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General physical properties of Group 1 elements: Alkali metals are good conductors of heat and electricity. Alkali metals have low melting and boiling points as compared to heavy metals such as copper and iron. Table compares the melting and boiling points of potassium (an alkali metal) and copper (a heavy metal).

Physical and Chemical Properties of Group 1 Elements - A ...

Physical Properties A physical property is a property that can be determined without changing the nature of the substance. Consider table sugar, or sucrose. You can see that it is a white solid at room temperature, so color and state are physical properties. It also has a gritty texture.

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Physical properties may be classified as either intensive or extensive. Intensive physical properties do not depend on the sample's size or mass. Examples of intensive properties include boiling point, state of matter, and density. Extensive physical properties depend on the amount of matter in the sample.

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These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

Mrs. J's Chemistry Page - Lecture Notes

Physical properties are not the only properties that describe matter. Chemical properties describe matter based on its ability to change into new matter. One chemical property of matter is reactivity. Reactivity is the ability of a substance to change into a new substance. One kind of reactivity is flammability.

1 SECTION 3 Chemical Properties - Lab RatKOS

Chemistry is the study of matter, and all matter is made up of atoms. We will learn about elements, atomic number and mass, isotopes, moles (chemistry moles, not the animal), and compounds. Learn for free about math, art, computer programming, economics, physics, chemistry, biology, medicine, finance, history, and more.

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