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Acids, Bases, and Solutions ANSWER KEY Acids, Bases, and Solutions Describing Acids and Bases Review and Reinforce 1. Sour 2. Bitter 3. Corrosive to magnesium, zinc, and iron; eats them away and produces bubbles of hydrogen gas 4. Doesn't react with metals 5. Produces carbon dioxide 6. Doesn't react with carbonates 7. Red 8. Blue 9.

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Solutions, Acids, and Bases Unit Test Review. STUDY. ... Thus the boiling point of a solution is higher than the boiling point of the pure solvent. Monoprotic Acids. any acid that contains one ionizable proton (hydrogen ion); nitric acid (HNO<sub>3</sub>) is a monoprotic acid. Diprotic Acids.

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