

## ***Holt Chemfile Problem Solving Workbook Stoichiometry Answers***

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Holt ChemFile: Problem-Solving Workbook 127 Percentage Yield Name Class Date Problem Solving continued Sample Problem 2 Acetylene,  $C_2H_2$ , can be used as an industrial starting material for the production of many organic compounds. Sometimes, it is first brominated to form 1,1,2,2-tetrabromoethane,  $CH_2Br_2$ , which can then be

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Holt ChemFile: Problem-Solving Workbook 191 Stoichiometry of Gases Name Class Date Problem Solving continued Rearrange the ideal-gas-law equation to solve for the unknown quantity,  $V$ .  $PV = nRT$   $V = \frac{nRT}{P}$  COMPUTE EVALUATE Are the units correct? Yes; units canceled to give liters of  $SO_2$ . Is the number of significant figures correct?

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Holt ChemFile: Problem-Solving Workbook 261 pH Name Class Date Problem Solving continued Sample Problem 1 A HCl solution has a concentration of 0.0050 M. Calculate  $[OH^-]$  and  $[H_3O^+]$  for this solution. HCl is a strong acid, so assume it is 100% ion-ized. Solution ANALYZE What is given in the problem? the molarity of the HCl solution, and the fact

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and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The ... Holt ChemFile: Problem-Solving Workbook 102 Stoichiometry Name Class Date Problem Solving continued

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Holt ChemFile: Problem-Solving Workbook 280 Titrations Name Class Date Problem Solving continued Sample Problem 3 A supply of NaOH is known to contain the contaminants NaCl and MgCl<sub>2</sub>. A 4.955 g sample of this material is dissolved and diluted to 500.00 mL with water. A 20.00 mL sample of this solution is titrated with 22.26 mL

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Holt ChemFile: Problem-Solving Workbook 58 Mole Concept Name Class Date Problem Solving continued CONVERTING NUMBER OF ATOMS OF AN ELEMENT TO MASS Sample Problem 4 uses the progression of steps 1 → 2 → 3 to convert from the mass of an element to the number of atoms.

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