

$$a = \frac{C_p - C_E}{C_{Sn} - C_E}$$

$$b = \frac{C_E - C_{Pb}}{C_{Sn} - C_{Pb}}$$

- a) Fraction of proeutectic phase (β)
 = Fraction of β just above 183° (T_E)

$$= \frac{80 - 61.9}{975 - 61.9} = \frac{18.1}{913.1} = 0.0198$$

- b) Fraction of Eutectic Mixture
 $= 1 - \text{fraction of proeutectic mixture}$
 $= 1 - \left(\frac{18.1}{35.6} \right) = 0.492$

c) $b(\mu) = \frac{C_B - C_0}{C_B - C_A} = \text{when } C_0 = 80^\circ \text{ to } C_B = 97.5^\circ, C_A = 18^\circ$

$$b(x) = \frac{97.5 - 80}{97.5 - 19} = 0.223$$

$$b_p = 1 - b_x = 0.777 \quad (4 \text{ mark})$$

- Just above the eutectic temp we have L + β mixture while on cooling slightly below eutectic temp we get ($\alpha + \beta$) and β mixture

d) Fraction of total β phase :-

Applying lever rule just below invariant line

$$f_{\beta} = \frac{C - C_{\alpha}}{C_{\beta} - C_{\alpha}} = \frac{20.0 - 19}{97.5 - 19} = \frac{61}{78.5} = \boxed{0.777}$$

e) Fraction of β phase that forms part of eutectic mixture

$$f_{\text{eutectic}} = f_{\text{total } \beta} - f_{\text{proeutectic } \beta}$$

$$= \frac{61}{78.5} - \frac{10.1}{35.6} = 0.777 - 0.508$$

$$\Rightarrow \boxed{0.269}$$