MCEN 3030 Computational Project 3 Due March 4<sup>th</sup> 2016 at start of class

Be sure to comment your code thoroughly and only submit one m file named cp3\_MEID (no dash) to D2L. Also publish your code and turn in a hard copy.

## **Newton's Law of Cooling**

The file cooling.txt contains 4 temperature measurements (°C) for each of the 6796 times (s) provided. These data describe the temperature of a glass of water as a function of time. You'll have to rearrange these data to make them useful. Try

$$T = [data(:,2); data(:,3); data(:,4); data(:,5)];$$

and something similar to create a single corresponding time vector of equal size. Plot temperature vs. time to check if it looks reasonable. Be sure to plot data points (not a line). The size of the data points can be reduced via

From our knowledge of heat transfer, we can expect the rate of heat loss (W) to be

$$q = -hA(T - T_{\infty})$$

where h is the convective heat transfer coef. (W/ m<sup>2</sup>K), A is the surface area, T = T(t) is the temperature of the water, and  $T_{\infty} = 26.5^{\circ}C$  is the temperature of the surroundings. The rate of change of T with time is related to the rate of heat loss via the specific heat capacity (J/kgK) and mass (kg) of water,

$$q = mc \frac{dT}{dt}$$
.

By integrating the above equations, you will be able to determine a model for T as a function of time with fitting parameters  $T_0$  (initial temperature) and  $\tau = \frac{hA}{mc}$  (thermal time constant). In this analysis, we will invoke lumped capacitance (our glass + water system has a homogenous T at

analysis, we will invoke lumped capacitance (our glass + water system has a homogenous I at any given time) and neglect the heat capacity of the glass. We will also neglect latent effects at the water's surface, which will introduce discrepancies at early times.

- 1. From the model you obtain, predict  $T_0$  and h based on a regression of the data. Do not use built-in functions. The cylindrical glass is 16 cm high and 7 cm wide.
- 2. Plot the model prediction out to 5 hours on top of the actual data and predict the system temperature at 10,000s.
- 3. Now use a Lagrange interpolating polynomial based on the first 25 data points to predict the system temperature at 1.5 seconds. Test that your prediction is working by checking that it accurately predicts the provided *T* at 1 and 2 seconds (since the polynomial is defined to pass through integer time values exactly). Explain. Would using Newton's polynomial improve the estimate? Spline interpolation?