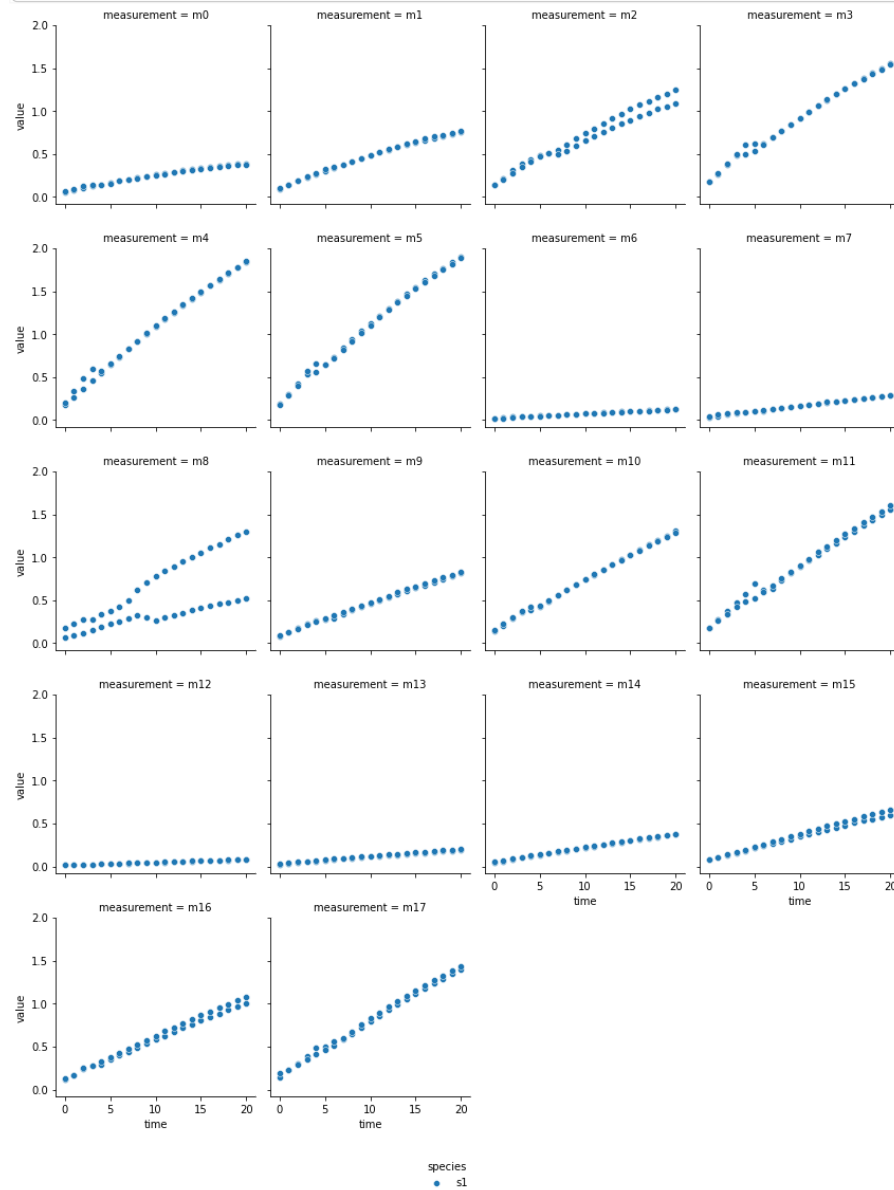
In the next cell, the blanked absorption data of each EnzymeML document is visualized with the `.visualize()`-method of `PyEnzyme` for quality control.

```

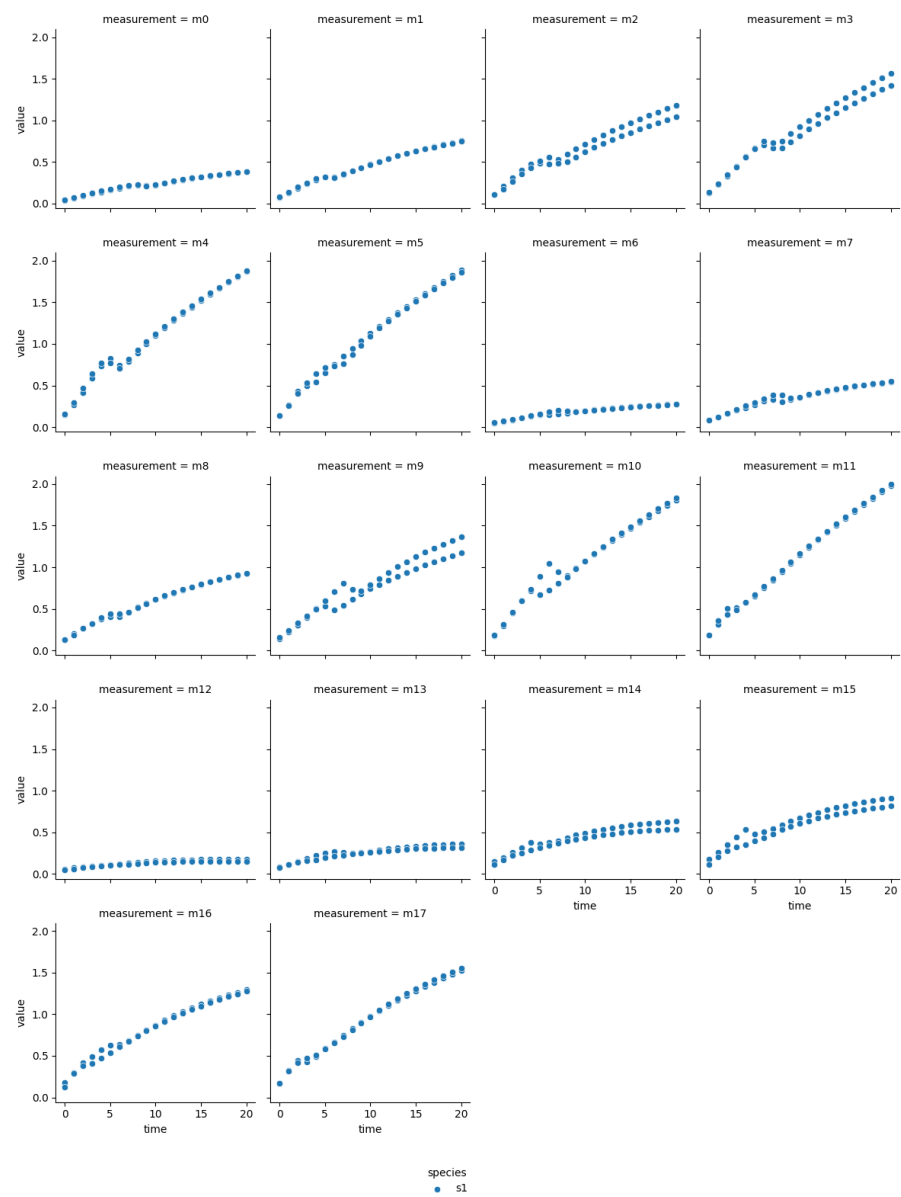
for doc in enzml_docs:
    print(doc.name)
    doc.visualize()
    plt.show()

```

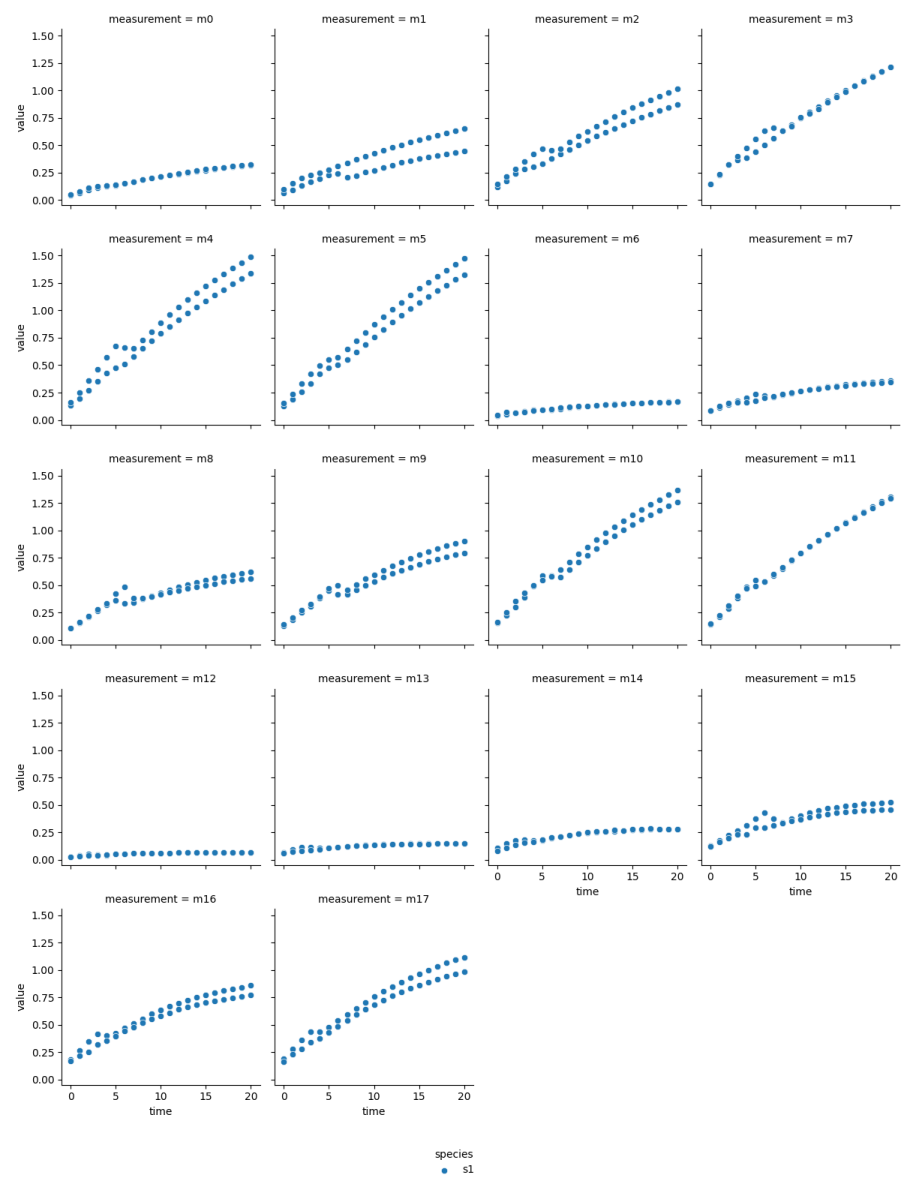
a-glucosidase inhibition by acarbose



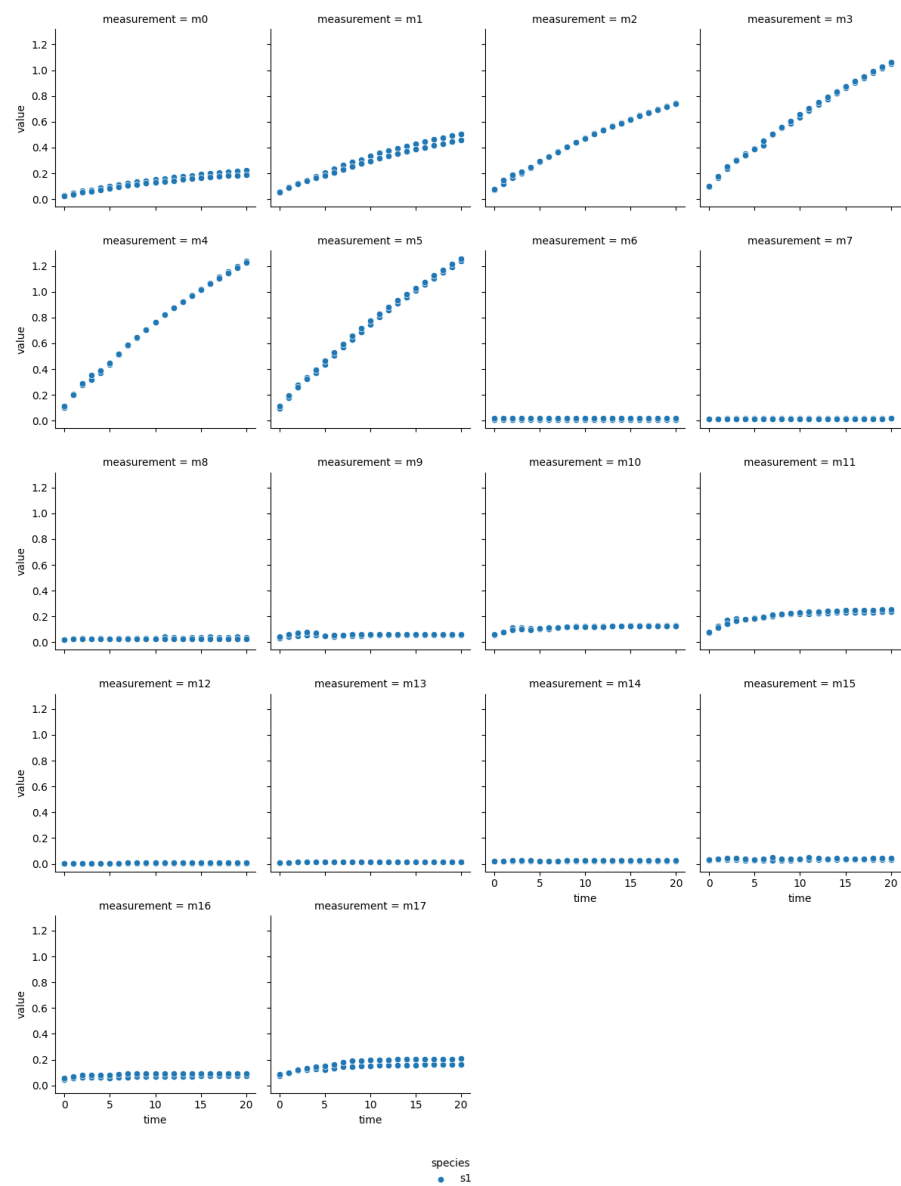
a-glucosidase inhibition by fucoidan from *E. maxima*



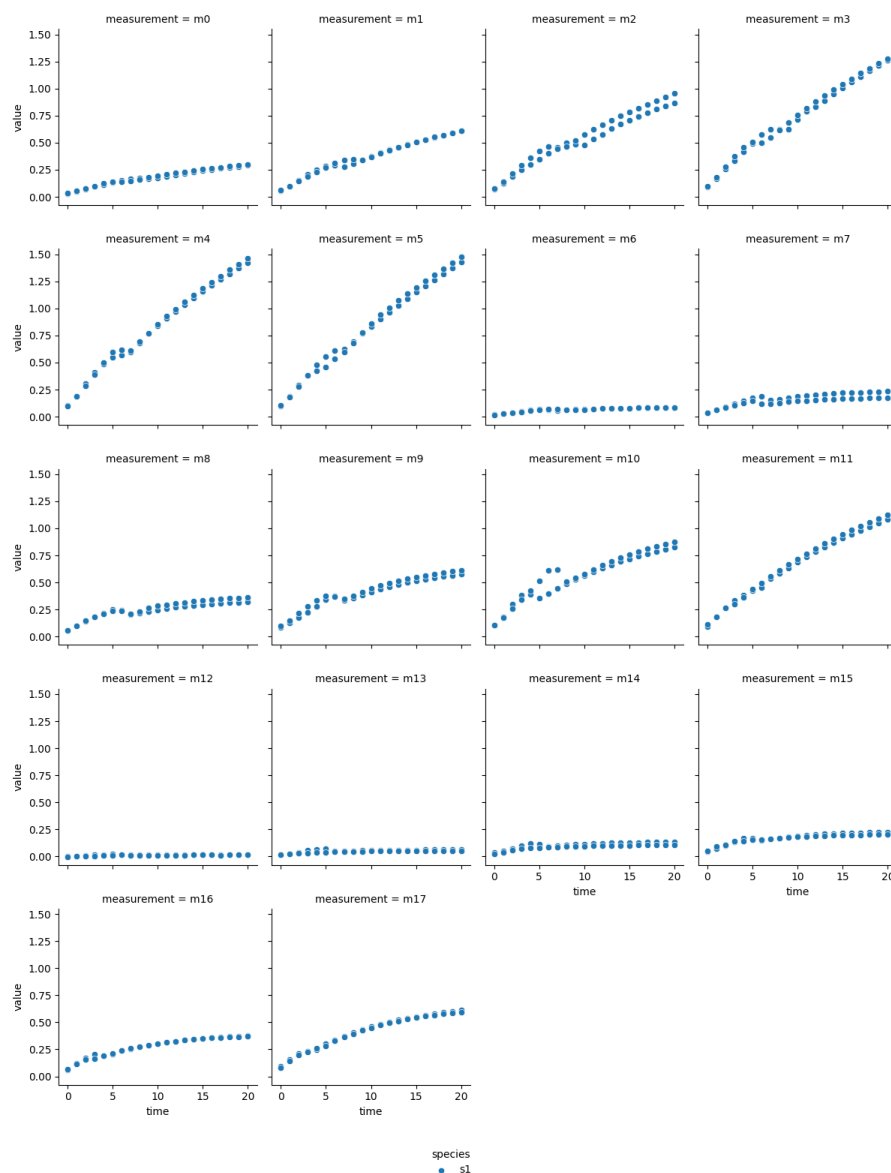
a-glucosidase inhibition by fucoidan from *E. radiata*



a-glucosidase inhibition by fucoidan from *F. vesiculosus*



a-glucosidase inhibition by fucoidan from *S. elegans*



The technical output of the cell above visualizes the blanked product absorbance data of each dataset. Therein, the individual measurements are labeled from m0 - m17, which represent individual experimental conditions. Thereby, measurements m0 - m5 are from reactions without inhibitor, m6 - m11 reactions with the lower inhibitor concentration, and m12 - m17 originate from reactions with the higher inhibitor concentration. Each subplot contains the data of two experimental repeats.

In most reactions a decrease followed by an increase of reaction rate is visible around minute 5. Since each dataset originates from a continuous photometric measurement carried out on a single MTP, the observed behavior likely sources from an analytical device malfunction. Additionally, one repeat of measurement 'm8' from the acarbose dataset shows low absorption, which is inconsistent with comparable measurements. Therefore, 'm8' was excluded from the dataset.

```
del enzml_docs[0].measurement_dict["m8"].getReactant("s1").replicates[0]
```

Concentration calculation

Standard data of p-NP was loaded from an excel file, and a standard curve was created. Then, the standard curve was applied to the EnzymeML documents.

```

path_calibration_data = "../../data/glucosidase_inhibition/p-NP_standard.xlsx"

product_standard = StandardCurve.from_excel(
    path=path_calibration_data,
    reactant_id="s1",
    sheet_name="csv",
    wavelength=405,
    concentration_unit = "mmole / l",
    cutoff_absorption=2)

product_standard.visualize("Quadratic")

# Apply calibration curves to absorption EnzymeML documents
for enzml_doc in enzml_docs:
    product_standard.apply_to_EnzymeML(enzml_doc, "s1", "Quadratic")

```

Calibration data was automatically blanked.

	AIC
3rd polynomial	-135
Quadratic	-135
Rational	-135
Linear	-125
Exponential	-11

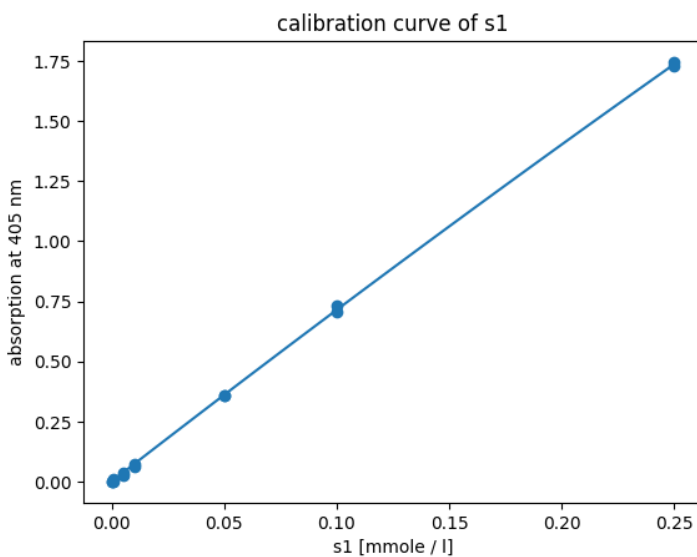


Fig. 5: Fitted quadratic calibration model to standard data of p-NP (s1).

Based on AIC, 3rd-degree polynomial, quadratic and rational calibration models describe the relation between absorption to concentration ratio of p-NP equally. Since the 3rd-degree polynomial model has one additional parameter compared to the quadratic model, the calibration model with less parameters is preferred, since the additional parameter did not give additional information. The fit between quadratic calibration model and calibration standard is visualized in Fig. 5.

Parameter estimation

Comparison between different inhibition models and thus determining the kinetic mechanism was not possible, as no model was consistently favored over the other based on AIC and standard deviations of the parameter estimates.

Thus, different inhibition models cannot be compared, since information provided by the initial slopes is insufficient to distinguish kinetic inhibition mechanisms by modeling. In consequence, kinetic parameters were estimated based on initial rates, assuming competitive inhibition, since this kinetic mechanism has been proposed in literature [Shan et al., 2016].

Additionally, the faulty measurement with an initial substrate concentration of 0.5 mM from the 'α-glucosidase inhibition by acarbose' data set was excluded from parameter estimation. Based on an initial parameter estimation run with subsequent visual analysis, several measurements from the *E. maxima* dataset were excluded. Thereby, measurements from both applied inhibitor concentrations and an initial substrate concentration of 2.5 mM and 5 mM showed similar activity compared to reactions without inhibitor. In contrast, the reactions with inhibitor present showed inhibition, according to the difference in slope.

```
# Discard measurement with 2.5 mM and 5 mM and lower inhibition concentration from
E. maxima dataset
del enzml_docs[1].measurement_dict["m10"]
del enzml_docs[1].measurement_dict["m11"]
del enzml_docs[1].measurement_dict["m16"]
del enzml_docs[1].measurement_dict["m17"]
```

```
# Run parameter estimation for all data sets
results=[]
for enzml_doc in enzml_docs:
    result = ParameterEstimator.from_EnzymeML(enzml_doc=enzml_doc, reactant_id="s1",
inhibitor_id="s2", measured_species="product")
    result.fit_models(enzyme_inactivation=False, display_output=False,
stop_time_index=3)
    results.append(result)

# Display results for competitive inhibition model
labels = np.array(["Acarbose", "Fucoidan (E. maxima)", "Fucoidan (E. radiata)",
"Fucoidan (F. vesiculosus)", "Fucoidan (S. elegans)"])
df = pd.concat([result.result_dict.loc["competitive inhibition"].to_frame().T for
result in results],\
keys=labels)\
    .reset_index()\
    .drop(columns=["level_1", "AIC", "Ki uncompetitive [g / l]"])\
    .set_index("level_0")\
    .rename_axis("Inhibitor", axis = 0)
print("Kinetic parameters based on competitive inhibition model:")
display(df.style.set_table_attributes('style="font-size: 12px"'))
```

Kinetic parameters based on competitive inhibition model:

Inhibitor	k_{cat} [1/min]	K_m [mmole / l]	k_{cat} / K_m [1/min * 1/mmole / l]	K_i competitive [g / l]
Acarbose	1.903 +/- 2.77%	0.341 +/- 10.79%	5.581 +/- 11.14%	0.112 +/- 10.80%
Fucoidan (E. maxima)	2.317 +/- 2.14%	0.406 +/- 7.64%	5.701 +/- 7.93%	0.874 +/- 14.25%
Fucoidan (E. radiata)	1.343 +/- 2.75%	0.180 +/- 14.57%	7.461 +/- 14.83%	0.028 +/- 19.29%
Fucoidan (F. vesiculosus)	1.467 +/- 2.07%	0.415 +/- 7.37%	3.531 +/- 7.66%	0.003 +/- 7.80%
Fucoidan (S. elegans)	1.474 +/- 2.42%	0.251 +/- 10.79%	5.863 +/- 11.06%	0.016 +/- 11.88%

The output above displays estimated kinetic parameters for all data sets, assuming competitive inhibition between enzyme and inhibitor. As a result k_{cat} was estimated between $2.317 \text{ min}^{-1} \pm 2.14\%$ and $1.343 \text{ min}^{-1} \pm 2.75\%$, whereas K_m was estimated between $0.415 \text{ mM} \pm 7.37\%$ and $0.180 \text{ mM} \pm 14.57\%$. Since all measurements were conducted under equal conditions, k_{cat} and K_m should be similar across experiments. Deviations in parameter values might therefore result from pipetting or deviations in reaction temperature.

```
# Get K_i parameter results
def get_parameter(result, model_name: str, parameter: str):
    return result.models[model_name].result.params[parameter]

ki = []
ki_stderr = []
for result in results:
    ki.append(get_parameter(result, "competitive inhibition", "K_ic").value)
    ki_stderr.append(get_parameter(result, "competitive inhibition",
"K_ic").stderr)
ki = np.array(ki)
ki_stderr = np.array(ki_stderr)

# Visualize
kis = dict(zip(labels, ki))
f, (a0, a1) = plt.subplots(1, 2, gridspec_kw={'width_ratios': [3, 1.4]})
a0.bar(labels[[0, 2, 3, 4]], ki[[0, 2, 3, 4]], yerr=ki_stderr[[0, 2, 3, 4]],
align='center', capsize=3)
a0.set_xticklabels(labels[[0, 2, 3, 4]], rotation = 90)
a1.bar(labels[:2], ki[:2], yerr=ki_stderr[:2], align='center', capsize=3)
a1.set_xticklabels(labels[:2], rotation = 90)
a0.set_ylabel("$K_{i}$ [mg$ $mL$^{-1}$]")
plt.tight_layout()
```

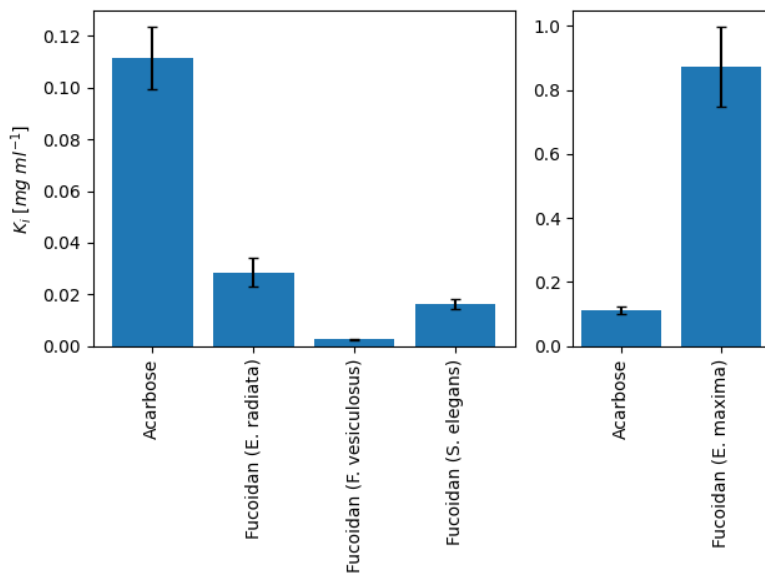


Fig. 6: Estimated inhibitory constants for α -glucosidase inhibition by fucoidan from different algae species and acarbose

Fucoidan from *E. radiata*, *F. vesiculosus*, and *S. elegans* all showed an lower K_i (28, 3, and 16 $\mu\text{g ml}^{-1}$ respectively) compared to the acarbose reference (112 $\mu\text{g ml}^{-1}$). This resembles a 37-fold higher binding affinity of fucoidan from *F. vesiculosus* to α -glucosidase compared to acarbose. Opposing to the previous study, fucoidan from *E. maxima* showed a more than 7-fold higher K_i (858 $\mu\text{g ml}^{-1}$) compared to acarbose. This likely originates from inconsistencies during experimental preparation. Additionally, 4 of 10 measurements with applied inhibitor were excluded from parameter estimation, since they showed higher activity compared to the respective reactions without inhibitor. Therefore, this measurement should be repeated. Estimated inhibition constants for each tested inhibitor are visualized in Fig. 6.

Project discussion

In general, the estimated parameters are not reliable on a quantitative level, since the malfunction of the analytical device undermines the validity of the whole dataset. Therefore, it cannot be ruled out, that some of the calculated concentrations deviate from the real concentrations. Furthermore, all estimated parameters are only estimated based on three time points of measure. In addition each experimental condition was applied as duplicate repeats, which additionally showed deviations. In conclusion the estimated parameters qualitatively confirm that fucoidan from different host organism, inhibits α -glucosidase more strongly, compared to acarbose.

Save modeling results

Lastly, the modeling results are written to the EnzymeML documents and the files are exported.

```

for result, enzml doc in zip(results, enzml_docs):

    # Write modeling results to kinetic parameters
    k_cat = pe.enzymeml.models.KineticParameter(
        name="k_cat",
        value=result.get_model_results()["k_cat"].value,
        unit=f"1 / {result.data.time_unit}")

    K_m = pe.enzymeml.models.KineticParameter(
        name="K_m",
        value=result.get_model_results()["Km"].value,
        unit=result.data.data_conc_unit)

    K_ic = pe.enzymeml.models.KineticParameter(
        name="K_ic",
        value=result.get_model_results()["K_ic"].value,
        unit=result.data.data_conc_unit)

    # Define kinetic model
    model = pe.KineticModel(
        name="competitive inhibition",
        equation="-k_cat * p0 * s0 / (K_m*(1+(s2 / K_ic))+s0)",
        parameters=[k_cat, K_m, K_ic])

    enzml doc.getReaction("r0").model = model

# Export EnzymeML documents
export = False
if export:
    enzml doc.toFile()

```

Scenario C: SLAC characterization

Data provided by Alaric Prins (Applied Microbial and Health Biotechnology Institute, Cape Peninsula University of Technology, Capetown, South Africa)

Project background

Laccases find industrial application in pulp and paper industry. Thereby, the enzyme is used for its polymerization or depolymerization capabilities through oxidation of phenoclic compounds [Widsten and Kandelbauer, 2008]. Something about SLAC...

Experimental design

In this scenario, the catalytic properties of the small laccase from *Streptomyces coelicolor* (SLAC) were investigated. Therefore, the enzymatic oxidation of 2,2'-Azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS) to its radical form ABTS^{•+} was studied in the pH range of pH 3 - pH 5.5 and temperature range of 25°C - 45°C. In total 30 kinetic enzyme assays in a substrate range between 0 - 200 µM of ABTS were conducted. Additionally, for each enzyme reaction with a given initial substrate concentration, a control reaction without enzyme was prepared.

For each pH - temperature condition an individual ABTS standard curve and absorption spectrum was recorded to account for varying ABTS absorption properties due to reaction conditions. Each enzyme reaction was followed for 15 min photometrically at two wavelengths, measuring substrate depletion and product accumulation simultaneously. Preliminary experiments confirmed, that ABTS absorbs at 340 nm, whereas the ABTS^{•+} absorbs at 420 nm in contrast to the substrate. Furthermore, cross absorbance of product at the substrate detection wavelength and vice versa was ruled out.

Data management

Overall, the dataset consists of more than 100 000 individual absorbance reads. Thus, data preparation was automated by custom parser functions, which were tailored to the output of the used spectrophotometer. Consequently, the tedious and error-prone manual copying of raw data was avoided. Information on the involved reactants, and the enzyme was filled in an EnzymeML Excel spreadsheet, which served as a meta data container. All other information was parsed from the output of the spectrophotometer. Fig. 7 illustrates the schematic data flow of this project.

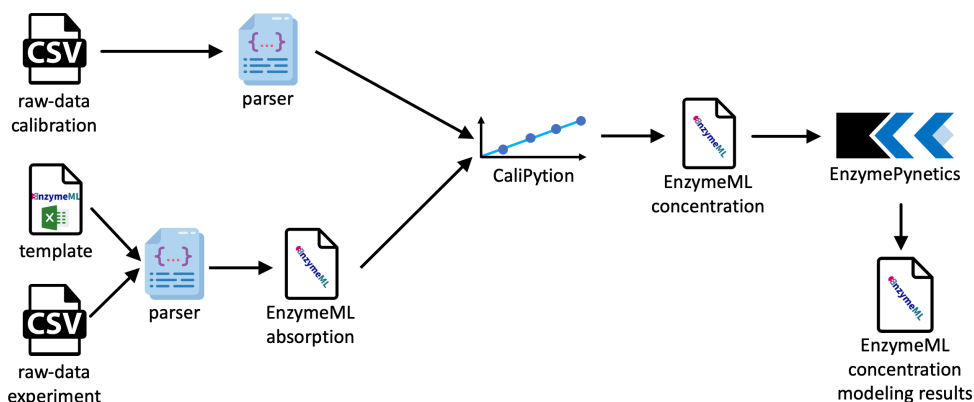


Fig. 7: Schematic data pipeline of the SLAC project.

Data preparation

Imports

```

from typing import Dict, List
import pyenzyme as pe
import numpy as np
import pandas as pd
import os
import matplotlib
import matplotlib.pyplot as plt
import seaborn as sns
import copy
import string
from lmfit import Parameters, minimize
from scipy.stats import linregress, pearsonr
from joblib import Parallel, delayed
from CaliPyton.tools.standardcurve import StandardCurve
from EnzymePynetics.tools.parameterestimator import ParameterEstimator

# Custom functions for data mapping
from parser_functions import measurement_data_to_EnzymeML, plot
from parser_functions import read_measurement_data, read_calibration_data
from parser_functions import add_enzyme_inactivation_model

import warnings
warnings.filterwarnings('ignore')

```

Experimental data

Data from SLAC reactions was loaded from the output files of the photometer and written to individual EnzymeML documents. Then, information of the control reactions was used to subtract the absorption contribution from enzyme and buffer from the substrate and product signal.


```

# Specify the location of the data sets
directory_measurement_data =
    "../data/SLAC_kinetic_characterization/TimeCourseData"
directory_standard_data = "../data/SLAC_kinetic_characterization/StandardData"
directory_spectrum_data = "../data/SLAC_kinetic_characterization/SpectrumData"
path_EnzymeML_templates =
    "../data/SLAC_kinetic_characterization/EnzymeML_templates"

# Define IDs for species, listed in the EnzymeML Excel template
substrate_id = "s0"
product_id = "s1"
substrate_control_id = "s2"
product_control_id = "s3"
species_ids = [substrate_id, product_id, substrate_control_id, product_control_id]

# Parse measurement data from photometer output
raw_data_dict = {}
for path in os.listdir(directory_measurement_data):
    data = read_measurement_data(f"{directory_measurement_data}/{path}")
    pH = data["pH"]
    temp = data["temperature"]
    raw_data_dict[f"{pH} {temp}"] = data

EnzymeML_template_dict = {
    3.0: "EnzymeML_SLAC_pH3.xlsm",
    3.5: "EnzymeML_SLAC_pH3_5.xlsm",
    4.0: "EnzymeML_SLAC_pH4.xlsm",
    4.5: "EnzymeML_SLAC_pH4_5.xlsm",
    5.0: "EnzymeML_SLAC_pH5.xlsm",
    5.5: "EnzymeML_SLAC_pH5_5.xlsm",
}

# Write absorption data to EnzymeMLDocuments
absortion_enzymemldocs: List[pe.EnzymeMLDocument] = []
for name, data in raw_data_dict.items():
    pH = data["pH"]
    absortion_enzymemldocs.append(measurement_data_to_EnzymeML(
        template_path=f"{path_EnzymeML_templates}/{EnzymeML_template_dict[pH]}",
        measurement_data=data,
        species_ids=species_ids,
        data_unit="umole / l",
        time_unit="s"))

# Sort documents by ascending pH and temperature
absortion_enzymemldocs = sorted(absortion_enzymemldocs, key=lambda x:
    (x.getReaction("r0").pH, x.getReaction("r0").temperature))

# Blanc measurement data
for enzml doc in absortion_enzymemldocs:
    blanc_measurement =
    enzml doc.measurement_dict["m0"].getReactant("s0").replicates
    blanc = np.mean([repeat.data for repeat in blanc_measurement])

    for id, measurement in enzml doc.measurement_dict.items():
        for rep, replicate in enumerate(measurement.getReactant("s0").replicates):
            blanced_data = [value - blanc for value in replicate.data]
            enzml doc.measurement_dict[id].getReactant("s0").replicates[rep].data =
            blanced_data
        for rep, replicate in enumerate(measurement.getReactant("s2").replicates):
            blanced_data = [value - blanc for value in replicate.data]
            enzml doc.measurement_dict[id].getReactant("s2").replicates[rep].data =
            blanced_data
        for rep, replicate in enumerate(measurement.getReactant("s1").replicates):
            blanced_data = [value - blanc for value in replicate.data]
            enzml doc.measurement_dict[id].getReactant("s1").replicates[rep].data =
            blanced_data
        for rep, replicate in enumerate(measurement.getReactant("s3").replicates):
            blanced_data = [value - blanc for value in replicate.data]
            enzml doc.measurement_dict[id].getReactant("s3").replicates[rep].data =
            blanced_data

    # Delete control measurement 'm0'
    del enzml doc.measurement_dict["m0"]

```

Concentration calculation

Calibration data was loaded and converted into individual instances of the calibration data model. Some meta data of the calibration needed to be provided to the custom `read_calibration_data` function, since the output of the used spectrophotometer only contained a minimum of information. Thereafter, a `StandardCurve` was created for each calibration data set. Thereby, only absorption values below 3.2 were considered, since higher absorption values could not be converted into concentration accurately. Lastly, the fit of each standard curves was visualized.

```

# Load calibration raw data
calibration_data = []
standard_directory = np.sort(os.listdir(directory_standard_data))
spectrum_directory = np.sort(os.listdir(directory_spectrum_data))

for standard, spectrum in zip(standard_directory, spectrum_directory):
    standard = f"{directory_standard_data}/{standard}"
    spectrum = f"{directory_spectrum_data}/{spectrum}"
    result = read_calibration_data(
        path_standard=standard,
        path_spectrum=spectrum,
        species_id=substrate_id,
        wavelengths=[340, 420],
        concentrations=[0,5,10,15,25,50,75,100,125,150,175,200],
        concentration_unit="umole / l",
        device_manufacturer="Molecular Devices",
        device_model="SpectraMax i3x",
        spectrum_reactant_concentration=100
    )
    for pH in result.keys():
        calibration_data.append(result[pH])

# Sort calibration data by ascending pH and temperature
calibration_data = sorted(calibration_data, key = lambda x: (x.pH, x.temperature))

# Generate standard curves for ABTS calibration data
standard_curves: List[StandardCurve] = []
for calibration in list(calibration_data):
    standard_curves.append(StandardCurve(calibration_data=calibration,
        wavelength=340, cutoff_absorption=3.2, show_output=False))

# Sort standard curves by ascending pH and temperature.
standard_curves = sorted(standard_curves, key = lambda x: (x.calibration_data.pH,
x.calibration_data.temperature))

# Visualize all fitted standard curves
fig, axes = plt.subplots(6,5, figsize=(10, 13), sharey=True, sharex=True)
for i, (standard, ax) in enumerate(zip(standard_curves[:30], axes.flatten())):
    if not i%5:
        ax.set_ylabel("ABTS [uM]")
        standard.visualize(ax=ax)
        ax.set_title(f"pH {standard.calibration_data.pH},
{standard.calibration_data.temperature}°C")
        if i in [25,26,27,28,29]:
            ax.set_xlabel("time [s]")
plt.tight_layout()

```

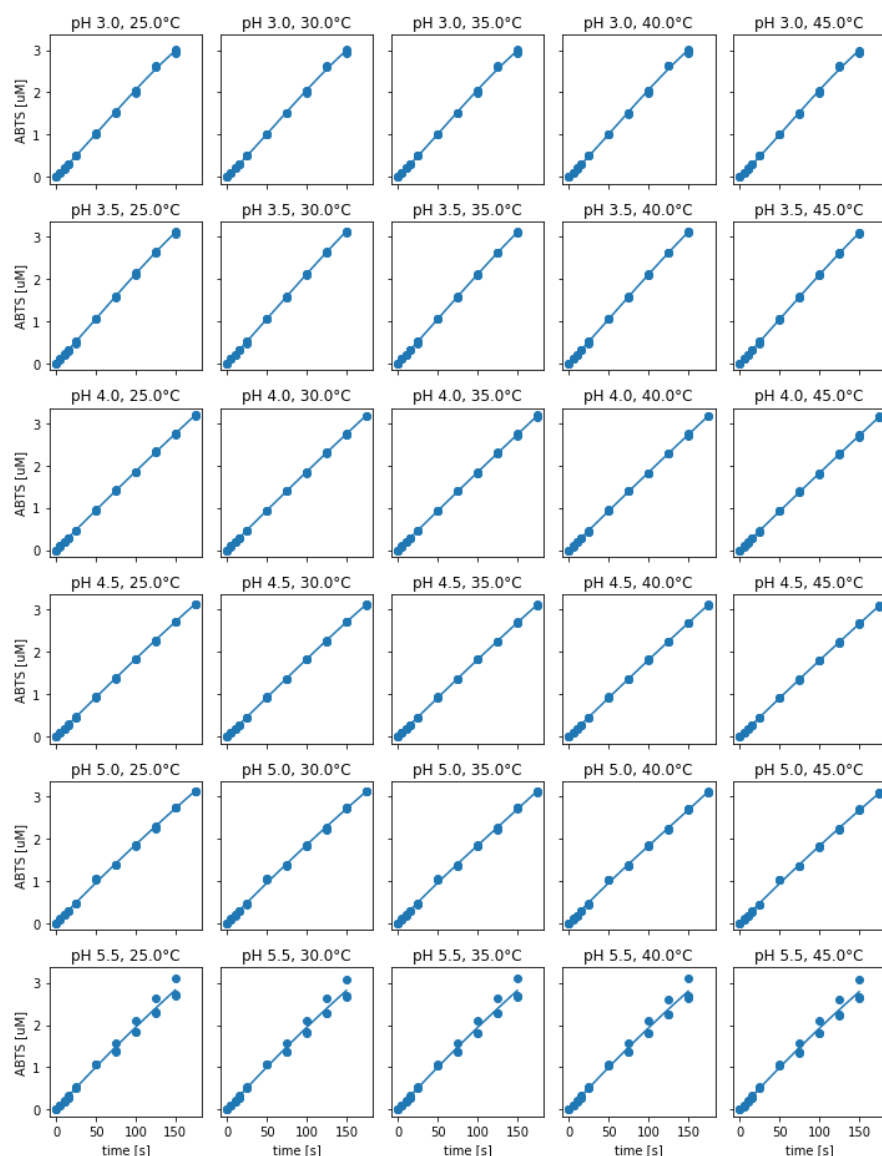


Fig. 8: ABTS standard curves for different experimental condition.

Since the absorption characteristics of ABTS change with pH, the calibration range differs between pH values, due to the upper absorption limit (Fig. 8). In consequence, the upper calibration limit for reactions at pH 3, pH 3.5, and pH 5.5 is at 150 μM of ABTS, whereas for all other pH values the upper limit is at 175 μM . This might source from the protonation state of ABTS, since the sulfonate groups of ABTS are deprotonated for less acidic pH values. Therefore, the absorption properties of ABTS can decrease. Calibration curve data at pH 5.5 showed larger variation between the repeats. In this case, pipetting of one of the three repeats differs from the other two, which should be considered for kinetic parameter estimation. In contrast to pH, the temperature during calibration affected the calibration curve only marginally.

The generated standard curves were used to convert the absorption measurement data into concentration data. Thereby, the respective concentration values were only calculated, if the measured absorption was within the respective calibration bounds to avoid extrapolation.

Since product and substrate of the SLAC reaction were simultaneously recorded, mass balance analysis was conducted as a control of quality. By assuming mass conservation, the following concentration balance can be established:

$$0 = S_{(t)} + P_{(t)} - S_0 \quad (14)$$

Thereby, S_0 denotes the initial substrate concentration, whereas $S_{(t)}$ and $P_{(t)}$ describe the substrate and product concentration for each time point t . S_t and S_0 were individually measured. Thus each enzyme reaction with a given initial substrate concentration had a control reaction with identical substrate concentration. In contrast to the substrate, no calibration standard was available for the product. Therefore, an additional parameter k was introduced to the mass balance equation, assuming linear relationship between the product concentration and its signal:

$$0 = S_{(t)} + P_{(t)}k - S_0 \quad (15)$$

k was determined for each data set individually by a minimization algorithm. The minimization objective was to find the optimal k , which minimizes all slopes of an experiment. If the slopes are zero, mass conservation is given over the observed time-course. Mass balances of all measurements are visualized in Fig. 9.

```

# Defenition of parameter 'k'
params = Parameters()
params.add("k", value=30, min=0, max=200)

# Target function for the minimizer
def residual(params, x):
    k=params["k"]

    substrate, product, control = x
    slopes = substrate[:,0] * 0.0
    for i, (s, p, c) in enumerate(zip(substrate, product, control)):
        sub = np.mean(s, axis=0)
        prod = np.mean(p, axis=0)
        cont = np.mean(c, axis=0)

        model = sub + prod*k - cont

        slopes[i] = linregress(np.arange(len(model)), model)[0]

    return slopes.flatten()

f = []
fig, axes = plt.subplots(6,5, figsize=(12.5, 15), sharey=True, sharex=True)
for e, (doc, ax) in enumerate(zip(concentration_enzymemldocs, axes.flatten())):
    substrate = []
    product = []
    initial_substrate = []
    control = []
    for measurement in doc.measurement_dict.values():
        initial_substrate.append(measurement.getReactant(substrate_id).init_conc)
        for replicate in measurement.getReactant(substrate_id).replicates:
            substrate.append(replicate.data)
        for replicate in measurement.getReactant(product_id).replicates:
            product.append(replicate.data)
        for replicate in measurement.getReactant("s2").replicates:
            control.append(replicate.data)

    time = np.array(replicate.time)

    product = np.array(product).reshape(int(len((product))/3),3,11)
    substrate = np.array(substrate).reshape(int(len((substrate))/3),3,11)
    control = np.array(control).reshape(int(len((control))/3),3,11)
    initial_substrate = np.array(initial_substrate)

    x = (substrate, product, control)

    result = minimize(residual, params, args=(x,))

    factor = result.params["k"].value
    f.append(factor)

    for i, (s, p, c, l) in enumerate(zip(substrate, product, control,
initial_substrate)):
        init = np.mean(c, axis=0)
        sub = np.mean(s, axis=0)
        prod = np.mean(p, axis=0)
        cont = np.mean(c, axis=0)
        balance = (sub + prod*factor - cont)

        ax.plot(doc.getMeasurement("m1").getReactant("s0").replicates[0].time,
balance, label = l)
        if not i%5:
            ax.set_ylabel("mass balance [uM]")
            if i in [24, 25, 26, 27, 28, 29]:
                ax.set_xlabel("time [s]")
            pH = doc.getReaction("r0").ph
            temp = doc.getReaction("r0").temperature
            ax.set_title(f"pH {pH}, {temp}°C")#, factor: {factor:.0f}")
        if e == 0:
            handles, labels = ax.get_legend_handles_labels()
            fig.legend(handles, labels, loc="lower center", ncol=len(labels),
title="initial ABTS [uM]", bbox_to_anchor=(0.5,-0.03))
plt.tight_layout()

```

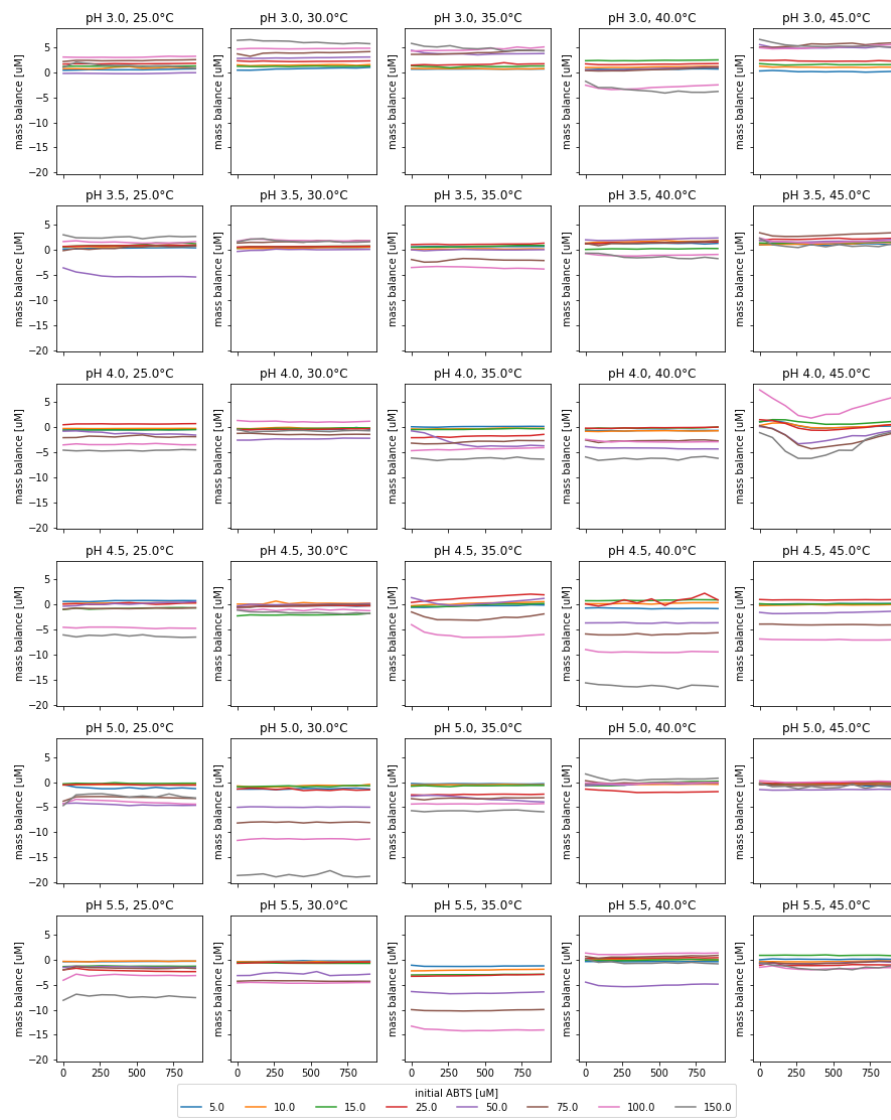



Fig. 9: Quality control of each experiment through modeled mass balance.

For all experiments, except for pH 4, 45°C and pH 4.5, 35°C, the product and substrate is in balance over the reaction time-course. Mass balances, differing from 0 μM originate from differing substrate concentration in the enzyme reaction and the substrate control. In the reactions at pH 4, 45°C and pH 4.5, 35°C the mass balance slopes are not linear. This indicates issues with the measurement. Therefore, the respective measurements likely result in wrong parameter estimations.

Kinetic parameter estimation

Choice of kinetic model

Model selection is vital for parameter estimation. Thus, different settings for the parameter estimator were tested. Firstly, product inhibition models were excluded, since in some of the experiments the measured initial substrate concentration was higher than the specified one, which is used to calculate product concentrations. This resulted in calculated product concentrations with negative values.

Secondly, substrate inhibition models were excluded, since the model was not able to describe the observed reaction kinetics. This was evident from a higher AIC as well as more than 100 % standard deviation on the estimated parameters.

Lastly, the irreversible Michaelis-Menten model with and without time-dependent enzyme inactivation was compared, since enzyme inactivation was observed in previous experiments. Estimated parameters and the fitted models are shown for the experimental data at pH 3 and 25°C under two scenarios - taking enzyme inactivation into consideration and, conversely, without enzyme inactivation (Fig. XXX).

```

fig, axes = plt.subplots(1,2, figsize=(10,5), sharey=True, sharex=True)
for i, ax in enumerate(axes.flatten()):
    ax.text(0, 1.1, string.ascii_uppercase[i], transform=ax.transAxes,
           size=20, weight='bold')
    if i == 0:
        print("Estimated parameters without time-dependent enzyme inactivation:")
        kinetics = ParameterEstimator.from_EnzymeML(concentration_enzymemldocs[0],
        "s0", "substrate")
        kinetics.fit_models(only_irrev_MM=True)
        kinetics.visualize(ax=ax, title="without enzyme inactivation")
        ax.set_ylabel("ABTS [uM]")
        print("\n")
    else:
        print("Estimated parameters with time-dependent enzyme inactivation:")
        kinetics = ParameterEstimator.from_EnzymeML(concentration_enzymemldocs[0],
        "s0", "substrate")
        kinetics.fit_models(enzyme_inactivation=True, only_irrev_MM=True)
        kinetics.visualize(ax=ax, title="with enzyme inactivation")
        ax.set_xlabel("time [s]")
handles, labels = ax.get_legend_handles_labels()
fig.legend(handles, labels, loc="lower center", ncol=len(labels), title="initial
ABTS [uM]", bbox_to_anchor=(0.5,-0.09))
plt.tight_layout()

```

Estimated parameters without time-dependent enzyme inactivation:
Fitting data to:
- irreversible Michaelis Menten

	AIC	Km [umole / l]	kcat / Km [1/s * 1/umole / l]	kcat [1/s]
irreversible Michaelis Menten	-162	35.473 +/- 5.52%	0.009 +/- 5.89%	0.308 +/- 2.07%

Estimated parameters with time-dependent enzyme inactivation:
Fitting data to:
- irreversible Michaelis Menten model

	AIC	Km [umole / l]	kcat / Km [1/s * 1/umole / l]	kcat [1/s]	ki time-dep enzyme-inactiv. [1/s]
irreversible Michaelis Menten	-476	35.137 +/- 3.00%	0.013 +/- 3.58%	0.460 +/- 1.95%	0.001 +/- 4.47%

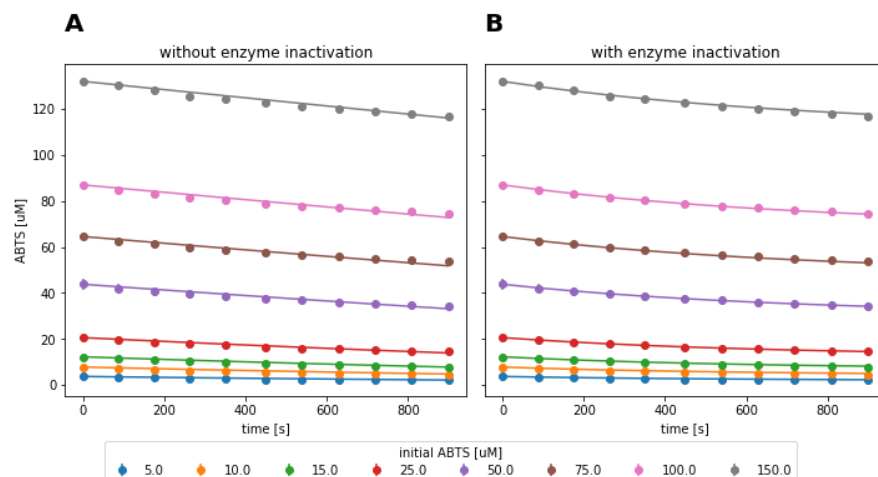


Fig. 10: Fitted irreversible Michaelis-Menten model to experimental data of SLAC reaction at pH 3 and 25°C with and without considering time-dependent enzyme inactivation.

The model with enzyme inactivation describes the data better (Fig. 10B). This is characterized through a lower AIC, as well as lower standard deviations on the estimated parameters. In absolute terms, the estimated K_m of both models is approximately identical, whereas the k_{cat} estimate is approximately 33% lower for the model without enzyme inactivation. Hence, the model without enzyme inactivation underestimates the turnover number of the enzyme. As a result, irreversible Michaelis-Menten model with time-dependent enzyme inactivation was selected for the parameter estimation for all data sets.

```

# Run parameter estimator for all datasets, utilizing multi-processing.
def run_ParameterEstimator(enzmldoc: pe.EnzymeMLDocument):
    kinetics = ParameterEstimator.from_EnzymeML(enzmldoc, "s0", "substrate")
    kinetics.fit_models(enzyme_inactivation=True, only_irrev_MM=True,
display_output=False)
    return kinetics

results = Parallel(n_jobs=8)(delayed(run_ParameterEstimator)(enzmldoc) for
enzmldoc in concentration_enzymemldocs)
results = sorted(results, key=lambda x: (x.data.pH, x.data.temperature))

# Visualize all fitted models
fig, axes = plt.subplots(6,5, figsize=(12.5, 15), sharey=True, sharex=True)
for i, (doc, ax) in enumerate(zip(results, axes.flatten())):
    pH = doc.data.pH
    if not i%5:
        ax.set_ylabel("ABTS [uM]")
    doc.visualize(ax=ax, title = f"pH {doc.data.pH}, {doc.data.temperature}°C")
    if i in [25,26,27,28,29]:
        ax.set_xlabel("time [s]")
    if i == 0:
        handles, labels = ax.get_legend_handles_labels()
        fig.legend(handles, labels, loc="lower center", ncol=len(labels),
title="initial ABTS [uM]", bbox_to_anchor=(0.5,-0.04))
plt.tight_layout()

```

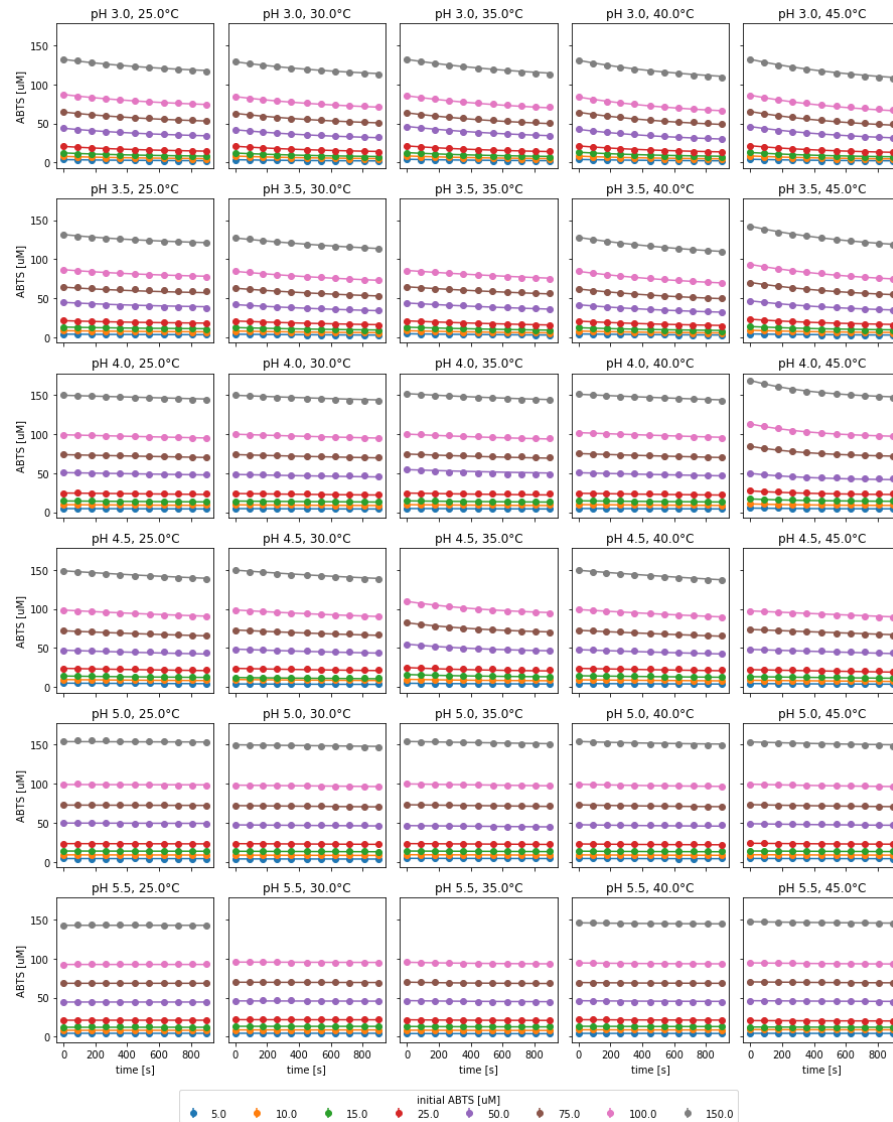


Fig. 11: Experimental data and fitted irreversible Michaelis-Menten model with time-dependent enzyme inactivation of SLAC reactions under various conditions.

Experimental data as well as the fitted model are visualized in Fig 11. Based on the reaction slopes, no catalytic activity was observed for reactions at pH 5 or higher. As in the mass balance analysis, the reactions at pH 4, 45°C and pH 4.5, 35°C differed from other experiments with identical pH. In both cases, all applied substrate concentration were higher than intended in the design of the experiment. Furthermore, the resulting parameter estimates have a high uncertainty. Therefore, the respective measurements were

excluded from further analysis. Additionally, the results from pH 5.5, 35°C were excluded, since the uncertainty of the parameters could not be estimated. All calculated kinetic parameters are listed in the table below.

```
# Extract kinetic parameters of all datasets
kcat = []
kcat_std = []
Km = []
Km_std = []
ki = []
ki_std = []
pH = []
temperature = []
corr_kcat_km = []
for result in results:
    params = result.get_parameter_dict()

    kcat.append(params["k_cat"].value)
    kcat_std.append(params["k_cat"].stderr)

    Km.append(params["Km"].value)
    Km_std.append(params["Km"].stderr)

    ki.append(params["K_ie"].value)
    ki_std.append(params["K_ie"].stderr)

    correlation = params["k_cat"].correl
    if correlation == None:
        corr_kcat_km.append(float("nan"))
    else:
        corr_kcat_km.append(correlation["Km"])

    pH.append(result.data.pH)
    temperature.append(result.data.temperature)

# Organize kinetic and experimental parameters in a 'DataFrame'
df = pd.DataFrame.from_dict({
    'pH':pH,
    'temperature [C]':temperature,
    'kcat [1/s]':kcat,
    'kcat stderr':kcat_std,
    'Km [uM]':Km,
    'Km stderr':Km_std,
    'Enzyme inactivation [1/s]':ki,
    'Enzyme inactivation std':ki_std,
    'correlation kcat/Km':corr_kcat_km})

df["kcat/Km [1/s * uM]"] = df["kcat [1/s]"] / df["Km [uM]"]
kcat_km_stderr = ((df["kcat stderr"]/df["kcat [1/s]"])**2 + (df["Km stderr"]/df["Km [uM]"])**2)**0.5 * df["kcat/Km [1/s * uM]"]
df["kcat/Km stderr"] = kcat_km_stderr

# Excluded results from failed experiments
df = df.drop(index=25) # reached parameter boundaries --> inactive
#df = df.drop(index=12) #Km really high stddev
df = df.drop(index=17) #kcat outlier, mass balance outlier
df = df.drop(index=14) #kcat outlier, mass balance outlier
df["zeros"] = np.zeros(27)

# Calculate halflife of enzyme
def calculate_halflife(x):
    return np.log(2)/x

df["half life [min]"] = df["Enzyme inactivation [1/s]"].apply(calculate_halflife)/60
df["half life std"] = df["Enzyme inactivation std"] / df["Enzyme inactivation [1/s]"] * df["half life [min]"]

# Delete inactivation rates
df = df.drop(columns=["Enzyme inactivation [1/s]", "Enzyme inactivation std"])
df
```

	pH	temperature [C]	kcat [1/s]	kcat stderr	Km [uM]	Km stderr	correlation kcat/Km	kcat/Km [1/s * uM]	kcat/Km stderr	zeros	half life [min]	half life :
0	3.0	25.0	0.459962	0.008966	35.137357	1.053968	0.535141	0.013090	0.000468	0.0	9.139562	0.4085
1	3.0	30.0	0.470366	0.009855	32.114083	1.037088	0.522963	0.014647	0.000564	0.0	9.522170	0.4761
2	3.0	35.0	0.546029	0.016598	37.454496	1.765891	0.558865	0.014578	0.000818	0.0	10.304462	0.7761
3	3.0	40.0	0.712488	0.021754	46.441229	2.155285	0.608947	0.015342	0.000852	0.0	9.127765	0.6008
4	3.0	45.0	0.815724	0.022451	47.967746	2.047138	0.623270	0.017006	0.000864	0.0	9.087343	0.5305
5	3.5	25.0	0.420480	0.045642	89.752447	13.848584	0.746575	0.004685	0.000884	0.0	8.824513	1.6803
6	3.5	30.0	0.399649	0.007455	61.224383	1.674235	0.666682	0.006528	0.000216	0.0	14.545045	0.8041
7	3.5	35.0	0.246773	0.009247	28.237534	1.572755	0.554406	0.008739	0.000587	0.0	18.269277	2.7668
8	3.5	40.0	0.650031	0.012397	79.432405	2.189303	0.731830	0.008183	0.000274	0.0	11.026573	0.4531
9	3.5	45.0	0.978983	0.026556	103.784356	4.040850	0.774126	0.009433	0.000448	0.0	9.110325	0.4253
10	4.0	25.0	0.115133	0.008752	65.852866	2.476592	0.247717	0.001748	0.000148	0.0	111.702355	222.0642
11	4.0	30.0	0.192374	0.020265	95.314777	5.569713	-0.718208	0.002018	0.000243	0.0	23.819987	10.8746
12	4.0	35.0	0.282991	0.029279	100.647793	11.576143	0.608286	0.002812	0.000435	0.0	14.929497	4.9742
13	4.0	40.0	0.225429	0.012766	97.966631	1.677116	0.168743	0.002301	0.000136	0.0	29.441552	12.2889
15	4.5	25.0	0.368730	0.025801	110.902372	0.830267	-0.736267	0.003325	0.000234	0.0	22.536023	6.2472
16	4.5	30.0	0.486402	0.036084	148.193267	3.440830	0.816213	0.003282	0.000255	0.0	19.046651	4.1091
18	4.5	40.0	0.549232	0.021351	178.481651	2.544896	0.084106	0.003077	0.000127	0.0	34.946634	11.9310
19	4.5	45.0	0.235716	0.013338	64.718239	0.037679	-0.113239	0.003642	0.000206	0.0	108.053681	160.9705
20	5.0	25.0	0.008085	0.006871	15.067578	20.447918	0.276523	0.000537	0.000859	0.0	115.522134	2593.0861
21	5.0	30.0	0.040665	0.005659	50.772187	3.784251	0.009319	0.000801	0.000126	0.0	115.512956	453.0162
22	5.0	35.0	0.061677	0.006158	68.874565	0.267258	0.299400	0.000895	0.000089	0.0	115.519891	286.7058
23	5.0	40.0	0.131196	0.017131	91.314407	2.756719	-0.426692	0.001437	0.000193	0.0	8.399863	2.5198
24	5.0	45.0	0.129399	0.014577	96.735544	2.562607	-0.696719	0.001338	0.000155	0.0	11.438272	2.9869
26	5.5	30.0	0.014374	0.007402	96.125085	0.393705	-0.200038	0.000150	0.000077	0.0	115.523248	1471.9722
27	5.5	35.0	0.124370	0.023968	81.996103	22.257271	0.771045	0.001517	0.000505	0.0	5.841192	1.4001
28	5.5	40.0	0.055892	0.016036	48.740460	19.833207	0.552798	0.001147	0.000571	0.0	6.132460	2.9573
29	5.5	45.0	0.075249	0.015052	115.390172	27.541849	0.604743	0.000652	0.000203	0.0	10.834329	5.3431

High correlation between k_{cat} and K_m

The parameter estimates for k_{cat} and K_m showed correlation. For experiments at pH 3 and pH 3.5, the parameters were correlated between $r^2 = 0.5 - 0.85$. This indicates, that the highest initial substrate concentration, which was applied for parameter estimation was not sufficiently high. Ideally, the highest initial substrate concentration applied for a kinetic experiment should be 10-fold higher than K_m to allow independent estimates for k_{cat} and K_m [Tipton et al., 2014]. In this scenario, only reactions with an initial substrate concentration up to 175 uM were used for parameter estimation, due to the limited photometric detection range of ABTS. Hence, ABTS was only applied 1.5-fold to 6-fold of the estimated K_m . For reactions at pH 4 and above, positive as well as negative correlations were observed. This might result from wrong parameter estimates, due to minimal enzyme activity at the given reaction conditions.

Catalytic efficiency for all experimental conditions is visualized in Fig. XXXB, whereas the pH is color-coded. For reactions with identical pH values, k_{cat} and K_m both increase with temperature. This might be attributed to the high correlation between the parameters. Thus, the true change of k_{cat} or K_m through pH or temperature cannot be assessed with certainty. Therefore, the catalytic efficiency $\frac{k_{cat}}{K_m}$, was a better measure to assess the enzyme activity under different reaction conditions.

For reactions at pH 5 and above, almost no catalytic activity was observed. Therefore, data of these experimental conditions is excluded from further analysis.

Parameter estimates for k_{cat} and K_m

```
# Visualize estimated parameters
fig, axes = plt.subplots(1,2, figsize=(12.8, 4.8), sharey=False, sharex=False)
for i, ax in enumerate(axes.flatten()):
    ax.text(0, 1.1, string.ascii_uppercase[i], transform=ax.transAxes,
           size=20, weight='bold')
    if i==1:
        plot(df, xdata="Km [uM]", ydata="kcat [1/s]", xerror="Km stderr",
             yerror="kcat stderr", colors="pH", ax=ax)
    else:
        plot(df, xdata="pH", ydata="kcat/Km [1/s * uM]", xerror="zeros",
             yerror="kcat/Km stderr", colors="temperature [C]", ax=ax)
```

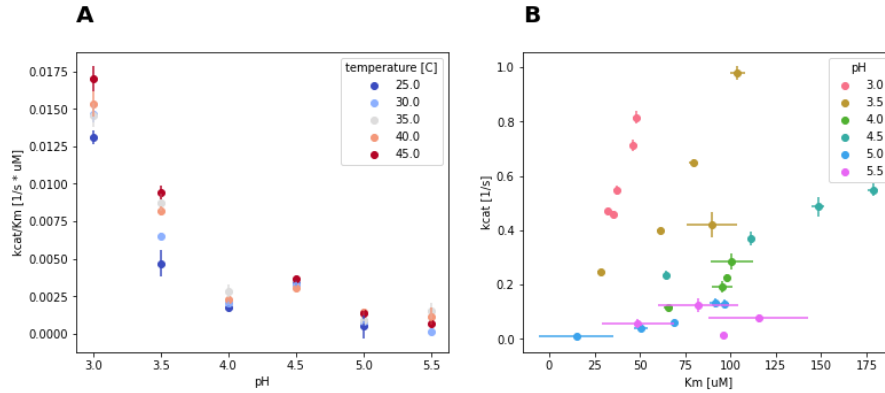


Fig. 12: A: Catalytic efficiency of SLAC under different pH and temperature conditions. B: Kinetic parameters k_{cat} and $\{K_m\}$ estimates for SLAC reactions at different pH values._

$\frac{k_{cat}}{K_m}$ in relation to reaction pH is visualized in Fig. XXXA. The highest catalytic efficiency was observed at pH 3 and 45°C. Increasing the pH by 0.5 reduced $\frac{k_{cat}}{K_m}$ approximately by half. For higher pH values, $\frac{k_{cat}}{K_m}$ is even further reduced. The catalytic efficiency is therefore highly sensitive to the pH. This might source from changes in the protonation state of either substrate or enzyme, hindering the efficient formation of the enzyme-substrate complex. In terms of temperature, higher $\frac{k_{cat}}{K_m}$ was achieved at higher temperatures. SLAC might even be more active above 45°C and under pH 3, since in previous studies SLAC was the most active at 95°C [Prins et al., 2015].

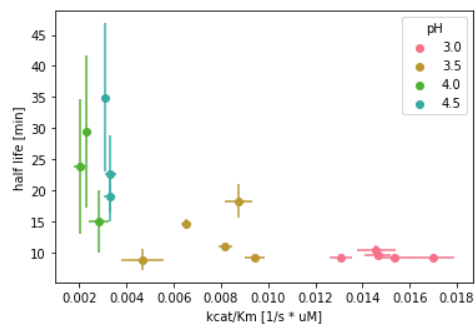
Time-dependent enzyme inactivation

The half life of SLAC was calculated with equation (16) [Buxbaum, 2015].

$$t_{\frac{1}{2}} = \frac{\ln(2)}{k_{inact}} \quad (16)$$

```
# Excluded inactive enzyme reactions above pH 4.5 and discard measurements with
# more than 100 % standard deviation on enzyme inactivation parameter
df = df.loc[:19]
df = df.drop(index=10) # more than 100 % standard deviation on enzyme inactivation
# parameter
df = df.drop(index=19) # more than 100 % standard deviation on enzyme inactivation
# parameter

# Plot half life over catalytic efficiency
plot(df, xdata="kcat/Km [1/s * uM]", ydata="half life [min]", xerror="kcat/Km
stderr", yerror="half life std", colors="pH")
plt.show()
```



The enzyme's calculated half life was between 8 - 18 min for reactions at pH 3 and pH 3.5, whereas enzyme reactions at higher pH values showed a half life between 15 - 35 min. Generally, reactions with higher catalytic efficiency showed a shorter half life compared to reactions with lower catalytic efficiency. In order to check for corrections between k_{inact} , k_{cat} , K_m , $\frac{k_{cat}}{K_m}$, as well as the reaction temperature and pH, a correlation analysis was conducted across all datasets.

```
# Calculate correlations
df_corr = df.drop(columns=["kcat stderr", "Km stderr", "correlation kcat/Km",
"Kcat/Km stderr", "zeros", "half life std"])
rho = df_corr.corr()

# Visualize correlation heatmap
#sns.heatmap(rho,
#            xticklabels=rho.columns,
#            yticklabels=rho.columns,
#            cmap = sns.color_palette("vlag", as_cmap=True))
#plt.show()

# Calculate confidence interval for correlations
pval = df_corr.corr(method='lambda x, y: pearsonr(x, y)[1]) - np.eye(*rho.shape)
p = pval.applymap(lambda x: ''.join(['*' for t in [.05] if x<=t]))
rho.round(2).astype(str) + p
```

	pH	temperature [C]	kcat [1/s]	Km [uM]	kcat/Km [1/s * uM]	half life [min]
pH	1.0*	-0.15	-0.39	0.88*	-0.89*	0.8*
temperature [C]	-0.15	1.0*	0.6*	0.07	0.27	0.03
kcat [1/s]	-0.39	0.6*	1.0*	-0.03	0.58*	-0.49
Km [uM]	0.88*	0.07	-0.03	1.0*	-0.75*	0.68*
kcat/Km [1/s * uM]	-0.89*	0.27	0.58*	-0.75*	1.0*	-0.71*
half life [min]	0.8*	0.03	-0.49	0.68*	-0.71*	1.0*

The above table shows correlation between between kinetic parameters and experimental conditions. Thereby, significant ($p < 0.05$) correlations are labeled with '*'. The half life of SLAC is significantly correlated to $\frac{k_{cat}}{K_m}$ as well as to pH. Due to correlation between k_{cat} and K_m , the enzyme's half life is also correlated to K_m . Multiple reasons might be the reason for this observation. On the one hand, the observed correlation might be of technical origin. As such, the system of ordinary differential equations, describing the change in substrate and enzyme concentration, are not an accurate model for the reaction system. Therefore, high cross-correlation occurs, since individual observations cannot be attributed to individual parameters. On the other hand, enzyme activity and enzyme inactivation might be causally related. In this case, the formed ABTS radical might inactivate the enzyme by potential suicide inhibition, or the enzyme deteriorates through catalysis.

Conclusion

Data management

The established data pipeline allows scalable data preparation and analysis of enzyme kinetics experiments. In future experiments this workflow could be used to further expand the experimental parameter space. As a result, different enzymes with different substrates in different buffers at different temperatures and pH values could be analyzed.

Correlation between parameters

The highest catalytic efficiency of SLAC was observed at pH 3 at 45°C and therefore on the edge of the investigated parameter space. Optimal reaction conditions might therefore be at an even lower pH and higher temperature. k_{cat} and K_m were correlated, which is likely the result of a too low initial substrate concentration in relation to the true K_m of the enzyme at a given experimental condition. Hence, in adjoining experiments the product signal could be used for parameter estimation, since the concentration to absorbance ratio is lower compared to the one of the product. However, a reliable method for product quantification would need to be established first.

Enzyme inactivation

SLAC showed a short half life of approximately 10 min at reaction conditions with the highest catalytic efficiency, which was correlated to the catalytic efficiency. Further experiments are required to investigate the observed inactivation. This might be done by comparing the half life of SLAC with different substrates as well as higher enzyme concentrations.

Save modeling results

Lastly, the modeling results are written to the EnzymeML documents and the files are exported.

```
for result, enzmldoc in zip(results, concentration_enzymemldocs):

    # Write modeling results to kinetic parameters
    k_cat = pe.enzymeml.models.KineticParameter(
        name="k_cat",
        value=result.get_model_results()["k_cat"].value,
        unit=f"1 / {result.data.time_unit}")

    K_m = pe.enzymeml.models.KineticParameter(
        name="K_m",
        value=result.get_model_results()["Km"].value,
        unit=result.data.data_conc_unit)

    # Define kinetic model
    model = pe.KineticModel(
        name="irreversible Michaelis-Menten",
        equation="-k_cat * p0 * s0 / (K_m + s0)",
        parameters=[k_cat, K_m])

    enzmldoc.getReaction("r0").model = model

    add_enzyme_inactivation_model(
        result=result,
        enzmldoc=enzmldoc)

# Export EnzymeML documents
export = False
if export:
    enzmldocToFile()
```

Scenario D: Time-dependent enzyme inactivation

Data provided by Paulo Durão (Microbial & Enzyme Technology, Instituto de Tecnologia Química e Biológica, Oeiras, Portugal)

Project background

All investigated enzyme reaction without inhibitor applied to the reaction showed progress curve behavior, which were not explainable by irreversible Michaelis-Menten kinetics. All experiments had in common, that enzyme reactions were carried out in 96-well polystyrene micro titer plates (MTP), whereas the change in substrate or product absorption was monitored photometrically.

One hypothesis for the suspected enzyme inactivation was hydrophobic interaction between the enzyme and the MTP surface. Thereby, hydrophobic regions of the enzyme's surface might interact with the polystyrene reaction vessel, potentially preventing substrate access to the active site of the enzyme. Another hypothesis for the observed gradual activity decrease was inactivation which is linked to the catalytic activity, as observed in the precious scenario.

In order to test the hypothesis, an experiment was designed in order to detach enzyme inactivation from enzyme activity. Therefore, enzyme was incubated in MTP wells prior to reaction start for different periods. Thereby, the enzyme activity should decrease depending on prior incubation time. If the hypothesis is correct, the calculated half life from the adsorption experiment should match the half life of an enzyme kinetics experiment which was conducted in parallel and also considered an inactivation term.

Experimental design

Determination of enzyme inactivation trough adsorption

The inactivation experiment was conducted by incubating CotA laccase from *Bacillus subtilis* in individual MTP wells up to 1 h prior to reaction start. Then, enzyme reactions were started in 10 min intervals by transferring 2 µL of incubated enzyme in individual MTP wells. Each proceeding enzyme reaction contained 256 nM CotA, 1 mM ABTS and was buffered in acetate buffer at pH 4. Product formation was followed

photometrically at 420 nm and 25°C for 5 min, whereas concentrations were calculated assuming an extinction coefficient of $\epsilon = 36000 \text{ M}^{-1}\text{cm}^{-1}$ for the ABTS radical product. Each experiment was performed as technical triplicates.

Enzyme kinetics experiment

Enzymatic oxidation of ABTS to its radical form was followed photometrically at 420 nm at 25°C for 70 min. Thereby, ABTS was applied in a range from 0.01 mM - 2 mM. Each proceeding enzyme reaction contained 256 nM CotA, and was buffered in acetate buffer at pH 4. Each experimental condition was repeated as technical quadruplicates. Concentrations were calculated via the provided extinction coefficient of the ABTS radical ($\epsilon = 36000 \text{ M}^{-1}\text{cm}^{-1}$).

Experimental data

Measurement data was provided as an Excel file, containing time-course absorption data. Meta data was written into the EnzymeML Excel template. Then, the experimental data was written to an EnzymeML document by a parser function.

Adsorption of enzyme to micro titer plate surface

Imports

```
import numpy as np
import pandas as pd
from scipy.stats import linregress
import pyenzyme as pe
from EnzymePynetics.tools.parameterestimator import ParameterEstimator
import matplotlib.pyplot as plt
import matplotlib.cm as cm
from IPython.display import display
import string
from parser_functions import add_enzyme_inactivation_model

import warnings
warnings.filterwarnings('ignore')
```

Calculation of inactivation rate

Experimental data was loaded into a Pandas DataFrame and the slopes of each incubation condition were calculated through linear regression. The resulting initial rates were then used to calculate the rate by which the enzyme activity decreases. Lastly the half life of the enzyme was calculated with equation [\(16\)](#).

```

# Load excel
path = '../data/enzyme_inactivation/Slide 2 - Activity effect of incubating
CotA in MTP.xlsx'
df = pd.read_excel(path, sheet_name='csv').set_index('time (min)')

# replace values of '0*' with nan-values, since the measurement is incorrect
df['0*'] = np.nan

# Get data from Excel file
columns = [int(x) for x in list(df.columns) if str(x).endswith("0")]
time = df.index.values
absorption = df.values.T.reshape(7,3,22)

# Calculate concentrations
extinction_coefficient = 36 # (1/mM * 1/cm)
optical_length = 0.65 # cm

def absorption_to_concentration(abso):
    return abso / (extinction_coefficient*optical_length)

concentration = absorption_to_concentration(absorption)
concentration_mean = np.nanmean(concentration, axis = 1)
concentration_std = np.nanstd(concentration, axis = 1)

# Linear regression
slopes = []
intercepts = []
stderrs = []
for time_set in concentration:
    mask = ~np.isnan(time_set)
    time_regression = np.tile(time, 3).reshape(mask.shape)[mask].flatten()
    data_regression = time_set[mask].flatten()
    slope, intercept, _, stderr = (linregress(time_regression, data_regression))
    slopes.append(slope)
    intercepts.append(intercept)
    stderrs.append(stderr)

# Regression of the slopes between 10 and 60 min
slope_of_slopes, intercept_of_slopes, _, _ = linregress(columns[1:], slopes[1:])

# Calculate time at which enzyme activity is reduced by half
half_activity = ((slopes[1]/2 - slopes[1]) / slope_of_slopes)

# Plot results
print(f"Calculated time after enzyme activity is halved based on linear
regression: {half_activity:.2f} min")

colors = cm.tab10(np.linspace(0, 1, len(absorption)))
fig, axes = plt.subplots(1,2, figsize=(12.8,4.8), sharey=False, sharex=False)

for mean, std, label, slope, intercept, color in zip(concentration_mean,
concentration_std, columns, slopes, intercepts, colors):
    axes[0].errorbar(time, mean, std, fmt='o', label=label, alpha = 0.4,
color=color)
    axes[0].plot(np.array(time), np.array(time)*slope+intercept, '--',
color=color)
    axes[0].legend(title="CotA incubation\nin MTP-well [min]")
    axes[0].set_ylabel('ABTS radical [mM]')
    axes[0].set_xlabel('time [min]')
    axes[0].set_title('Experimental data with calculated reaction rate by linear
regression')
    axes[0].text(-0.1, 1.1, string.ascii_uppercase[0],
transform=axes[0].transAxes,
size=20, weight='bold')

axes[1].errorbar(columns, slopes, stderrs, fmt='o', label="initial rates within
the first 5 min")
axes[1].set_xlabel("CotA incubation time\nin MTP-well [min]")
axes[1].set_ylabel('initial rate [mM min$^{-1}$]')
axes[1].plot(np.array(columns[1:]),
slope_of_slopes*np.array(columns[1:])+intercept_of_slopes, "--", color=colors[0],
label=f"regression")
axes[1].legend()
axes[1].set_title('Initial rates of CotA reaction')
axes[1].text(1.1, 1.1, string.ascii_uppercase[1], transform=axes[0].transAxes,
size=20, weight='bold')

plt.show()

```

Calculated time after enzyme activity is halved based on linear regression: 105.09 min

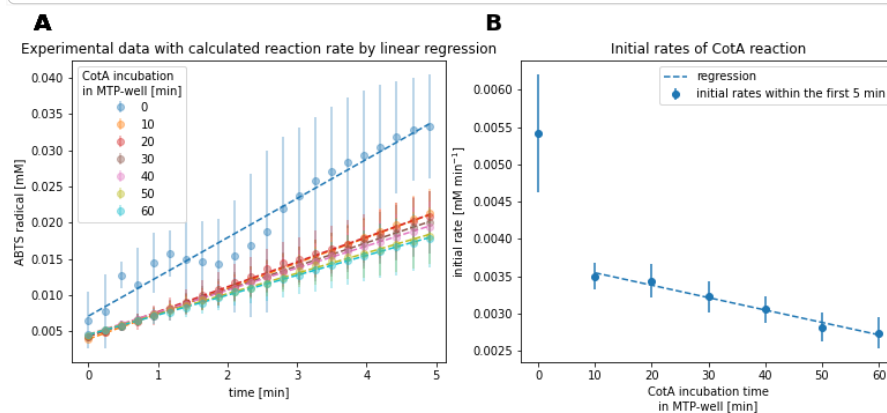


Fig. 13: Experimental data and regression results of CotA reaction with different enzyme incubation times.

Enzyme reactions without prior incubation showed the fastest reaction rate, whereas all reactions with prior incubation showed an approximately halved reaction rate (Fig. 13A). Thereby, reactions with longer incubation time showed gradually reduced reaction rates with increased incubation time. Enzyme reactions, especially reactions without incubation, showed large deviations across replicates. This presumably resulted from inhomogeneous mixing or deviating enzyme concentration due to the small volume of enzyme solution, which was used to start the reactions.

Since the true reaction rate of reactions without prior incubation was highly uncertain and showed mixing issues, the respective data was not used to estimate the enzyme's time-dependent inactivation rate. Based on linear regression of the initial reaction rates, excluding the reaction without prior incubation, the enzyme activity is approximately halved after 105 min (Fig. 13B). This assumes validity of extrapolation.

Enzyme kinetics experiment

Experimental data

Experimental data was provided as an Excel file, whereas meta data of the experiment was filled in to an EnzymeML Excel spreadsheet. Measurement data was written to the EnzymeML document by a parser function. Kinetic parameters were estimated with and without considering enzyme inactivation.

```

# Load experimental data from excel file
df = pd.read_excel("../data/enzyme_inactivation/Repetition CotA ABTS kinetics
higher volumes 2nd time.xlsx", sheet_name="csv").set_index("Time(min)")
data = df.values.T.reshape(8,4,72)
time = df.index.values

# Calculate concentrations
extinction_coefficient = 36 # (1/mM * 1/cm)
optical_length = 0.65 # cm

def absorption_to_concentration(abso):
    return abso / (extinction_coefficient*optical_length)

concentration_data = absorption_to_concentration(data)

# Parser function
def data_to_EnzymeML(
    template_path: str,
    measurement_data: np.ndarray,
    species_id: str,
    data_unit: str,
    time_unit: str
) -> pe.EnzymeMLDocument:

    enzmldoc = pe.EnzymeMLDocument.fromTemplate(template_path)
    for IDs, concentration in zip(enzmldoc.measurement_dict.keys(),
measurement_data):
        for counter, replicate in enumerate(concentration):
            rep = pe.Replicate(
                id=f"Measurement{counter}",
                species_id=species_id,
                data=list(replicate),
                data_unit=data_unit,
                time=list(time),
                time_unit=time_unit)
            enzmldoc.getMeasurement(IDs).addReplicates(rep, enzmldoc)
    return enzmldoc

# Write experimental data to EnzymeML document vis parser function
enzmldoc = data_to_EnzymeML(
    template_path="../data/enzyme_inactivation/EnzymeML_CotA.xlsm",
    measurement_data=concentration_data,
    species_id="s1",
    data_unit="mmole / l",
    time_unit="min")

# Visualize experimental data
fig, axes = plt.subplots(1,2, figsize=(12.8, 4.8), sharey=False, sharex=False)
for measurement in enzmldoc.measurement_dict.values():
    concentration = []
    product = measurement.getReactant("s1")
    init_substrate=measurement.getReactant("s0").init_conc
    for replicate in product.replicates:
        concentration.append(replicate.data)
    axes[0].errorbar(time, np.mean(concentration, axis=0), np.std(concentration,
axis=0), label=init_substrate,\
        fmt=".", alpha=0.5)
    axes[1].errorbar(time, np.mean(concentration, axis=0), np.std(concentration,
axis=0), label=init_substrate,\
        fmt="o")

axes[0].legend(title="Initial ABTS [mM]")
axes[1].legend(title="Initial ABTS [mM]")
axes[0].set_ylabel("ABTS radical [mM]")
axes[0].set_title("ABTS radical concentration over CotA reaction time-course ")
axes[1].set_title("ABTS radical concentration within the first 5 min")
axes[1].set_ylabel("ABTS radical [mM]")
axes[0].set_xlabel("time [min]")
axes[1].set_xlabel("time [min]")
axes[1].set_xlim([-0.2,5.2])
axes[1].set_ylim([0,0.01])
axes[0].text(-0.1, 1.1, string.ascii_uppercase[0], transform=axes[0].transAxes,
        size=20, weight='bold')
axes[1].text(1.1, 1.1, string.ascii_uppercase[1], transform=axes[0].transAxes,
        size=20, weight='bold')

fig.show()

```

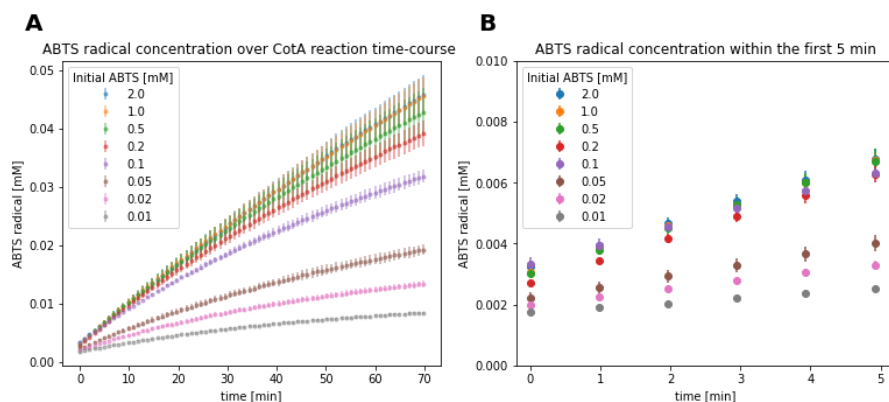


Fig 14. ABTS radical concentration over the time-course of CotA reaction.

The provided reaction data showed large deviations between individual repeats of reactions with identical initial ABTS concentrations. Especially for initial ABTS concentrations above 0.1 mM (Fig. 14A). Furthermore, visual analysis of the initial measurement points indicated that reactions with an initial substrate concentration of 0.1 mM had an increased enzyme concentration compared to others. This is evident, because the mentioned reactions show the highest concentration at the initial measurement time (Fig. 14B). The order of initial measurement points is expected to reflect the initial substrate concentrations, since the substrate slightly absorbs at the product detection wavelength. The substrate of the mentioned reaction was likely applied correctly, since the final concentration at 70 min is between reactions with the next lower and next higher initial substrate concentration.

Kinetic parameter estimation

Quality control through modeling

An initial parameter estimation considering enzyme inactivation was conducted in order to check further systematic deviations. As a result, reactions with an initial ABTS concentration of 0.05 mM showed to deviate from all other reactions, since the respective product concentrations is lower than the model predicted. The mentioned measurements were the only ones where the measured concentration was distinctly and systematically below that of the model (Fig. 15A). Hence indicating that either less substrate or enzyme than defined in the protocol was applied to the respective reactions. In consequence, data from reactions with an initial substrate concentration of 0.05 and 0.1 mM were excluded from parameter estimation, because the identified systematic deviations in reaction conditions would distort the results.

```
# Parameter estimation considering time-dependent enzyme inactivation
CotA_kinetics_first_run = ParameterEstimator.from_EnzymeML(enzmldoc, "s1",
"product")
CotA_kinetics_first_run.fit_models(enzyme_inactivation=True, display_output=False,
initial_substrate_concs=[0.01, 0.02, 0.05, 0.1, 0.2, 0.5, 1.0, 2.0])

# Parameter estimation considering time-dependent enzyme inactivation
CotA_kinetics_with_inactivation = ParameterEstimator.from_EnzymeML(enzmldoc, "s1",
"product")
CotA_kinetics_with_inactivation.fit_models(enzyme_inactivation=True,
display_output=False, initial_substrate_concs=[0.01, 0.02, 0.2, 0.5, 1.0, 2.0])
df_inactivation = CotA_kinetics_with_inactivation.result_dict.drop(columns=["kcat
/ Km [1/min * 1/mMole / l]"])
df_inactivation.insert(1, "Enzyme inactivation model", "True")

# Visualize fit of the models
fig, axes = plt.subplots(1,2, figsize=(12.8,4.8), sharey=True, sharex=True)
CotA_kinetics_first_run.visualize("irreversible Michaelis Menten",ax=axes[0],
alpha =.2,\
title="Parameter estimation based on full dataset")
CotA_kinetics_with_inactivation.visualize("irreversible Michaelis
Menten",ax=axes[1], alpha =.2,\
title="Parameter estimation based on data subset")

axes[0].set_ylabel("ABTS radical [mM]")
axes[0].set_xlabel("time [min]")
axes[1].set_xlabel("time [min]")
axes[0].legend(title="initial ABTS [mM]")
axes[1].legend(title="initial ABTS [mM]")
axes[0].text(-0.1, 1.1, string.ascii_uppercase[0], transform=axes[0].transAxes,
size=20, weight='bold')
axes[1].text(1.1, 1.1, string.ascii_uppercase[1], transform=axes[0].transAxes,
size=20, weight='bold')
plt.tight_layout()
```

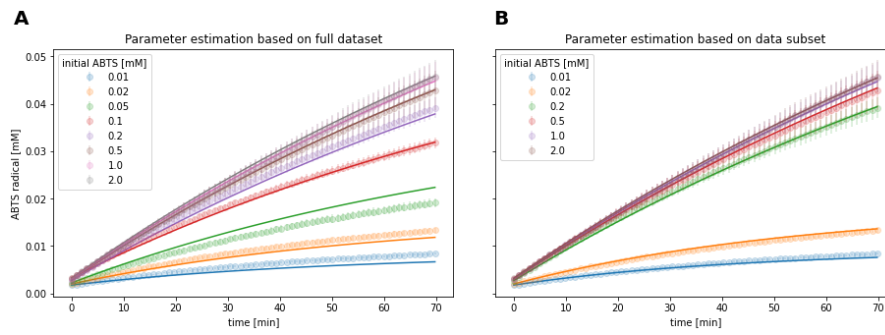


Fig 15: Identification of systematic deviations through kinetic modeling.

By excluding the mentioned measurements with systematic issues, the model fit was improved (Fig 15B). Additionally, the correlation between parameter estimates for k_{cat} and K_m was reduced from $r^2 = 0.518$ to 0.204.

Parameter estimation with and without considering enzyme inactivation

```
# Parameter estimation without time-dependent enzyme inactivation
CotA_kinetics = ParameterEstimator.from_EnzymeML(enzmldoc, "s1", "product")
CotA_kinetics.fit_models(enzyme_inactivation=False, display_output=False,
initial_substrate_concs=[0.01, 0.02, 0.2, 0.5, 1.0, 2.0])
df = CotA_kinetics.result_dict.drop(columns=["kcat / Km [1/min * 1/mmol / l]"])
df.insert(1, "Enzyme inactivation model", "False")

results = df.append(df_inactivation).sort_values("AIC")
display(results.style.set_table_attributes('style="font-size: 12px"'))
```

	AIC	Enzyme inactivation model	kcat [1/min]	Km [mmole / l]	Ki competitive [mmole / l]	Ki uncompetitive [mmole / l]	ki time-dep enzyme-inactiv. [1/min]
irreversible Michaelis Menten	-22202	True	3.003 +/- 0.77%	0.030 +/- 1.99%	-	-	0.007 +/- 4.48%
competitive product inhibition	-22202	True	2.999 +/- 0.78%	0.029 +/- 3.85%	0.221 +/- 92.18%	-	0.007 +/- 5.06%
substrate inhibition	-22202	True	3.021 +/- 0.86%	0.031 +/- 2.38%	-	208.577 +/- 67.12%	0.007 +/- 4.48%
uncompetitive product inhibition	-22199	True	3.012 +/- 1.23%	0.030 +/- 4.72%	-	2.096 +/- 361.49%	0.006 +/- 15.30%
non-competitive product inhibition	-22198	True	3.012 +/- 2.44%	0.029 +/- 6.56%	0.167 +/- 132.63%	1.113 +/- 466.95%	0.006 +/- 40.69%
non-competitive product inhibition	-22185	False	3.241 +/- 1.40%	0.033 +/- 5.05%	0.038 +/- 22.41%	0.073 +/- 6.86%	nan
uncompetitive product inhibition	-22156	False	3.268 +/- 1.43%	0.041 +/- 2.70%	-	0.066 +/- 6.35%	nan
competitive product inhibition	-21840	False	2.586 +/- 0.35%	0.020 +/- 6.10%	0.016 +/- 15.88%	-	nan
irreversible Michaelis Menten	-21739	False	2.529 +/- 0.27%	0.030 +/- 2.30%	-	-	nan
substrate inhibition	-21737	False	2.537 +/- 0.52%	0.030 +/- 2.71%	-	352.069 +/- 134.63%	nan

""#TODO AIC no more informative value goodness of fit is attributed to additional paramteres -> additional variable is uninformative [Arnold, 2010]

Kinetic parameters were estimated with and without considering time-dependent enzyme inactivation. Modeling results are listed in the table above. All models with an additional parameter for enzyme inactivation resulted in lower AIC values, indicating a better fit of the experimental data to inactivation models. When comparing models, an additional parameter in a model is informative, if the difference in AIC between the two models is larger than 2. Accordingly two additional parameters in a model compared to another model are informative, if the the difference in AIC is larger than 4 [Arnold, 2010]. Thus, the additional parameters in inhibition models considering enzyme activation were uninformative and did not enhance the fit quality. Additionally, the mentioned models showed high standard deviations above 80% for

the respective K_i estimates. Without considering enzyme inactivation, product inhibition models were favoured over the irreversible Michaelis-Menten model, whereas irreversible Michaelis-Menten with enzyme inactivation showed the best fit statistic compared to all other models (Fig XXXA). Thereby, k_{cat} was estimated to $3.003 \text{ min}^{-1} \pm 0.77\%$ whereas $30 \text{ } \mu\text{M} \pm 1.99\%$ was estimated for K_m . The respective half life of CotA was estimated at $95 \text{ min} \pm 4 \text{ min}$ using equation (16), whereas k_{cat} and k_{inact} were highly correlated ($\text{corr} > 0.95$). Irreversible Michaelis-Menten model without enzyme concentration did not fit the measurement data, since the reaction rate is initially underestimated, and overestimated in later stages of the reaction (Fig xxxB).

```
fig, axes = plt.subplots(2,2, figsize=(12.8,9.2), sharey=True, sharex=True)
CotA_kinetics_with_inactivation.visualize("irreversible Michaelis
Menten",ax=axes[0][0], alpha =.2,\
    title="irrev. Michaelis-Menten model with time-dependent enzyme inactivation")
CotA_kinetics.visualize("irreversible Michaelis Menten",ax=axes[0][1], alpha =.2,\
    title="irrev. Michaelis-Menten model")
CotA_kinetics.visualize("non-competitive product inhibition",ax=axes[1][0], alpha
=.2,\
    title="non-competitive product inhibition model")
CotA_kinetics.visualize("uncompetitive product inhibition",ax=axes[1][1], alpha
=.2,\
    title="uncompetitive product inhibition model")

axes[0][0].set_ylabel("ABTS radical [mM]")
axes[1][0].set_ylabel("ABTS radical [mM]")
axes[1][0].set_xlabel("time [min]")
axes[1][1].set_xlabel("time [min]")
[ax.legend(title="initial ABTS [mM]") for ax in axes.flatten()]

[ax.text(0, 1.1, string.ascii_uppercase[i], transform=ax.transAxes, size=20,
weight='bold') for i, ax in enumerate(axes.flatten())]

# axes[0]
# axes[1].text(1.1, 1.1, string.ascii_uppercase[1], transform=axes[0].transAxes,
# size=20, weight='bold')
plt.tight_layout()
```

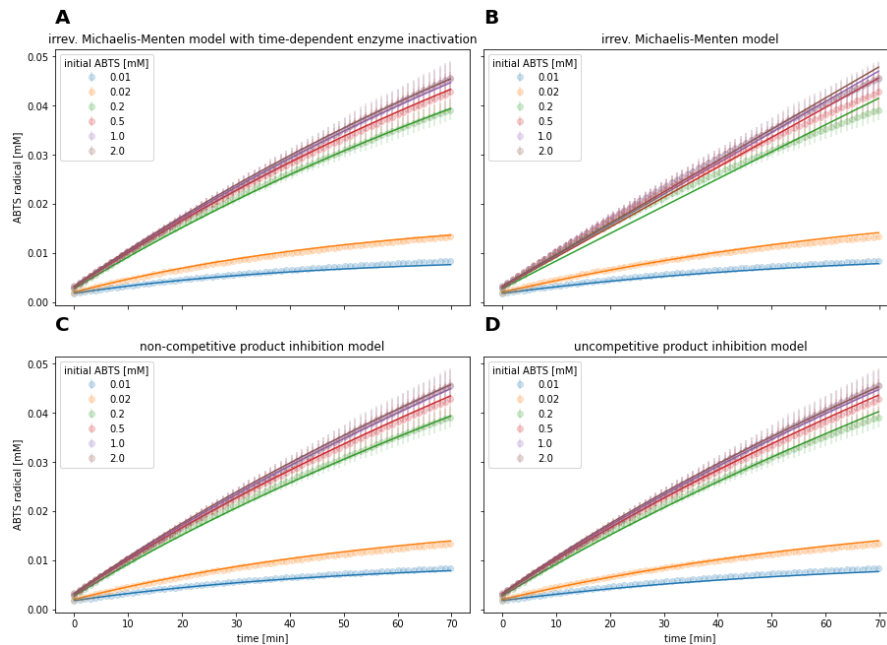


Fig. 16: Different kinetic models fitted to CotA reaction data

Out of the models without enzyme inactivation, product inhibition models showed the lowest AIC. Qualitatively, the inhibition models described the progress curve of the reaction (Fig. 16C and 16D) similarly to the best fitting model (Fig. 16A)

Project discussion

Data quality

Both datasets, for the absorption experiment and the enzyme kinetics experiment, showed quality issues. In the adsorption experiment the initial measurement showed large uncertainties as well as mixing issues, indicated by an apparent fluctuation in reaction rate. In the enzyme kinetics experiment, individual reactions showed to have from the assay protocol deviating substrate and enzyme concentrations. Furthermore, the

measured product concentration showed large deviations among experimental repeats, indicating unequal enzyme concentrations. Hence, larger pipetting volume for the enzyme solution was suggested to reduce pipetting errors due to avoidable small pipetting volumes. Furthermore, the product partners were advised to repeat the problematic experiments.

Enzyme inactivation

The observed time-dependent decrease in enzyme activity can be explained by enzyme inactivation and product inhibition. In case of time-dependent enzyme inactivation, the catalytic activity decreases, since the catalyst becomes inactive. Thereby, the decrease in relative reaction rate is independent from product concentration. In case of product inhibition, the formed product reduces the reaction rate. Hence, the observed enzyme inactivation is dependent on product concentration. Since both phenomena express themselves in an apparent decrease of reaction rate, they bare the potential to be confused, as the results of the enzyme kinetics experiment show. Visually, product inhibition models without enzyme inactivation and irreversible Michaelis-Menten model with enzyme inactivation are able to qualitatively describe the observed progress curve (Fig XXXc, XXXd).

Both independent and methodologically different experiments resulted in an comparable time period, in which the enzyme's activity is halved. Whereas in the inactivation experiment the enzyme's activity was halved after approximately 105 min, half the activity was left in the enzyme kinetics experiment after 95 min. Due to the design of the incubation experiment, the loss of enzyme activity showed to be independent from the catalysis itself. In consequence, product inhibition is not a conclusive explanation for the observed behavior. Product inhibition is further disfavored based on the modeling results, since the AIC of all inhibition models was higher compared the irreversible Michaelis-Menten model with time-dependent enzyme inactivation.

In conclusion, the observed inactivation is likely caused by hydrophobic interaction between the reaction vessel and the enzyme. Interactions between hydrophobic surfaces and enzymes leading to decreased enzymatic activity were already reported for ribonuclease A [Wei et al., 2014]. In this work, enzyme inactivation was observed across multiple enzyme classes, whereas all reactions were carried out in MTPs. Therefore, MTPs might cause enzyme inactivation for a broad range of different enzymes.

Misinterpretation of the apparent decrease in enzymatic activity might lead to wrong conclusions and therefore negatively impact the research process. As already mentioned the activity decrease might be wrongfully attributed to product inhibition. This attribution might lead to wrong design decisions if the enzyme in question should be applied to an enzymatic cascade reaction or make the enzyme entirely unfavorable for this particular application.

Furthermore, screening applications might be impacted by enzyme inactivation. If yields of an enzymatic reactions should be determined over the time course of multiple hours. All determined yield might be estimated too low. Therefore, enzyme inactivation in MTPs should be investigated in future experiments and the results should be reported on.

Save modeling results

Lastly, the modeling results were written to the EnzymeML documents and the files were exported.


```

# Write modeling results to kinetic parameters
k_cat = pe.enzymeml.models.KineticParameter(
    name="k_cat",
    value=CotA_kinetics_with_inactivation.get_model_results()["k_cat"].value,
    unit=f"1 / {CotA_kinetics_with_inactivation.data.time_unit}")

K_m = pe.enzymeml.models.KineticParameter(
    name="K_m",
    value=CotA_kinetics_with_inactivation.get_model_results()["Km"].value,
    unit=CotA_kinetics_with_inactivation.data.data_conc_unit)

# Define kinetic model
model = pe.KineticModel(
    name="irreversible Michaelis-Menten",
    equation="-k_cat * p0 * s0 / (K_m + s0)",
    parameters=[k_cat, K_m])

enzmldoc.getReaction("r0").model = model

add_enzyme_inactivation_model(
    result=CotA_kinetics_with_inactivation,
    enzmldoc=enzmldoc)

# Export EnzymeML documents
export = False
if export:
    enzmldocToFile()

```

Discussion

Availability of raw data is the prerequisite to reproduce results of an data analysis process and additionally reinforces confidence in the validity of scientific findings. Hence, the analysis workflow was conceptualized starting from photometric measurement data. For that purpose, discussing the output format of the analytical device with the experimental partners proved to be valuable to plan the data acquisition process. Hence, a project specific parser function was established, which transferred the measurement data to the respective EnzymeML containing information on the measurement conditions. Thereby, error-prone and tedious manual copying between files was avoided, ensuring raw data integrity from the very start of the workflow.

Data integrity through accurate concentration calculation

Another aspect of data integrity is the accurate concentration calculation based on the measured absorption signal. The relation between concentration to absorption signal of an analyte can be linear as well as non-linear. Therefore, wrong assumption of a linear relationship albeit the relation is better described non-linearly, leads to avoidable inaccuracy in concentration calculation [Hsu and Chen, 2010] [Martin et al., 2017] and thus imprecise kinetic parameter estimates. Hence, concentration calculation based on extinction coefficients was discouraged, if the underlying relation between absorption and concentration of an analyte was not self-determined for the respective reaction system. To enable precise concentration calculation of linear and non-linear absorption to concentration relationships, CaliPytion was applied with analyte standards provided by the experimental partners. All of the analyzed standard curves, except for scenario C ABTS at pH 5, were best described by non-linear calibration equations based on their fit quality determined by AIC. Hence, uncertainties from concentration calculation were reduced and therefore their impact on the kinetic parameters reduced.

Quality control of experimental data through kinetic modeling

Besides data integrity, high quality data is required for accurate parameter estimates, whereas unnoticed experimental errors unintentionally distort the resulting kinetic parameters. Therefore, quality control of measurement data is an important aspect of every data analysis process. EnzymePyntics allows visualization of measurement data together with the fitted kinetic model. Ultimately, systematic deviations of single measurements were identified by simultaneously visualizing measurement data and the fitted model. By this method, errors with pipetting, mixing, and temperature equilibration during experimental preparation as well as issues with the experimental design itself were identified in different scenarios. Systematic pipetting errors were disclosed by systematic deviations between the measurement data and the fitted model. In scenario D individual enzyme reactions showed to have deviating substrate concentrations from the assay protocol, whereas modeling without the presumably erroneous measurements resulted in a good fit between data and model as well as low standard deviation on the parameter estimates. Hence the experimental partners were advised to repeat the deviating measurement. In two projects the progress-curve of the reactions indicate a lag phase after reaction start. In one case,

small pipetting volumes of enzyme to start the reaction led to inhomogeneous mixing and thus increasing reaction rates until the enzyme was distributed evenly. As a measure, the project partners were advised to increase the pipetting volume of the enzyme solution. In another case, temperature incubation effects in MTP caused by prolonged assay preparation times resulted in an initial lag phase. Due to small reaction volumes and low mass, MTPs have a low heat capacity and thus high susceptibility to temperature change. Hence, project partners were advised to pre-incubate the MTP within the photometer at reaction temperature for 5 min and then start the reaction by adding enzyme as quick as possible. In another project, modeling results of the estimated k_{cat} and K_m were used to iteratively improve the design of the enzyme assay. In addition, the correlation between the aforementioned parameters was used as a measure to ensure that the highest substrate concentration applied was sufficiently high and thus the kinetic parameters could be determined independently. Thereby, the appropriate enzyme concentration and substrate concentration range were identified through multiple round of lab experimental with subsequent kinetic modeling.

Assessing the data quality through modeling and enhancing assay designs proved to be a strength of the implemented progress curve method on which the parameter estimation of this workflow is based on. In contrast to the predominantly applied initial rates method [Tang and Leyh, 2010], the progress curve method offers intrinsic advantages with regard to methods reproducibility providing that continuous assay data exist. In initial rate kinetics no consensus on the linear reaction period on which the kinetic parameters are estimated exists. Hence, the linear period is manually determined or only assumed. Ultimately, the resulting parameters are influenced by the choice of linear period. For progress curve analysis the entire dataset of an enzyme assay is used. If the fit statistics reveal that the data is not in accordance with the model, either the assay should be repeated due to an underlying issue or the model should be questioned. Contrarily, initial rates method can always be applied by arbitrarily choosing any linear subset or only the initial two data points of an time course data set. In discussions with project partners this showed to be a common practice although scientifically at least questionable, depending on the circumstances.

fggf #TODO Importance of precise kinetic parameters for biocatalysis With upcoming big data technologies like

Cumulative error, which ultimately leads to unreproducible kinetic parameters

- initial rates vs progress-curve analysis

FAIRness of the parameter estimation workflow

fair data management comes along with fair science

An secondary necessity for reproducibility is documentation. Thereby, both laboratory procedures as well as of data analytic procedures need to be documented sufficiently to make experimental

Furthermore, progress curve analysis is more transparent about the underlying data compared

sharing the raw data and comprehensive description of individual data treatment steps,

Findable

Accessible

Interoperable

Reusable

- Packages are reusable
- Precision of parameters
- sophisticated methods, otherwise wrong conclusions (italian group)

Jupyter Notebook and Book

contemporary

Besides the data analysis, this entire thesis was conceptualized adhering to FAIR data principles . In consequence, this thesis was written as a Jupyter Book, which allow to combine multiple Jupyter Notebooks with text chapters in a structured document. Ultimately, this thesis is findable and accessible on

GitHub, interoperable through Binder and the Jupyter Notebook format itself. Hence, making the work of this thesis reusable in a state of the art manner.

Printed version is read twice, digital version unlimited hence optimized for digital experience

- Besides the methodology documentation is bad
- Need for documentation without limiting experimental possibilities -> Jupyter notebooks
- great documentation (FAIR)

In this project, a robust and FAIR workflow from analytical raw-data to kinetic parameters was established. Thereby, the output of a analytical device was used directly as the input of the modeling pipeline.

Methods reproducibility of the parameter estimation workflow

Methods reproducibility is given

- Lab vs computer documentation
- Strenda guidelines are insufficient?? focused on lab, neglecting data analytic aspect

Results reproducibility of the parameter estimation workflow

In sum minimizing errors copying method, manual handling errors and (calculation)

opportunities:

- how precise should parameters be
- randomized order in MTPs

Conclusion

In combination with jupyternotebook the developed workflow is fair. The EnzymeMLdocument itself is not

Analysis of the four scenarios proved the applicability and advantages of the developed workflow.

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