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FIGURE 3.6 Anodic and cathodic half-cell reactions present simultaneously on a. The current density (rate) of reduction must exceed the critical current density for passivation to ensure low corrosion rate in the passive state. 2. 4 / Passivity active state as the one most likely in cases of borderline passivity.

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Article: Principles and prevention of crevice corrosion

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CORROSION PRINCIPLES. Corrosion is defined as the damage or deterioration of a material (usually a metal) due to a reaction or interaction with the environment. The basic cause of corrosion is the instability of metal in its refined form. The process of corrosion is the tendency of a metal to revert to its natural state.

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