

You may have heard of 'Nano'

Nano-cars Nano-bots Nano-phones Nano-cameras

'Nano' has become a buzz word in popular culture to just mean small.

But what is it really?

What is Nano?

To understand nanotechnology we must first think about scale.

A metre is about the distance between the A metre is about the ond of your hand. A tip of your nose will be a second of your hand. nanometre is a billionth of a metre.

A nanometre is the unit of measurement used in the nanoscale.





Quite a few nanometre here!

A human hair is 40,000 -200,000 nanometre wide.

A man's beard grows a nanometre per second.

When a seagull lands on an oil tanker the ship dips down one nanometre.

A nanometre is about the width of 6 bonded carbon Nanoparticles have at least atoms.

one dimension within the range from 1 to 100 nanometres.

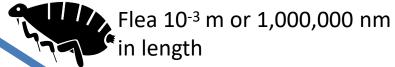


C₆₀ is a nanoparticle its diameter is Just over 1 nm

The Nanoscale



Dog 10^o m or 1,000,000,000 nm in length.



Anything Nano is defined as having One of its dimensions smaller than 100 nanometres (nm).



Red Blood cells 10⁻⁵ m or 6000 nm in diameter

> DNA 10⁻⁸ m or 2.5nm in diameter.

Nanoparticle 10⁻⁹ m. One dimension between 1



A water molecule which measures approximately 0.5 nm is about as big in relation to an apple as an apple is to planet Earth.

apple

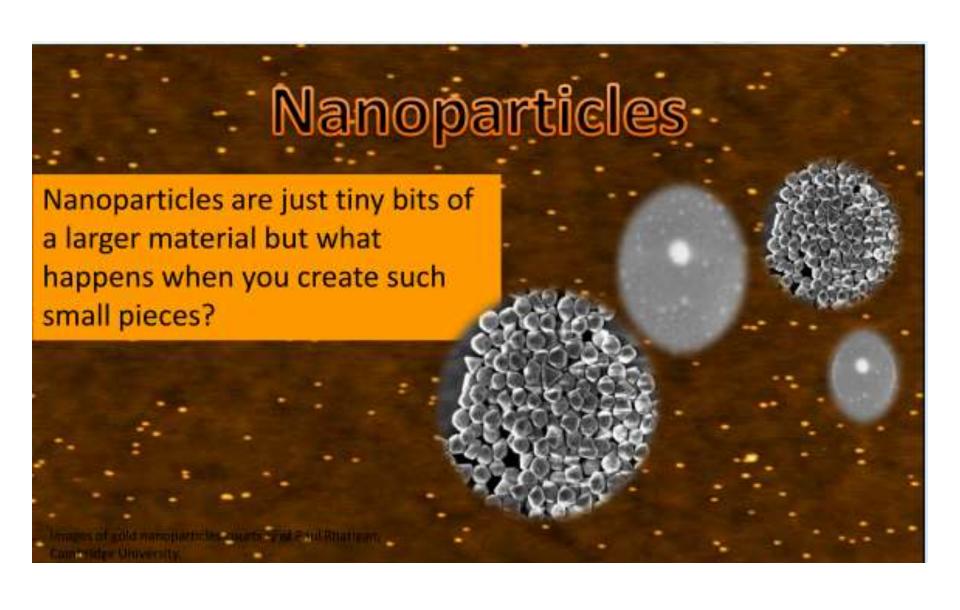


It's a long way down to the nanoscale!



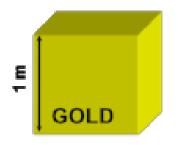


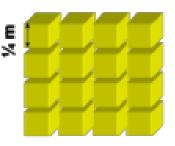


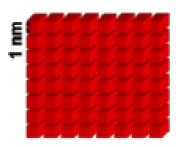


Tiny pieces = more......

Surface area.







Each side = 1 meter

Mass ≈ 43,000 lb

Surface Area (SA) = 6 m²

≈ 8 ft x 8 ft room

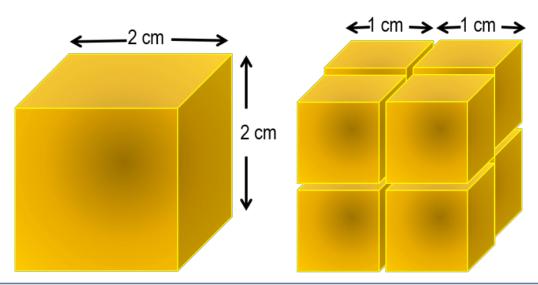
Each side = 1/4 meter Mass ≈ 43,000 lb SA = 24 m²

Each side = 1 nanometer

Mass ≈ 43,000 lb

SA = 6 billion m² ≈ 2500 miles²

Area of Delaware = 2490 miles²



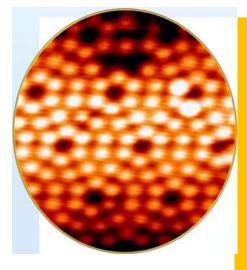
# of Cube (s)	Dimensions (cm)	Surface Area (I x I x 6 cm ²)	Volume (I x I x I cm³)	Surface Area/Volume ratio
1	2 x 2	$(2 \times 2 \times 6) = 24$	$(2 \times 2 \times 2) = 8$	3
8	1 x 1	8 (1 x 1 x 6) = 48	8 (1 x 1 x 1) = 8	6

<u>More surface area</u> <u>= more reactivity</u>

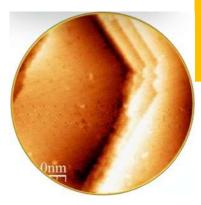
Nanoparticles have more <u>surface area</u>. This makes them more reactive since chemical reactions happen on the surface.

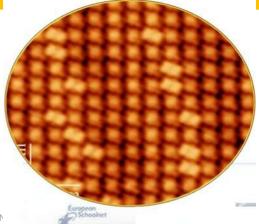
More reactive means potentially more <u>useful</u>.





In a cubic centimetre of material <u>one in 10</u> million atoms are on the surface but for a cubic nanometre, <u>80%</u> of atoms are on the surface and potentially ready to react!







Images courtesy of Dr Colm Durkan, Cambridge University.

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Size is the key here!

At the Nanoscale a material's properties can change dramatically. These could be their boiling points, solubility or catalytic activity.

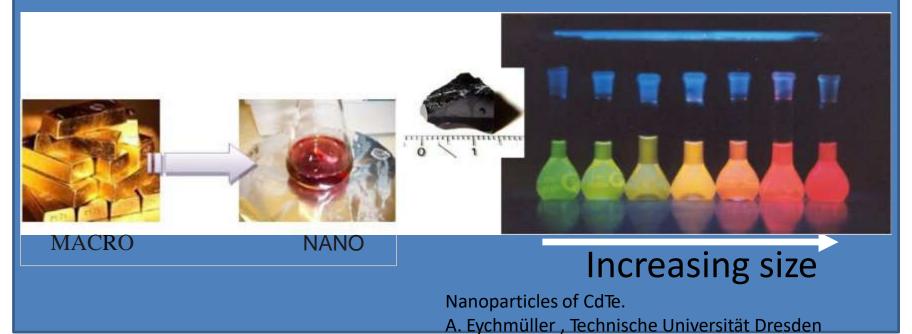
With only a reduction in size materials can exhibit new properties, properties they do not possess when they are on a larger or macro scale.

The normal 'classical' laws of physics no longer apply!

Colour changes

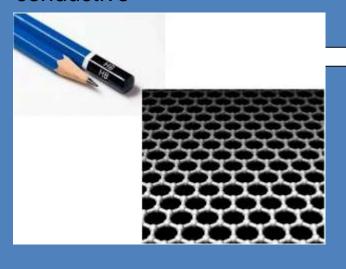
• For example, sometimes just changing the size of a particle can drastically alter its colour.

Cadmium Telluride



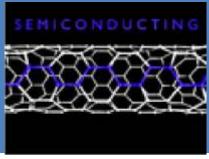
Even mechanical and electrical properties can be influenced by size!

Graphene is brittle and non-conductive



Carbon nanotubes are like graphene sheets but rolled up...however they have totally different properties.





Did you know? Carbon nanotubes are much stronger than steel yet much lighter, and they can be conductive.

So how do scientists work on such a tiny scale?

Seeing Nano

Mainly because microscopes have become more sophisticated.



Micrographia 1665

The first microscopes to be developed in around 1665 opened up a whole new world for scientists. For the first time cells and structures of nature that we are familiar with now became visible. Public opinion was fearful of this voyage into the unseen world of the microscopic.

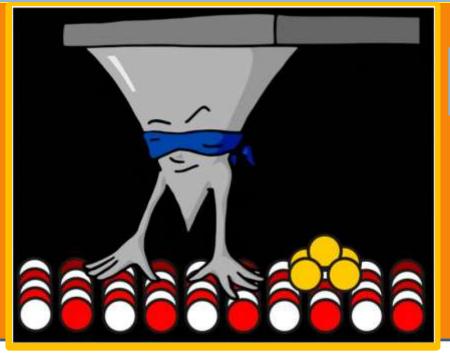


Now we have microscopes that can see even further into this world, actually to the atoms that are the very building blocks of our living world.

These more sophisticated microscopes are known collectively as Scanning Probe Microscopes or SPMs.



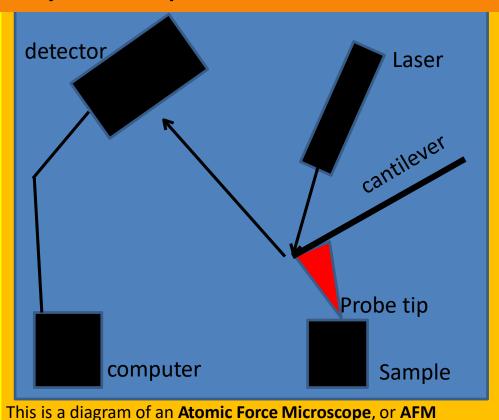
A Scanning Probe Microscope uses an extremely fine probe tip (sometimes ending in only a few atoms) and runs over the surface feeling for contours and shapes.



Probe tip feeling for shapes.

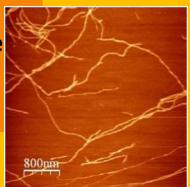
Examples include:
Atomic Force Microscope,
Scanning Tunnelling
Microscope

A cantilever, similar to a diving board, is attached to the tip of **Atomic Force Microscope or AFM**. The whole thing move up and down as the tip moves over the atomic–scale hills and valleys of a sample's surface.



A laser reflects off the back of the cantilever. When the cantilever deflects, so does the laser beam. A detector in a computer records the movement of the laser and translates that data into an image, such as the images on the right.

Images courtesy of T. Oppenh
Cambridge University



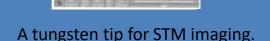




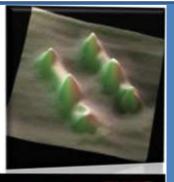
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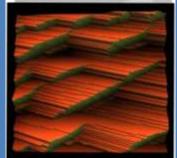
Scanning tunnelling microscope.

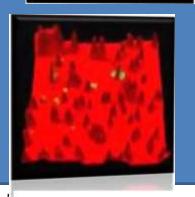
A scanning tunnelling microscope works by having a tiny probe which scans across a surface picking up variations in the current that runs between itself and the atoms on the surface. This probe is made of conducting material (usually Tungsten) and the very end of it is only 1 atom wide!



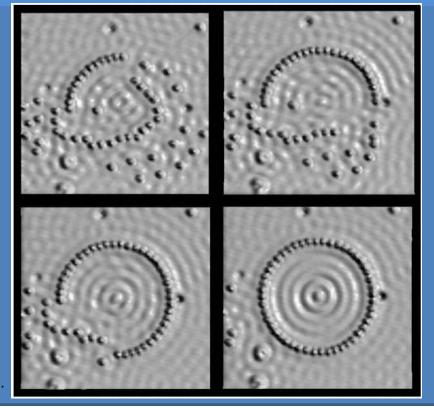
These variations in current are translated into images such as the ones to the right.







The probe tips can also be used to move individual atoms.

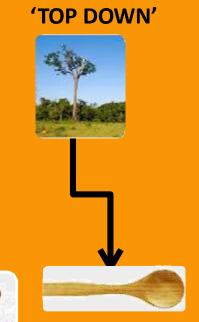


The image shows iron atoms which have been moved individually.

Image originally created by Don Eigler, IBM corporation.

Making Nano

This idea of building things from the atoms and working up is interesting for researchers.



Computers and phones are getting smaller yet more powerful. This trend demands smaller and smaller components, getting down to the nanoscale. It makes sense to be able to create these nano-devices from the 'bottom up'. Rather than the more traditional 'top down' approach to manufacturing.

Manufacturing from the 'bottom up' would mean less energy and less waste.

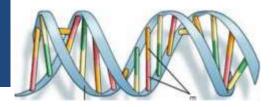
Nanoyou Consortoum

Moving atoms individually is still not a viable technique for creating nanostructures because it's too slow. However there is another process called 'SELF ASSEMBLY', often seen in nature, which scientists can exploit.

Atoms, molecules or nanoparticles will arrange themselves into larger structures if they have the right properties and environments.

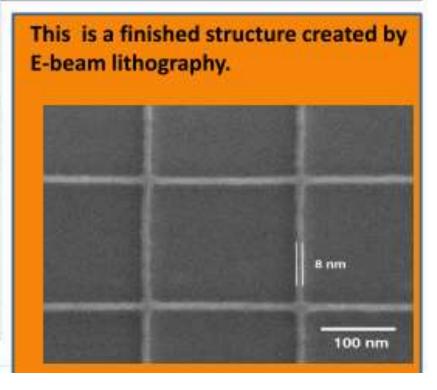
This process plays a central role in the construction of DNA, cells, bones and viruses. These all self assemble without assistance.





Another way scientists can create nano-sized devices is a process called <u>lithography</u>. It is often used to make <u>computer chips</u>. It works in a very similar way to printing or if you were spray painting over a stencil. But <u>lithography</u> uses light or electrons instead of ink or paint.

Photolithography uses light and structures as small as around 20 nm can be made. If scientists want to make even smaller structures (around 5 nm), they use electrons instead of light, this is called E-beam lithography.





In this image each letter is made up of hundreds of nanotubes!

They are <u>stronger</u> yet considerably <u>lighter</u> than steel and flexible. They can behave like metals but also semiconductors, they are great at transmitting heat and they are assembled from carbon atoms.

Are an exciting discovery which could revolutionise materials of the future.

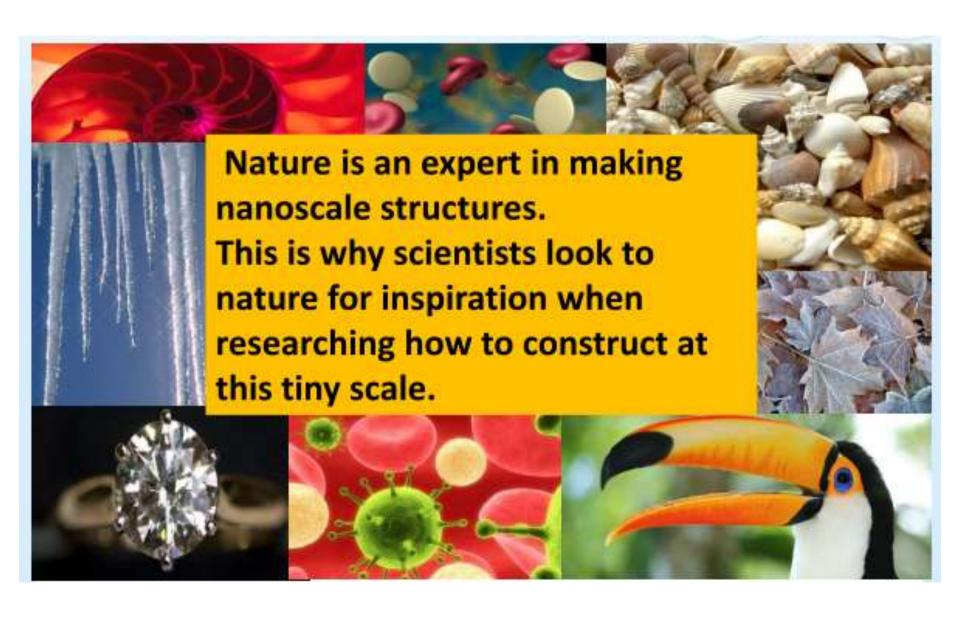




Courtesy of Stephan Hoffman, Cambridge University.

Even when nanotubes are fully grown, each entity is smaller than a human blood cell.

With properties like this they could play a pivotal role in the development of new materials and future electronics.



Nature's Nano

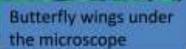
Biomimicry

Biomimicry is the term scientists copying nature. By studying Lotus leaves and their structure scientists have designed non-wetting materials and self-cleaning windows.

By studying butterfly wings and their nanostructures scientists have explained how light can interact differently with surfaces. This has helped design security holograms and hair products!

Nameyou Consumbure





Even the nano structure of a Toucan's beak has offered insight into designing ultra-light aircraft components.

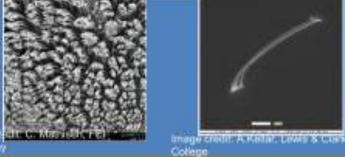






Geckos' feet have nanostructures.





Geckos' feet are covered in tiny hair-like structures called setaes. These structures can get so close to a surface that weak 'sticky' interactions between molecules become significant. The result is strong adhesion which is entirely due to Van der Waals forces. Scientists have been inspired by this nanostructure to create internal bandages which still adhere even in a wet environment.

Nature is experienced in working on the nanoscale and we can learn a lot from its success at creating natural nanotechnology.

Nano and you.

Nanotechnology is already in our lives.





Nanoelectronics has enabled the miniaturisation of electronic gadgets in daily use.



Some tennis balls stay bouncy for longer thanks to a nano-structured inner layer.



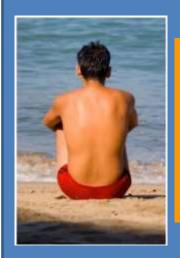


Socks that contain antibacterial nanoparticles of silver are supposed to keep your feet healthy and odour free. Some tennis racquets are made from a carbon-based nanocomposite making them stronger and lighter than ever before.

Nanotechnology is already in our lives.



Some textiles are made very hydrophobic using liquid repellent nanostructures which enable them to be water and stain resistant.



Cosmetics and sunscreens which contain nanoparticles can promote better protection and comfort.

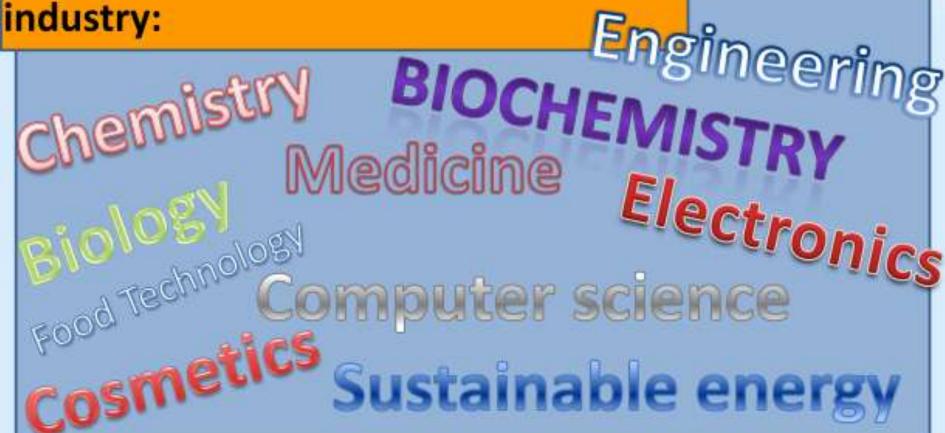


Sun glasses which have a nano-structured coating are easier to clean, harder to scratch, antistatic,anti-misting and anti-bacterial.



Studying Nano

Nanotechnology is a multidisciplinary subject as it incorporates many different areas of science and industry:



Nano future



Information Technology e.g. faster computers

Energy Solutions e.g. such as more economical fuel cells and solar cells

Areas which have an impact on all our lives.





