

1) Fermi Functions

a) equilibrium and $T > 0\text{K}$ @ fermi level

$$\hookrightarrow E = E_F$$

$$\therefore \frac{1}{1 + e^{(E - E_F)/kT}} = \frac{1}{2} \quad \left. \begin{array}{l} \text{the probability of an electronic state} \\ \text{being occupied is } 50\%. \end{array} \right\}$$

b) $E_F = E_C ; E = E_C + kT$

$$\hookrightarrow E - E_F \Rightarrow (E_C + kT) - E_C = kT \quad \left. \begin{array}{l} \frac{1}{1 + e^{kT/kT}} = \frac{1}{e} \end{array} \right\}$$

c) $f(E_C + kT) = 1 - f(E_C + kT) \therefore f(E_C + kT) = 0.5 = 50\%$

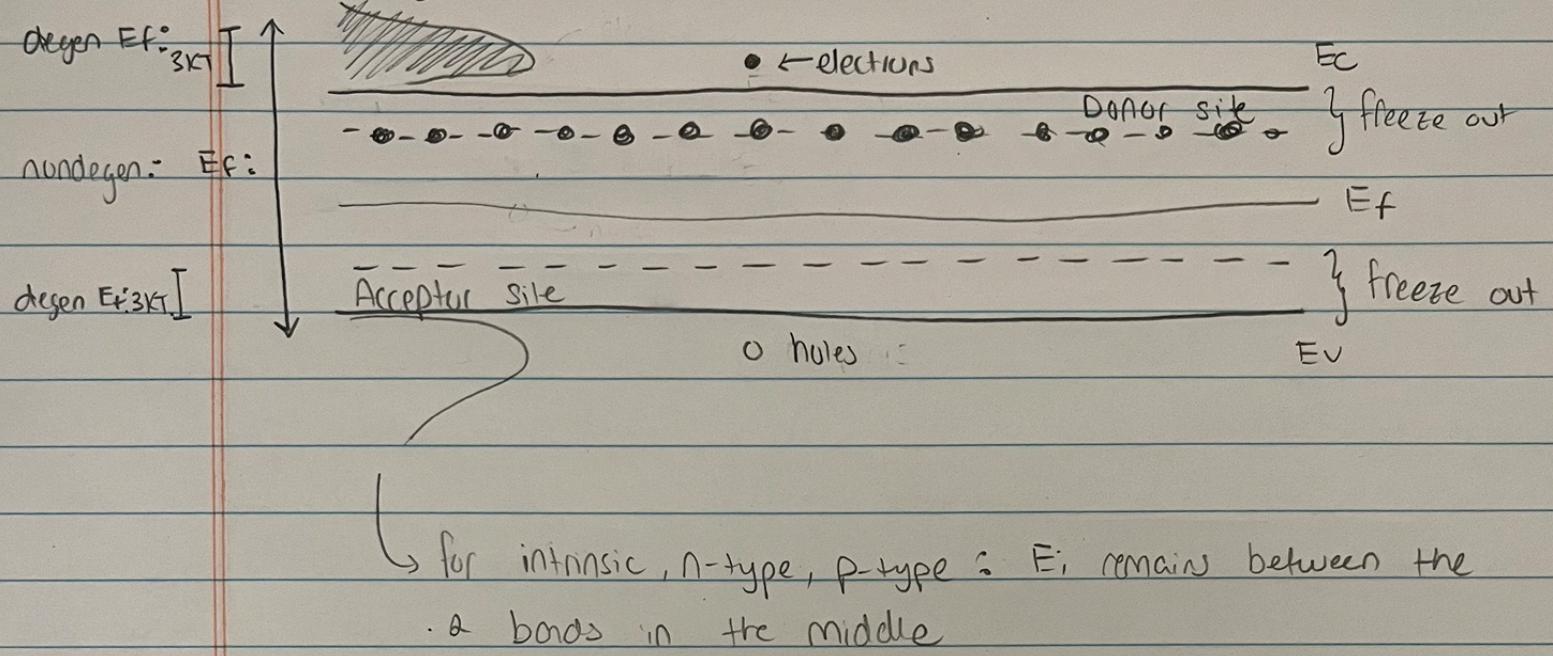
\hookrightarrow the fermi level is located at $E_C + kT$

d) $E = E_F - kT$

$$\hookrightarrow \text{hole probability} : 1 - \left(\frac{1}{1 + e^{(E - E_F)/kT}} \right)$$

$$\therefore 1 - \left(\frac{1}{1 + e^{-1}} \right) = 0.269$$

2) Energy Bond Model - Pierret Problem 2.3



E_F : for n-type is closer to E_C

p-type is closer to E_V

3) Pierret 2.7 : Prove the peak to be at $E_C + \frac{kT}{2}$ & $E_V - \frac{kT}{2}$

$$\hookrightarrow \text{Carrier distribution: } \int_{E_C}^{E_{\text{top}}} g_c(E) f(E) dE \Rightarrow g_c(E) f(E)$$

$$\text{Where } g_c(E) = \frac{m^* n \sqrt{2m^*} n}{\pi^2 \hbar^3} \cdot \frac{1}{1 + e^{\frac{E - E_F}{kT}}} = \frac{M^* n \sqrt{2m^*}}{\pi^2 \hbar^3} \sqrt{E - E_C} e^{-\frac{(E - E_F)}{kT}}$$

$$\frac{d}{dE} (g_c(E) f(E)) = 0 \text{ to get where the peak will occur}$$

$$= \left(\frac{m^* n \sqrt{2m^*}}{\pi^2 \hbar^3} \right) e^{-\frac{(E - E_F)}{kT}} - \left(\frac{m^* n \sqrt{2m^*}}{\pi^2 \hbar^3} \right) \sqrt{E - E_C} e^{-\frac{(E - E_F)}{kT}} = 0$$

\hookrightarrow Solving this gives the peak to be at $E = E_C + \frac{kT}{2}$

• Pierret → Problem 2.16

a) Uniformly doped p type $\rightarrow N_A = 10^{15}/\text{cm}^3$; At $T \approx 0\text{K}$

↳ equilibrium hole and electron concentration? w/ $T \approx 0\text{K}$, $n \approx p$ will approach 0

b) $N > N_i$ & all impurities are ionized; $n \approx N$ & $p \approx N_i^2/N$

↳ Since $N > N_i$ & $n \approx N$, we can say the impurity is a donor

c) Si @ $T \approx 300\text{K}$; $n = 10^5/\text{cm}^3$

$$\hookrightarrow n_p = N_i^2 \text{ & } n = 10^5/\text{cm}^3 \rightarrow p = N_i^2 / N_D = (10^{10}/\text{cm}^3)^2 / 10^5/\text{cm}^3 = 10^{13}/\text{cm}^3$$

↳ non-degenerate assumption

c) $N_i = 10^{13}/\text{cm}^3$, $n = 2p$, $N_A = 0$; $n + N_D = ?$

$$\hookrightarrow n_p = N_i^2 \text{ where } n = 2p \therefore N_i^2 = \frac{n^2}{2} \Rightarrow n = \sqrt{2} N_i$$

$$\left. \begin{array}{l} \text{charge neutrality: } p - n + N_D - N_A = 0 \\ \frac{n}{2} - n + N_D - 0 = 0 \end{array} \right\} N_D = \frac{n}{2} = \sqrt{2} N_i = 0.707 \times 10^{13}/\text{cm}^3$$

• Pierret → Problem 2.17

c) $T = 300\text{K}$, $N_A = 9 \times 10^{15}/\text{cm}^3$, $N_D = 10^{16}/\text{cm}^3$

↳ (2.28): $N_i >> |N_D - N_A| \therefore n \approx p \approx N_i \rightarrow n \approx N_i = N_D - N_A = 10^{15}/\text{cm}^3$

$$\hookrightarrow p = N_i^2 / (N_D - N_A) \approx (10^{10})^2 / 10^5 = 10^5/\text{cm}^3$$

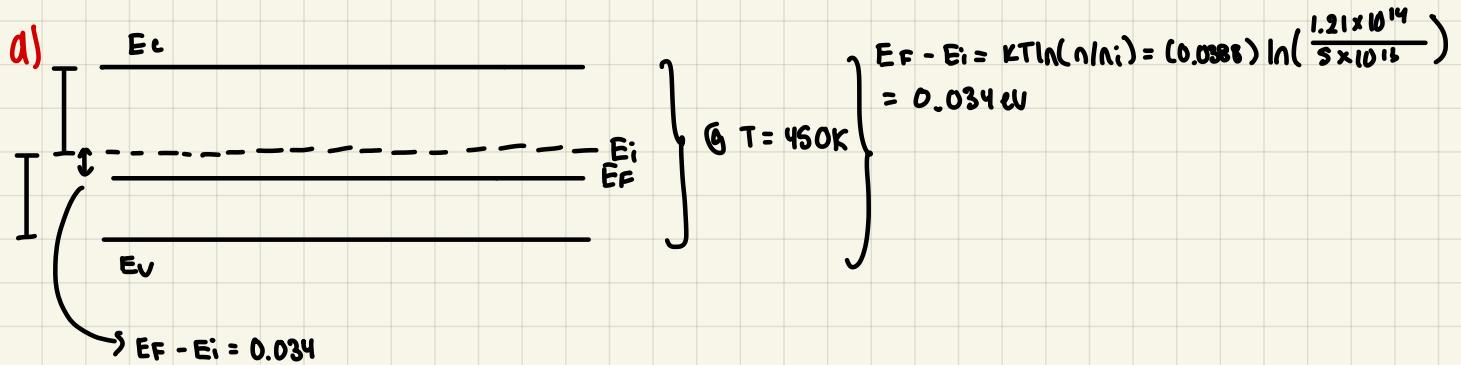
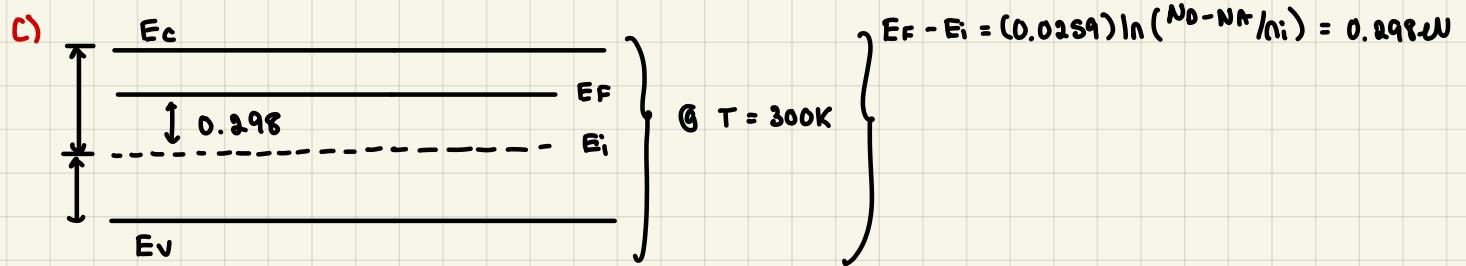
a) $T = 450\text{K}$; $N_A = 0$; $N_D = 10^{14}/\text{cm}^3$

$$\hookrightarrow n_{\text{Si}} \approx 5 \times 10^{13}/\text{cm}^3 \rightarrow n = \frac{N_D - N_A}{2} + \left[\left(\frac{N_D - N_A}{2} \right)^2 + N_i \right]^{1/2}$$

$$n = \frac{10^{14}/\text{cm}^3}{2} + \left[\left(\frac{10^{14}/\text{cm}^3}{2} \right)^2 + 5 \times 10^{13}/\text{cm}^3 \right]^{1/2} = 1.21 \times 10^{14}/\text{cm}^3$$

$$p = \frac{N_i^2}{n} \rightarrow \frac{(5 \times 10^{13}/\text{cm}^3)^2}{1.21 \times 10^{14}/\text{cm}^3} = 2.07 \times 10^{13}/\text{cm}^3$$

• Pierret → Problem 3.18



- **Activity 1**

1. The 2 photon energies that an electron in the ground state of a H atom can absorb are 12.1eV and 10.1eV
2. The shortest wavelength it can radiate is 99nm ~ 100nm

- **Activity 2**

1. Bonding energy → As the hydrogen atoms get closer to each other, the energy exponentially decreases

Proton - Proton Repulsion Energy → As the hydrogen atoms grow closer, the repulsion energy exponentially increases

2. The bond length is determined by the shared energy levels, the size of the atoms, and etc.

- **Activity 3**

1. 3.41×10^{20} energy states exist between 0.7eV and 0.72eV assuming the middle of the bandgap is at 0 energy
2. 4.43×10^{20} energy states exist between 0.8eV and 0.82eV assuming the middle of the bandgap is at 0 energy