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Chemical Thermodynamics Example 9.2 The element mercury, Hg, is a silvery liquid at room temperature. The normal freezing point of mercury is -38.9°C , and its molar enthalpy of fusion is $\Delta H_{\text{fusion}} = 2.29 \text{ kJ/mol}$. What is the entropy change of the system when 50.0 g of Hg(l) freezes at the normal freezing

Chapter 19 Chemical Thermodynamics

The ceremony of the "2019 Excellence Award in Thermodynamics and Transport Properties" of the EFCE (European Federation of Chemical Engineering)) will be held at the Thermodynamics'2019 in Punta Umbría (Huelva). Call for nominations for the 2019 Excellent Award in now open.

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10 INTRODUCTION TO CHEMICAL ENGINEERING THERMODYNAMICS INTRODUCTION where m is the mass of the piston, pan, and weights, g is the local acceleration of gravity, and A is the cross-sectional area of the piston. Gauges in common use, such as Bourdon gauges, are calibrated by comparison with dead-weight gauges.

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