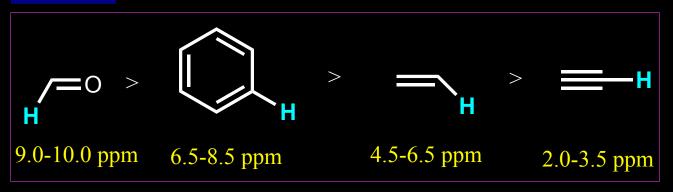
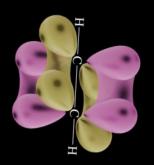
- 1) Reference in proton NMR
- 2) Chemical Shift
- 3) Integration of Signal
- 4) Identification of signal
- 5) Anisotropic effect
- 6) Splitting of signals

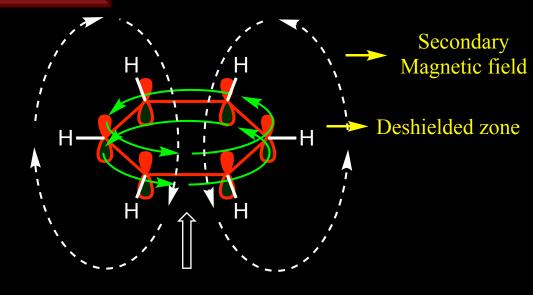
★ Note-3:



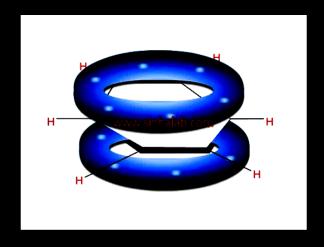
Reason: Anisotropic effect



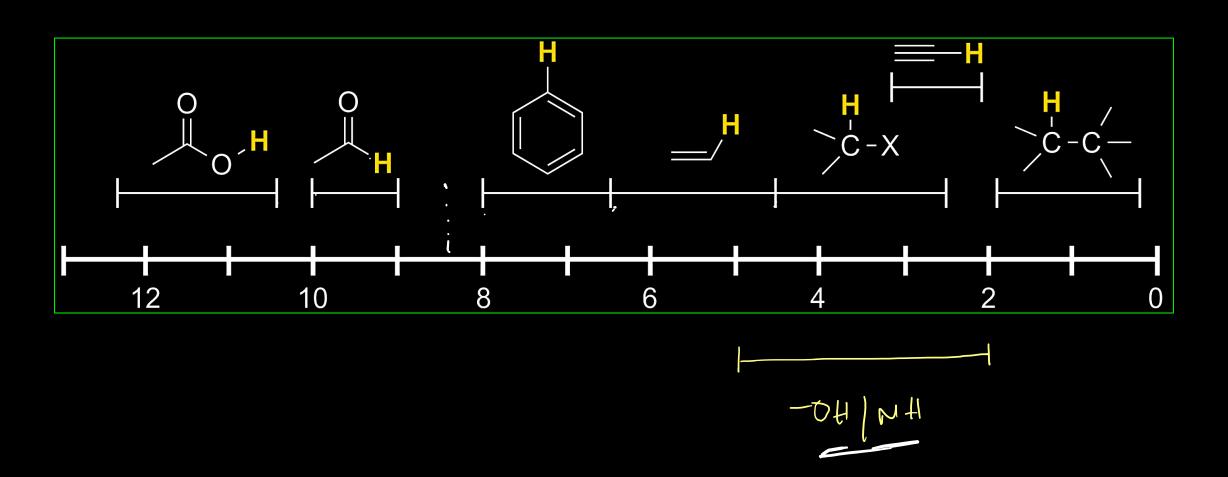
The circulating π -electrons induce a magnetic field that adds to the applied field which Causes them to experience a stronger net field and therefore resonate at higher frequency



Applied Magnetic Field

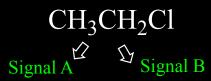


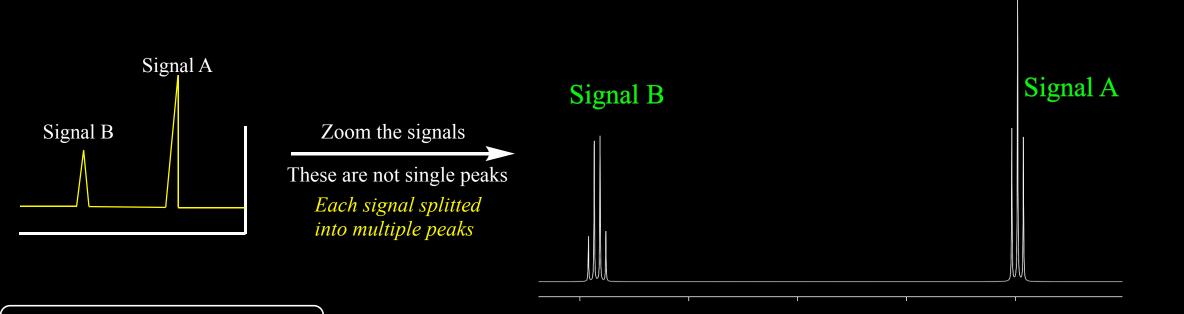
Your simple guide to chemical shifts



- 1) Reference in proton NMR
- 2) Chemical Shift
- 3) Integration of Signal
- 4) Identification of signal
- 5) Anisotropic effect
- 6) Splitting of signals







N+1 Rule

N= number of neighboring / different hydrogens

CH₂ Signal

Neighboring to CH₂ contains carbon with 3 H's

$$N+1 = 3+1 = 4$$

CH₂ signal will split into 4 peaks

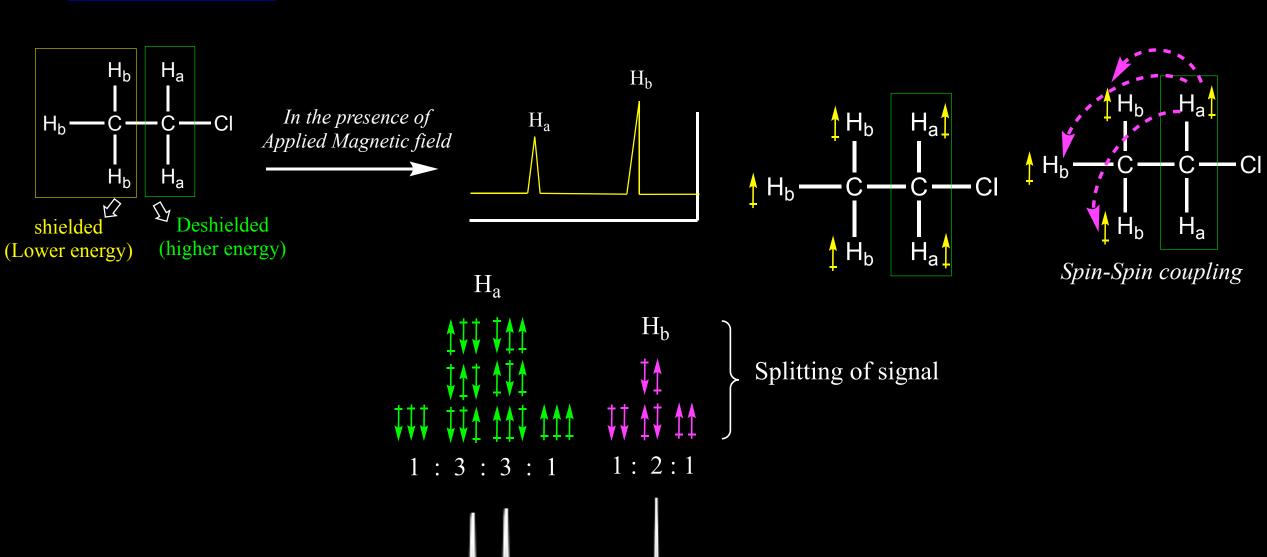
CH₃ Signal

Neighboring to CH₃ contains carbon with 2 H's

$$N+1 = 2+1 = 3$$

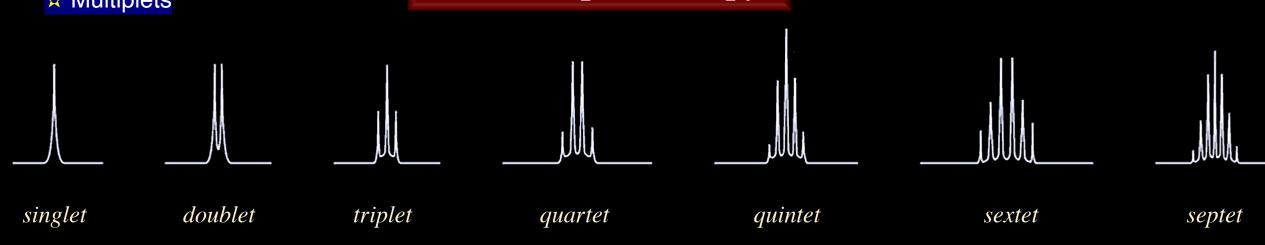
CH₃ signal will split into 3 peaks

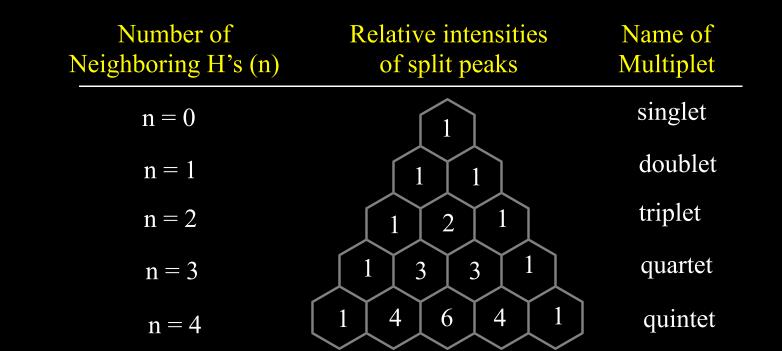
★ Splitting of signal



★ Multiplets

¹H NMR Spectroscopy





• Cl-CH₂-CH₂-Br

• 2 chemically non equivalent protons

• two signals

Cl-CH₂ Signal

Neighboring to CH₂ contains carbon with 2 H's

$$N+1 = 2+1 = 3$$

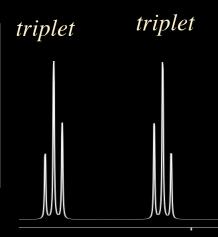
CH₂ signal will split into 3 peaks

CH₂-Br Signal

Neighboring to CH₃ contains carbon with 2 H's

$$N+1 = 2+1 = 3$$

CH₂ signal will split into 3 peaks



 $\begin{array}{cccc} & \text{CH}_3 & \text{CI} \\ \text{I} & \text{C} & \text{CI} \\ \text{H}_3\text{C} - \text{C} - \text{C} & \text{I} & \text{H}_2 \\ & \text{CH}_3 & \text{CH}_3 & \text{CI} \end{array}$

• 2 chemically non equivalent protons

• two signals

CH₂ Signal

Neighboring to CH₂ contains carbon with 0 H's

$$N+1 = 0+1 = 1$$

CH₂ signal will appear as singlet

CH₃ Signal

Neighboring to CH₃ contains carbon with 0 H's

$$N+1 = 0+1 = 1$$

CH₃ signal will appear as singlet

singlet

singlet