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Modern Chemistry 109 Chapter Test Chapter: Solutions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.
_____ 1. Molecules whose water solutions can carry electric current a. are nonpolar. b. ionize in water. c. do not dissolve in water. d. decompose in water. _____ 2.

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Solution Concentrations Molality (m) is the concentration of a solution expressed in moles of solute per kilogram of solvent. $m = \frac{\text{mol}}{\text{kg}}$ Do practice problems #1 & #2 on page 424. Solution Concentrations Percent composition by mass (%) is a concentration that expresses the percent of solute in a solution.

Modern Chemistry Chapter 12- Solutions

These unit notes correspond with Holt's Modern Chemistry. These unit notes cover a wide variety topics that will help you prepare for the AP Chemistry Exam or any other Chemistry test.

Holt's Modern Chemistry Textbook | CourseNotes

the solution. This works because the voltage is proportional to the hydro-nium ion concentration. 27. The pH changes slowly at first, then rapidly through the equivalence point, then slowly again. 28. $\text{HIn} + \text{H}_2\text{O} \leftrightarrow \text{H}_3\text{O}^+ + \text{In}^-$ In or $\text{HIn} \rightleftharpoons \text{H}^+ + \text{In}^-$ In acidic solutions, the H_3O^+ ions in solution drive the equation toward the nonionized form. HIn is present in

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