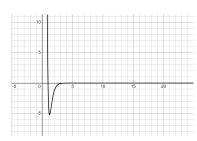
# Computer Modelling: Exercise 3

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#### Introduction 1



Exercise 3 asks to simulate 2 oscillating particles using the Morse potential (see equation 1) and the force on the particles  $\mathbf{F} = -\nabla U_M$ .

$$U_M(\mathbf{r_1}, \mathbf{r_2}) = D_e([1 - e^{-\alpha(r_{12} - r_e)}]^2 - 1)$$
 (1)

$$\mathbf{F_1}(\mathbf{r_1}, \mathbf{r_2}) = 2\alpha D_e [1 - e^{-\alpha(r_{12} - r_e)}]^2 e^{-\alpha(r_{12} - r_e)} \hat{\mathbf{r_{12}}}$$
 (2)

Where  $r_{12} = |\mathbf{r_2} - \mathbf{r_1}|$ ,  $\hat{\mathbf{r_{12}}} = \frac{\mathbf{r_{12}}}{r_{12}}$  and  $\mathbf{F_1} = -\mathbf{F_2}$  due to Newton's third law.  $\alpha$ ,  $D_e$  and  $r_e$  control the depth and curvature of the potential minimum, shown in figure 1 [1].

Figure 1: Desmos plot of Morse potential for  $O_2$  pa-

 $T = 1.02 \times 10^{-14} \ s.$ 

As the code simulates atoms, for the code to be sensible we use incredibly small units, like eV, A and amu. The rameters. time units in the code are not seconds but are shown in equation 3, giving a value of

$$T = \mathring{A}\sqrt{\frac{amu}{eV}} \tag{3}$$

In this code we are measuring the wavenumber of oscillation of the particles and the energy variation of the simulation. Wavenumber is related to period by  $\bar{v} = \frac{v}{c} = \frac{1}{Tc}$ similar to  $k = \frac{2\pi}{Tc}$ . Energy variation increases with time-step (seen in figure 2b) and oscillates along with position.

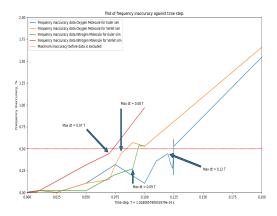
#### Calculating Maximum Time-Step 2

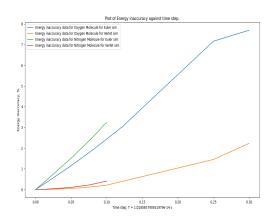
Table 1 shows the maximum time-step for each simulation. These values are also shown in figure 2. The values were calculated by comparing the frequency of each simulation to the same simulation at a time step,  $dt = 10^{-5}T$ . Equation 4 shows the calculation for the frequency inaccuracy, where  $v_0$  is the frequency at  $dt = 10^{-5}T$ .

$$\frac{\Delta v}{v_0} \equiv \frac{\mid v - v_0 \mid}{v_0} \tag{4}$$

Table 1: Table showing maximum time-steps for each simulation method and molecule simulated. Units given in Figure 2.

Simulation	Molecule	Maximum Time-step (T)
Symplectic Euler	Oxygen	0.12
Symplectic Euler	Nitrogen	0.09
Velocity Verlet	Oxygen	0.08
Velocity Verlet	Nitrogen	0.07





(a) Plot of frequency inaccuracy against time-step (b) Plot of energy inaccuracy against time-step for  $O_2$ .

Figure 2: Plots showing inaccuracy of frequency and energy as time-step increases for the Euler and Verlet simulation of  $O_2$  and  $N_2$ .

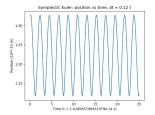
### 3 Results

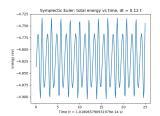
Table 2 shows the wavenumber measurements for all simulations with the maximum time-step for each simulation. These values are an underestimate on the experimental values. The spinned simulations (2D velocity) show this significantly more than the 1D oscillations. Other than that both simulation methods (Euler and Verlet) agree with each other so a place where the simulations go wrong could be initialising the particles or calculations of the force. The code is a good simulator for the Morse potential between 2 particles. Symplectic Euler simulation plots can be seen in figure 3, Verlet simulations are in figure 4. The code measures the position as the distance between particle 1 and particle 2. Applying a spin to both particles produces rotational frequency on top of the vibrational frequency, complicating the period analysis, thus giving an inaccurate wavenumber measurement. The code could be improved to find the distance between each particle and a common center.

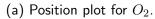
Velocity Verlet integration method is much better at keeping energy constant which can be seen in figure 2b as the Verlet lines have a smaller gradient than the Euler lines. However, in the way that Verlet is better at keeping energy constant, the Euler method produces more accurate frequencies, as shown in table 1 and figure 2a, as Euler simulations have a higher maximum time-step and generally lower frequency inaccuracies than Verlet simulations.

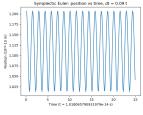
Table 2: Table of wavenumbers for all simulations Including experimental values for  $\mathcal{O}_2$  and  $\mathcal{N}_2$  from [1].

Simulation	Wavenumber Measured $cm^{-1}$
1D Symplectic Euler $O_2$	1532
2D Symplectic Euler $O_2$	1386
1D Velocity Verlet $O_2$	1532
2D Velocity Verlet $O_2$	1384
Experimental $O_2$	1580
1D Symplectic Euler $N_2$	2293
2D Symplectic Euler $N_2$	2198
1D Velocity Verlet $N_2$	2297
2D Velocity Verlet $N_2$	2201
Experimental $N_2$	2359

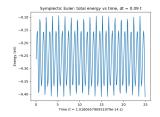








(b) Energy plot for  $O_2$ .



(c) Position plot for  $N_2$ .

(d) Energy plot for  $N_2$ .

Figure 3: Plots of Symplectic Euler simulation.

# References

1. Zuntz, J. Computer Modelling Exercise 3 https://www.learn.ed.ac.uk/ultra/courses/\_111189\_1/outline/file/\_9269517\_1 (Nov. 7, 2023).

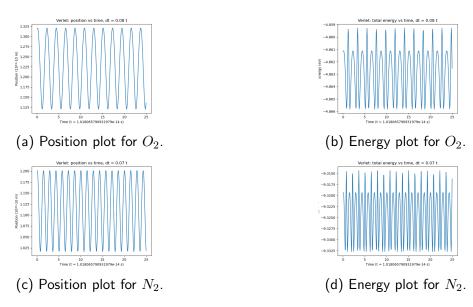


Figure 4: Plots of Velocity Verlet simulation.