$$\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$$

$$\frac{310 \times 10^{3} (0.025)}{273.15 + 25} = \frac{P_{2}(0.025)}{273.15 + 50}$$

P2 = 335993.6274Pa

$$P_2 - P_1 = 335993.6274 - 310 \times 10^3$$

= 25993.62737Pa
 $\approx 26kPa$

$$26 \times 10^{3} \left(0.025\right) = m\left(\frac{8.314}{29}\right) \left(273.15 + 50\right)$$

$$m = 7.0144.03018$$

$$m = 7.014403018$$

$$\approx 0.007 \text{kg}$$

2a) When the piston begins to move down,
$$P_{air} - P_{piston} - P_{atm} = 0$$

$$P_{air} = \frac{50 \times 9.81}{(0.1)^2 \pi} + 100 \times 10^3$$

$$\frac{P_1}{T_1} = \frac{P_2}{T_2}$$

$$\frac{250 + 10^3}{273.15 + 300} = \frac{162452.3997}{T_2}$$

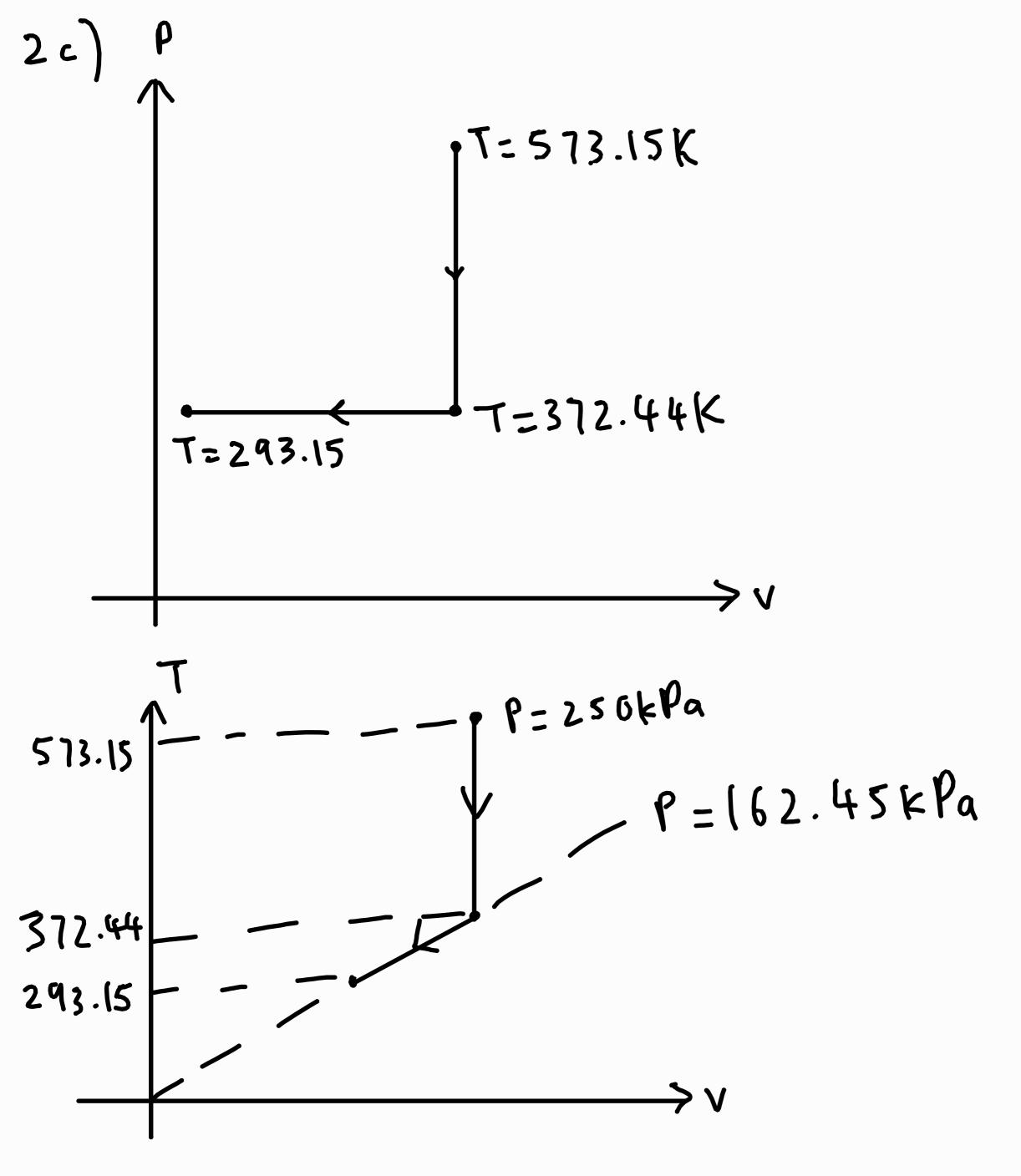
$$T_2 = 372.4383715K$$

 $2372.44K$

$$\frac{(0.1)^{2} \pi (25 \times 10^{-2})}{372.4383715} = \frac{(0.1)^{2} \pi h}{273.15 + 20}$$

h = 19.67775224cm

~ 5.3 cm



$$3a)$$
 Initial pressure
= $\frac{mR_mT}{V}$
= $0.1(\frac{8.3144}{28.97} \times 10^3)$ (273.15 + 35)

= 98265.7293Pa

Increase in pressure = 250×103 - 98265.7293 = 151734.2707Pa

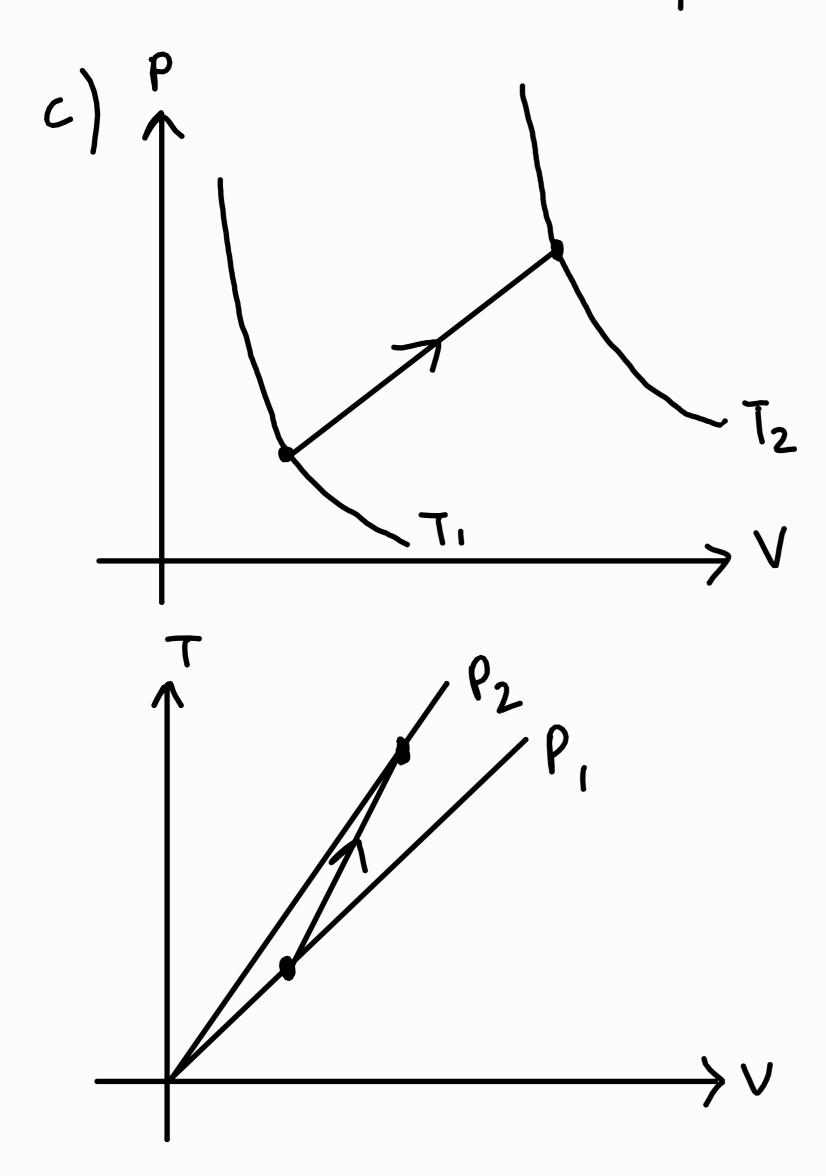
Distance travelled by the piston = $\frac{151734.2707 \times 0.05}{15 \times 10^3}$ = 0.5057809023

P, V, = P2 \frac{1}{7}

 $\frac{98265.7293(0.09)}{273.15+35} = \frac{250\times10^{3}(\frac{0.09}{0.05}+0.506)(0.05)}{T_{2}}$

T2 = 1004.4 K

3b) Distance travelled by the piston = 0.506m



3d)
$$W = \frac{98265.7293*250 \times 10^3}{2} (0.506)(0.05)$$

= 4405.561476
\$\times 4.4(k)

$$Q = M_{air} (7, -7_2) + W$$

$$= 0.1 (0.718) (1004.4 - 273.15-35)$$

$$+ 4.41$$

$$= 54.4 \text{ K}$$