

HOW MUCH DO YOU NEED TO STABILIZE TO GET RATE INCREASE?



$$P_{\text{T.S.}} = \frac{[\text{T.S.}]}{[\text{S}] + [\text{T.S.}]}$$

IF $\Delta G \gg 0$, $[\text{T.S.}] \ll [\text{S}]$

$$P_{\text{T.S.}} = \frac{[\text{T.S.}]}{[\text{S}]} = K$$

$$\Delta G = -RT \ln(K)$$

$$\therefore P_{\text{T.S.}} = \exp(-\Delta G/RT)$$

THIS GIVES PROBABILITY. WHAT ABOUT RATE?

DIE: $P_6 = \frac{1}{6}$ HOW OFTEN I GET 6: $\text{HOW OFTEN I ROLL} \times P_6$

FOR T.S.: $k = A \cdot \exp(-\Delta G/RT)$

PREFACTOR: TRIES PER UNIT TIME. \uparrow
 DIFFUSION (USUALLY) \uparrow
 IN BIOLOGY, s^{-1}

k \uparrow RATE CONSTANT

A \uparrow E_a

IN BIOLOGY, ENZYMES INCREASE RATES DRAMATICALLY. TEMPS \uparrow GIVES 10" SPEEDUP!

OK, SO WHAT CHANGE IN E_a WOULD LEAD TO 10" SPEED UP FOR TRYPSIN?

$$k_{ENZ} = A \exp(-E_{a,ENZ}/RT) \leftarrow \text{CAT RATE}$$

$$k_{H_2O} = A \exp(-E_{a,H_2O}/RT) \leftarrow \text{UNCAT RATE}$$

$$\frac{k_{ENZ}}{k_{H_2O}} = 10 = \frac{A \exp(-E_{a,ENZ}/RT)}{A \exp(-E_{a,H_2O}/RT)}$$

$$= \exp(-E_{a,ENZ}/RT) \cdot \exp(E_{a,H_2O}/RT)$$

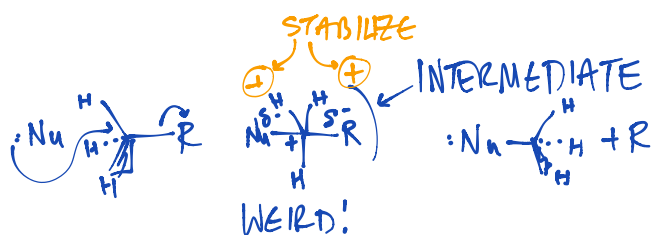
$$= \exp(-E_{a,ENZ}/RT + E_{a,H_2O}/RT)$$

$$= \exp((E_{a,H_2O} - E_{a,ENZ})/RT) \quad \Delta E_a = E_{a,ENZ} - E_{a,H_2O}$$

$$= \exp(-\Delta E_a/RT)$$

$$-RT \ln(10) = \Delta E_a = -300 \cdot 0.0083 \cdot \ln(10) = -63 \text{ kJ/mol}$$

SMALL ΔE_a LEADS TO HUGE RATE INCREASE

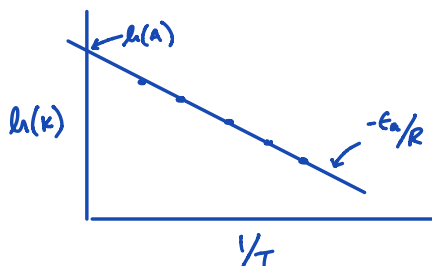


ENZYMES PREPARE TO STABILIZE TS. BY HAVING UNFULFILLED INTERACTIONS THAT CAN ONLY BE FULFILLED BY INTERMEDIATE.

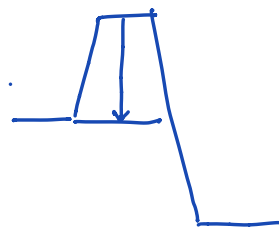
HOW CAN WE MEASURE THIS?

$$\ln(k) = \ln(A) - E_a/RT$$

$$\ln(k) = \frac{-E_a}{R} \cdot \left(\frac{1}{T}\right) + \ln(A)$$



WHAT IF E_a GETS REALLY SMALL?



$$E_a = 0 : k = A e^{-0/RT}$$

$$k = A$$

SUMMARY:

- ① RATE IS DETERMINED BY HOW OFTEN WE "TRY" REACTION TIMES
PROBABILITY OF SUCCESS $(k = Ae^{-E_a/RT})$
- ② SMALL CHANGE IN ENERGY BARRIER, HAKE CHANGE IS RATE
- ③ CAN BE MEASURED BY PLOTTING $\ln(k)$ AGAINST $1/T$
- ④ FOR "BARRIERLESS" TRANSITIONS, LIMIT IS PREFACTOR (DIFFUSION).