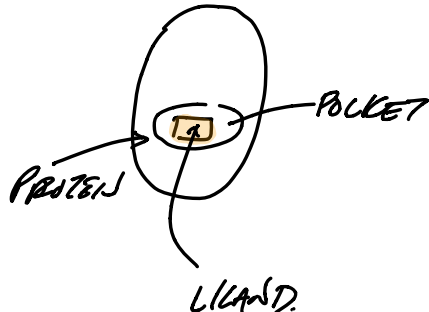


ELECTROSTATICS I: 10/26/19

IMAGINE WE HAVE A HYDROPHOBIC LIGAND IN A HYDROPHOBIC POCKET:

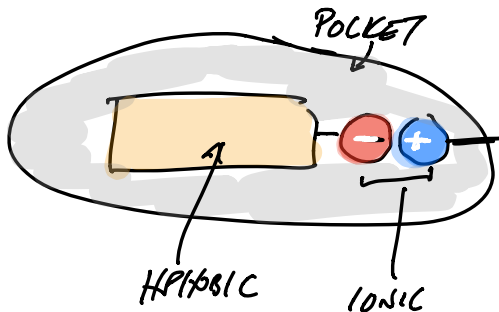


HOW DO WE CALCULATE ΔG_{BIND} ?

1. $\Delta G(\Delta ASA)$: ASA OF POCKET AND LIGAND UNBOUND VS ASA OF COMPLEX.
2. FULL MD SIMULATION WITH ALL ATOMS.

GOING TO USE ASA FOR THIS TALK.

NOW ADD IONIC INTERACTION:



HOW DO WE CALCULATE ΔG_{BIND} ?

$$\Delta G_{BIND} = \Delta G_{ASA} + \Delta G_{COULOMB}$$

$$U_{COULOMB} = \frac{q_i \cdot q_j}{r_{ij} \cdot \epsilon} \cdot \alpha$$

$$\alpha = 1389 \text{ Å} \cdot \text{kJ/mol}$$

$$r = \text{DIST IN Å}$$

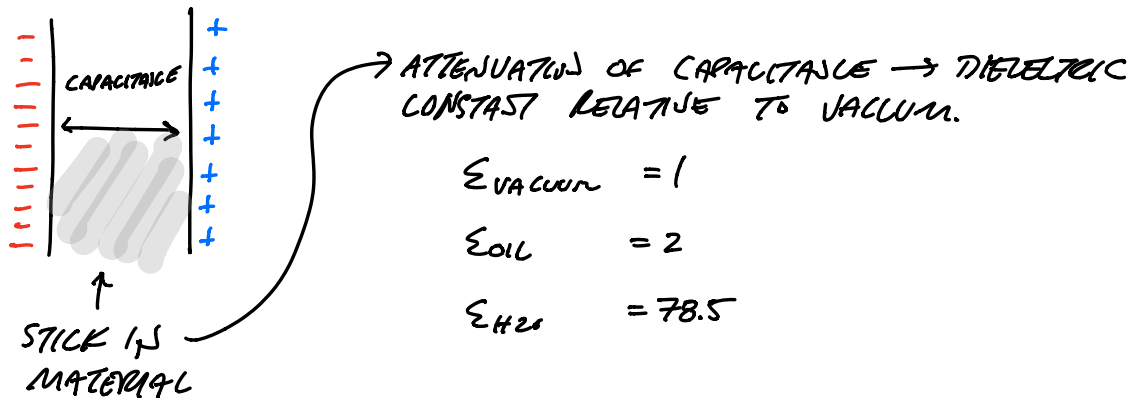
$$q_x = \text{CHARGE}$$

DIELECTRIC CONSTANT (ϵ)
MEASURES HOW GOOD ENVIRONMENT IS AT ATTENUATING ELECTROSTATIC INTERACTIONS.

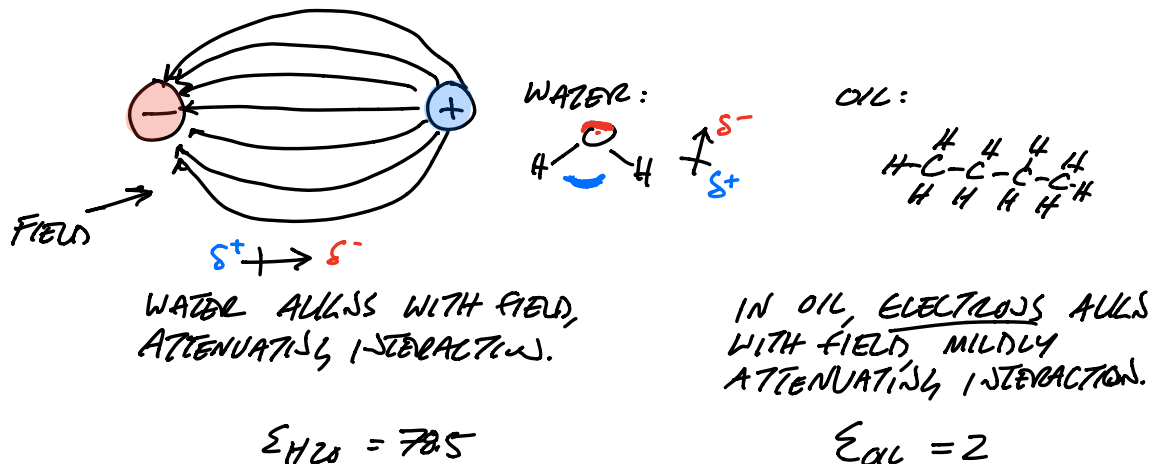
ENERGY TO TAKE CHARGE FROM INFINITE DISTANCE TO DISTANCE " r ".
THIS IS BASICALLY A BINDING REACTION,
SO

$$\Delta G_{COUL} = U_{COUL}$$

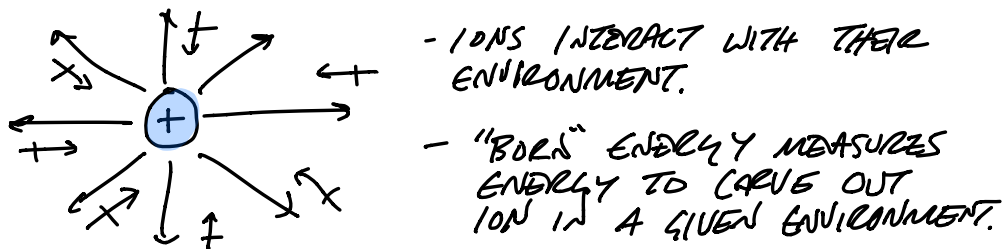
WHAT IS A DIELECTRIC CONSTANT, (IF MEASURED) EXPERIMENTALLY?



WHAT IS A DIELECTRIC CONSTANT MOLECULARLY?



BUT WHAT ABOUT INDIVIDUAL IONS? DON'T THEY HAVE FIELDS TOO?



$$G_{\text{BORN}} = \frac{q^2}{r_{\text{ion}} \cdot \epsilon} \cdot 2$$

\leftarrow ENERGY TO BRING BITS OF CHARGE ONTO SURFACE OF SPHERE OF RADIUS r_{ion} .

$$\Delta G_{\text{Born}} = \frac{q^2}{8\pi\epsilon_0} \left(\frac{1}{\epsilon_{\text{prot}}} - \frac{1}{\epsilon_{\text{H}_2\text{O}}} \right)$$

ENERGY TO BRING ION FROM WATER INTO PROTEIN WITH DIELECTRIC OF ϵ_{prot} .

WHAT IS "RIGHT" VALUE OF ϵ_p ?

- DEPENDS ON EXACT ENVIRONMENT.
- $\epsilon_{\text{prot}} \approx 4$ DEEP IN INTERIOR
- $\epsilon_{\text{prot}} \approx 15$ ON PROTEIN SURFACE.
- HARD PROBLEM: HOW TO ATTENUATE ELECTROSTATICS.

$$\Delta G_{\text{BIND}} = \Delta G_{\text{ASA}} + \Delta G_{\text{coul}} + \Delta G_{\text{Born}} + \dots$$