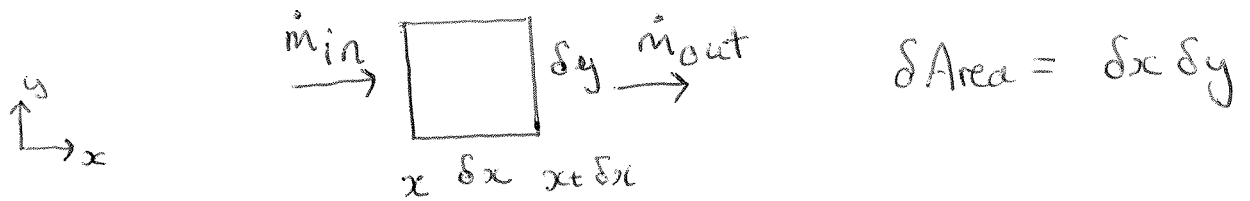


# Derivation of general continuity equation

①



$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out} \quad \leftarrow \text{Change in mass.}$$

Change in concentration:  $C = \text{mass / volume (area)}$

$$\delta C = \delta m / \delta A$$

$$\therefore \frac{\delta C \delta A}{\delta t} = \frac{\delta C \delta x \delta y}{\delta t} = \dot{m}_{in} - \dot{m}_{out}$$

$j$  = flux of mass = mass per unit length per unit time.

$$\frac{\delta C \delta x \delta y}{\delta t} = j(x) \cdot \underset{\substack{\downarrow \\ \text{length of surface}}}{\delta y} - j(x + \delta x) \cdot \delta y$$

$$= - (j(x + \delta x) - j(x)) \delta y$$

Divide by  $\delta x \delta y$

$$\frac{\delta C}{\delta t} = - \frac{(j(x + \delta x) - j(x))}{\delta x}$$

delta y cancels.

As  $\delta C, \delta x, \delta y, \delta t \rightarrow 0$

$$\Rightarrow \frac{\partial C}{\partial t} = - \frac{\partial j}{\partial x}$$

## Flux operators

(2)

$$\frac{\partial c}{\partial t} + \nabla \cdot \mathbf{j} = 0$$

general form where

$$\nabla = \left( \frac{\partial}{\partial x}, \frac{\partial}{\partial y}, \frac{\partial}{\partial z} \right)$$

(tutorial exercise)

$\mathbf{j}$  = total flux.

### Diffusive Flux.

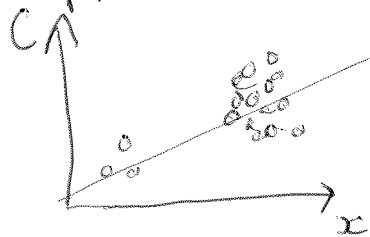
Fick's first law  $\rightarrow$  molecular diffusion.

$\rightarrow$  quantity moves in proportion to local gradient

$\rightarrow \uparrow$  the gradient,  $\uparrow$  the flux.

$$\mathbf{j}_{\text{diff}} = -D \nabla c$$

$\nabla$  negative because flux is in opposite direction to the gradient.



+ve gradient

-ve flux.

Advective flux :- flux due to flow of fluid.

$$\mathbf{j}_{\text{advective}} = \underline{\mathbf{v}} c$$

$$= \frac{L}{T} \cdot \frac{M}{L^3} = \frac{M}{L^2 T} = \text{mass} / (\text{area} \times \text{time})$$

Put together

(3)

$$\frac{\partial c}{\partial t} + \nabla \cdot (j_{\text{diff}} + j_{\text{advec}}) = 0$$

$$\frac{\partial c}{\partial t} + \nabla \cdot (-D \nabla c + \underline{v} c) = 0$$

Expanding

$$\frac{\partial c}{\partial t} + \nabla \cdot (\underline{v} c) = \nabla \cdot (D \nabla c)$$

Common simplifications.

Fluid is incompressible i.e.  $\nabla \cdot \underline{v} = 0$

Diffusion coefficient is a <sup>constant</sup> scalar  $\nabla D = 0$

$$\Rightarrow \frac{\partial c}{\partial t} + \underline{v} \cdot \nabla c + \underbrace{c \nabla \cdot \underline{v}}_{=0} = D \nabla^2 c + \underbrace{\nabla c \cdot \nabla D}_{=0}$$

$$\Rightarrow \frac{\partial c}{\partial t} + \underline{v} \cdot \nabla c = D \nabla^2 c$$

## Source / Sinks

(4)

Mass flux in/out not only way to increase mass.  
<sup>src</sup>Source / <sup>-ve</sup>Sinks also do this.

$$\frac{\partial c}{\partial t} + \nabla \cdot \mathbf{j} = S$$

### Mass source

Straight forward  $f = \left( \frac{\text{mass}}{\text{volume/time}} \right) = f(\underline{x}, t)$

e.g. modelling pollution from a chimney in environmental fluid mechanics.

### Reaction

Source proportional to value of  $c$ .

$$S = \lambda c$$

$\lambda > 0$  = source  $\rightarrow$  produces

$\lambda < 0$  = sink  $\rightarrow$  destroys.

Used often in chemical reaction models.

### Final Equation

$$\frac{\partial c}{\partial t} + \underline{v} \cdot \nabla c = D \nabla^2 c + \lambda c + f$$