

$$K_D = \frac{[M][L]}{[ML]}$$

$$\theta = \frac{[ML]}{[M] + [ML]} = \frac{1}{1 + K_D/[L]}$$

$$0.1 = \frac{1}{1 + K_D/[L]}$$

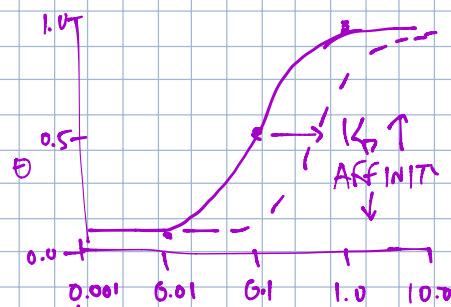
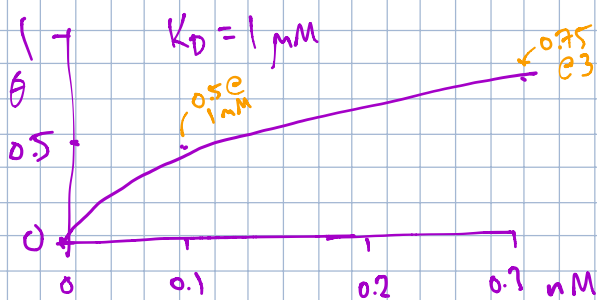
$$0.1(1 + K_D/[L]) = 1$$

$$0.1 + 0.1 K_D/[L] = 1$$

$$0.1 K_D/[L] = 1 - 0.1 = 0.9$$

$$K_D/[L] = 9$$

$$K_D/[L] = 11$$



IF I MOVE CURVE TO RIGHT  
DO I HAVE HIGHER AFFINITY?

$$\Delta G_{EST} = -RT \ln(K_D)$$

$$= -0.0083 \cdot 300 \cdot \ln(10^{-10})$$

$$= -57 \text{ kJ/mol}$$

$$\Delta G_{ITÉS} = -23 \text{ kJ/mol} \quad \Delta \Delta G = -34 \text{ kJ/mol}$$

$$0.2 \text{ mM}$$

$$0.2 - 2 \text{ mM}$$

$$\theta_{0.2} = \frac{1}{1 + 0.1/0.2} = 0.66$$

$$\theta_{2.0} = \frac{1}{1 + 0.1/2} = 0.95$$

RESPONDS